Research Article

Cationic Polybutyl Cyanoacrylate Nanoparticles for DNA Delivery

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To enhance the intracellular delivery potential of plasmid DNA using nonviral vectors, we used polybutyl cyanoacrylate (PBCA) and chitosan to prepare PBCA nanoparticles (NPs) by emulsion polymerization and prepared NP/DNA complexes through the complex coacervation of nanoparticles with the DNA. The object of our work is to evaluate the characterization and transfection efficiency of PBCA-NPs. The NPs have a zeta potential of 25.53 mV at pH 7.4 and size about 200 nm. Electrophoretic analysis suggested that the NPs with positive charges could protect the DNA from nuclease degradation and cell viability assay showed that the NPs exhibit a low cytotoxicity to human hepatocellular carcinoma (HepG2) cells. Qualitative and quantitative analysis of transfection in HepG2 cells by the nanoparticles carrying plasmid DNA encoding for enhanced green fluorescent protein (EGFP-N1) was done by digital fluorescence imaging microscopy system and fluorescence-activated cell sorting (FACS). Qualitative results showed highly efficient expression of GFP that remained stable for up to 96 hours. Quantitative results from FACS showed that PBCA-NPs were significantly more effective in transfecting HepG2 cells after 72 hours postincubation. The results of this study suggested that PBCA-NPs have favorable properties for nonviral delivery.

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1. Introduction

Hepatocellular carcinoma (HCC) is a liver cancer prevalent in Asia, especially in the Mainland China [1, 2]. As an established human hepatocarcinoma cell line with epithelial morphology, HepG2 cells are used routinely for a variety of biochemical and cell biological studies [3–5].

The barrier to gene delivery, nuclease degradation in the lysosomal compartment, has been the focus of many investigators. Important research has been moving toward the development of polycation-based gene-delivery systems (such as polylysine conjugates) designed to minimize nuclease degradation through the design of vectors with the capacity to escape the endosomal-lysosomal pathway [6–8]. Behr and others introduced the concept of the “proton sponge” and hypothesized that polymers with buffering capacities between 7.2 and 5.0, such as polyethylenimine (PEI) and imidazole-containing polymers, could buffer the endosome

and potentially induce its rupture. Protein-expression levels mediated by the polycationic proton-sponge polymer, PEI, were at least 10-fold greater than polylysine alone [9–12]. However, protein expression after delivery of plasmid DNA to the cell nucleus depends on the processes of transcription and translation. Cytotoxic gene-delivery systems may compromise these processes and potentially limit protein expression. These situations are perhaps the most prevalent in the nonviral polycationic gene-delivery class in which the polycationic nature of the delivery system can lead to cytotoxicity. To drive gene therapy ultimately into the clinic, improved delivery systems, or vectors, must deliver DNA to the cell in a transcriptionally active form and must fulfill all regulatory agency mandates to be considered safe for use in humans [13–16].

PBCA particles are extensively investigated for gene and drug delivery, but now most investigated stabilizers for PBCA lead to negatively charged particles [17]. In an

attempt to provide a positive charge to a colloidal system, chitosan has been used in the preparation and stabilization of polyester nanocapsules [18, 19], nanoparticles [20], submicron-sized emulsions [18], microcapsules [21], and liposome [22]. Chitosan is a cationic high-molecular-weight heteropolysaccharide composed mainly of b-(1,4)-2-deoxy-2-amino-D-glucopyranose units and partially of b-(1,4)-2-deoxy-2-acetamido-D-glucopyranose. Because of favorable biological properties such as biodegradability, biocompatibility, and nontoxicity, chitosan has attracted great attention in pharmaceutical and biomedical fields [23–28]. Chitosan also increases the transcellular and paracellular transport across epithelium [29].

The aim of the study was to develop positively charged polybutyl cyanoacrylate (PBCA)-NPs in the presence of chitosan for the potential use as a targeting gene delivery system. Physicochemical characteristics of the prepared nanoparticles were examined by dynamic light scattering, transmission electron microscopy, and Fourier transform infrared (FT-IR) spectroscopy. The transfection efficiency of the prepared NPs into HepG2 cells was examined in vitro using digital fluorescence imaging microscopy system and fluorescence-activated cell sorting (FACS).

