Comparative Phytochemical Analysis and Antioxidant Activities of Tamalakyadi Decoction with Its Modified Dosage Forms

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Received 1 February 2019; Revised 22 March 2019; Accepted 16 April 2019; Published 2 May 2019

Guest Editor: Jesus R. R. Amado

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Background and Objective. Tamalakyadi decoction (TD) is a classical formulation mentioned in authentic traditional medicine text Sarasankshepaya under nasal diseases and used as a remedy for allergic rhinitis. It consists of 12 plant ingredients. Decoction preparations are widely used in Sri Lankan traditional system and considered effective and safe for treating many disorders. However, decoctions have to be used only in fresh state due to shorter shelf life and loss of stability. This decoction preparation method leads to decreasing the patient compliance and is also time consuming. Hence, the objective of the present study was to convert TD to consumer friendly novel dosage form, namely, freeze dried, spray dried, and traditional ganasara forms. Methodology. Therefore, we compared the phytochemical constituents and antioxidant activities of TD with its modified dosage forms. The chemical comparison of four dosage forms comprises phytochemical screening, TLC and HPTLC fingerprint profiles and the antioxidant activities by DPPH free radical scavenging activity, Ferric reducing antioxidant power (FRAP), total polyphenol content (TPC), and total flavonoid content (TFC). Results. Phytochemical screening revealed the presence of alkaloids, saponins, tannins, steroids, flavonoids, phenols, and terpenoids in all dosage forms. However, the saponins, alkaloids, flavonoids, terpenoids, and steroids were more prominent in TD and freeze dried preparation than the other two preparations. HPTLC fingerprint pattern of freeze dried dosage was more similar with HPTLC fingerprint pattern of TD in terms of number of peaks and their intensity compared to that of spray dried and ganasara dosage forms. Antioxidant activities such as DPPH, FRAP, TPC, and TFC were higher in decoction and freeze dried preparation than in spray dried and ganasara preparation. Conclusion. Freeze dried TD is the most suitable ready to use preparation having similar chemical properties and antioxidant activities to TD.

1. Introduction

Tamalakyadi decoction (TD) is an effective herbal decoction used for allergic rhinitis since long time. It is mentioned in authentic traditional medicine text, Sarasankshepaya, under nasa roga (nasal diseases) [1]. Allergic rhinitis is an IgE mediated immune response of the nasal mucosa against inhaled allergens and defined as symptoms of sneezing, rhinorrhea, nasal congestion, and itching of the nose and eyes. It is commonly defined as seasonal or perennial, depending upon whether symptoms are manifested at defined yearly intervals or throughout the year, respectively [2]. This condition is the most common allergic disorder and the prevalence of allergic rhinitis is estimated in the range from 9 % to 42 % [3]. The symptoms of allergic rhinitis may significantly affect a patient’s quality of life and can be associated with conditions such as fatigue, headache, cognitive impairment, and sleep disturbances. Appropriate management of allergic
rhinitis is an important component in effective management of coexisting or complicated respiratory conditions such as asthma, sinusitis, and sleep apnea [4].

In Ayurveda system of medicine, allergic rhinitis is described as Apeenasa or Peenasa and the concept of allergy is explained under “Asatmyaja vyadhi” (allergic disorders), while its effects are explained in hereditary, Viruddhahara (polluted substances or allergic agents) and Ritu sandhi (seasonal changes) [5].

Effective therapeutic methods for allergic rhinitis including internal as well as external treatments are described in Sri Lankan traditional system of medicine and in Ayurveda medicine. TD is one of the effective decoctions used for allergic rhinitis. It includes 12 ingredients which are mentioned in Table 1. Among those 12 ingredients, Clerodendrum serratum (L.) Moon plant and Solanum indicum L. plant were replaced by Premna herbacea Roxb. and Solanum melongena L. plants, respectively, for many years.

Decoction is a basic Ayurveda dosage form which is one of the most commonly used and considered as very effective dosage form in system of traditional medicine. However, the decoction preparations have some drawbacks such as dosage form in system of traditional medicine. However, the decoction preparations have some drawbacks such as difficulties in their busy lifestyles. Hence in this study an approach was made to prepare ready to use user friendly modified dosage forms.

