Nanofibers Offer Alternative Ways to the Treatment of Skin Infections

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Injury to the skin causes a breach in the protective layer surrounding the body. Many pathogens are resistant to antibiotics, rendering conventional treatment less effective. This led to the use of alternative antimicrobial compounds, such as silver ions, in skin treatment. In this review, nanofibers, and the incorporation of natural antimicrobial compounds in these scaffolds, are discussed as an alternative way to control skin infections. Electrospinning as a technique to prepare nanofibers is discussed. The possibility of using these structures as drug delivery systems is investigated.

1. Introduction

Severe skin damage and loss of the protective layer exposes underlying tissue to secondary infections [1, 2]. In the United States, treatment of fire and burn wound infections amount to more than 7.5 billion US dollars per annum [3]. From one million patients, an estimated 10,000 die from secondary microbial infections [2, 4–9]. If wounds are not treated effectively, pathogens form biofilms, rendering them resistant to antibiotics [10–13]. In severe cases, biofilms need to be surgically removed to prevent further infection [14].

Staphylococcus aureus, including methicillin resistant strains (MRSA), are the most prevalent in skin infections. Pseudomonas aeruginosa, Escherichia coli, Acinetobacter spp., and coagulase-negative staphylococci have also been isolated from skin lesions [15–19]. More than 95% of S. aureus strains are resistant to penicillin and 60–70% are resistant to methicillin [20–22]. Methicillin resistance is attributed to the mec A gene encoding penicillin binding proteins [20, 23]. These proteins occur in the cell wall and play a role in the synthesis of peptidoglycan and are usually inactivated by beta-lactam antibiotics. However, in MRSA the mec A gene encodes a low-antibiotic affinity penicillin binding protein, known as PBP2a, conferring methicillin resistance to the cells [20]. Some MRSA strains are also resistant to tetracyclines, sulphonamides, trimethoprim, macrolides, aminoglycosides, mupirocin, mafenide acetate, silver sulphadiazine, bacitracin, ciprofloxacin, and vancomycin [24–29]. Vancomycin-resistant enterococci and P. aeruginosa strains resistant to several antibiotics have also been reported [30].

This review focuses on the treatment of skin infections and open wounds and the biomedical application of electrospun nanofibers, with specific emphasis on antimicrobial delivery systems. Antimicrobials other than antibiotics are discussed.

2. Current Treatment of Skin Infections and Drawbacks

Most data on skin infections are from publications on burn wounds. Several treatments have been proposed. Silver sulfadiazine, a combination of silver nitrate and sodium sulfadiazine, has been used to treat invasive burn wound sepsis [6, 14, 32, 33]. With prolonged use, silver ions may be toxic, as it binds to DNA and prevents replication [34, 35]. Furthermore, some pathogens have developed resistance to silver [36–38].

Mafenide acetate and chlorexidine digluconate creams have been used to treat burn wound infections [14, 39].
The disadvantage of these topical creams is that they have to be applied twice daily. If incorporated in wound dressings, the bandages have to be changed daily, which may expose the wound to further infection [40]. As in the case of many other antimicrobials, pathogens with resistance to mafenide acetate have been reported, especially when used over an extended period [27]. Mupirocin has also been used in the treatment of burn wound infections [41].

Nanotechnology offers the latest alternatives to wound dressings, of which Acticoat A.B, Silverlon, and Silvasorb with nanoparticles are good examples. Supporters of this technology claim that the nanocrystalline silver particles are released in a controlled manner, inhibiting the growth of a broad spectrum of pathogens [43–45]. Entrance of the nanoparticles into cells and their mode of action is summarized in Figure 1. Endosomes, filled with silver nanoparticles, lysosomes, and silver nanoparticles in the nucleus of treated cells, have been observed. Cells may also take up silver nanoparticles via endocytosis.

