Research Article

Structural, Elastic, and Electronic Properties of Antiperovskite Chromium-Based Carbides $ACCr_3$ (A = Al and Ga)

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We theoretically investigated antiperovskite chromium-based carbides $ACCr_3$ through the first-principles calculation based on density functional theory (DFT). The structure optimization shows that the lattice parameter of $ACCr_3$ is basically proportional to the radius of A-site elements. The calculated formation energies show that $AlCCr_3$ and $GaCCr_3$ can be synthesized at ambient pressure and are stable with nonmagnetic ground states. Based on the calculation of elastic constants, some elastic, mechanical, and thermal parameters are derived and discussed. $AlCCr_3$ and $GaCCr_3$ show ductile natures and may have similar thermal properties. From the analysis of the electronic structures, it was found that there are electron and hole bands that cross the Fermi level for $AlCCr_3$ and $GaCCr_3$, indicating multiple-band natures. The Fermi level locates at the vicinity of the density of states (DOSs) peak, which leads to a large DOS at Fermi level dominated by Cr-3d electrons. The band structures of $AlCCr_3$ and $GaCCr_3$ are very similar to those of the superconducting antiperovskite $MgCNi_3$. The similarity may make $AlCCr_3$ and $GaCCr_3$ behave superconductively, which needs to be further investigated in theoretical and experimental studies.

1. Introduction

Recently, antiperovskite compounds AXM_3 (A, main group elements; X, carbon, boron, or nitrogen; M, transition metal) have attracted considerable attention. Due to the high concentration of transition metals in a cell, it can be deduced that interesting properties will be found in the family of compounds. In the antiperovskites family, nickel-based and manganese-based antiperovskites were extensively studied. Abundant physical properties were shown in the two kinds of compounds, such as superconductivity [1–3], giant magnetoresistance (MR) [4, 5], large negative magnetocaloric effect (MCE) [6, 7], giant negative thermal expansion [8, 9], magnetostriction [10], and nearly zero temperature coefficient of resistivity [11, 12]. But there are only a few reports about other 3d-metal-based antiperovskites so far. The difficulty restricting researchers is the exploration of new materials that can be experimentally synthesized. Therefore, theoretical investigations on these potential 3d-metal-based antiperovskites are useful to find the easily prepared stable materials and explore the new physical properties.

In the earlier years, researchers have found that in chromium compounds there are varieties of interesting physical properties. Many of chromium alloys such as Cr-Ru, Cr-Rh, and Cr-Ir alloys show superconductivity [13]. And it was found there is spin density wave antiferromagnetism that coexists with superconductivity in $Cr_{1-x}Re_x$ ($x > 0.17$) [14], $Cr_{1-x}Ru_x$ ($x > 0.17$) [15], and $(Cr_{1-x}Mo_x)_{75}Ru_{75}$ ($x = 0, 0.03, 0.06$, and 0.10) [16]. For chromium-based antiperovskites, Wiendlocha et al. [17] discussed the possibility of superconductivity in $GaNCr_3$ and $RhNCr_3$, and recent phonon and electron-phonon coupling calculation of $RhNCr_3$ [18] supports Wiendlocha et al.'s prediction. In the present work, we theoretically investigated antiperovskite chromium-based carbides $ACCr_3$ through the first-principles calculation based on density functional theory (DFT). The optimized lattice parameter of $ACCr_3$ is basically proportional to the radius of A-site elements. From the analysis of formation energies, we predict that only $AlCCr_3$ and $GaCCr_3$ can be synthesized at ambient pressure, and they are stable with nonmagnetic ground states. The elastic and electronic properties of the two compounds are specifically discussed.

TABLE 1: The coefficient A_2 in (1) as combinations for second order elastic constant for cubic crystal. η_α denotes the Lagrangian strain tensors in terms of ξ.

