Research Article

Nature and Evolution of the Ore-Forming Fluids from Nanmushu Carbonate-Hosted Zn-Pb Deposit in the Mayuan District, Shaanxi Province, Southwest China

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Received 5 May 2017; Revised 21 August 2017; Accepted 24 October 2017; Published 21 November 2017

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The Nanmushu carbonate-hosted Zn-Pb deposit is located in the Mayuan district of Shaanxi Province, a newly discovered metallogenic district next to the Sichuan Basin, in the northern margin of the Yangtze Block, which is the largest and the only one that is currently mined in this district. The \( \delta^{34} \)S values of sulfides are characterized by positive values with a peak around +18‰, and the reduced sulfur may have derived from reduction of \( \text{SO}_4^{2-} \) from paleoseawater or evaporitic sulfates that have possibly been leached by basinal brines during mineralization stage. Detailed fluid inclusion study shows two types of fluids in the sphalerite, quartz, dolomite, calcite and barite, that is, aqueous-salt dominant inclusions (type I) and hydrocarbon-bearing inclusions (type II). The Laser Raman spectroscopy study shows occurrence of certain amount of \( \text{CH}_4 \), \( \text{C}_4 \text{H}_6 \), and bitumen. The salinities show similar values around 6 to 12 wt% NaCl equivalent but a decreasing temperature from early to late stages (typically 200°C to 320°C in stage I, 180°C to 260°C in stage II, and 140°C to 180°C in stage III). These features may be related to basinal brines mixing between an external higher salinity \( \text{CaCl}_2 \pm \text{MgCl}_2 \)-rich fluid and a local \( \text{H}_2\text{O-NaCl} \) methane-rich fluid.

1. Introduction

Carbonate-hosted Zn-Pb deposit is one of the most important Pb and Zn producers in the world; however, the detailed ore-forming processes are still ambiguous, especially when carbonate-hosted Zn-Pb deposit is spatially associated with organic matter in or next to sedimentary basins. The close relationship between carbonate-hosted Zn-Pb deposits and hydrocarbon reservoirs in a sedimentary basin has been widely noted [1–4], but the detailed metallogenic mechanism remains poorly understood.

The Mayuan district, in the northern margin of the Yangtze Block (Figure 1), next to the Sichuan Basin, is a newly discovered Zn-Pb ore concentrated area with large economic significance. Since the discovery of these deposits in 2004, only few researches have been carried out and the ore genesis remains controversial. Qi et al. [5] suggested a stratabound syn-sedimentary hydrothermal origin for these Zn-Pb deposits, according to geological characteristics of all orebodies hosted in the Neoproterozoic Dengying Formation dolomite. Chen et al. [6] claimed that the Dengying Formation dolomite may have been influenced by hydrothermal fluids on account of trace elements and REEs data. Other researchers [7–11] suggested that these deposits belong to the Mississippi Valley-type (MVT) Zn-Pb deposit based on the fluid inclusion and isotope geochemical data. Previous microthermometry measurements show that fluid inclusions of hydrothermal epoch homogenized in temperatures between 98 and 337°C [8]. Based on \( \text{H-O-C-He-Ar-Sr} \) isotope data, Gao et al. [11] proposed that the Zn-Pb deposits in the Mayuan district may have precipitated from a mixture of crustal fluid and meteoric water. Gao et al. [11] put forward a hypothesis of organic fluid origin for ore fluids based on the low \( \delta^{13} \text{C} \) values (from −36.0 to −28.3‰ for \( \delta^{13} \text{C}_{\text{CH}_4} \))
and from −27.7 to −22.4‰ for δ13C(CH2)6, resp.) of fluid inclusion in the sphalerite, which are similar to the δ13C values of the gas reservoir in the Dengying Formation. In this deposit, hand specimen and microscopic-scale thin sections of the sulfide ores show a spatial association with bitumen [7, 8, 10]. However, the characteristics of the organic-rich fluids, the link between the organic-rich fluids, mechanisms by which the ore-forming fluids make sphalerite and galena precipitated, and the controlling factors of mineralization are still ambiguous.