Some concerns exist regarding the medical use of silver particles, as they form reactive oxygen species (ROS), reduce ATP, and damage mitochondria and DNA [31, 46]. Furthermore, silver nanoparticles caused an inflammatory response in a murine model system [47]. Silver nanoparticles are, however, less toxic compared to gold nanoparticles, as observed in experiments with J774 A1 murine macrophages [48].

### 3. Production of Nanofibers

Nanofibers are produced from polymers treated in a specific manner to form threads of a few micrometers to nanometers in diameter. The large surface to volume ratio, and manipulation of surface properties, renders nanofibers the ideal matrix to develop super fine structures [49–51]. The possibility to immobilize antibiotics, enzymes, antimicrobial peptides, and growth hormones to nanofibers, or encapsulation into fiber matrices, opens a new field in biomedical engineering [52–60].

Several methods are used to produce ultra fine fibers, for example, self assembly of polymers, template synthesis, phase separation, and electrospinning [49, 62–66]. Electrospinning, schematically presented in Figure 2, is the most cost effective and easiest way to produce large volumes of nanofibers. One electrode is placed in a polymer solution and the other electrode is linked to a collector, which is usually a stationary or rotating metal screen, plate, or wheel. The electrically charged polymer forms a Taylor cone at the tip of the needle and is ejected at a specific charge. As the polymer solution accelerates, the solvent evaporates and nanofibers are formed. Fibers are aligned by using a rotating collector, an auxiliary electrical field, or a rotating collector with a sharp edge and a rapidly oscillating frame [67–71]. Coaxial electrospinning (Figure 3(a)) is used to produce nanofibers with a core-shell structure (Figure 4(a)), which is ideal for encapsulating hydrophilic molecules. Coaxial spun fibers have a high loading efficiency [56, 72]. Emulsion electrospinning is also used to produce core-shell-structured nanofibers (Figures 3(b) and 4(b)). An emulsion is prepared by emulsifying an aqueous phase, which contains a hydrophilic polymer or molecule to be encapsulated, into an organic phase containing a polymer that forms the shell [42, 61]. The emulsion is then electrospun into core-shell-structured nanofibers (Figure 4(b)).

### 4. Parameters That Influence Nanofiber Formation and the Electrospinning Process

The quality and characteristics of the final product are determined by the temperature, viscosity, elasticity, conductivity, and surface tension of the solution, strength of the electric field, distance between the needle tip and collector, and humidity [49, 73, 74]. Larger fibers (bigger diameter) are obtained by increasing the concentration of the polymer in solution. Polyvinylpyrrolidone (PVP) at 4% (w/w) in a 50:50 (w/w) dimethylformamide:ethanol solution is used to produces fibers of 20 nm in diameter [75]. However, PVP at 8% (w/w) in the same solution produces fibers of 50 nm in diameter, and PVP at 10% (w/w) produce fibers of 300 nm in diameter. Electrospinning different concentrations of poly L-lactic acid (PLLA) in a chloroform solution produce nanofibers with different morphologies (Figure 5). PLLA of 1% (w/w) produces a “bead on a string” structure whereas 3% (w/w) PLLA forms nanofibers with a smooth structure [61].
Solvents influence the surface tension and viscosity of the solution and affect the morphology of fibers [75, 76]. PVP (4%, w/w) dissolved in dichloromethane (MC) forms fibers with spindle-like beads and hollow or solid structures whereas the same concentration PVP in N,N-dimethylformamide (DMF) forms sphere-like beads with solid structures. PVP dissolved in ethanol yields smooth fibers with a diameter ranging from 100 to 625 nm [75]. The PVP/DMF and PVP/MC solutions have a high surface tension (47.1 and 38.7 centipoise, resp.) and low viscosity (9.8 and 13.0 centipoise, resp.). A PVP/ethanol solution, on the other hand, has a low surface tension (29.3 centipoise) and high viscosity (17.3 centipoise). Different solvents evaporate at different speeds and affect the structure of the fibers [76]. Changes in current may also affect the morphology of fibers, as observed with polyethylene oxide (PEO). An increase from 5.5 kV to 9.0 kV changed the morphology of the PEO fibers from smooth to a “bead on a string” structure [74]. The importance of processing variables that influence fiber morphology during electrospinning is reviewed by Deitzel et al. [74].