Strain type	A_2
$\eta_1 = (\xi, 0, 0, 0, 0, 0)$	C_{11}
$\eta_2 = (\xi, \xi, 0, 0, 0, 0)$	$2C_{11} + 2C_{12}$
$\eta_3 = (\xi, \xi, \xi, 0, 0, 0)$	$C_{11} + 4C_{44}$

$AlCCr_3$ and $GaCCr_3$ show ductile natures and may have similar thermal properties. The electron and hole bands cross the Fermi level, implying the multiple-band nature of $AlCCr_3$ and $GaCCr_3$. The Fermi level locates at the vicinity of the density of states (DOS) peak, which leads to a large DOS at Fermi level dominated by Cr-3d electrons. The bands properties of $AlCCr_3$ and $GaCCr_3$ are very similar to those of superconducting antiperovskite $MgCNi_3$. The similarity may make $AlCCr_3$ and $GaCCr_3$ show superconductivity.

2. Computational Details

The calculations were performed by projected augmented-wave (PAW) [19, 20] method using the ABINIT code [21–23]. The PAW method can lead to very accurate results comparable to other all-electron methods. For the exchange-correlation functional, the generalized gradient approximation (GGA) according to the Perdew-Burke-Ernzerhof [24] parametrization was used. Electronic wavefunctions are expanded with plane waves up to an energy cutoff of 1200 eV. Brillouin zone sampling is performed on the Monkhorst-Pack (MP) mesh [25] of $16 \times 16 \times 16$. The self-consistent calculations were considered to be converged when the total energy of the system was stable within 10^{-6} Ha. Nonmagnetic, ferromagnetic (FM), and antiferromagnetic (AFM) states were tested in the study. For AFM states, we only considered the simplest case: the spins of Cr atom in [111] layers are parallel with each other in the same layer and are antiparallel with spins in the neighboring layers.

The elastic constants are evaluated according to the finite-strain continuum elasticity theory [26, 27]. The strain energy per unit mass can be expressed as a polynomial of the strain parameter ξ [28]:

$$\phi(\xi) = \frac{1}{2}A_2\xi^2 + \frac{1}{6}A_3\xi^3 + O(\xi^4), \tag{1}$$

where the coefficients A_2 and A_3 are combinations of second- and third-order elastic constants of the crystal, respectively. For a cubic structure, there are three independent second-order elastic constants (C_{11}, C_{12}, and C_{44}). Therefore, here we introduced 3 Lagrangian strain tensors labeled as η_α in terms of ξ. η_α and the corresponding coefficients A_2 are listed in Table 1.

Once the single-crystal elastic constants are calculated, the related elastic and mechanic parameters may also be evaluated. Using the Voigt-Reuss-Hill approximation [29],

bulk modulus B and shear modulus G of cubic crystal can be defined as

$$B = \frac{C_{11} + 2C_{12}}{3}, \tag{2}$$

$$G = \frac{G_V + G_R}{2}, \tag{3}$$

where the subscripts V and R denote the Voigt, and Reuss average, respectively. In (3),

$$G_V = \frac{C_{11} - C_{12} + 3C_{44}}{5},$$

$$G_R = \frac{5(C_{11} - C_{12})C_{44}}{4C_{44} + 3(C_{11} - C_{12})}. \tag{4}$$

Thus Young's modulus E and Poisson's ratio σ can be obtained using the relations

$$E = \frac{9BG}{3B + G},$$

$$\sigma = \frac{3B - 2G}{2(3B + G)}. \tag{5}$$

Using the calculated bulk modulus B, shear modulus G, and Young's modulus E, the Debye temperature Θ_D can be determined as

$$\Theta_D = \frac{h}{k_B}\left(\frac{3n\rho N_A}{4\pi M}\right)^{1/3} v_m, \tag{6}$$

where h, k_B, and N_A are Planck's, Boltzmann's constants, and Avogadro's number, respectively. ρ is the mass density, M is the molecular weight, and n is the number of atoms in the unit cell. The mean sound velocity is defined as