There are three ore zones (North, Middle, and South) in the Mayuan district [5]. Metallogenic prospecting of the North and Middle zones is 1.0 Mt and 1.5 Mt, respectively. Further prospecting work has not started yet in the North and Middle zones. The South zone is the only one that is being currently mined in the Mayuan district. There are a series of occurrences of the Zn-Pb deposits in the South zone, among which the Nanmushu deposit is the largest one (Figure 1). The study of ore-forming fluid is an effective way to trace the ore-forming process and can provide effective information about mineralization [12–15]. In this paper, the Nanmushu deposit was selected for a detailed fluid inclusion study, and our results demonstrate a carbonate-hosted Zn-Pb deposit may have formed by mixing of high salinity brines into a local H2O-NaCl methane-rich fluid. In addition, here we also present a detailed geological and sulfur isotope study and, combined with fluid inclusion study for the Nanmushu deposit, it can provide a better understanding of the nature and origin of the ore-forming fluids and the ore genesis. These results can also provide new insights into the mechanisms that formed the ore deposits with hydrocarbon-bearing fluids. Although this approach does not attempt to explain all the deposit features in this district, it will possibly provide a reasonably detailed new understanding for a typical deposit in the Mayuan district.

2. Geological Background

2.1. Regional Geology. The Yangtze Block is separated from the North China Block by the Qinling Orogenic Belt (Figure 1(a)). The Yangtze Block is composed of Archean to Early Proterozoic basements, Mesoproterozoic folded basement, and Neoproterozoic to Mesozoic cover sequences [18–21]. The thick cover of Neoproterozoic to Permian strata is over 9 km, which comprises glacial deposits, clastic, metavolcanic rocks, and carbonate. Abundant carbonate-hosted Zn-Pb deposits are located in the northern belt of the Yangtze Block [22].

The Mayuan district is a newly discovered Zn-Pb metallogenic district, located in the northern margin of Yangtze Block, next to Sichuan Basin. This district is also located at the junction of the Qinling Orogenic Belt to the South of Yangtze Block and along the eastern margin of the Micangshan fold-and-thrust belt (Figure 1(a)). The Mayuan Zn-Pb metallogenic district includes three ore zones, namely, the South, Middle, and North zones (Figure 1(b)). The South ore zone extends over 20 km and broadens 20 to 120 m with approximately 2.1 Mt Zn (with a grade of 1.2–4.5 wt.%) and 0.1 Mt Pb (with a grade of 0.6–3.5 wt.%). This ore zone contains five deposits including Kongxigou, Lengqingpo, Nanmushu, Jiulingzi, and Jiandongzi from SWW to NEE [17, 23].

2.2. Geology of the Nanmushu Zn-Pb Deposit. The Nanmushu Zn-Pb deposit is located in the central part of the South
ore zone (Figure 1(b)). In the deposit, the outcropped strata consist of Mesoproterozoic Huodiya Group, Neoproterozoic Dengying Formation, and the Lower Cambrian Guojiaba Formation (Figure 2). The Huodiya Group is mainly composed of marbles, pyroclastic rocks, and siliceous slates, and the most common rocks of the Huodiya Group in this area are marbles. This group is unconformably overlain by the Neoproterozoic Dengying Formation carbonate rocks which mainly include dolomites with minor amount of quartz sandstones and limestones, whereas the Lower Cambrian Guojiaba Formation is characterized by argillaceous limestones and carbonaceous slates. The Dengying Formation in the study area can be subdivided into the lower member ($Z_2dn_1$) and the upper member ($Z_2dn_2$–$4$) (Figure 2). $Z_2dn_1$ (31 to 50 m thick) is characterized by clastic rocks, sandstone, and conglomerate. $Z_2dn_2^{1–4}$ can be further subdivided into four beds from the bottom upward: the first bed ($Z_2dn_2^1$, 350 to 450 m thick) consists of off-white fine-crystalline dolomite, sandstone, and pebbly sandstone; the second bed ($Z_2dn_2^2$, 40 to 80 m thick) contains gray laminar algal dolomite and gray-white laminar dolomite; the third bed ($Z_2dn_2^3$, 78 to 120 m thick) comprises banded dolomite, breccia dolomite, and dolorudite, which constitute major orebodies in the Nanmushu deposit; the fourth bed ($Z_2dn_2^4$, 35 to 50 m thick) is dominated by chert-bearing banded dolomite.