5. Electrospun Nanofibers in Biomedical Engineering as Drug Delivery Vehicles

Natural and synthetic polymers have been spun into nanofibers for potential use in biomedical engineering [53, 54, 61, 76–79]. Chitin, a structural polysaccharide from arthropods, yielded fibers ranging from 40 to 600 nm in diameter [79]. A combination of water-soluble carboxyethyl chitosan and poly-vinyl alcohol (PVA) was electrospun to produce a wound dressing. The nanofibers revealed no toxicity when tested with a mouse fibroblast L929 cell line, and promoted cell attachment and proliferation [78]. Chitosan acetate bandages proved effective as an antimicrobial dressing when tested on BALB/c mice with burn wounds that have been infected with *P. aeruginosa* and *Proteus mirabilis* [80].

Hydrophilic and hydrophobic polymer blends have also been spun into biodegradable nanofibers. The hydrophobic polymer provides the structure or “backbone” and degrades over a long period whereas the more hydrophilic polymer degrades or dissolves faster. The choice of polymer or polymer blends play an important role in devices aimed at controlled release. Examples of using hydrophilic and/or hydrophobic polymers for the controlled release of molecules, for example, antibiotics, plasmids, growth factors, proteins, silver particles, bacteria and viruses will be discussed in more detail [54, 61, 82, 83, 89, 93].

5.1. Antibiotics. Rifampin, encapsulated in PLLA during electrospinning, and incubated in a 0.05 M Tris-HCl buffer, was only released when proteinase K was added to the solution. This suggests that the release of rifampin was initiated by the degradation of PLLA and not by normal diffusion [81]. In another experiment, doxorubicin hydrochloride and paclitaxil were encapsulated into PLLA nanofibers [81]. Doxorubicin hydrochloride was detected on the surface of the nanofibers but paclitaxil remained encapsulated. Rifampin and paclitaxil were more soluble in the chloroform/acetone solvent compared to doxorubicin hydrochloride. The solubility of the molecule to be encapsulated in the polymer solvent plays an important role in its distribution throughout the nanofibers. Tetracycline hydrochloride (5%, w/w) encapsulated in poly-ethylene-co-vinyl acetate (PEVA), or in a blend of PEVA and PLLA, has a relatively slow and consistent release rate [60]. The PEVA and PEVA/PLLA blend containing 5% (w/w) tetracycline hydrochloride had
plasmid (pCMVβ), encoding β-galactosidase. The majority of plasmid DNA was released over 20 days. The bioactivity of pCMVβ was evaluated by conducting transfection experiments with preosteoblastic MC3T3 cells. The β-galactosidase gene was successfully expressed by preosteoblastic MC3T3 cells that have taken up the plasmid.

5.3. Growth Factors. Human β-nerve growth factor (NGF) was encapsulated into fibers consisting of a copolymer of poly ε-caprolactone and ethyl ethylene phosphate (PCLEEP) through electrospinning [54]. The bioactivity of NGF was evaluated by incubating rat pheochromocytoma (PC 12) cells in the supernatant of nanofibers containing encapsulated NGF and searching for differentiation into neurons. Bioactivity was recorded for up to three months. Human glialcell-derived neurotrophic factor (GDNF), encapsulated into PCLEEP nanofibers, was released in active form for up to two months [84]. A 15 mm nerve lesion was made in the left sciatic nerve of 3.5 month-old Sprague–Dawley rats. The rats then received longitudinally aligned fibers impregnated with GDNF. Longitudinally or circumferentially aligned fibers with no GDNF encapsulated within served as controls. In rats that received encapsulated GDNF, a bridge formed across the lesion and the nerve was regenerated after three months. However, the nerve system in only half the number of rats that received control fibers was regenerated. The stimulation of bone regeneration in nude mice that have been treated with bone morphogenetic protein-2 (BMP-2) encapsulated in PLGA-hydroxyapatite (HAp) nanofibers was also shown in [85]. Bioactive BMP-2 was released from the nanofibers over four weeks.

