Research Article

Structural, Linear, and Nonlinear Optical and Mechanical Properties of New Organic L-Serine Crystal

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Received 29 November 2013; Accepted 31 December 2013; Published 11 February 2014

Academic Editor: Iwan Kityk

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Nonlinear optical single crystal of organic amino acid L-Serine (LS) was grown by slow evaporation technique. Solubility study of the compound was measured and metastable zone width was found. Single crystal X-ray diffraction study was carried out for the grown crystal. The linear and nonlinear optical properties of the crystal were confirmed by UV-Vis analysis and powder SHG tester. FT-IR spectrum was recorded and functional groups were analyzed. Vickers’ microhardness studies showed the mechanical strength of the grown crystal. Laser damage threshold value of the crystal was calculated. Photoconductivity studies reveal the conductivity of the crystal.

1. Introduction

Nonlinear optical (NLO) materials, with high frequency conversion efficiency have the most required properties for optical communication and optical storage devices, and numerous device applications [1, 2]. Amino acids family crystals possess high NLO efficiency because of their non-centrosymmetric space group and chiral carbon atom [3, 4]. Second order molecular nonlinearity can be enhanced by large delocalized \( \pi \)-electron system with strong donor and acceptor groups [5, 6]. Amino acids contain a proton donor carboxyl acid (\( \text{coo}^- \)) group and the proton acceptor (\( \text{NH}_2^+ \)) with them. Molecular hyperpolarizability \( \beta \) is the basis of a strong second harmonic generation response. Organic molecules usually exhibit large \( \beta \) value are certainly potential candidate molecules for the NLO materials [7]. Many organic amino acid single crystals, L-Arginine Acetate (LAA) [8], L-Arginine Phosphate (LAP) [9], L-Alanine Acetate, and L-Histidine Acetate, are reported earlier.

L-Serine is an organic amino acid and exists in a zwitterionic form. The crystal structure of L-Serine has been reported [10]. Some of L-Serine based crystals are L-Serine sodium nitrate (LSSN) [11], L-Serine formate (LSF) [12], L-Serine hydrochloride (LSHCL), and L-Serine acetate (LSA) [13], and they have been reported in recent years.

In the present study, a systematic investigation has been, carried out on the growth of L-Serine (LS) single crystal. Lattice parameter values of L-Serine were carried out. Solubility of LS in different solvents is described and the metastable zone width is found out from the curve. FT-IR reveals the various functional groups present in the crystal system. UV-Vis studies were carried out for the grown crystal and the absorbance of the crystal in the visible region is measured. Microhardness, photoconductivity, laser damage threshold (LDT), and NLO efficiency of the grown crystal were also measured.

2. Materials and Methods

2.1. Synthesis and Crystal Growth. Five grams of L-Serine (Merck) salt was taken and dissolved in double ionized distilled water and saturated solution of L-Serine was prepared. The pH value of the saturated solution is 3.7 and it was allowed to evaporate at room temperature. Good quality, highly transparent crystal of L-Serine was harvested in a period of 3 weeks. The photograph of as-grown L-serine single crystal is shown Figure 1.
3. Characterization Studies

3.1. Solubility and Metastable Zone Width Measurement. The solubility of L-Serine was determined in the temperature range from 30°C to 60°C using water, acetone, and methanol as solvents. Initially, the temperature is fixed as 30°C and the LS salt is added step by step into 100 mL of deionized water and in other two solvents. The solution is stirred continuously till the supersaturation is achieved. Similarly, the amount of LS salt dissolved in 100 mL of water, acetone, and methanol at various temperatures has been determined. The solubility curves of the LS crystal were drawn and presented in Figure 2(a). It is concluded that solubility of LS in water is much higher than that of acetone and methanol solvents.

Metastable zone width is an essential parameter for the growth of large single crystal from solution. The prepared solution was stirred for about 6 hrs continuously and the temperature is reduced at a rate of 5°C/hr. The temperature at which the first nucleation appeared is noted as nucleation temperature. The difference in nucleation temperature and saturated temperature is regarded to be the metastable zone width of the crystal [14].

The metastable zone width of LS in water solvent is given in Figure 2(b). From the graph, it is clear that metastable zone width decreases with the increase in temperature. Larger zone width at lower temperature shows that the LS crystal is a suitable material to grow by slow evaporation technique.

3.2. Single Crystal XRD Studies. The grown crystal of L-serine was subjected to single crystal XRD analysis to confirm the crystalline nature and to find the cell parameters by employing Enraf Nonius CAD 4 single crystal XRD diffractometer with MoKα (0.71073 Å). From the XRD data, it is observed that LS crystallizes in orthorhombic crystal system with the space group P2₁2₁2₁. The lattice parameter values are in good agreement with the data reported earlier [10]. Table 1 shows the lattice parameter values of LS along with the reported values.

3.3. FT-IR Spectrum. Fourier Transform Infrared analysis was carried out for the grown LS crystal to find the functional groups of the crystal. A fine powdered sample of LS was used for the FT-IR spectral analysis and the spectrum was recorded in the range 450 cm⁻¹-4000 cm⁻¹ using KBr pellet technique. The recorded spectrum is shown in Figure 3.

The absorption peak at 3091 cm⁻¹ is an indication of the presence of NH₃⁺ groups in the crystal. The peaks at 2733 cm⁻¹ and 2566 cm⁻¹ are attributed to the C-H stretching mode vibration. The COO⁻ symmetric stretching vibration modes are confirmed at the peaks 1402 cm⁻¹, 1336 cm⁻¹, and 1371 cm⁻¹. Asymmetric deformation of NH₃⁺ is confirmed by the peak at 1597 cm⁻¹. The strong absorption peak at 1630 cm⁻¹ indicates the presence of primary amino acid group. The multiple combination and overtone bands extended the absorption of hydrogen bonded N–H stretching vibration to 2000 cm⁻¹ [15].

