

Research Article

Electronic and Magnetic Properties Studies on Mn and Oxygen Vacancies Codoped Anatase TiO₂

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The electronic and magnetic properties of Mn and oxygen vacancies codoped anatase TiO₂ were investigated. The calculated results showed that the TiO₂ codoped with Mn and oxygen vacancies have a magnetic moment value of 3.415 μ_B per Ti₃₁MnO₆₃ supercell. Furthermore, Ti₃₁MnO₆₃ gets the lowest energy with a geometrical optimization where the Mn ions locate at the nearest-neighbor sites of the oxygen vacancy. And experimental results indicated the magnetism is associated with the defects of Mn ions and oxygen vacancies induced by the Mn doping, which is consistent with the calculation results.

1. Introduction

Diluted magnetic semiconductors (DMSs), which are desirable for spintronic applications, have been extensively investigated for their magnetic and transport properties. Oxygen vacancies (V_Os) are important intrinsic defect in TiO₂ which can be manipulated by oxygen gas pressure in the process of the fabrication, and they could alternate the materials' properties significantly [1–10]. The origin of the observed ferromagnetism has been considered to be related closely to the V_Os instead of the cation vacancies reported by Wang et al. [11]. The magnetization enlarges as V_O content increases in oxygen-deficient TiO₂ (TiO_{2-δ}) films observed by Yoon et al. [8]. And the carrier is n-type and the resistivity increases upon cooling [12] according to the transport measurements. Several theoretical investigations employed by first principle have been reported, most of which focus on cation vacancy in TiO₂ bulk materials [13–28]. However, less attention has been drawn to the V_O doped anatase TiO₂ and V_O-Mn codoped anatase TiO₂ in those researches.

There are three questions that should be figured out: (1) what is the electronic orbital composition of a V_O-induced

state in detail for anatase TiO₂? (2) What is the electronic orbital geometry for the V_O-Mn induced states in V_O-Mn codoped anatase TiO₂? (3) What is the most stable structure for V_O-Mn codoped anatase TiO₂? It is necessary to bring further insight into the origin of the ferromagnetism in the TiO₂ with the transition-metal doping, where the V_Os play an important role in promoting long-range ferromagnetic order [29]. In this paper, we investigated the electronic and magnetic properties of Mn and V_Os codoped anatase TiO₂ (Ti₃₁MO₆₃) using the first-principles method based on the density functional theory (DFT). The experimental results are consistent with that of the first principles calculations, and the magnetism is associated with the V_O defect electrons induced by the Mn doping. The connections between the doped Mn ions and V_Os are discussed to explain the ferromagnetism observed in these materials.

2. Experiments and Model Calculations

The undoped TiO₂ films were fabricated by sol-gel spin coating. Step one: Mn-doped TiO₂ sol-gel solution preparation. In this step, titanium isopropoxide (99%), absolute ethanol and

hydrochloric acid were used as starting materials. Firstly, 10 mL titanium isopropoxide was dissolved in 70 mL absolute ethanol solution. The mixture was stirred for 5 min. Then 2 mL of 2 mol/L hydrochloric acid was added to the mixture and was stirred for 5 min. And then 30 mL absolute ethanol solution and proper weight of MnCl_2 (to obtain a doping concentration of 1 mol%, 3 mol%, 5 mol%, 7 mol%, and 9 mol% resp.) were added to the mixture and then were stirred for 24 h. Subsequently, the solution was kept stirred for 2 h with a heating magnetic stirrer until the sol-gel formed. Step two: the preparation of the Mn-doped anatase TiO_2 films using a spin coating method. Firstly, the fluorine-doped tin oxide (FTO) substrates were cleaned in de-ion water and in absolute ethanol solution for 5 min with ultrasonic cleaner, respectively. Then, the anatase TiO_2 films were prepared using TiO_2 sol-gel by the spin coating method. The spin rate was 400 rpm for 10 s and then was extended to 3000 rpm for 25 s. This spin coating procedure was repeated for five times to get the ideal film thickness of 800 nm. After coating, these samples were treated at 450°C in atmosphere for 2 h and cooled down naturally. All the reagents of acetone, absolute ethanol, titanium isopropoxide, manganese dichloride, and hydrochloric acid were of analytical grade without further purification. The surface morphologies of TiO_2 films were observed by the scanning electron microscopy (SEM, S-4800). X-ray photoelectron spectroscopy (XPS, VG Multilab 2000x) was utilized to qualify and the chemical compositions and identify electronic structures. The binding energy of the XPS spectra was calibrated with the reference to the C 1s peak at 284.6 eV. The magnetic properties were investigated using a vibrating sample magnetometer (VSM) equipped in the physical property measurement system (PPMS-9, Quantum Design).