5.4. Proteins. Lysozyme was encapsulated in biodegradable poly-ε-caprolactone (PCL) and PEO fibers [53]. The highest release of lysozyme (87% over 12 days) was recorded in a PEO/PCL nanofiber with a 90/10 ratio. The released lysozyme maintained 90% of its catalytic activity. Cytochrome C has been encapsulated in nanofibers by emulsion electrospinning [61]. This was done by emulsifying an aqueous solution of cytochrome C in a chloroformPLL. High encapsulation efficiencies (85% to 95%) were recorded after spinning. However, low levels of cytochrome C were released. Controlled release was obtained when PLLA was blended with poly(L-lysine) (PLL) and poly(ethylene imine) (PEI), hydrophilic polymers. A blend containing 50% PLL released most of the protein (75%) with a high initial burst release.

5.5. Bacteria and Viruses. Escherichia coli, Staphylococcus albus, and bacteriophages T7, T4, and λ were encapsulated in PVA nanofibers with water as solvent [88]. The encapsulated cells and bacteriophages survived the electrospinning process and remained viable for three months at −20 and −55°C. M13 viruses were encapsulated into PVP nanofibers [91]. The fibers were dissolved in a tris-buffered saline solution (pH 7.5). The released viruses were still able to infect bacterial cells. Micrococcus luteus and E. coli were encapsulated into PEO nanofibers with water as solvent [89]. Up to 74% of the M. luteus cells, but only 0.1%
of the E. coli cells, survived the electrospinning process. We recently reported on the encapsulation of a probiotic lactic acid bacterium in electrospun PEO nanofibers [90]. Only 2% of the Lactobacillus plantarum cells survived the electrospinninig process. However, the cells that survived were still able to produce the antimicrobial peptide (bacteriocin) and inhibited the growth of E. faecium HKLHS that served as target strain.

5.6. Silver Nanoparticles. Silver nanoparticles were incorporated in cellulose acetate nanofibers by electrospinning cellulose acetate with 0.5 wt% AgNO₃ [94]. Silver particles were generated on the surface of the nanofibers after irradiation at 245 nm. Almost all viable cells (99.9%) of Gram-positive bacteria, E. coli, Klebsiella pneumoniae, and P. aeruginosa were killed after 18 hours of exposure to the encapsulated silver particles. Silver-loaded zirconium phosphate was spun into poly ε-caprolactone fibers [86]. Growth inhibition of up to 99.27% of S. aureus and up to 98.44% of E. coli was recorded when the strains were cultured in the presence of these nanofibers. Human dermal fibroblasts that attached to the nanofibers continued to proliferate, suggesting that the fibers may be used as wound dressings. Similar findings have also been reported by Rujitanaroj et al. [95]. However, some authors have reported that silver may elicit toxic side effects on human cells as discussed elsewhere in this paper. Table I summarizes various molecules that have been encapsulated into synthetic and natural polymers, or blends thereof, by electrospinning to facilitate their release.

6. Natural Alternatives to Antibiotics

Lactic acid bacteria (LAB) are a diverse group of organisms with GRAS (generally regarded as safe) status and have been consumed over decades [96]. Most species produce bacteriocins, that is, ribosomally synthesized proteins or protein complexes with bacteriostatic or bactericidal activity against closely related species [97, 98]. The peptides have a net positive charge (cationic) and are amphiphilic or rather more hydrophobic. They intercalate into the cell membrane of sensitive cells, form pores and disrupt the proton motive force (PMF) [99–101].

Bacteriocins are classified into two major classes. Class I contains the lantibiotics that are small peptides that undergo posttranslational modifications and have lanthionine or β-methyllanthionine residues, for example, nisin, merscadin [102, 103]. Class II contains the nonlanthionine-containing bacteriocins that are small (<10 kDa) heat-stable
Table 1: Molecules and organisms encapsulated into electrospun nanofibers.

