Research Article

Tuning Rheological Performance of Silica Concentrated Shear Thickening Fluid by Using Graphene Oxide

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The addition of a small amount of graphene oxide into a traditional colloidal silica-based shear thickening fluid (STF) can lead to a significant change in viscosity, critical shear rate, storage modulus, and loss modulus of STF. This finding provides an effective way to prepare stronger and light-weight STFs.

1. Introduction

Graphene oxide (GO) is a graphene sheet functionalized with various oxygen-containing functional groups, which inherits many merits from graphene, including exceptionally high mechanical strength, and high aspect ratio/specific surface area but it can be well dispersed in some solvents including water [1–3]. GO has attracted tremendous interest in the last few years, because of not only its use as a precursor for graphene but also its unique colloidal behaviours, such as amphiphilic property, liquid crystal, and viscoelastic behaviour [1, 4–8]. In particular, the unique viscoelastic behavior of GO dispersion is fundamentally different from traditional viscoelastic materials such as polymers, wherein GO is able to form a gel at very low concentrations as the outstanding reinforcing effect is ascribed to a mechanical percolation phenomenon [2, 9–11]. Understanding the viscoelastic behaviour of GO is both of theoretical and practical importance in synthesis of graphene-based materials, including graphene fibers/foams and its composite materials [12–16]. Despite the significant work that has been done on exploring the colloidal behaviour of GO itself [11], the understanding on rheological behaviour of GO in complex colloidal systems remains largely unexplored. Shear thickening is one of the most interesting rheological phenomena of colloidal suspensions observed over the past decades [17–20]. Shear thickening, by definition, is a dramatic increase in viscosity when the shear rate of certain mechanical deformation/perturbation increases to a critical value. This unique feature enables the use of STFs in body armours and energy absorbers, as it can provide advantages of light-weight and flexibility compared with conventional fibre-based body armours [21]. The shear thickening phenomenon has previously been observed in various fluids or suspensions and several mechanisms have been formulated involving the order-disorder transition [22, 23], impact activated solidification [24], and hydrodynamic formation of clusters [19, 25–27]. For the reversible shear thickening of colloidal suspensions, the widely accepted mechanism is the shear-induced hydroclusters transitionally sized together by hydrolubrication forces [27, 28]. As shown in Figures S1(a) and (b) in Supplementary Material available online at http://dx.doi.org/10.1155/2015/734250, shear thickening behaviour is caused by the formation of a clustering structure at the higher strain rate, hindering the movement of free particles through it [18, 27]. At present, the most studied STF is based on the suspension of silica in polyethylene glycol (PEG) matrix. The hydrophilic fumed silica nanoparticles are stably dispersed in PEG through hydrogen bonding to obtain the silica concentrated STF [28]. It is noteworthy that the shear thickening behaviour is strongly dependent on the particle size and volume fraction [27]. In this work, we explore
how the introduction of two-dimensional nanoparticle, GO sheets, can change the rheological properties of the silica-based STF.

2. Materials and Methods

2.1. Materials. Fumed silica nanoparticles (S5505) and polyethylene glycol (PEG-200) were purchased from Sigma-Aldrich Company. The fumed silica particles were hydrophilic, containing the hydroxyl groups on the surface with crystal size of 14 nm in diameter. The surface area of these nanoparticles is 200 m$^2$/g ± 25 m$^2$/g. The silica primary particles were irreversibly fused into large aggregated structures that cannot be disrupted by shear. GO dispersions were synthesized by using a similar procedure, reported by our group previously [1, 7]. Before the experiments, the GO was ultrasonicated for 10 mins to achieve a uniform dispersion.

2.2. Preparation of STFs. Silica/PEG samples were prepared by adding two different weight percentages of silica (15% and 20%) into PEG-200 and a combination of manual mixing and ultrasonication processes were used. The mixture was stirred manually for 10 minutes firstly and then put in the ultrasonicator for 3 hours until a homogenous mixture was achieved. Silica/GO/PEG samples were prepared by adding different concentrations (weight concentration) of the exfoliated GO to the prepared silica/PEG. The mixture of GO and silica concentrated STF experienced a second dispersion cycle by ultrasonication. After proper mixing, the excess water was removed via an evaporation process by heating up the samples in the oven at 80°C overnight. It was found that the suspensions showed no visible evidence of phase separation, precipitation, or increases in turbidity over long periods of time (four months).

2.3. Characterization. The rheometer (ARES) was employed to determine the rheological performance of silica-based suspension. Both steady shear and dynamic shear methods were carried out at the room temperature. The steady test was conducted by cone-and-plate geometry with 0.02 rad in angle and 25 mm in diameter and the dynamic test was performed by two parallel plates with 25 mm in diameter and 500 mm in gap. All the samples were presheared at shear rate of 1 S$^{-1}$ for 180 s and then were relaxed for 300 s before both steady and dynamic tests. A preshear process is necessary for achieving a better dispersion.

Scanning electron microscope images were captured by JEOL 7001. The samples prepared for SEM were spun from the dilute STF (with and without GO). All samples were deposited on the high conductive silicon wafer.

3. Results and Discussion

The silica-based STF was prepared by dispersing the fumed silica in PEG followed by mechanical stirring. GO was preexfoliated in an aqueous solution with the same method reported in our previous paper [29], followed by mixing with the silica-based STF. After mixing, the water was removed by evaporation and GO sheets were found to be well dispersed in the PEG matrix. The prepared suspension is rather stable under the ambient conditions and no significant colour change and aggregation of GO were observed over four months.