First-principles calculations based on spin-polarized density-functional theory and projector augmented wave (PAW) pseudopotential technique are performed as implemented within the Vienna Ab-Initio Simulation Package (VASP) [30, 31]. The generalized gradient approximation (GGA-PBE) for the wave functions is used with a cutoff of 400 eV to model the exchange and correlation functional [32]. The calculations have been carried out for two cases: (1) an oxygen atom is substituted by a V_O ; (2) a titanium (Ti) atom and an oxygen atom are substituted by a Mn atom and a V_O . The Monkhorst-Pack scheme k -points grid sampling was set to be $2 \times 2 \times 5$ for the 95-atom anatase supercell, $\text{Ti}_{31}\text{MnO}_{63}$ (corresponding to doping content of Mn \sim 3.13%). The valence electrons configuration for the O, Ti, and Mn are $2s^2 2p^4$, $3s^2 3p^6 3d^2 4s^2$, and $3d^5 4s^2$, respectively. All the atomic positions are fully optimized until the atom forces drop below the value $0.02 \text{ eV}/\text{\AA}$. The DFT method has been proven to be one of the most accurate methods for the computation of the electronic structure of solids [4–7].

3. Results and Discussions

3.1. Electronic Properties and Magnetism of a V_O Doped TiO_2 . According to the crystal field theory, due to the hybridization of Ti–O, the Ti 3d orbitals split into two parts, the t_{2g} (d_{xy} , d_{xz} , d_{yz}) and e_g (d_{z^2} , $d_{x^2-y^2}$) states. Geometrically, t_{2g} orbitals

have lobes pointing between oxygen atoms, while e_g orbitals point toward oxygen. The O 2p orbitals split into p_π and p_s states. O 2p and the t_{2g} (d_{xz} , d_{yz}) of Ti 3d devoted to the valence band (p_π devote to the top of valence band), while the conduction band was contributed by the d_{xy} and e_g of Ti 3d (d_{xy} devotes to the bottom of conduction band) [33]. In the ionic limit, each Ti gives four electrons to two oxygen atoms, resulting in the nominal charges of Ti^{4+} and O^{2-} [12]. When the V_O is introduced into the lattice, each V_O is assumed to donate two electrons. For Figure 1(b) (I), the TDOS of V_O -doped anatase TiO_2 , there is a spin-split lower around the Fermi level illustrating the existence of magnetism. Such a state is stabilized in the spin-polarized DFT calculation (with a lower free energy of 0.0042 eV than the result of non-spin-polarized DFT calculation). The calculated net magnetic moment is about $0.533 \mu_B$. Figure 1(a) (II, III, and IV) shows the orbital decomposed density of states for three Ti 3d electrons nearest to the V_O . Three Ti atoms are labeled as Ti-II, Ti-III, and Ti-IV, with Ti-III and Ti-IV being equivalent with respect to the vacancy. It can be seen from the figure that, for Ti-II atom, the exchange spin-splitting at the bottom of the conduction band is mainly derived from the one e_g orbital (d_{xy}) and two t_{2g} orbitals (d_{yz} and d_{x^2}), while the exchange spin-splitting at the bottom of the conduction band is mainly derived from the one e_g orbital (d_{z^2}) and two t_{2g} orbitals (d_{yz} and d_{x^2}) for Ti-III and Ti-IV atom. The orbitals overlap between the three Ti atoms nearest to the V_O . The hybridization between these three Ti 3d electron states leads to the electrons of Ti 3d spin-polarized. The charge distribution is illustrated in Figure 1(b). It can be seen that the state spread over all three Ti atoms adjacent to the vacancy and the local orbitals are the mixture of e_g and t_{2g} . The charge distribution is similar to the charge distribution of the hybrid state in [12]. In the calculations, the results are obtained in a $\text{Ti}_{32}\text{O}_{63}$ anatase supercell with lowest total free energy for different V_O sites. The two electrons donated by one V_O are shared by the three Ti ions with up-spin forming Ti^{3+} ion to get the lowest energy. It is guessed that the appearance of the net magnetic moment in anatase TiO_2 may come from the larger crystal lattice distortion compared to rutile TiO_2 .