$$v_m = \left[\frac{1}{3}\left(\frac{2}{v_t^3} + \frac{1}{v_l^3}\right)\right]^{1/3}, \tag{7}$$

where v_l and v_t are the longitudinal and transverse sound velocities, which can be obtained from bulk modulus B and shear modulus G:

$$v_l = \left(\frac{3B + 4G}{3\rho}\right)^{1/2},$$

$$v_t = \left(\frac{G}{\rho}\right)^{1/2}. \tag{8}$$

3. Results and Discussion

3.1. Ground State and Formation Energy. The structures were fully optimized with respect to lattice parameter and atomic positions. The optimized lattice parameters of $ACCr_3$ for different A-site elements are listed in Table 2. The results show that the lattice parameter of $ACCr_3$ is basically proportional to the radius of A-site elements (see Figure 1).

The common technique for producing antiperovskite carbides is the solid-state synthesis from the parent materials.

TABLE 2: Lattice parameter a, magnetic moments per Cr atom M, and the formation energies ΔE of ACCr$_3$.

	a (Å)	ΔE_{NM} (eV/atom)	ΔE_{FM} (eV/atom)	M_{FM} (μ_B/Cr)	ΔE_{AFM} (eV/atom)	M_{AFM} (μ_B/Cr)
CuCCr$_3$	3.869	0.214	0.212	0.396	0.205	1.006
ZnCCr$_3$	3.894	0.125	0.126	0.127	0.115	1.020
GaCCr$_3$	3.891	−0.019	−0.019	0.001	−0.019	0.099
MoCCr$_3$	3.898	0.105	0.105	0.053	0.105	0.240
AlCCr$_3$	3.887	−0.0479	−0.0478	0.034	−0.0478	0.212
AgCCr$_3$	3.936	0.286	0.292	0.544	0.268	1.649
CdCCr$_3$	3.973	0.256	0.257	0.341	0.213	2.102
SnCCr$_3$	3.981	0.082	0.0786	0.400	0.082	0.000
MgCCr$_3$	3.952	0.220	0.219	0.173	0.195	1.824
InCCr$_3$	3.999	0.187	0.189	0.406	0.188	0.006
YCCr$_3$	4.088	0.269	0.268	0.611	0.270	0.000
LaCCr$_3$	4.122	0.496	0.362	1.8101	0.497	0.002
CaCCr$_3$	4.083	0.539	0.541	0.3311	0.539	0.001
SrCCr$_3$	4.155	0.795	0.795	0.402	0.796	0.002
BaCCr$_3$	4.246	1.166	0.998	1.957	1.166	0.002

TABLE 3: Elastic constants and derived quantities for ACCr$_3$ (A = Al and Ga).

	C_{11} (GPa)	C_{12} (GPa)	C_{44} (GPa)	A	C_p (GPa)	B (GPa)	G (GPa)	B/G	E (GPa)	σ	Θ_D (K)
AlCCr$_3$	448.18	297.98	68.31	0.91	229.67	348.04	70.95	4.90	199.30	0.40	532.43
GaCCr$_3$	480.81	314.43	64.63	0.78	249.80	369.89	71.51	4.40	201.53	0.41	497.12
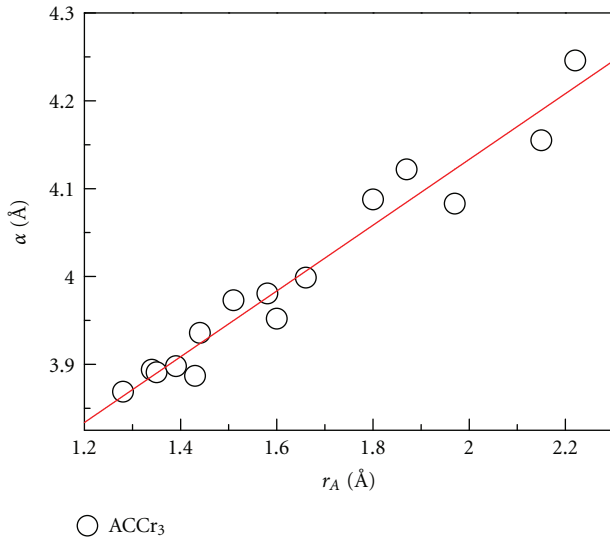
FIGURE 1: Relation between the lattice parameter a and the radius of A-site elements r_A.