2.2. Preparation Method of Tamalakyadi Decoction (TD).

TD was prepared according to the traditional decoction preparation method [7]. Five grams was taken from each ingredient of the formulation and boiled with 1920 ml of water under mild flame to reduce the volume up to 240 ml. Then the decoction was filtered through a single folded cotton cloth and collected to a separate vessel. Same procedure was repeated for eight times and pooled decoction (240 ml × 8) was divided equally (240 ml × 2) into four portions. The fist portion was labeled as TD and others were subjected to prepare modified dosage forms.

2.3. Preparation of Freeze Dried Form of Tamalakyadi Decoction (FDF-TD). TD (240 ml × 2) was freeze dried using a freeze dryer (Telstar LyoBeta) with the temperature – 45°C to 40°C and kept in a refrigerator (at 4°C) until used.

2.4. Preparation of Spray Dried Form of Tamalakyadi Decoction (SDF-TD). TD (240 ml × 2) was spray dried using a spray dryer (Mini Spray drier B-290 BUCHI) with 180°C inlet temperature, 102°C outlet temperature, and 50 kg of feed pressure.

2.5. Preparation of Ganasara Form of Tamalakyadi Decoction (GSF-TD). TD (240 ml × 2) was subjected to mild heat, converted to semisolid form, and oven dried (at 105°C) to prepare the GSF-TD [8].

2.6. Phytochemical Screening. Phytochemical screening was carried out according to the methods described by Goveas [9] and Joanne and coworkers [10] with some modifications. In brief, freshly prepared TD (240 ml × 2) and the FDF-TD, SDF-TD, and GSF-TD samples dissolved in hot water (240 ml × 2) separately were subjected to phytochemical screening studies as follows.

2.7. Test for Saponins. Five milliliters of extract and 2.5 ml of water were added to a test tube, shaken vigorously, and kept

<table>
<thead>
<tr>
<th>Plant name</th>
<th>Family</th>
<th>Sinhala name used in Sri Lanka</th>
<th>Sanskrit name</th>
<th>Used part</th>
</tr>
</thead>
<tbody>
<tr>
<td>Phyllanthus niruri L.</td>
<td>Phyllanthaceae</td>
<td>Pitawakka</td>
<td>Tamalaki</td>
<td>Whole plant</td>
</tr>
<tr>
<td>Terminalia chebula Retz.</td>
<td>Combretaceae</td>
<td>Aralu</td>
<td>Haritaki</td>
<td>Fruit cover</td>
</tr>
<tr>
<td>Premna herbacea Roxb.</td>
<td>Lamiaceae</td>
<td>Siritekku</td>
<td>Bharangi</td>
<td>Roots</td>
</tr>
<tr>
<td>Piper retrofractum Vahl</td>
<td>Piperaceae</td>
<td>Siviya</td>
<td>Chavya</td>
<td>Roots</td>
</tr>
<tr>
<td>Piper longum L.</td>
<td>Piperaceae</td>
<td>Tippili</td>
<td>Pippali</td>
<td>Fruits</td>
</tr>
<tr>
<td>Solanum trilobatum L.</td>
<td>Solanaceae</td>
<td>Wel Tibbatu</td>
<td>Vallikantakari</td>
<td>Whole plant</td>
</tr>
<tr>
<td>Solanum melongena L.</td>
<td>Solanaceae</td>
<td>Rasakinda</td>
<td>Guduchi</td>
<td>Stem</td>
</tr>
<tr>
<td>Zingiber officinalis Roscoe</td>
<td>Zingiberaceae</td>
<td>Inguru</td>
<td>Shunti</td>
<td>Dried Rhizome</td>
</tr>
<tr>
<td>Piper nigrum L.</td>
<td>Piperaceae</td>
<td>Gammiris</td>
<td>Maricha</td>
<td>Fruits</td>
</tr>
<tr>
<td>Solanum melongena L.</td>
<td>Solanaceae</td>
<td>Elabatu</td>
<td>Vruhati</td>
<td>Roots</td>
</tr>
<tr>
<td>Solanum xanthocarpum L.</td>
<td>Solanaceae</td>
<td>Katuwellatu</td>
<td>Kantakari</td>
<td>Whole plant</td>
</tr>
<tr>
<td>Justicia adhatoda L.</td>
<td>Acanthaceae</td>
<td>Adathoda</td>
<td>Vasa</td>
<td>Whole plant</td>
</tr>
</tbody>
</table>
for 10 minutes. Then the froth was mixed with 3 drops of olive oil and shaken vigorously and the formation of emulsion was observed. The presence of stable froth indicates that saponins are found in the extract.