3.2. Electronic Properties and Magnetism of V_O -Mn Codoped TiO_2 . Magnetic properties of $\text{Ti}_{31}\text{MnO}_{63}$ were investigated using the first-principles methods based on the DFT. The lowest energies are achieved after geometrical optimization of $\text{Ti}_{31}\text{MnO}_{63}$ for each V_O -Mn codoping models, where doped Mn ions are located at the nearest-neighbor site of V_O . The result is similar to that of the report done recently [34].

In order to understand the nature of ferromagnetism, the electronic properties and magnetism of V_O -Mn codoped TiO_2 are investigated in detail. The lowest energy is obtained in the spin-polarized DFT calculation. Figure 2 shows the spin-resolved total density of states for the $\text{Ti}_{31}\text{MnO}_{63}$. The calculations show ferromagnetism with clear majority-spin defect states localized in the band gap for V_O -Mn codoped TiO_2 . The major magnetic moment originates from the d electron of Mn ions. For V_O -Mn codoped TiO_2 , the valence electrons configuration of doped Mn atom is $3d^5$. And the

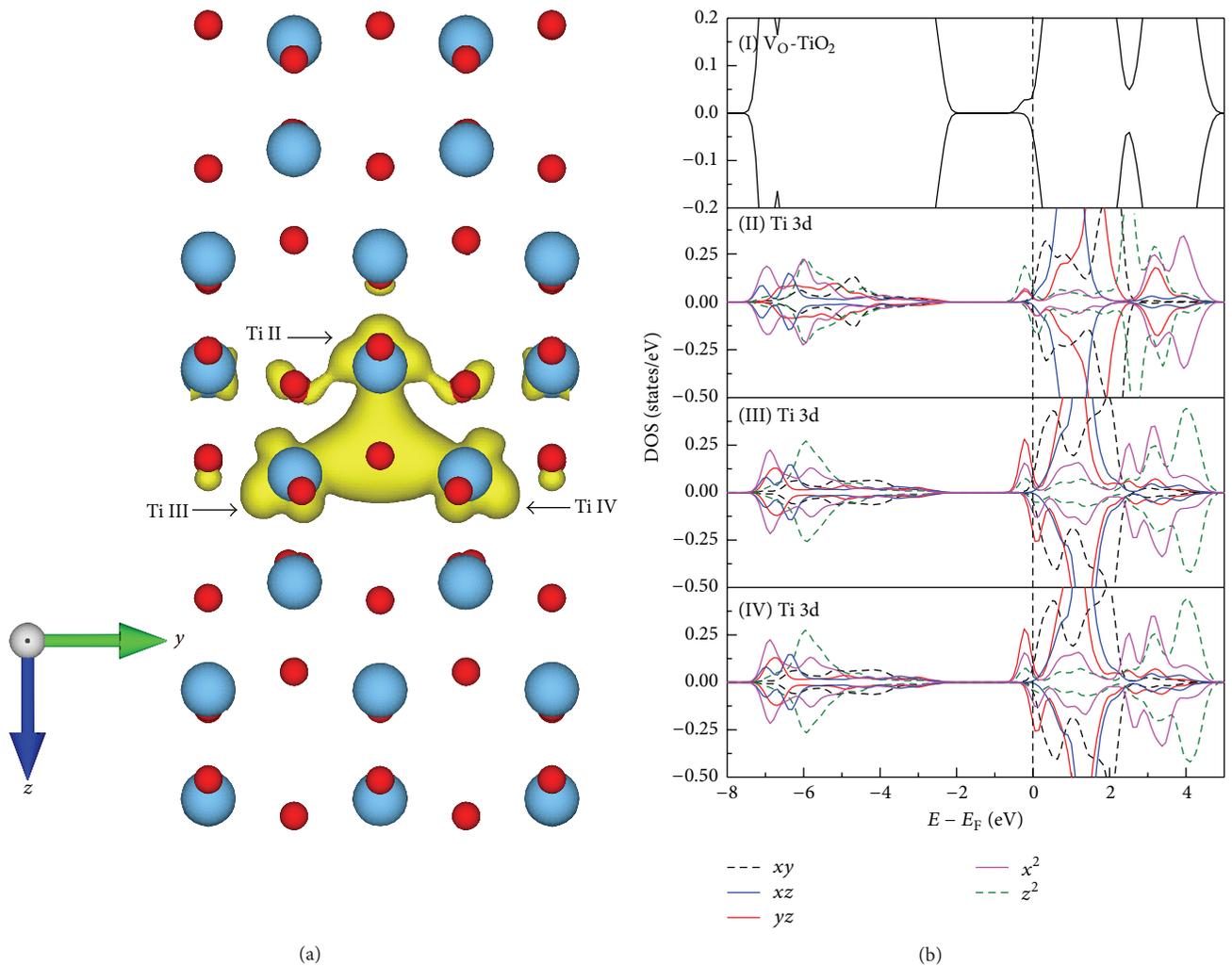


FIGURE 1: (a) Charge distribution of V_O -doped anatase TiO_2 . (b) TDOS of V_O - TiO_2 and the orbitally decomposed density of states for three Ti 3d electrons nearest to V_O . The olive and black dashed curves represent the e_g bands; the red, blue, and magenta solid curves represent the t_{2g} bands, respectively.

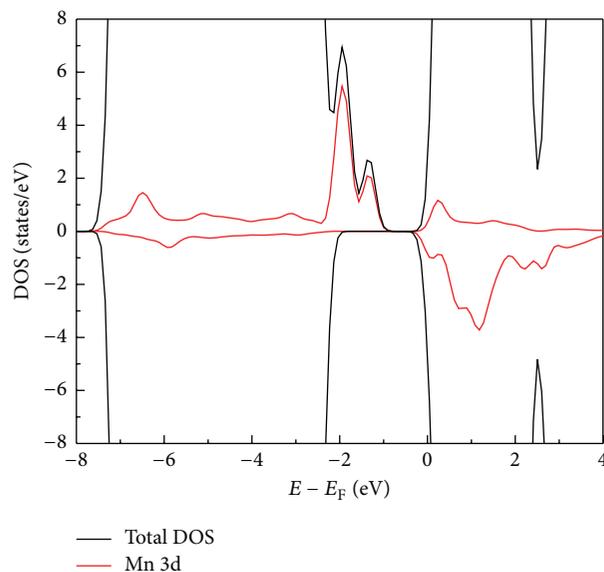


FIGURE 2: Spin-resolved total density of states for the Mn codoped TiO_2 in combination with an oxygen vacancy and spin- and orbital-resolved PDOS for the doped Mn atom.