Therefore we assumed that the antiperovskite chromium-based carbides ACCr$_3$ could be synthesized in the same routine. We analyze the stability of compounds by calculating the formation energies ΔE, which is defined as the difference between the total energies of initial reagents (elements) and final compounds:

$$\Delta E = E_{tot}^{ACCr_3} - \left(E_{tot}^{A} + E_{tot}^{C} + 3E_{tot}^{Cr} \right). \quad (9)$$

Here, E_{tot}^{A}, E_{tot}^{C}, and E_{tot}^{Cr} are the total energies of the initial reagents.

According to the definition of ΔE, negative values indicate that the formation of compounds from the initial components is energetically favorable and the compounds are stable with respect to a mixture of the initial elements. Conversely, if $\Delta E > 0$, the compounds are unlikely to be synthesized under normal conditions.

The calculated ΔE for the series of A-site elements with different magnetic configurations are listed in Table 2. Except for AlCCr$_3$ and GaCCr$_3$, all other compounds show positive ΔE. For AlCCr$_3$ and GaCCr$_3$, the values of ΔE for different magnetic configurations are very close, and the magnetic moments per Cr atom for the FM and AFM states are very small, which implies the nonmagnetic ground states of AlCCr$_3$ and GaCCr$_3$. From the analysis of formation energies, it seems that ACCr$_3$ can be synthesized only when A = Al and Ga. The elastic and electronic properties of the two compounds will be discussed later.

3.2. Elastic and Mechanical Properties. In order to obtain accurate elastic constants, ξ is varied with a step of 0.0025 in every case for η_α. As an example, the strain energies for AlCCr$_3$ and the fitted polynomials are shown in Figure 2. The calculated elastic constants at ambient pressure and other derived mechanical parameters are listed in Table 3. From the calculated values of the elastic constants, one can find that they satisfy the mechanical stability criteria for a cubic crystal, that is, $C_{11} > C_{12}$, $C_{44} > 0$, and $C_{11} + 2C_{12} > 0$ [30], which indicates that the compounds are mechanically stable.
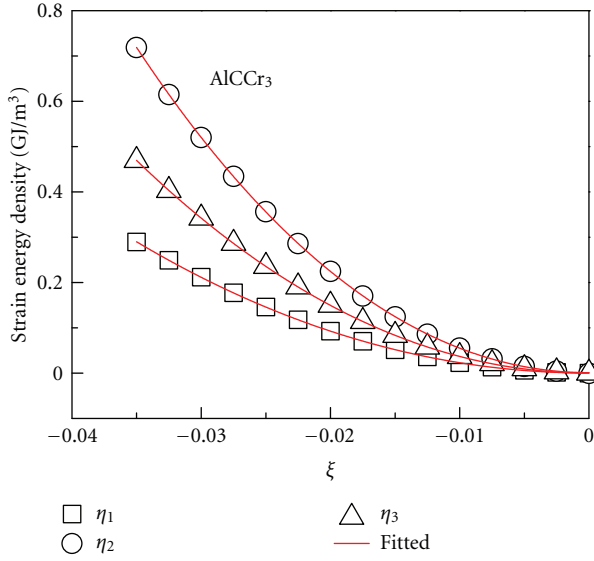
FIGURE 2: The strain-energy relation of AlCCr$_3$.