2.8. Tests for Tannins

(a) Ferric chloride test: five drops of FeCl₃ was added to each extract and mixed well. Appearance of a black precipitate indicates the presence of tannins.

(b) Lead acetate test: three drops of Pb(OAc)₂ was added to 5 ml of extract and mixed well. Formation of a yellow precipitate is indicative of tannins.

(c) Vanillin test: few drops of 10 % vanillin in ethyl alcohol and conc. HCl were added to each extract and mixed well. Appearance of red color indicates the presence of tannins.

2.9. Test for Phenols

(a) Vanillin test: few drops of 10 % vanillin in ethyl alcohol and conc. HCl were added to 2 ml of extract. Appearance of red color indicates the presence of phenols.

(b) Lead acetate test: three drops of Pb(OAc)₂ was added to 5 ml of extract and mixed well. Formation of yellow precipitate is indicative of tannins.

(c) Wagner reagent test: two drops of Wagner reagent was added to 2 ml of extract and mixed well. Appearance of a reddish color indicates the presence of alkaloids.

2.10. Test for Alkaloids

(a) Picric acid test: few drops of picric acid was added to 5 ml of extract and mixed well. Formation of a yellow color crystalline precipitate indicates the presence of alkaloids.

(b) Tannic acid test: few drops of tannic was added to 5 ml of extract and mixed well. Formation of a yellow color crystalline precipitate indicates the presence of alkaloids.

(c) Wagner reagent test: two drops of Wagner reagent was added to 2 ml of extract and mixed well. Appearance of a reddish color indicates the presence of alkaloids.

2.11. Test for Flavonoids

(a) Five milliliters of dilute ammonia solution was added to 5 ml of extract followed by the addition of conc. H₂SO₄. Appearance of yellow color indicates the presence of flavonoids.

(b) Five milliliters of extract was added to a test tube containing piece of metallic mg and 3 drops of conc. HCl and heated. Flavonoids give a red-orange color.

2.12. Test for Terpenoids

(a) Salkowski test: extract (5 ml) was mixed with 2 ml of chloroform in a test tube and 3 ml of conc. H₂SO₄ was added along the sides of the test tube. Formation of reddish brown color is an indicative of presence of terpenoids.

(b) Test for sesquiterpenes: one milliliter of conc. H₂SO₄ was added to 2 ml of extract and mixed well. A reddish brown color indicates the presence of terpenoids.

2.13. Test for Steroids

(a) Five milliliters of acetic anhydride and 5 ml of conc. H₂SO₄ were added to the 5 ml of extract and mixed well. A color change from violet to blue or green color indicates the presence of steroids.

(b) Lieberman Burchard test reaction: two milliliters of acetic anhydride and 2 ml of conc. H₂SO₄ were added to 2 ml of extract and mixed well. Formation of a dark bluish green color indicates the presence of steroids.

2.14. Test for Cardiac Glycosides. One milliliter of glacial acetic acid was added to 3 ml of extract and conc. H₂SO₄ acid was introduced to the bottom of the tube. A reddish brown or violet brown ring at the interface of the two liquids indicates the presence of cardiac glycosides.

2.15. Development of Thin Layer Chromatography (TLC) and High Performance Thin Layer Chromatography (HPTLC) Fingerprints. Freshly prepared TD (100 ml) and FDF-TD, SDF-TD and GSF-TD dosage forms dissolved in hot water (100 ml from each) were added separately to a separating funnel containing 50 ml of dichloromethane, mixed well and kept for 20 min. After that, dichloromethane layer was separated. This was done thrice and collected dichloromethane fractions were pooled and evaporated to dryness. Dried dichloromethane fractions of TD, FDF-TD, SDF-TD, and GSF-TD dosage forms were redissolved in 5 ml of dichloromethane separately and spotted on a TLC plate. TLC fingerprint profile was developed for all fractions using dichloromethane, ethyl acetate, and cyclohexane in a ratio of 3:0.5:1.5 v/v. The plate was visualized under UV radiation (both 254 nm and 366 nm) and HPTLC fingerprint patterns were observed by using CAMAG - HPTLC scanner.