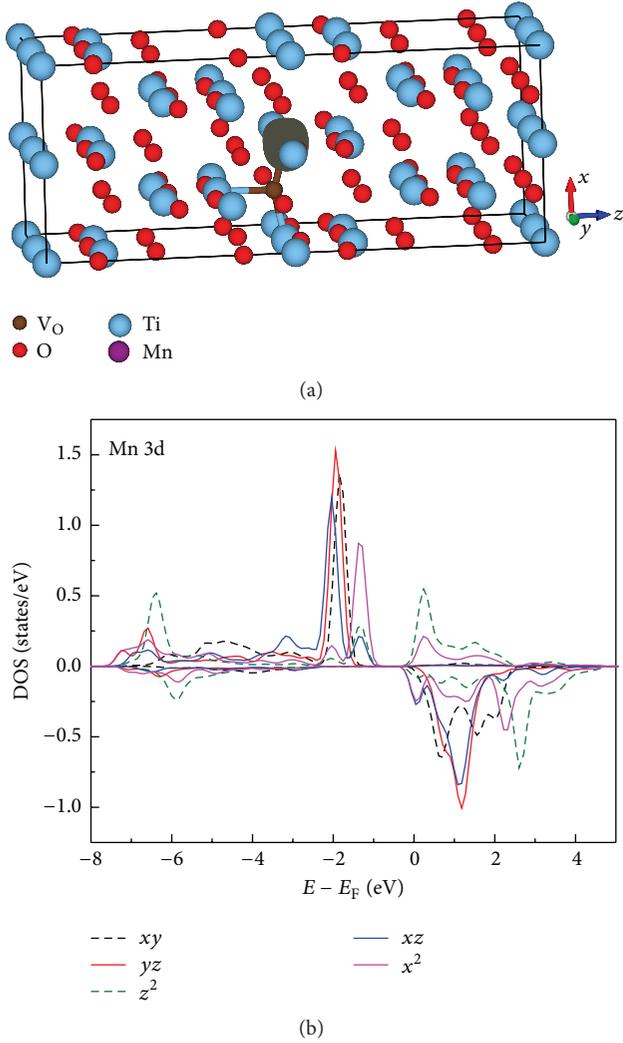


FIGURE 3: Isosurface plot of magnetization density (majority-spin minus minority-spin) for (a) $\text{Ti}_{31}\text{MnO}_{63}$ supercell; (b) the orbitally decomposed density of states projected on Mn ion.

calculated net magnetic moment is about $3.415 \mu_B$, which is mainly contributed by doped Mn ion (calculated net magnetic moment is about $3.323 \mu_B$).

The isosurface plot of magnetization density (majority-spin minus minority-spin) for $\text{Ti}_{31}\text{MnO}_{63}$ supercell and the orbital decomposed density of states projected on Mn ion are shown in Figure 3. Spin density distribution of $\text{Ti}_{31}\text{MnO}_{63}$ supercell is shown in Figure 3(a). It can be noticed that the spin densities are mainly distributed on the doped Mn ion. It holds the spin with up direction. The values of calculated bond lengths for V_O -Mn, V_O -Ti (III), and V_O -Ti (IV) are 1.8073 Å, 2.09822 Å, and 2.16686 Å, respectively. The lowest energy is achieved after geometrical optimization of $\text{Ti}_{31}\text{MnO}_{63}$, where one Mn ion is located at the nearest-neighbor site of V_O .

The orbital decomposed density of states projected on Mn ion is presented in Figure 3(b). There is a complete exchange spin-splitting of Mn 3d appearing within the Fermi level.