Figure 1 presents the steady shear test results. The silica-based STF follows the typical shear thickening behaviour, experiencing a slight decrease of viscosity with shear rate at the low shear rate region, followed by a dramatic increase of viscosity when the shear rate is increased to a certain point or so-called critical shear rate, after which the viscosity is decreased with the shear rate. The rheological behaviours of STF strongly are dependent on the particle shape, concentration, and the interaction between these elements; hence tuning the silica content is a conventional way to change the rheological behaviour of STF [17, 28]. Figure 1 also illustrates the effect of an increasing silica fraction on the shear rate dependence of viscosity for the silica-based STFs. At the beginning of the steady shear test, the viscosity increases from 16 Pa·s in 15% (w/w) silica sample to 35 Pa·s in 20% (w/w) silica sample. In addition, at 15% (w/w) silica in PEG, a continuous shear thickening region appears at the higher critical shear rates of 16 S$^{-1}$. When the silica fraction is increased to 20% (w/w), the critical shear rate shifts to 10 S$^{-1}$. As the fraction of particles is increased, this shear thickening region progressively shifts to lower shear rates and the increase in viscosity becomes more dramatic. These trends have been observed to be consistent in a range of suspensions and a high concentration of particles is seen as a requirement observation of shear thickening at achievable shear rates [30, 31].

Compared with the silica-only STF, the introduction of GO exhibits a similar viscosity trend as a function of shear rate, but it has led to remarkable changes in the viscosity and critical shear rate with significantly lower concentrations.
As the GO concentration increased from 0% to 0.3%, the viscosity of the fluid increases over one order of magnitude from 16 Pa·s to 168 Pa·s (for the sample containing 15% (w/w) silica). In addition, the critical shear rate, one of the most important parameters in STF, also exhibits a dramatic shift towards the lower critical shear rate in the fluid incorporated with GO. The critical shear rate of 15% (w/w) silica in PEG is around 16 S⁻¹. In comparison, the corresponding critical shear rates of 15% (w/w) silica in PEG with the additional 0.1% (w/w), 0.2% (w/w), and 0.3% (w/w) GO were 3 S⁻¹, 2.5 S⁻¹, and 1 S⁻¹, respectively. The critical shear rate is the onset of the viscosity increase, indicating the formation of a hydrocluster structure which hinders the free particles passing through it. Apparently, the increasing GO content results in the formation of a hydrocluster structure at a lower shear rate.

We have also investigated the dynamic shear behaviour and the results are presented in Figure 2. Both the storage and loss modulus of the STFs are tested as a function of frequency. Comparing the fluid containing only 15% (w/w) silica in PEG with the samples incorporated by small amounts of GO, both the storage and loss moduli are increased over the whole frequency range tested. Particularly, the moduli of the STF with 0.3% (w/w) GO are increased by nearly an order of magnitude. The use of GO in the fluid also shifts the onset of shear thickening behaviour towards the lower frequency compared with the bare fluid of 15% (w/w) silica in PEG. The rapid increase of modulus at the lower frequency can be viewed as being equivalent to the rapid viscosity increase at the critical shear rate under the steady shear sweep according to the equation of Cox-Merz [32].

Both the steady and dynamic rheological tests have indicated that by adding only 0.2% GO into the STF containing 15% of silica, the viscosity, storage, and loss modulus of the fluid can be greatly enhanced to be close to that of the fluid containing 5% more silica (i.e., 20% silica in total). This finding suggests that at least 5% (w/w, calculated from the difference between 15% and 20%) of silica particles can be replaced by a very small amount of GO (0.2%) to achieve a similar rheological performance. This is remarkable as this will lead to the reduction of total weight of the STF, which is favourable to many applications such as body armours.

To further investigate the role of GO in the STFs, we have also examined the viscoelastic behaviour of pure GO dispersions. Our experiment showed that the GO dispersion itself is a shear-thinning fluid (Figure S3), which is in agreement with the literature [33–35]. Thus, it is unlikely that the remarkable change of the rheological behaviour of GO-added STFs is caused by the interaction between GO sheets themselves. The interaction between GO sheets and silica nanoparticles and particularly the large aspect ratio of GO sheets could account for the significant change of the rheological behaviour of GO-added STFs is caused by the interaction between GO sheets themselves. The interaction between GO sheets and silica nanoparticles and particularly the large aspect ratio of GO sheets could account for the significant change of the rheological performance. This is remarkable as this will lead to the reduction of total weight of the STF, which is favourable to many applications such as body armours.
peak power as well; mathematically representation is the product of viscosity times the square of shear rate for the shear thickening fluid and eventually depends on the elastic modulus and strength of transitionally formed hydrocluster structure. Therefore the effects of graphene oxide on the shear thickening fluid is envisaged to be twofold; one is the ability to initiate the shear thickening at a reduced overall weight; the other is similar to that of filler in the polymer composite, the capability to strengthen the transitionally formed hydrocluster structure in order to keep the materials structural integrity after several impacts.

4. Conclusions

In summary, the addition of a small amount of GO into silica/PEG-based STFs can lead to remarkable increase of the viscosity and shift of the shear thickening onset towards lower shear rate. It can also enhance the materials’ storage modulus. Our finding suggests that GO can be utilized as a very efficient nanofiller to modify silica-based STF and provides a new route to design and fabrication of light-weight and high-strength energy absorbers. It also opens a new route to tune the onset of shear thickening to preferred regimes for certain applications. However, further experimental and theoretical study will be required to better understand and interpret the influence mechanism of GO on the rheological properties of the relevant silica suspensions.

Conflict of Interests

The authors declare that there is no conflict of interests regarding the publication of this paper.

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References


Figure 3: SEM images of spin-casted STFs on Si wafer: (a) 15% silica + 0% GO and (b) 15% silica + 0.2% GO.