2. Materials and Methods

2.1. Materials. Chitosan (degree of deacetylation = 90%) was purchased from (Shanghai Bio Life Science & Technology Co., Ltd. Shanghai, China). Butyl cyanoacrylate (BCA) monomer was synthesized by (Guangzhou Baiyun Medical Adhesive Co., Ltd. Guangzhou, China). HepG2 cell lines were provided by ourselves. All other chemicals used were of analytical reagent grade and without further purification. Ultrapure water was used for the preparation of all solutions.

2.2. PBCA Nanoparticles. PBCA-NPs were prepared by emulsion polymerization as described in detail elsewhere [30–35]. Briefly, the desired amount (100 μ L) of BCA monomer was dropped into an acidic solution of chitosan adjusted with 1 mol/L hydrochloric acid at room temperature, pH 1.5, under constant magnetic stirring, stirring was maintained for at least 6 hours until the polymerization was complete. The colloidal suspension obtained was brought to pH value of 5.5 by adding 0.5 mol/L NaOH.

2.3. Fourier Transform Infrared (FT-IR) Spectrometer. FT-IR spectra were recorded on a spectrophotometer TENSOR 27 (Bruker, Billerica, Mass, USA). Test samples used for the FT-IR analysis first were dried and ground into a powder. The powder then was mixed with KBr and pressed into a disk. The samples were scanned from 400–4000 cm^{-1} .

2.4. Particle Size and Zeta Potential. The size of PBCA-NPs was assessed using a dynamic light scattering spectrophotometer Zetasizer 1000HS_A (Malvern Instruments Ltd., Worcestershire, UK). The colloidal suspension of the NPs was diluted with deionized distilled water, and the particle size analysis was carried out at a scattering angle of 90°C and a temperature of 25°C.

The zeta potential was measured on Zetasizer Nano system (Malvern Instruments Ltd., Worcestershire, UK). The measurement was done in 10 mmol/L NaCl using disposable zeta cells using the general purpose protocol at 25°C. The instrument was calibrated routinely with a –50 mV latex standard. The mean zeta potential was determined using phase analysis light scattering technique.

2.5. Transmission Electron Microscopy (TEM) Observation. Particle morphology of PBCA-NPs was observed by transmission electron microscopy (TEM) using a Hitachi H-600 at 50 kV. 1 mL dispersion was diluted with 1 mL demineralized water and a drop of it was placed onto a collodion support on copper grids (200 mesh). About 2 minutes of deposition, the grid was tapped with a filter paper to remove surface water and negatively stained by using a sodium phosphotungstate for 5 seconds. The grid was allowed to dry further for 10 minutes and was then examined with the electron microscope.

2.6. Preparation of NP/DNA Complexes. EGFP-N1 driven by immediate early promoter of cytomegalovirus (CMV) was purchased from (Clontech Laboratories, Palo Alto, Calif, USA). The plasmids were propagated in *Escherichia coli*, extracted by the alkali lysis technique [36], and purified by an EZ-10 Spin Column Plasmid Mini-Preps kit (Bio Basic Inc, Calif, USA) according to the manufacturer's instruction. DNA purity was determined by measuring the optical density (OD), and DNA with OD₂₆₀/OD₂₈₀ $\geq$ 1.8 was used.

NP/DNA complexes were prepared through the complex coacervation of the cationic polymer (PBCA-NPs) with the EGFP-N1 plasmid. Microliters of pDNA (100 μ g/mL) were added into PBCA-NPs solution, which had been filtrated through a 0.45 μ m filter and mixed with them in 10 mmol/L phosphate buffer (pH 7.4). The mixed liquor was standing for 30 minutes at room temperature and the complexes were used for further study. Complexes formation was confirmed by electrophoresis on a 1.0% agarose gel with TBE buffer (89 mmol/L Tris, 89 mmol/L boric acid, and 2 mmol/L EDTA pH 8.0) at 75 V for 90 minutes. DNA was visualized by staining gels with ethidium bromide (0.5 μ g/ μ L). Images were acquired using an UV transilluminator (VILBER LOURMAT, France).