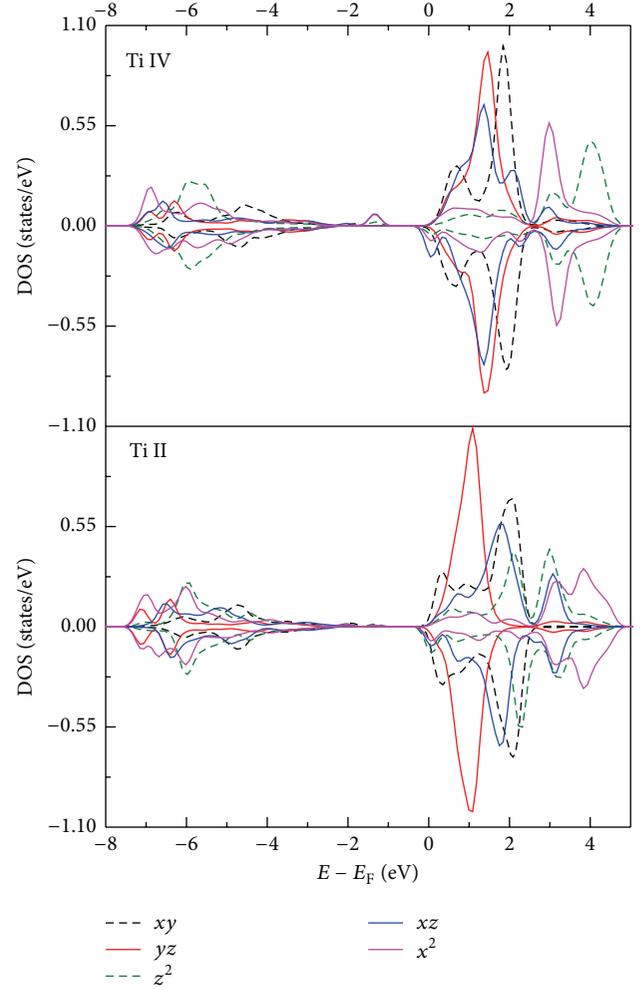


FIGURE 4: The orbital decomposed density of states projected on Ti II 3d electrons and Ti IV 3d electrons nearest to V_O for $\text{Ti}_{31}\text{MnO}_{63}$ supercell.

The results indicate that there are t_{2g} and e_g states of Mn 3d electrons with up spin within the Fermi level. The valence electrons configuration on the Mn is $3s^2 3p^6 3d^5$. The orbital decomposed density of states projected on Ti II 3d electrons and Ti IV 3d electrons nearest to V_O for (a) $\text{Ti}_{31}\text{MnO}_{63}$ supercell is noticed in Figure 4. As it can be seen, there is almost no spin-splitting of Ti II 3d electrons and Ti IV 3d electrons nearest to V_O within the Fermi level. The results indicate that there are no electron states of Ti 3d electrons appearing within the Fermi level, which is consistent with the result of Figure 3. That just says that, due to the difference in the electronegativity and valence between Ti^{4+} ion and Mn^{2+} ions, the effect of the substitution is that no electron is donated to the TiO_2 lattice instead of two. Two electrons donated by one V_O are captured by doped Mn ion and two electrons donated by the doped Mn^{2+} ion will be captured by five O ions nearest. As a result, the total net magnetic moment of V_O -Mn codoped TiO_2 is $3.415 \mu_B$.