Cracks in crystals are directly related to the anisotropy of thermal and elastic properties [31]. The elastic anisotropy factor A of cubic crystal can be described as

$$A = \frac{2C_{44}}{C_{11} - C_{12}}. \qquad (10)$$

For a complete isotropic system, the anisotropy factor A takes the value of unity, and the deviation from unity measures the degree of elastic anisotropy [32]. The elastic constant C_{44} represents the shear resistance in a [100] direction, which is related to the C–Cr bonding. The values of C_{44} for the two compounds are almost the same, which implies the similar C–Cr bonding nature of the two compounds. $C_{11} - C_{12}$ turns out to be the stiffness associated with a shear in a [110] direction, which can be connected with the bonding between A–Cr bonding. Although A–M bonding is usually very weak in antiperovskite ACM$_3$, it still can influence the anisotropy of the material. The anisotropic factors A are listed in Table 3. The different A–Cr bondings of the two compounds lead to different anisotropy. For AlCCr$_3$, A = 0.91 indicates that AlCCr$_3$ is almost isotropic. However, GaCCr$_3$ seems like more anisotropic (A = 0.78).

Young's modulus E defines the ratio between the linear stress and strain. The larger the value of E, the stiffer is the material. In general, as Young's modulus increases, the covalent nature of the compound also increases, which further has an impact on the ductility of the compounds. The values of E for AlCCr$_3$ and GaCCr$_3$ are very close to each other, suggesting the similar covalent nature of the two compounds.

The Cauchy's pressure C_p is defined as the difference between the two particular elastic constants $C_p = C_{12} - C_{44}$. It is considered to serve as an indication of ductility: $C_p > 0$ suggests that the material is expected to be ductile; otherwise, the material is expected to be brittle [33]. As shown in Table 3, the values of C_p for AlCCr$_3$ and GaCCr$_3$ are positive,

which is a clear indication for the compounds to be ductile. Another index of ductility is the ratio B/G. According to Pugh [34], the ductile/brittle properties of materials could be related empirically to the ratio B/G. If $B/G > 1.75$, the material would be ductile; otherwise the material behaves in a brittle manner. For ACCr$_3$ (A = Al and Ga), all the calculated B/G ratios are much larger than 1.75, which clearly highlights the ductile nature of ACCr$_3$ (A = Al and Ga). Poisson's ratio generally quantifies the stability of the crystal against shear and takes the value between −1 and 0.5, which are the lower and the upper bounds. The lower bound is where the material does not change its shape, and the upper bound is where the volume remains unchanged. The calculated σ of ACCr$_3$ (A = Al and Ga), listed in Table 3, are very close to the typical σ value of 0.33 for ductile metallic materials [35]. All the parameters clearly show the ductility of ACCr$_3$ (A = Al and Ga).

Using bulk modulus B, shear modulus G, and Young's modulus E, the Debye temperature Θ_D can be calculated. As a matter of fact, a higher Θ_D would imply a higher thermal conductivity associated with the material [36]. For the present calculation, the Θ_D are estimated to be 532.43 K and 497.12 K for AlCCr$_3$ and GaCCr$_3$, respectively. The similar values of Θ_D suggest the similar thermal characteristics of the two compounds.

3.3. Electronic Properties. The calculated electronic band structures along the high symmetry directions in the Brillouin zones together with the total and site-projected l-decomposed DOS at equilibrium lattice parameters for ACCr$_3$ (A = Al and Ga) are shown in Figures 3 and 4. The band structures of the two compounds are very similar. Bands from −13 eV to −10 eV are mainly from the C-2s and Cr-3d states with strong hybridizations characteristic. From −8 eV to 2 eV, there are hybridizations between C-2p and Cr-3d states, and the Cr-3d states contribute dominantly to the bands, which suggests the itinerant nature of Cr-3d electrons. From Figure 4, it can be seen the Fermi level locates at the vicinity of the DOS peak, which leads large DOS at the Fermi level $N(E_F)$ with values of 4.79 states/eV for AlCCr$_3$ and 5.62 states/eV for GaCCr$_3$.