2.7. Stability of NP/DNA Complexes. Nanoparticles suspension or naked DNA was incubated with 4 μ L DNase I (1 U/ μ L) in DNase I Buffer, which consisted of 100 mmol/L Tris-HCl, 25 mmol/L MgCl₂, and 1 mmol/L CaCl₂ at 37°C for 1 hours. Samples of naked DNA, undigested and digested complexes were loaded onto a 1.0% agarose gel in Tris-borate EDTA buffer at pH 8.0. The samples were run on the gel at 75 V for 90 minutes; the gel was stained with ethidium bromide (0.5 μ g/ μ L).

2.8. Cell Uptake Study. HepG2 cells were incubated in Dulbecco's modification of eagle's medium (DMEM, Beijing Solarbio Science & Technology Co., Ltd, Beijing, China) at

FIGURE 1: (A) Fourier transform IR spectra of chitosan; (B) chitosan-coated PBCA-NPs; (C) PBCA-NPs.

37°C, 5% CO₂ with 10% fetal bovine serum (FBS) and 100 U/mL penicillin-streptomycin.

A confluent layer of approximately 8000 HepG2 cells per well was incubated with 20 microliters fluorescent NP suspension, the final concentration of NPs in the serum-free medium was 210 µg/mL, in the 6-well black plates at 37°C. After 6 hours of incubation, the cells were washed three times with cold PBS (pH 7.4) to eliminate excess particles which were not entrapped by the cells. The uptake of NPs by the HepG2 cells was examined with fluorescence microscopy Leica DM LB2 (Leica Microsystems Wetzlar GmbH, Germany). In the control experiment, there were no fluorescent NPs in the medium.

To preparing for flow cytometry, after digested with the lyse solution containing 0.25% trypsin, the cocultivated cells were harvested and rinsed with cooled D-Hank's solution and centrifuged (1000 rpm, 5 minutes) to discard cell fragments. The cells were further washed three times with PBS and centrifuged (1000 rpm, 5 minutes) to remove the uncombined nanoparticles. The obtained cells were diluted to single-cell suspension stabilized with 1% formaldehyde at 4°C. Processed single-cell suspensions were analyzed on an FACS Calibur flow cytometer (BD, Franklin Lakes, NJ, USA) using a 488 nm argon laser and FL1 bandpass emission (530 ± 20 nm) for the green fluorescence. HepG2 cells were gated by sideward scatter versus forward scatter (SSC/FSC) plots. Fluorescence measurements were done in the FL1 channel.

2.9. Cytotoxicity Evaluation by MTT Assay. Cytotoxicity of the NPs was determined by (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) MTT assay in HepG2 cells. This assay is based on the ability of living cells to reduce a water-soluble yellow dye, MTT, to a purple colored water-insoluble formazan product by mitochondrial enzyme succinate dehydrogenase. The cells were maintained in DMEM supplemented with 10% FBS in 5% CO₂ incubator at 37°C. Eight thousand cells were seeded per well in 96-well microtiter plates followed by incubation for 24 hours. The growth medium was replaced with a fresh one containing 0.1–8.0 µg/µL NPs. After 24-hour incubation, 20 µL of MTT solution (5 mg/mL in PBS, pH 7.4) was added to cells followed by further incubation for 4 hours. Thereafter, the

media was removed and cells were rinsed with PBS. The formazan crystals formed were dissolved using dimethyl sulfoxide (DMSO) (150 µL/well) and absorbance was read at 570 nm on a multiskan spectrum (Electro Thermo, Milford, Mass, USA). Cell viability was determined as a percentage of the negative control (untreated cells).