2.16. Extracts for In Vitro Antioxidant Assays. The powders obtained from freeze drying, spray drying, and ganasara methods were dissolved in methanol to prepare methanolic extracts. Liquid form decoction was dried by evaporation using rotary evaporator and redissolved in methanol.

2.17. Antioxidants Assay. The antioxidant activities of these four preparations were assessed by using DPPH free radical scavenging activity, Ferric reducing antioxidant power (FRAP), total polyphenol content (TPC), and total flavonoid content (TFC).

2.18. DPPH Free Radical Scavenging Activity. The DPPH free radical scavenging assay was performed according to the method described by Blois, [11] with some modifications in 96-well microplates. Reaction mixture of 200 µl, containing 150 µl of DPPH solution and 50 µl of each extract (dissolved
in methanol) of decoction or freeze dried or spray dried or ganasara was incubated at room temperature (25 ± 2°C) for 10 minutes in dark and the absorbance was recorded at 517 nm. Five different concentrations of Trolox (2.5, 5, 10, 20, 30 μg/ml) were used to construct the standard curve. Results were expressed as IC₅₀; μg/ml.

2.19. Ferric Reducing Antioxidant Power (FRAP). The assay was carried out according to the Benzie and Strain [12] with some modifications in 96-well microplates. The working FRAP reagent was prepared by mixing 300 mM acetate buffer (pH 3.6), 10 mM TPTZ solution and 20 mM FeCl₃ 6H₂O (10:1:1 v/v/v) just before use and incubated at 37°C for 8 minutes. Reaction mixtures of 200 μl containing 150 μl FRAP reagent, 30 μl of acetate buffer, and 20 μl of four extracts (120 μg/ml) were incubated at room temperature (25 ± 2°C) for 8 minutes and the absorbance was recorded at 700 nm. Six different concentrations of Trolox (10.3125, 20.625, 41.25, 83.5, 167 μg/ml) were used to construct the standard curve. Results were expressed as mg TE/g of extract.

2.20. Total Polyphenol Content (TPC). Total polyphenol content of four extracts was determined by the Folin-Ciocalteu spectrophotometric method adopted from Singleton and Rossi [13] by using gallic acid as standard phenolic compound using 96-well microplates. Twenty microliters of four extracts, each dissolved in distilled water (150 μg/ml), were added to 110 μl of ten times diluted freshly prepared Folin-Ciocalteu reagent and incubated with 70 μl of 10% sodium carbonate solution at room temperature (25 ± 2°C) for 30 minutes and the absorbance was recorded at 765 nm. Five different concentrations of gallic acid (0.78, 1.562, 3.125, 6.25, 12.5, 25, and 50 mg/ml) were used to construct the standard curve. Total Polyphenol Content was expressed as mg Gallic Acid Equivalents (GAE)/g of extract.

2.21. Total Flavonoid Content of (TFC). Total flavonoid content of four samples was determined by Aluminium chloride method [14]. One hundred microliters of 2% Aluminium chloride in methanol solution was incubated with 100 μl of four samples dissolved in methanol (120 μg/ml) at room temperature (25 ± 2°C) for 10 minutes and absorbance was recorded at 415 nm. Six different concentrations of Quercetin (1, 2, 4, 8, 16, 32 μg/ml) were used to construct the standard curve. Total Flavonoid Content was expressed as mg Quercetin Equivalents (QE)/g of extract.

2.22. Statistical Analysis. All the assays were performed four times and the absorbance was presented as Mean ± SEM. Analysis of variance was performed using SPSS procedures. The level of significance was used for comparison at 0.05 levels. SPSS t-test was used for testing significance level between in other.