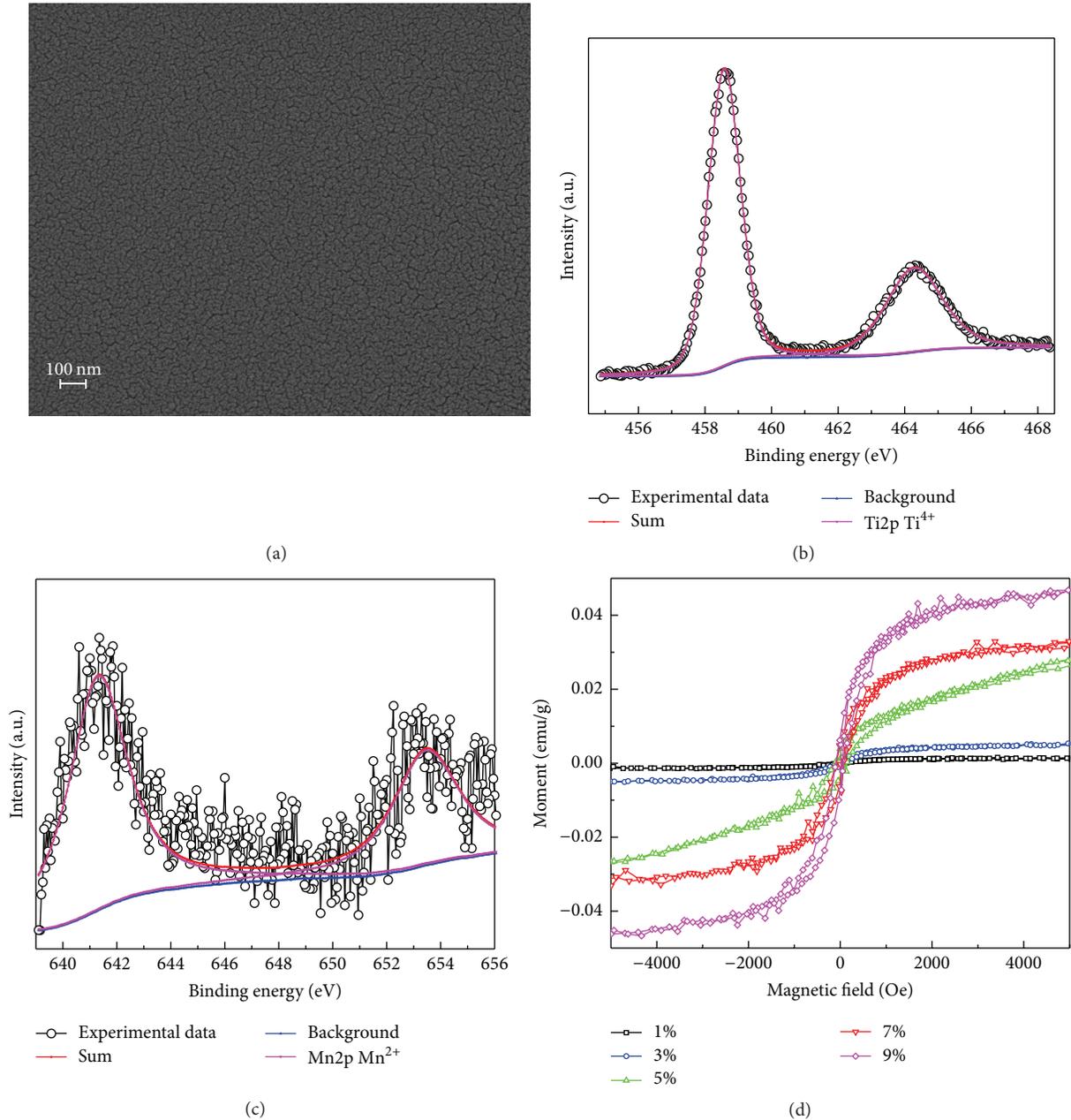


FIGURE 5: (a) SEM image of $\text{Ti}_{0.97}\text{Mn}_{0.03}\text{O}_2$ film; (b) XPS spectra of the core-level signal of Ti-2p and (c) Mn-2p of $\text{Ti}_{0.97}\text{Mn}_{0.03}\text{O}_2$ film; (d) magnetic hysteresis curves of the Mn-doped TiO_2 films with 1%, 3%, 5%, 7%, and 9% doping content measured at 300 K.

3.3. Experiment Results. The calculation results are confirmed by experimental data obtained for Mn-doped TiO_2 films. The Mn-doped TiO_2 films (with 1%, 3%, 5%, 7%, and 9% doping content) were grown by prepared by sol-gel spin coating. Figure 5(a) demonstrates the SEM images of the sample, which suggests that the film is smooth and the size of particles is about 10 nm.