2.10. In Vitro Transfection. For the qualitative analysis of transfection, HepG2 cells were grown in 6-well culture plates containing a Corning circular glass cover-slip with seeded 1×10^5 cells seeded per well and incubated until the cover rate reached 60–80%. The NP/DNA complexes were dispersed in serum-free medium and untreated cells were used as a negative control. After filtration, the NP/DNA complexes were added at a concentration equivalent to 2 µg of EGFP-N1 plasmid DNA per well, and incubated with the cells at 37°C and 5% CO₂ atmosphere for 6 hours. The culture was removed from each well and replaced with regular growth medium including 10% FBS. The cover slips were washed three times with PBS at 96 hours posttransfection and mounted on to microscopic slides containing a drop of antifade mounting medium (Beyotime Institute of Biotechnology, Shanghai, China). The expression of GFP in the cells was observed by a fluorescence microscope.

For the quantitative analysis of transfection, HepG2 cells were grown to semiconfluence in a 6-well cell culture plate and transfected with EGFP-N1 plasmid DNA encapsulated PBCA-NPs and compared with cells treated with serum-free media alone. The NP/DNA complexes were dispersed in serum-free media and added to each of the six wells at a concentration of 2 µg of plasmid DNA per well. The cells are allowed to incubate for a period of 6 hours at 37°C in a 5% CO₂ incubator and then replaced by fresh growth medium. The adherent cells are trypsinized at time intervals of 24, 48, 72, and 96 hours postincubation, centrifuged, and fixed with 100 µL of 4% formalin buffer solution. The fluorescence of the GFP produced in the transfected cell was detected by a flow cytometer equipped with an argon 488 laser. The FL1 channel was used to detect the cells expressing GFP fluorescence.

3. Results and Discussion

3.1. Preparation and Characterization of PBCA-NPs. We synthesized the nanoparticles by dropping monomer into an acidic solution of chitosan. We obtained uniform particles by adding BCA monomer, pH 1.5, at room temperature, to 0.5% (w/v) chitosan during high-speed vortexing.

FT-IR analysis was performed to confirm the chitosan-graft-PBCA. Figure 1 shows a comparison between the FT-IR spectra of chitosan (A), chitosan-coated PBCA (B), and PBCA (C). After chitosan-stabilized NPs had been extracted, the residue still possessed a -CN stretch (2253.84 cm⁻¹) and a carbonyl stretch (1752.93 cm⁻¹) corresponding to those of PBCA and a hydrogen-bonded OH stretch (3461.64 cm⁻¹) corresponding to that of chitosan. The amide band at 1640.57 cm⁻¹ that was observed in chitosan shifts to 1635.89 cm⁻¹. These observations suggest that chitosan, especially the NH₂ group, may initiate the BCA monomer

FIGURE 2: (a) Size distribution and (b) zeta potential of the prepared chitosan-coated PBCA-NPs.

FIGURE 3: Transmission electron micrograph of the prepared chitosan-coated PBCA nanoparticles.

in the acidic polymerization medium, leading to chitosan chemically coupled to PBCA to form chitosan-stabilized NPs.

The size distribution and zeta potential of the NPs in NaCl solution were investigated by dynamic light scattering. Dynamic size distribution and zeta potential may play important roles in determining the fate of NPs after administration [37]. Figure 2(a) shows a unimodal particle size distribution (number average) between 100 and 200 nm. It was reported that large particles hardly reach the cells [38], while smaller particles tended to accumulate in the tumor sites due to the enhanced permeability and retention (EPR) effect and a greater internalization was also observed [6]. Additionally, gene carriers with a diameter larger than 200 nm are readily scavenged nonspecifically by monocytes and the reticuloendothelial system [39].

The results of Figure 2(b) indicated that PBCA-NPs carry positive charge with a zeta potential of about 25 mV. PBCA-NPs have a very negative zeta potential because of the adsorption of anions from the aqueous polymerization medium, but chitosan is a polycationic biopolymer, positive charge PBCA-NPs may have been produced by coating them with chitosan [40, 41].