In the overtone region, there is a strong absorption peak at 2040 cm⁻¹, which is assigned to the combination of asymmetrical NH₃⁺ bending vibration (1630 cm⁻¹) and torsional oscillations of NH₃⁺ group (516.92 cm⁻¹).

3.4. Linear and Nonlinear Optical Studies. The optical absorbance studies of LS crystal have been carried out between 190 nm and 500 nm ranges using UV-Vis spectrophotometer. The optical absorbance spectrum of L-Serine crystal is presented in Figure 4. From the spectrum, it is clear that the LS crystal is highly transparent in the entire visible region with the lower cutoff wavelength of 208 nm. Transmittance of the crystal in the entire visible region suggests its suitability for the second harmonic generation [16].

The second harmonic efficiency of L-Serine crystal has been measured by Kurtz-Perry powder technique [17]. The SHG signal was confirmed by the emission of green light (λ = 532 nm) from the sample. The output signal of 46 mV for LS crystal was obtained for an input of 30 mJ/pulse compared with the SHG output signal of standard KDP crystal of 55 mV for the same input energy. The SHG efficiency of the grown LS crystal is 0.84 times that of KDP crystal.
3.5. Microhardness. Micro hardness test is the suitable method to find the mechanical property of the materials [18]. Mechanical strength of the crystal has been carried out using Vickers’ micro hardness tester. Smooth surface of LS crystal is subjected to hardness studies at room temperature. For a time interval of 5 s, different load is applied to the crystal. The microhardness number \( H_v \) is calculated using the relation

\[
H_v = \frac{1.855P}{d^2} \text{ in kg/mm}^2.
\]

A graph between hardness number \( H_v \) and applied load is drawn and shown in Figure 5. From the graph, it can be observed that the hardness increases with the increase in load. The hardness number varies for the planes (1 0 0) and (0 1 0) of the same applied load. The maximum \( H_v \) for (1 0 0) plane is 42.7 kg/mm\(^2\) for the load of 50 g and the \( H_v \) for (0 1 0) plane is 57.8 kg/mm\(^2\) for the load of 70 g.

The result shows that the hardness for (0 1 0) plane is higher than (1 0 0) plane. Thus, it is clear that the LS crystal possesses anisotropy property. The anisotropy coefficient was calculated from the formula

\[
A = \frac{\nabla H_v}{H_v},
\]

where \( \nabla H_v \) is the difference in hardness number and \( H_v \) is the maximum hardness number [19]. The anisotropy coefficient of the crystal is calculated as 26.12%. By plotting \( \log d \) versus \( \log P \), the working hardness coefficient “\( n \)” of the crystal is to be found, where \( P \) is the applied load and \( d \) is the average diagonal length. From the slope, “\( n \)” value is calculated as
1.53. According to Onitsch, $n$ lies between 1 and 1.6 for hard materials and $n$ is greater than 1.6 for soft materials [20]. Thus, LS belongs to hard material category (Figure 6).

3.6. Laser Damage Threshold (LDT) Studies. Laser damage is one of the main criteria for selecting NLO material, since in nonlinear process high intensities are involved. Well-polished surface of the crystal was subjected to Q switched Nd:YAG laser for 20 ns pulse width to do laser damage threshold study.

The output laser intensity is delivered to the sample. The energy density of the laser input beam for which the crystal gets damaged was recorded. The surface power density of the crystal was calculated using the following relation:

$$P_{(d)} = \frac{E}{\tau A},$$

where $P_{(d)}$ is the power density, $E$ is energy of the laser beam in mJ, $\tau$ is the pulse width, and $A$ is the area of the circular spot size of the crystal. The calculated LDT value for the grown LS crystal is 2.17 GW/cm$^2$. The laser damage threshold of L-Serine is higher than KDP and Urea [21] and the values are shown in Table 2.

3.7. Photoconductivity Studies. Photoconductivity studies were carried out for the LS crystal using Keithley 485 picoammeter at room temperature. The dark current ($I_d$) of the sample was measured using DC power supply and picoammeter. Photo current of the sample was measured using halogen lamp containing iodine vapor. DC supply is increased step by step from 10 V to 100 V and the photo current ($I_p$) was measured.

Figure 7 shows the variation of photo current and dark current as a function of applied field. It is observed from the plot that dark current ($I_d$) and photo current ($I_p$) of the sample increase linearly with the applied field and the photo current is always greater than the dark current. Thus, the material exhibits positive photoconductivity [22].

4. Conclusion

Single crystal of L-Serine was grown by slow evaporation technique and the cell parameters of the crystal confirmed the material. LS crystallizes in orthorhombic crystal system. Solubility study and metastable zone width of the crystal have been found. Linear optical studies show the transparency of
the crystal in the entire visible region. SHG studies confirm
the nonlinearity of the grown crystal by the emission of
green light and the SHG efficiency of the grown LS crystal is
found to be 0.84 times that of KDP crystal. FT-IR spectrum
of the crystal elucidates the presence of various functional
groups in the crystal. The Vickers micro hardness study of
the crystal was carried out and the crystal is found to be of hard material category. The grown crystal exhibits
anisotropy. Photo conductivity studies reveal the positive
photoconductivity of the crystal. The laser damage threshold
of L-Serine is higher than KDP and Urea.

Conflict of Interests

The authors declare that there is no conflict of interests
regarding the publication of this paper.

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