The binding state of Ti and Co in the as-synthesized cobalt incorporated TiO_2 is analyzed with XPS. XPS analysis is performed after sputtering with Ar^+ . The Ti-2p and Co-2p core levels spectra of the $\text{Ti}_{0.97}\text{Mn}_{0.03}\text{O}_2$ doped sample are

shown in Figures 5(b) and 5(c). Figure 5(b) displays the Ti-2p core level XPS spectrum for $\text{Ti}_{0.97}\text{Al}_{0.03}\text{O}_2$ film sample. The open circles are the experimental data, the blue line is the background, the red line is the sum, and the peaks related to Ti^{4+} are fitted with magenta lines, where Ti 2p $_{3/2}$ core level spectrum was fitted with combined Gaussian-Lorentzian functions. The main peak of Ti 2p $_{3/2}$ situates around at 458.5 eV and a low intensity minor peak at 464.2 eV is attributed to Ti 2p $_{1/2}$, close to the binding energies of the core-level of Ti^{4+} ion, which indicates the existence of Ti^{4+} . No signal of Ti^{3+} was detected. The Ti^{3+} ions were also not

observed in other Mn-doped films, which is not like the case of Nb- or Fe-doped TiO₂ films, due to the difference in the electronegativity and valence between Mn ion, Nb ion, and Fe ion [35, 36]. Figure 5(c) shows the Mn-2p core-level XPS spectrum for Ti_{0.97}Mo_{0.03}O₂ films. The open circles are the experimental data, the blue line is the background, the red line is the sum, and the peaks related to Mn²⁺ ion are fitted with magenta lines. The main peak of Mn 2p^{3/2} situates around at 641.2 eV and a low intensity minor peak at 653.6 eV is attributed to Mn 2p^{1/2}, respectively, close to the binding energies of the core level of Mn²⁺ ion and lower to the binding energies of the core-levels of Mn³⁺ ion and Mn⁴⁺ ion, which indicates the existence of Mn²⁺.

Figure 5(d) indicates the magnetic hysteresis (M-H) curves of Mn-doped samples (with 1%, 3%, 5%, 7%, and 9% doping content) at room temperature. All the Mn-doped samples showed ferromagnetism at room temperature (RT). The saturation magnetization M_S increases rapidly as the Mn content increases from 1% to 9%, alike other reports [37]. The magnetic moment per doped Mn atom exhibits more or less constant values for the 1% to 9% samples, up to the present experimental data.

Combining the calculation and experimental results presented above, we introduce a defect electron based model for the observed ferromagnetism. The magnetic moment associated with a Mn²⁺/V_O/Ti⁴⁺ complex is spread over two or more neighboring Ti sites. These moments are aligned ferromagnetic. The major contribution to the magnetic moment of our Mn-doped TiO₂ films comes from the unpaired d electron of Mn²⁺, consistent with our first-principles calculation.

4. Conclusions

Electronic and magnetic properties of Ti₃₁MnO₆₃ have been studied combined with experiments and the first-principles calculations. Magnetic moment value is 3.415 μ_B per supercell for the V_O-Mn codoped anatase TiO₂. The lowest energy is achieved after geometrical optimization of Ti₃₁MnO₆₃, where doped Mn atoms are located at the nearest-neighbor site of V_O. Four electrons donated by the the V_O doped Mn²⁺ ion are shared by the nearest five O ions. The major magnetic moment originates from the d electron of Mn. Ferromagnetic behavior in V_O-Mn doped TiO₂ is corresponding to the strong Mn d-shell contribution and V_Os. The calculation results are confirmed by experimental data of Mn-doped TiO₂ films.

Competing Interests

The authors declare that there is no conflict of interests regarding the publication of this paper.

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