FIGURE 4: Agarose gel electrophoresis of NP/DNA complexes. Lane 1: DNA molecular weight marker; lane 2: naked plasmid DNA; lane 3: chitosan-coated PBCA-NPs; lanes 4–7 correspond to DNA with progressively increasing proportions of NPs at the charge ratio of 5:1, 10:1, 15:1, and 30:1 (w/w), respectively.

In theory, more pronounced zeta potential values, being positive or negative, tend to stabilize particle suspension due to the electrostatic repulsion forces between particles with the same electric charge prevents the aggregation of the spheres [42]. Consequently, the presence of the positively charged groups on the particles could lead to electrical charge interactions between the negatively charged surface cells and the particles [43].

TEM images shown in Figure 3 reveal the surface morphology and spherical shape of the prepared particles. The actual diameter of the nanoparticles observed by TEM was approximately 200 nm and was found to be similar to the values obtained by measurements based on light scattering. The plasmid is protected from DNase degradation in this formulation [44].

3.2. Complex Formation of PBCA-NPs With pEGFP-N1 Plasmid DNA. pDNA complex with previously prepared nanoparticles was achieved by the formation of ion-pairs between the positively charged amino groups located on the

FIGURE 5: Electrophoretic mobility analysis NP/DNA complexes following DNase I digestion. Lane 1: EGFP-N1 plasmid DNA; lane 2: lane 1 + digestion; lane 3: NP/DNA complex at the charge of 5:1; lane 4: lane 3 + digestion; lane 5: NP/DNA complex at the charge of 15:1; lane 6: lane 5 + digestion.

particle surface and the negatively charged phosphodiester backbone of the plasmid DNA [45]. The electrophoretic mobility of DNA, complex with PBCA-NPs, decreases gradually on the agarose gel with increasing charge ratio (Figure 4). However, in the control experiment, there was no DNA in agarose gel. This method also allowed showing the dependency of this interaction on pH, reflecting the deprotonation of the modified silica particles at alkaline pH (data not shown). The particle-DNA binding was also inhibited by the presence of high salt concentrations (data not shown). This suggests that this interaction is based on electrostatic forces, resulting in interpolyelectrolyte complexes.

3.3. Enzyme Protection. Behr introduced the concept of the “proton sponge” and hypothesized that polymers with buffering capacities between 7.2 and 5.0 could buffer the endosome and potentially induce its rupture [46]. As the chitosan molecule is a weak base with a pKa value of the D-glucosamine residue about 6.2–7.0 [47], we hypothesize NP/DNA complexes formation can resist enzymatic degradation. The results were analyzed by using agarose gel electrophoresis. Figure 5 shows that free-plasmid DNA was degraded completely into small fragments and could not be visualized in the agarose gel, however, NP/DNA complexes digested by DNase I were still visualized around the sample pore.

3.4. Cytotoxicity Evaluation of NP/DNA Complexes. In vitro toxicity of NP/DNA complexes was evaluated by MTT assay in HepG2 cells using increasing doses of NPs. MTT assay shows that the cytotoxicity of the NPs depended on their concentration ranging from 0.1–8.0 $\mu\text{g}/\mu\text{L}$ (Figure 6). The statistic analysis demonstrates that there are significant differences between concentrations (one-way ANOVA, $P < .001$). These data can indicate that the NPs below 2 $\mu\text{g}/\mu\text{L}$ had little adverse effect on the HepG2 cells viability, suggesting that the doses of in vitro HepG2 cells uptake study less

FIGURE 6: Cytotoxicity of cationic chitosan coated PBCA-NPs analyzed by MTT assay, 24 hours posttreatment ($n = 8$). Columns, mean; bars, SD, $P < .001$ as evaluated by one-way ANOVA using SPSS 17.0 version.

than 0.5 $\mu\text{g}/\mu\text{L}$ presented no toxicity. This is important because most of the cationic polymers and lipids, which are commonly used for gene transfection, have toxic effects on cells at a higher concentration due to electrostatic interaction with negatively charged cellular membrane [48, 49]. Thus, the PBCA-NPs is a promising carrier for gene delivery with low cytotoxicity. However, its long-term in vivo toxicity and immunogenicity should be further investigated.