3. Results and Discussion

In Ayurveda system of medicine we can identify various medicinal preparations mentioned under Bhaishajya Kalpana [15]. Decoctions (kashaya), vati (pills), powders (churna), oils (tails), and arishta-asava (fermented preparations) are few examples for them. These drug preparations can be classified into two: primary preparations and secondary preparations. Panchavidha Kashaya Kalpana is considered as primary preparations which include five types of liquid preparations that are therapeutically effective. These primary preparations are commonly used as the initial dosage forms in treatment and as the base for the different medicinal preparations.

Decoction is one of the effective dosage forms widely used in Ayurveda treatment and the shelf life of this preparation is 24 hours, which means, in the treatment, patient should prepare the decoction everyday [16]. If we are able to develop novel products from decoctions having long shelf life, that would be convenient for people. However, in order to fulfill this requirement, potency of the preparation should be same as the traditional formulation. Potency of a medicine is critical for its efficacy. When modifying the preparation to an easy to use dosage form with appropriate shelf life, active principles or phytochemicals of the drug have to be protected as the traditional preparation.

In this study qualitative phytochemical analysis was done to detect and compare the chemical constituents of TD and its modified dosage forms. Most of the phytochemicals including saponins, alkaloids, flavonoids, phenols, terpenoids, tannins, and steroids were present in all four types of preparations (Table 2). However, saponins, alkaloids, flavonoids, terpenoids, and steroids were more prominent in both traditional TD and FDF-TD than the SDF-TD and GSF-TD. Plant secondary metabolites such as phenols, flavonoids, tannins, and saponins are responsible for many activities including antioxidants, anti-inflammatory, antibacterial, antiasthmatic, immunomodulatory actions etc. [17]. Prolonged administration of saponin from Clerodendrum serratum plant has been reported to exhibit antihistaminic and antiallergic activity [18, 19]. C. serratum is one of the ingredients in TD and high content of saponins was found in both TD and FDF-TD. This factor helps to prove the effectiveness of TD and FDF-TD in the treatment of allergic rhinitis which is characterized by nasal congestion, watery nasal discharge, itching of the nose, and sneezing [20]. Therefore, the above properties of the drug could overcome the symptoms of allergic rhinitis. Further this is the first attempt taking place to screen possible phytochemicals present in TD.

TLC and HPTLC techniques are used for quality assessment in Ayurvedic preparations. These methods are widely employed in pharmaceutical industry in process of identification, development and quality control of herbal products [21]. However, HPTLC technique is more advanced than TLC and used for quantification purpose. When considering the TLC fingerprint patterns almost similar TLC profiles were observed in all four dosage forms bearing R₅ values of 0.12, 0.32, 0.43, 0.59, 0.70, and 0.93 (at 245 nm). However, one additional spot was observed in TD bearing R₅ value of 0.26 (at 366 nm) (Figure 1).
### Table 2: Phytochemical screening of Tamalakyadi decoction and its modified dosage forms.