The Fermi surfaces of the two compounds are similar (see Figure 5). There are four bands that cross the Fermi level. Hole pockets surrounding Γ point come from the lower two bands (denoted with red and green colors in Figure 3). Electron pockets surrounding R point come from the upper two bands (denoted with blue and purple colors in Figure 3). The differences of the Fermi surface of the two compounds come from the band denoted with blue color in Figure 3. It crosses the Fermi level at the vicinity of M point for AlCCr$_3$, which makes the electron pockets surround the corners of the Brillouin zones connected to each other at M point. However, for GaCCr$_3$, it crosses the Fermi level in the place between Γ and M points, which forms a cubic cage-like electron pocket surrounding Γ point and connecting with the corner-centered electron pockets. The presence of electron and hole bands crossing the Fermi level indicates the multiple-band natures for AlCCr$_3$ and GaCCr$_3$.
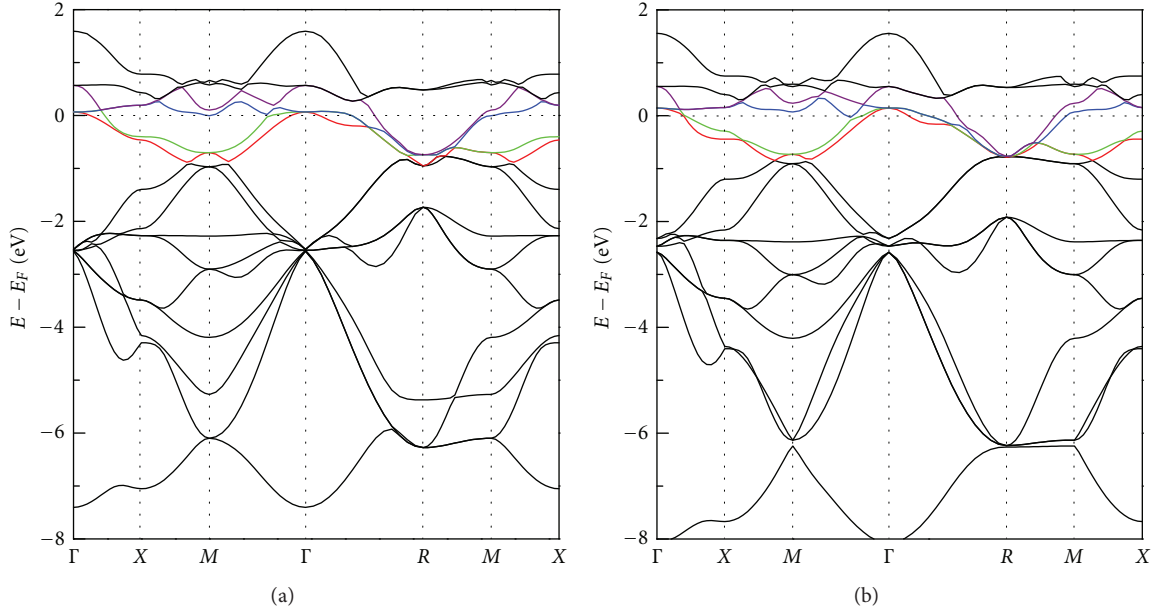
FIGURE 3: The band structures for (a) AlCCr$_3$ and (b) GaCCr$_3$. The colored lines denote the bands crossing the Fermi level.