<table>
<thead>
<tr>
<th>Encapsulated molecules</th>
<th>Polymer/polymer blends</th>
<th>Reference</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Antibiotics</strong></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Rifampin</td>
<td>PLLA</td>
<td>[81]</td>
</tr>
<tr>
<td>Doxorubicin hydrochloride</td>
<td>PLLA</td>
<td>[81]</td>
</tr>
<tr>
<td>Paclitaxil</td>
<td>PLLA</td>
<td>[81]</td>
</tr>
<tr>
<td>Tetracycline hydrochloride</td>
<td>PEVA</td>
<td>[60]</td>
</tr>
<tr>
<td>Mefoxin</td>
<td>PLGA/PEG-b-PLLA</td>
<td>[82]</td>
</tr>
<tr>
<td><strong>Plasmid DNA</strong></td>
<td></td>
<td></td>
</tr>
<tr>
<td>pCMVb encoding a β-Galactosidase</td>
<td>PLGA and PLA-PEG</td>
<td>[83]</td>
</tr>
<tr>
<td><strong>Growth Factors</strong></td>
<td></td>
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<tr>
<td>Human β-nerve growth factor (NGF)</td>
<td>PCLEEP</td>
<td>[54]</td>
</tr>
<tr>
<td>Human glialcell-derived neurotrophic factor (GDNF)</td>
<td>PCLEEP</td>
<td>[84]</td>
</tr>
<tr>
<td>Bone morfogenetic protein-2 (BMP-2)</td>
<td>PLGA-Hap</td>
<td>[85]</td>
</tr>
<tr>
<td><strong>Antimicrobial compounds</strong></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Silver nanoparticles</td>
<td>Cellulose acetate</td>
<td>[72]</td>
</tr>
<tr>
<td></td>
<td>PCL/silver-loaded zirconium phosphate</td>
<td>[86]</td>
</tr>
<tr>
<td><strong>Proteins</strong></td>
<td></td>
<td></td>
</tr>
<tr>
<td>BSA</td>
<td>PEO</td>
<td>[87]</td>
</tr>
<tr>
<td>Lysozyme</td>
<td>PCL/PEO</td>
<td>[53]</td>
</tr>
<tr>
<td>Lysozyme</td>
<td>PCL/PEG</td>
<td>[77]</td>
</tr>
<tr>
<td>Cytochrome C</td>
<td>PLLA/PLL</td>
<td>[61]</td>
</tr>
<tr>
<td><strong>Bacteria</strong></td>
<td></td>
<td></td>
</tr>
<tr>
<td><em>Eschericia coli</em></td>
<td></td>
<td></td>
</tr>
<tr>
<td><em>Staphylococcus albus</em></td>
<td>PVA</td>
<td>[88]</td>
</tr>
<tr>
<td><em>Micrococcus luteus</em></td>
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</tr>
<tr>
<td><em>Eschericia coli</em></td>
<td>PEO</td>
<td>[89]</td>
</tr>
<tr>
<td><em>Lactobacillus plantarum</em></td>
<td>PEO</td>
<td>[90]</td>
</tr>
<tr>
<td><strong>Bacteriophages</strong></td>
<td></td>
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<tr>
<td>T7</td>
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<td>T4</td>
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<tr>
<td>λ</td>
<td>PVA</td>
<td>[88]</td>
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<tr>
<td>M13</td>
<td>PVP</td>
<td>[91]</td>
</tr>
</tbody>
</table>

PLLA: poly (L-lactide); PEVA: poly-ethylene-co-vinyl acetate; PLGA: poly-lactate-co-glycolide; PEG: polyethylene glycol; PLGA/PEG-b-PLLA: poly-lactate-co-glycolide/polyethylene glycol-block-poly(L-lactide); PCL: poly ε-caprolactone; PCLEEP: poly ε-caprolactone ethyl ethylene phosphate; Hap: Hydroxyapatite; PEO: polyethylene oxide; PDLA: poly(D,L-lactide); PLL: poly(L-lysine); PVA: poly(vinyl alcohol); PVP: polyvinylpyrrolidone.