3.5. Uptake of the Functionalized PBCA-NPs. In order to study cellular uptake of NPs in vitro, the use of fluorescently or radioactively labeled NPs is the most common experimental approach found in the literature. Fluorescent labeling, which we used the prepared fluorescent positive PBCA-NPs in this study, was chosen for the present study to avoid exposure of the samples to radioactive materials. Fluorescent labeling makes cellular uptake of NPs readily detectable by fluorescence microscopy. The extent of particle uptake can then be determined by flow cytometry. After incubation of HepG2 cells with the fluorescent PBCA-NPs for 6 hours, the FACS measurements showed significantly higher mean fluorescence intensity of 75.06% for cells incubated with the positively charged fluorescent PBCA-NPs with respect to the control particles (see Figure 7).

Figure 8 shows fluorescent microscopic images of HepG2 cell monolayers after the NPs uptake experiments, which strongly support the previous quantitative measurements of the cellular uptake of the NPs by showing strong fluorescence in the cell. However, no fluorescence can be detected from the image of the control cells.

3.6. In Vitro Gene Transfection and Expression in HepG2 Cells. In order to evaluate the effect of PBCA-NPs on transferring DNA, we investigated the transfection efficiency of complexes containing pDNA in HepG2 cells to find gene

FIGURE 7: (a) Quantitative analysis intracellular fluorescence of HepG2 cells; (b) endocytosised with fluorescent chitosan-coated PBCA-NPs by flow cytometry.

FIGURE 8: Biocompatibility of functionalized chitosan-coated PBCA-NPs ($\times 200$). (a) HepG2 cell endocytosised fluorescent chitosan-coated PBCA-NPs; (b) HepG2 cell without fluorescent chitosan-coated PBCA-NPs.

expression level. The concentration of PBCA-NPs used in this experiment was much less than $0.5 \mu\text{g}/\mu\text{L}$, which would not affect the cell viability obviously according to the result mentioned in cell viability assay.

The pictures of fluorescence microscope to detect the green fluorescence sent out by transfected cells indicated relatively higher transfection efficiencies of NP/DNA complexes (Figure 9(a)). To determine the gene transfer capability of NPs in vitro exactly, flow cytometry experiments were conducted to determine the EGFP transfection levels in HepG2 cells. It was found that the transgene expression levels increased with the time, and reached a climax at 72 hours, then decreased. The results of the complexes are fused with cell membrane easily through electrostatical interaction can be explained by the fact that complexes possess the positive potential in neutral conditions.

4. Conclusions

From these investigations, it is evident that this method forms uniform cationic PBCA nanospheres which can bind DNA readily by electrostatic interaction. Chitosan cannot only chemically couple to PBCA giving the net positive surface charge, but also produced particles with uniform size and spherical shape, as observed by TEM. By applying emulsion polymerization technique, it is possible to prepare fluorescent dye labeled PBCA-NPs. In the present study, we identify that chitosan-coated PBCA-NPs are positive charge and can be used as nonviral carrier for gene transfection.

Compared with negatively charged NPs, positively charged NP/DNA complexes might prevent destabilization by cation adsorption, give a better controlled release of plasmid, favorably interact with negatively charged tissues and provide site-specific target in vivo. Therefore, positively charged targeting gene delivery vehicles may be considered as very promising gene carriers for various administration routes.

FIGURE 9: (a) Qualitative evaluation of EGFP-N1 transgene expression in HepG2 cells with chitosan-coated PBCA-NPs, fluorescent image ($\times 200$) of the cells was obtained after 96 hours upon incubation. (b) Quantitative evaluation of EGFP-N1 transgene expression in HepG2 cells as a function of time measured by flow cytometric analysis with chitosan-coated PBCA-NPs. HepG2 cells without any treatment were used as control.

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