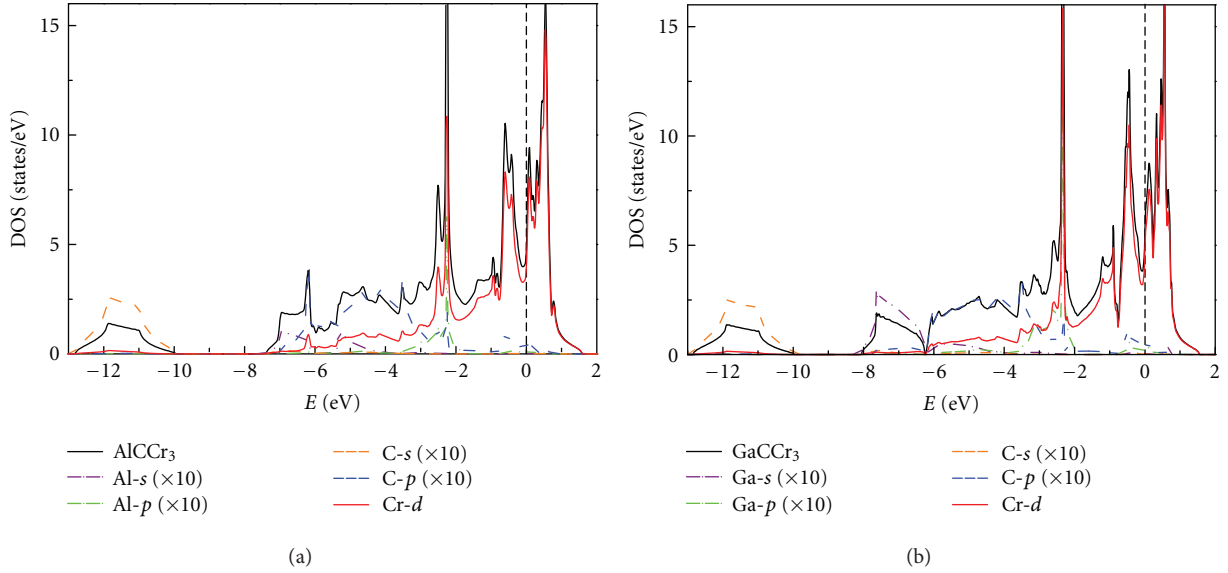
FIGURE 4: The density of states of (a) AlCCr$_3$ and (b) GaCCr$_3$.

In order to understand the bonding nature among the atoms in AlCCr$_3$ and GaCCr$_3$, we plot the contour maps of the charge density of the two compounds in Figure 6. The C–Cr bonds are very strong, which coincides with the strong hybridization between C-2p and Cr-3d electrons shown in DOS figures. The similarity of the bonding nature for the two compounds coincides with the similar Young's modulus E discussed before.

The electronic properties of AlCCr$_3$ and GaCCr$_3$ are very similar to those of the superconducting antiperovskite MgCNi$_3$ [37], for which the Fermi level locates at the vicinity of the DOS peak as well, and there is multiple-band

