Research Article

Optimization of Cultural Conditions for Production of Extracellular Polymeric Substances (EPS) by Serpentine Rhizobacterium Cupriavidus pauculus KPS 201

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Received 30 March 2013; Revised 10 October 2013; Accepted 11 October 2013

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Extracellular polymeric substances (EPS) are complex biopolymers produced by a wide array of microorganisms for protection against dessication, aggregation, adhesion, and expression of virulence. Growth associated production of EPS by Ni-resistant Cupriavidus pauculus KPS 201 was determined in batch culture using sodium gluconate as the sole carbon source. The optimum pH and temperature for EPS production were 6.5 and 25°C, respectively. Optimal EPS yield (118 μg/mL) was attained at 0.35% Na-gluconate after 72 h of growth. Cupriavidus KPS 201 cells also utilized glutamate, acetate, pyruvate, fumarate, malate, malonate, formate, citrate, and succinate for EPS production. Although EPS production was positively influenced by the increase of nitrogen and phosphate in the growth medium, it was negatively influenced by nickel ions. Compositional analysis of the purified EPS showed that it is a homopolymer of rhamnose containing uronic acid, protein, and nucleic acid. Presence of lipids was also detected with spectroscopy. Non-destructive EPS mediated biofilm formation of KPS 201 was also visualized by epifluorescence microscopy.

1. Introduction

Extracellular polymeric substances (EPS) are biosynthetic polymers of microbial origin produced in natural as well as artificial environments by single species or in heterogeneous communities. Irrespective of their origin, EPS are localized at or outside the bacterial cell surface and composed of a variety of high molecular weight organic macromolecules such as polysaccharides, proteins, nucleic acids, and phospholipids [1]. These exopolymers result from microbial processes like active secretion, shedding of cell surface materials, cell lysis, and adsorption from the environment. Based on their physical state, EPS can be distinguished as bound and soluble forms. The bound EPS include microbial sheaths, capsular polymers, condensed gel, and so forth while the soluble EPS include soluble macromolecules, colloids and slimes. According to the metabolism of bacterial cells, the soluble EPS are considered to be actively secreted by bacteria and are biodegradable, while the bound ones remain attached to active and inert biomass or are molecules resulting from cell lysis.

The production of EPS by bacteria in culture depends on phases of growth, nutritional status, and the environmental conditions. The production of water-soluble EPS by Rhodobacter capsulatus was influenced by the C:N ratio and polymer production was found to be dependent on the nature of carbon sources used [2]. Similarly, in anaerobically growing Shewanella spp., the terminal electron acceptors were found to influence the quantity and composition of capsular exopolymers. Enhanced production of EPS was induced under stressful culture conditions in Enterobacteriaceae members where compositional variation was noted due to varying proportions of different polysaccharides synthesized within the biofilm. However, for a number of bacterial species, increased EPS synthesis was found to be genetically controlled and differentially expressed genes in biofilm state regulate its composition and development.
EPS are considered to be the key components that determine the structural and functional integrity of microbial aggregates (cohesion) and for anchoring the biofilms to the substratum (adhesion). A large amount of work has been conducted on characterizing the EPS of various pure and mixed cultures [3, 4], activated sludge [5], biogranules [6], and biofilm [7, 8]. The EPS may not be essential for growth and viability in free-living bacterial cultures but protect cells from hostile environments. They are also involved in the degradation of particulate substances, sorption of heavy metals, leaching of minerals from sulphidic ores and biocorrosion, thus aiding in heavy metal bioremediation [9, 10].

Cupriavidus pauculus KPS 201 is a Ni-resistant rhizobacterium isolated from nickel percolated serpentine outcrops of Andaman Islands, India [11]. The bacterium was found to accumulate intracellular biopolymer, the polyhydroxalkanoates under nutrient limited conditions as well under metal stress [12]. The aim of our present study was to evaluate whether this bacterium also produces extracellular polymeric substances and also to elucidate the influence of metal ions on polymer production.