Table 2: Classification system for bacteriocins [92].

<table>
<thead>
<tr>
<th>Classes</th>
<th>Characteristics</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Class I</strong></td>
<td></td>
</tr>
<tr>
<td>Class Ia</td>
<td><em>Lantibiotics</em></td>
</tr>
<tr>
<td>Small (19–38 amino acids), elongated, positively charged peptides that form pores</td>
<td></td>
</tr>
<tr>
<td>Class Ib</td>
<td>Globerular peptides that interfere with essential enzymes</td>
</tr>
<tr>
<td><strong>Class II</strong></td>
<td><em>Nonlantionine containing-bacteriocins</em></td>
</tr>
<tr>
<td>Class Ia</td>
<td>Pediocin-like peptides that contain the YNGVXCVXXXXXXV consensus sequence in their N-terminal</td>
</tr>
<tr>
<td>Class Ib</td>
<td>Two-peptide bacteriocins, require both peptides for activity</td>
</tr>
<tr>
<td>Class IIC</td>
<td>Cyclic peptides, N- and C-terminal of peptides are covalently linked</td>
</tr>
<tr>
<td>Class IId</td>
<td>Single non-pediocin like peptides</td>
</tr>
</tbody>
</table>

Some bacteriocins, such as mersacidin, have shown activity towards MRSA that have been associated with various hospital acquired infections [108]. Nisin F was also investigated as treatment for subcutaneous skin infections.
caused by *S. aureus* [109]. Bacteriocins are thus attractive natural alternatives to antibiotics, which can be used in the treatment of bacterial infections. A localized delivery system is, however, required to control the level and rate of bacteriocins delivered to the wound. A novel approach would be to encapsulate bacteriocins into electrospun nanofibers and use this as wound dressings for burned victims.

The feasibility of encapsulating bacteriocins into electrospun nanofibers was recently reported in [90]. The bacteriocin plantaricin 423 retained activity after electrospinning and inhibited the growth of *E. faecium* HKLHS and *Lactobacillus sakei* DSM 20017 that served as target strains. Nisin, a bacteriocin produced by *Lactococcus lactis*, was successfully loaded in PLLA nanoparticles by using semicontinuous compressed CO2 antisolvent precipitation [110]. Nisin was released in the active form and exerted its antibacterial activity up to 45 days, when incubated with a culture of *Lactobacillus delbrueckii*.

7. Conclusion and Future Trends

Electrospinning is a versatile and relatively easy technique to produce large amounts of nanofibers with diverse molecules encapsulated within. The large surface to volume ratio of nanofibers allows the encapsulation of high concentrations of bacteriocins and direct delivery to sites of skin infection. The use of bacteriocins to control infections may help to prevent further increase in antibiotic resistance amongst bacteria and may prevent negative side effects some current medication has on patients. Release of bacteriocins from nanofibers can be controlled by selecting polymers of specific composition. Furthermore, specific nanofiber scaffolds can be designed that are oxygen permeable and structurally similar to the extracellular matrix (EM) in skin.

Ideally, nanofiber wound dressings should not only contain antimicrobial agents, but a combination of compounds that would accelerate the healing process and alleviate discomfort. Such compounds may include anti-inflammatory and tissue repairing drugs. Although anti-inflammatory drugs have been encapsulated into nanofibers, no reports have been published on the encapsulation of combined compounds.

Future research has to focus on developing nanofiber wound dressings containing a combination of antimicrobial compounds, anti-inflammatory drugs, and painkillers. Furthermore, the nanofiber scaffold has to be designed to allow controlled release of the drugs over an extended period to avoid frequent changes of wound dressings. The toxicity of nanofibers needs to be researched, that is, much more *in vivo* studies need to be performed.

References


[64] M. J. Pender and L. G. Sneddon, “An e-