nature present. The similarity possibly makes AlCCr$_3$ and GaCCr$_3$ potential superconductors. According to McMillan's coupling theory [38], the electron-phonon coupling constant λ can be calculated from the following expression $\lambda = \sum_\alpha (\eta_\alpha / M_\alpha \langle \omega_\alpha^2 \rangle)$, where M_α is the atomic mass and $\langle \omega_\alpha^2 \rangle$ is the averaged phonon frequency and can be approximated as $\langle \omega_\alpha^2 \rangle = \Theta_D^2 / 2$. For superconducting MgCNi$_3$, there is a peak (a van Hove singularity) in DOS just below Fermi level, which leads to a large $N(E_F)$. The $N(E_F)$ is not large enough to lead magnetic instability, but it can lead to a large electron-phonon coupling constant [39]. To the best of our knowledge, except for the superconducting nickel antiperovskites, there are no
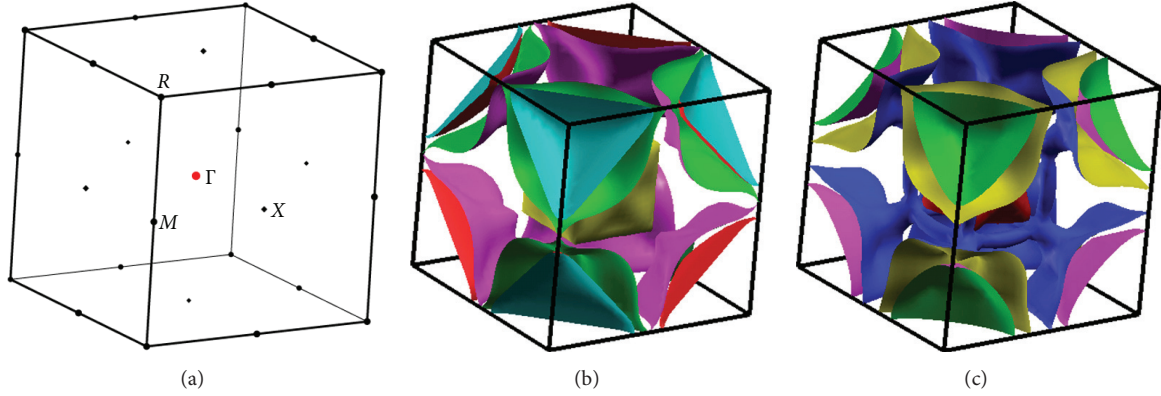
FIGURE 5: (a) Brillouin zone of ACCr$_3$. Fermi surfaces of (b) AlCCr$_3$ and (c) GaCCr$_3$.
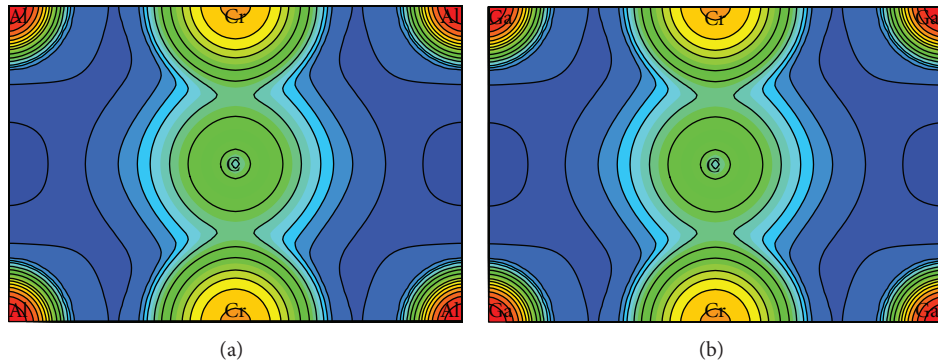

FIGURE 6: The contour map of electron charge density of (a) AlCCr$_3$ and (b) GaCCr$_3$.

other antiperovskites that have such a nature. Wiendlocha et al. [17] pointed out that in chromium systems, chromium atoms have very large values of the McMillan-Hopfield parameters [38], which may lead to a very strong electron-phonon coupling. The recently reported work about RhNCr$_3$ [18] supports Wiendlocha et al.'s prediction. The phonon and electron-phonon coupling calculations show that RhNCr$_3$, which has a large $N(E_F)$, is a strong coupling superconductor with Tc above 16 K. Therefore, we consider that there is possibility for superconductivity appearing in AlCCr$_3$ and GaCCr$_3$. Phonon and electron-phonon coupling calculations will be carried out in the future to confirm the possibility.

4. Conclusion

In conclusion, we theoretically investigated the antiperovskite chromium-based carbides ACCr$_3$ through the first-principles calculation based on density functional theory. The optimized lattice parameter of ACCr$_3$ is basically proportional to the radius of A-site elements. Only AlCCr$_3$ and GaCCr$_3$ have negative formation energies, implying that the two compounds can be synthesized at ambient pressure and are stable with nonmagnetic ground states. AlCCr$_3$ and GaCCr$_3$ show ductile natures, and they may have similar thermal properties. Similar to superconducting antiperovskite MgCNi$_3$, there are electron and hole bands that cross the Fermi level for AlCCr$_3$

and GaCCr$_3$, indicating multiple-band natures. The Fermi level locates at the vicinity of the DOS peak, which leads to a large $N(E_F)$ dominated by Cr-3d electrons. These similarities possibly make AlCCr$_3$ and GaCCr$_3$ show superconductivity, which needs to be further investigated from theoretical and experimental studies.

Acknowledgments

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