2. Materials and Methods

2.1. Bacterial Strain and Cultural Conditions. Cupriavidus pauculus KPS 201 (MTCC 6280; GeneBank Accession no. AM418462), used throughout the present study, was previously isolated from rhizosphere of Rinorea bengalensis, a Ni-hyperaccumulator growing in serpentine outcrops of Saddle hills in north Andaman Islands, India [11]. The organism was grown on slopes of Tryptic soy agar medium at 28°C for 24 h and regularly subcultured at 15-day interval.

2.2. Growth and EPS Production. Production of extracellular polymeric substances (EPS) by Cupriavidus KPS 201 cells was determined during growth in batch culture using Tris-minimal medium supplemented with sodium gluconate as the sole carbon source. The low phosphate Tris-buffered minimal medium (pH 7.0 ± 0.2) contained (g/L) Tris-base 6.06, NaCl 4.68, KCl 1.49, NH₄Cl 1.07, NaSO₄ 0.43, MgSO₄·6H₂O 0.2, CaCl₂·2H₂O 0.03, Na₂HPO₄·12H₂O 0.02, and 0.4% Na-gluconate. Erlenmeyer flasks (250 mL) containing 50 mL media were inoculated with 18 h grown bacterial culture and agitated on a rotary shaker (120 rpm) at 28°C for 76 h. Samples were withdrawn at definite time intervals for determination of cell number and EPS production. Growth was estimated by total count method following centrifugation and washing twice in normal saline. The total number of cells in 1 mL of culture was estimated using a haemocytometer (Neubauer, Fein-Optik Jena, Germany) and a phase contrast microscope (Zeiss Winkel Model no. 148786, Germany).

2.3. Estimation of EPS. The fermented medium was first centrifuged (10,000 xg, 4°C for 10 min) to separate the supernatant and cell pellet. The soluble EPS in the supernatant was precipitated with 2 volumes of prechilled acetone, kept overnight at 4°C and further separated by centrifugation (12,000 xg, 4°C, 20 min). Subsequently, the precipitated EPS was dissolved in distilled water and quantified following the method of Dubois et al. [13]. To 1 mL of EPS solution, 1 mL of 5% (w/v) phenol solution was added and mixed thoroughly. To the reaction mixture, 5 mL of concentrated H₂SO₄ was purged in and the optical density was measured at 490 nm. The amount of EPS was determined from the calibration curve using glucose as the standard.

2.4. Purification and Compositional Analysis of EPS. The crude EPS thus obtained was purified by repeated dissolution in sterile distilled water and reprecipitation by prechilled acetone. The precipitate was lyophilized and the white powdered polymer obtained was analyzed. Total sugar content was estimated following the anthrone method of Dubois et al. [13] using glucose as standard. Estimation of sugar monomers was achieved by gas chromatography (Shimadzu GC-17A, Japan) following methanolysis of EPS. Total protein content was estimated by Folin–phenol reagent using bovine serum albumin as standard [14]. Nucleic acid was estimated by measuring the absorbance at 260 and 280 nm using purified DNA sample as standard. Quantitative estimation of the uronic acids was spectrophotometrically performed following the method of Blumenkrantz and Asboe-Hansen [15].

2.5. FTIR Analysis. The Fourier transform infrared (FTIR) spectra of the purified EPS were recorded in a Perkin Elmer RX-1 FTIR spectrometer. The samples prepared as KBr pellets were scanned from 500 to 4000 cm⁻¹.

2.6. Staining and Visualization of EPS with Fluorescently Labeled Lectins. The lectin staining technique based on the method of Sizemore et al. [16] was used for visualization of EPS production by Cupriavidus KPS 201 cells. Lectin was dissolved in phosphate buffered saline (PBS) containing, in grams per liter, NaCl, 4, KCl, 0.1, Na₂HPO₄·2H₂O, 0.72, and KH₂PO₄, 0.1, adjusted to pH 7.4. Sterile glass slides were immersed in bacterial suspension kept in Petri dishes for 3–7 days at 28°C for cell attachment and biofilm formation. Bacterial biofilms prepared on glass surface were washed with PBS once and covered with 50 μL DAPI solution (0.01%) for 10 min in the dark. The supernatant was removed, washed once with PBS, and stained with 50 μL fluorescein isothiocyanate-labeled lectin concanavalin A (0.1 g/L) for 4 min in the dark. Finally, the slide was washed with 1 mL PBS and 1 mL sterile distilled water and dried before observation under epifluorescence microscope (Carl Zeiss).