<table>
<thead>
<tr>
<th>Phytoconstituents</th>
<th>Test</th>
<th>TD</th>
<th>FDF-TD</th>
<th>SDF-TD</th>
<th>GSF-TD</th>
</tr>
</thead>
<tbody>
<tr>
<td>Saponins</td>
<td>Frothing test</td>
<td>+++</td>
<td>High</td>
<td>+</td>
<td>+++</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Tannins</td>
<td>FeCl₃ test</td>
<td>++(Blue black precipitate)</td>
<td>High</td>
<td>+++</td>
<td>+++</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>Pb(OAc)₂ test</td>
<td>+++(Yellow precipitate)</td>
<td>Negative</td>
<td>+++</td>
<td>+++</td>
</tr>
<tr>
<td>Phenols</td>
<td>Vanillin test</td>
<td>Negative</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>Pb(OAc)₂ test</td>
<td>+++(Yellow precipitate)</td>
<td>++(Yellow precipitate)</td>
<td>+++(Yellow precipitate)</td>
<td>+++(Yellow precipitate)</td>
</tr>
<tr>
<td>Alkaloids</td>
<td>Tannic acid test</td>
<td>+++(yellow precipitate)</td>
<td>+</td>
<td>+++(yellow precipitate)</td>
<td>+++(yellow precipitate)</td>
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<tr>
<td></td>
<td>Picric acid test</td>
<td>+++(yellow precipitate)</td>
<td>+++(yellow precipitate)</td>
<td>+</td>
<td>+++(yellow precipitate)</td>
</tr>
<tr>
<td></td>
<td>Wagner test</td>
<td>+++(red colour)</td>
<td>++(red colour)</td>
<td>+++(red colour)</td>
<td>+++(red colour)</td>
</tr>
<tr>
<td>Flavonoids</td>
<td>Test (a)</td>
<td>+++(Yellow colour)</td>
<td>+++(Yellow colour)</td>
<td>+</td>
<td>+++(Yellow colour)</td>
</tr>
<tr>
<td></td>
<td>Test (b)</td>
<td>++(Orange colour)</td>
<td>+++(Orange colour)</td>
<td>+</td>
<td>++(Orange colour)</td>
</tr>
<tr>
<td>Terpenoids</td>
<td>Salkowski test</td>
<td>+++(reddish brown colour)</td>
<td>+</td>
<td>+++(reddish brown colour)</td>
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<td></td>
<td>Sesquiterpenes test</td>
<td>+++(reddish brown colour)</td>
<td>+</td>
<td>+++(reddish brown colour)</td>
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<td>Steroids</td>
<td>Test (a)</td>
<td>+++(violet colour)</td>
<td>+</td>
<td>+++(violet colour)</td>
<td>+++(violet colour)</td>
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<tr>
<td></td>
<td>Liebermann Burchard Test</td>
<td>+++(Dark bluish green colour)</td>
<td>+</td>
<td>+++(Dark bluish green colour)</td>
<td>+++(Dark bluish green colour)</td>
</tr>
<tr>
<td>Cardiac glycosides</td>
<td></td>
<td>Reddish brown ring formed</td>
<td>Reddish brown ring formed</td>
<td>Reddish brown ring formed</td>
<td>Reddish brown ring formed</td>
</tr>
</tbody>
</table>

- -ve: negative, +: positive in low level, ++: positive in moderate level, +++: positive in high level.

**TD**: Tamalakyadi Decoction, **FDF-TD**: Freeze Dried Form of Tamalakyadi Decoction, **SDF-TD**: Spray Dried Form of Tamalakyadi Decoction, **GSF-TD**: Ganasara Form of Tamalakyadi Decoction.

HPTLC study was carried out to compare the area and intensity of the spots appeared in TLC profiles of four preparations. HPTLC fingerprint pattern of TD was similar to that of FDF-TD in terms of number of peaks and their intensity compared to that of SDF-TD and GSF-TD (Figure 2). This may be due to the temperature and time which affect chemical constituents of plant materials during drug preparation. Decomposition of chemical constituents or change in chemical structure or reduction of chemical constituents occurred when increasing the temperature and time [22–26]. TD is the traditional preparation and all the modified dosage forms are made out of it. Therefore, FDF-TD, SDF-TD, and GSF-TD initially subjected to 105°C. However, FDF-TD will not be exposed more than 105°C as we used the freeze drying process while SDF-TD will be exposed to 180°C during the preparation of modifies dosage form. When preparation of GSF-TD will not be exposed more than 105°C but it has to keep prolong time in 105°C. Therefore, heat labile compound/s in both SDF-TD and GSF-TD may be decomposed during the preparation of modified dosage forms. This may be the reason that the chemical profile of FDF-TD was similar to that of TD. In contrast, research
Figure 1: TLC fingerprint profiles of Tamalakyadi Decoction and its modified dosage forms. A: Tamalakyadi decoction (TD), B: Freeze Dried Form of Tamalakyadi Decoction (FDF-TD), C: Spray Dried Form of Tamalakyadi Decoction (SDF-TD), D: Ganasara Form of Tamalakyadi Decoction (GSF-TD).

Figure 2: HPTLC fingerprint profiles of Tamalakyadi decoction and its modified dosage forms.
findings of Singh and coworkers [27] showed that spray dried form of Lodhradi Kashaya was chemically similar to that of conventional dosage form which was prepared according to the classical method mentioned in Sharangadhar Samhita. Different temperatures used in spray drying in different studies may have accounted for this. The freeze dried dosage form is exposed to low temperature (-40°C) which may cause less damage to phytoconstituents. Research reports revealed, when compared to air drying/ovendrying methods, freeze drying improved the retention of phytochemicals during processing and in some cases it even increased the concentration of phytochemicals blue berry and raspberry [28,29].