2.7. Statistical Analysis. All experiments were carried out in triplicate and results represent mean ± standard error.

3. Results and Discussion

3.1. Growth Associated Production of EPS. Cells of Cupriavidus pauculus KPS 201 produced extracellular polymeric substance during growth in Tris-buffered minimal medium using sodium gluconate as the sole carbon source. Various
physical and chemical procedures for isolation of EPS were screened, amongst which centrifugation at 12,000 xg for 20 min at 4°C was found to be most suitable. Time course of growth revealed that EPS production started at late logarithmic phase and continued to increase till late stationary phase (Figure 1). The medium pH turned alkaline with growth and remained 7.7 after 76 h of incubation.

3.2. Optimization of Cultural Conditions for EPS Production. The optimum pH and temperature for EPS production during growth of KPS 201 cells were 6.5 and 25°C, respectively. However, the optimum conditions do not coincide with the optimum conditions for growth of *Cupriavidus* KPS 201 cells (Figure 2). Gradual increase in carbon source (Na-gluconate) from 0 to 0.5% led to steady increase in cell number along with EPS yield (Figure 3). However, growth attained a maximum at 0.3% gluconate and optimum EPS yield (118 μg/mL) was attained at 0.35% gluconate after 72 h growth. A similar phenomenon was observed by Sheng et al. [2], who reported that an adequate amount of carbohydrate source was essential for the EPS production by *Rhodopseudomonas acidophila*. EPS production by *Ralstonia eutropha* and *Pseudomonas* sp. was also enhanced in the presence of sufficient carbon sources [17].

*Cupriavidus* KPS 201 cells were tested for their ability to utilize nine different carbon sources like glutamate, acetate, pyruvate, fumarate, malate, malonate, formate, citrate, and succinate (as sodium salts) for EPS production apart from Na-gluconate. Amongst all carbon sources used (at 0.4 w/v), EPS yield was the highest when cells grew in sodium glutamate and EPS production was found to increase by 25.2% in case of glutamate grown cells (132.3 μg/mL) than those in Na-gluconate (105.6 μg/mL) (Table 1). However, cells failed to show any visible growth in presence of sodium malate and formate and hence produced no extracellular polymer. Different species of *Rhizobium* and *Agrobacterium* have also been reported to utilize various monosaccharides, disaccharides, polysaccharides as well as mannitol, glutamate, succinate, and so forth, for production of exopolysaccharides [18].

The concentration of nitrogen has a profound influence on production of EPS by the bacterium and growth as well as EPS production increased with increase in nitrogen content in the medium. EPS yield increased by 23% of control and amounted to 129.2 μg/mL at 1.75 g/L of nitrogen (Figure 4). The positive effect of nitrogen on EPS production has been reported previously for a number of bacterial cultures [19, 20]. However, for some bacteria, such as *Xanthomonas*, *Pseudomonas*, and *Rhizobium* sp., cultivation under nitrogen-limiting conditions results in an increasing EPS production [21].

Increase in phosphate concentration from 0.01–5.0 g/L also led to enhancement of growth and EPS production by *Cupriavidus* KPS 201 cells. Although, maximum growth was attained at 0.5 g/L of phosphate, the optimum EPS yield (258.7 μg/mL) was achieved at 0.75 g/L of phosphate. At this concentration of phosphate, more than twofold increase in EPS yield was obtained (Figure 5). However, there was no change in the final pH level of culture filtrate which remained at 7.5 to 7.7 at different phosphate concentrations. Similar reports on increase in EPS production influenced by high phosphate concentration have been reported in *Sinorhizobium meliloti* where 20–100 mM phosphate enhanced the production of succinoglycan which is required for invasion of root nodules in alfalfa [22].

3.3. Effect of Heavy Metals on EPS Production. In view of the ability of *Cupriavidus* KPS 201 to tolerate high nickel concentrations, the effect of presence of increasing nickel concentrations on EPS production was tested. Results show that with increase in Ni concentration in the medium, EPS yield gradually declined (Figure 6) illustrating that presence of Ni has a negative influence on EPS production in this metal-resistant bacterium. Hence, production of EPS may not be a possible mechanism of Ni resistance in *Cupriavidus* KPS 201. In contrast, several bacterial isolates resistant to metals show production of high amount to exopolymers during growth in presence of heavy metals [23, 24] and in some cases the available metals get bound to the carbohydrate or protein moiety of the EPS remaining outside the bacterial cell [9].

3.4. Compositional Analysis of EPS. Compositional analysis of the purified EPS obtained from KPS 201 cells showed that it is a homopolymer of rhamnose. Apart from this, a small
amount of uronic acid (105.5 mg/g), protein (139.5 mg/g), and nucleic acid (1.8 mg/g) was also detected. FTIR spectrum of purified EPS prepared as KBr pellet is illustrated in Figure 7. The spectrum shows characteristic absorption peaks at wavenumbers 3,305, 2,961, 1,655, and 1,070 cm\(^{-1}\). The broad O–H stretching absorption can be observed at 335.6 cm\(^{-1}\) and a weak C–H aliphatic stretch was observed at 2961.3 cm\(^{-1}\) indicating presence of lipids. The medium stretching bands at 1655, 1547.2, and 1070 cm\(^{-1}\) may be assigned to the N–H bending and C–N amine group of proteins. The weak bands between 1300 and 900 may be consigned to carbohydrates and nucleic acids [25].

3.5. Visualization of EPS with Fluorescently Labeled Lectins. EPSs are known to be the main components behind structural and functional properties of bacterial biofilms. Fluorescent microscopy allows nondestructive visualization of the biofilm EPS which get disturbed during isolation and extraction process. In the present study, glass surface was used as the substratum for biofilm development. The DNA and EPS of...
Figure 3: Effect of gluconate concentration on growth and EPS production by Cupriavidus KPS 201.

Figure 4: Effect of nitrogen concentration on growth and EPS production by Cupriavidus KPS 201.

Figure 5: Effect of phosphate concentration on growth and EPS production by Cupriavidus KPS 201.

Figure 6: Effect of increasing Ni concentration on growth and EPS production by Cupriavidus KPS 201.
bacterial cells (KPS 201) were stained with 4′,6-diamidino-2-phenylindol (DAPI) and the fluorescein isothiocyanate-labeled lectin concanavalin A (FITC-ConA), respectively, and visualized by epifluorescence microscopy (Figure 8). The use of fluorescently labeled lectins allowed the microscopically in situ detection of EPS (in green filter) and their distribution in pure culture biofilms (cells visualized under blue filter) of different bacteria [26].

4. Conclusion

The present study demonstrates growth associated production of extracellular polymeric substances (EPS) by serpentine rhizobacterium, *Cupriavidus pauculus* KPS 201, and its role in the biofilm formation. Optimization of cultural conditions shows that EPS production is dependent on temperature, initial pH, and concentrations of carbon, nitrogen, and phosphate in the medium. EPS production, however, is negatively influenced by the presence of nickel in the growth medium revealing that EPS has no role in conferring high degree of resistance to nickel in the serpentine isolate. Nevertheless, the interaction of purified polymer with heavy metals needs to be studied extensively for bioremediation approaches.

Acknowledgments

The authors thank Professor Wolfgang Sand, University of Hamburg, Germany (presently in University of Duisburg, Germany) for providing necessary help and suggestion in this research. The authors acknowledge Prof. A. N. Patra, Department of Chemistry, University of Calcutta, Kolkata, for FTIR spectroscopy analysis. Financial support to A. Pal from German Academic Exchange Service (DAAD), Bonn, Germany, is duly acknowledged.

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