Antioxidants are compounds that inhibit or delay onset of oxidation and may be classified as natural or synthetic [30]. There is an increasing demand for natural antioxidants for curing and prevention of diseases indicating that compounds in natural formulations are more active than their isolated form [31]. During the recent years many changes have occurred in the management of allergic rhinitis by using medicines in various traditional medicinal systems [32, 33]. A number of scientific investigations have proven the association between antioxidants and allergic diseases and antioxidant intake seems to have a protective effect on allergic diseases like rhinitis [34–36]. Hence in this study we had examined the antioxidant activities of four preparations to detect most similar antioxidant activity with the TD (Table 3). TD exhibited highest antioxidant activity in terms of capability of scavenging DPPH radicals, ferric reducing antioxidant power, and total phenolic and flavonoid contents (Table 3). Further, the temperature had an effect on total flavonoids and phenolic contents of the plant which have major contribution to the radical scavenging activity [38–40]. Among the four preparations, total flavonoids and phenolic contents are high in TD and therefore it indicated more DPPH radical scavenging activity than the three modified preparations. Also antioxidant assay results showed that the FRAP value is highest in FDF-TD. Similar results were observed in freeze dried samples of leaves and berries of Cayratia trifolia [41] and spearmint leaves [42]. This may be due to the formation of ice crystals within the tissue matrix during the freeze drying process which can rupture the cell structure, which allows the exit of cellular components and the access of solvent [43]. Hence it can perform more antioxidant potency in the media.

### 4. Conclusion

Phytochemical studies, HPTLC patterns, and antioxidant studies showed that the FDF-TD is more similar to TD which was prepared according to traditional method. Therefore, FDF-TD can be used as a novel dosage form to treat allergic rhinitis. However, clinical evaluation is needed for further confirmation.

### Data Availability

The data used to support the findings of this study are available from the corresponding author upon request.

### Conflicts of Interest

The authors declare that they have no conflicts of interest.

### Acknowledgments

This project was funded by the University Grant Commission, Sri Lanka (UGC/ VCDRIC/PG2016(II)/IIM/02, 02.02.2017).

<table>
<thead>
<tr>
<th>Extract</th>
<th>DPPH free radical scavenging activity (IC$_{50}$/μg/mL)</th>
<th>FRAP (mg TE/g of extract)</th>
<th>TPC (mg GAE/g of extract)</th>
<th>TFC (mg QE/g of extract)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Tamalakyadi Decoction</td>
<td>8.2 ± 0.1$^a$</td>
<td>572.5 ± 2.3$^{ab}$</td>
<td>206.0 ± 2.3$^{ab}$</td>
<td>8.2 ± 0.5$^{ab2}$</td>
</tr>
<tr>
<td>Freeze Dried Form of Tamalakyadi Decoction</td>
<td>10.6 ± 0.3$^b$</td>
<td>634.2 ± 1.3$^{bb}$</td>
<td>148.2 ± 0.7$^{bb1}$</td>
<td>6.2 ± 0.3$^{bb2}$</td>
</tr>
<tr>
<td>Spray Dried Form of Tamalakyadi Decoction</td>
<td>20.8 ± 0.3$^c$</td>
<td>154.8 ± 1.9$^{cc}$</td>
<td>63.8 ± 2.0$^{cc1}$</td>
<td>2.9 ± 0.1$^{cc2}$</td>
</tr>
<tr>
<td>Ganasara Form of Tamalakyadi Decoction</td>
<td>17.9 ± 0.6$^d$</td>
<td>222.4 ± 1.0$^{dd}$</td>
<td>69.5 ± 0.4$^{dd1}$</td>
<td>2.9 ± 0.1$^{dd2}$</td>
</tr>
<tr>
<td>Trolox (standard)</td>
<td>5.35 ± 0.25$^e$</td>
<td>-</td>
<td>-</td>
<td>-</td>
</tr>
</tbody>
</table>

Results are presented as Mean±SEM (n=4). Values with the different scripts are significantly different P< 0.05 from each other.

TE: Trolox Equivalents, GAE: Gallic Acid Equivalents, QE: Quercetin Equivalents.
References