Han et al. [10] described five major faults in the Nanmushu deposit. The $F_1$ separated the Dengying Formation and the Huodiya Group with strikes 150° to 165° and dips 20° to 50°. The $F_2$ separated upper member of Dengying Formation into two parts, that is, $Z_2dn_2^2$ and $Z_2dn_2^3$. This fault is 8 to 15 m wide with strikes 155° to 165° and dips 15° to 55°. The $F_3$ is the ore-controlling fault in the mining district, which crosscuts the Dengying Formation dolomites and generally parallels to $F_2$. The $F_3$ fault is a thrust fault with strikes 150° to 165° and dips 25° to 55°. This fault is marked by a 3 to 30 m wide fracture and breccia zone which probably acted as conduits for ore-forming hydrothermal fluids and closely related to the mineralization. The $F_4$ and $F_5$ are post-ore faults that cut the orebodies.

In the Nanmushu deposit, there are five zinc orebodies and three lead-zinc orebodies (Figure 2; Table 1). Five zinc orebodies occur within $Z_2dn_2^3$ of Dengying Formation, of which single orebody is 100–2560 m in length and 0.5–32.5 m in thickness, with zinc grades at 1.3–13.1 wt.%. Three lead-zinc orebodies are 86–160 m in length and 0.8–8.4 m in thickness, with zinc grades at 1.1–6.4 wt.% and 0.9–3.3 wt.%, respectively (Table 1).

Mineralization throughout the Nanmushu Zn-Pb deposit is characterized by open space filling and occurs as breccias, veins, stockworks, and massive ores, which appear to be
controlled by the third bed of the Dengying Formation ($Z_3dN_3$). It can be considered that the cavity filling was formed mainly by hydrothermal mineralization. Hou et al. [19, 20] described the Znl as a major ore shoot, which extends 2560 m and broadens 1.5–32.5 m (average thickness at 7.6 m), with zinc grade at 1.5–11.4 wt.% (average value at 4.5 wt.%) (Table 1), with the ore shoot occupying a dilational flexure in the 1030 m height (Figure 3). Hydrothermal alteration in the Nanmushu Zn-Pb deposit, comprising silicification in the open pit at elevations of 1030 m, 1070 m, and 1170 m, respectively. Polished blocks and thin sections were initially examined by reflected and transmitted light microscopy to characterize the mineralogy, textures, and parageneses.

Twenty-nine doubly polished-thin sections (100 µm thick) of quartz, dolomite, barite, calcite, and sphalerite were prepared from representative samples of the three hydrothermal mineralization stages (I, II, and III). Petrographic examination of fluid inclusions was conducted using a Nikon E801 microscope with an ultraviolet (UV) attachment using medium-width bandpass excitation filters (330–380 nm). Oil-bearing fluid inclusions were identified by their fluorescence under UV excitation. Microthermometric measurements were conducted using a Linkham THMS-600 Heating-Freezing Systems (from -196° to 550°C) equipped with a Nikon E801 microscope at China University of Geosciences, Wuhan, China. Thermocouples were calibrated at -56.6°C, 0.0°C, and 374°C using synthetic FIs supplied by FLUID INC. The heating/freezing rate is generally 0.2° to 5°C/min, but it was reduced to <0.2°C/min near the phase transformation. The uncertainties for the measurements are ±0.5°, ±0.2°C, and ±2°C for runs in the range of -120° to -70°C, -70° to +100°C, and +100° to +600°C, respectively. Ice-melting temperatures were observed at a heating rate of less than 0.1°C/min and homogenization temperatures at rate of ≤1°C/min.

Fluid inclusions were carefully observed to identify their genetic and compositional types, vapor-liquid ratios, spatial clustering, and the species. The fluid inclusion petrographic study was based on the concept of fluid inclusion assemblages (FIAs) as described by Goldstein [24]; an approach that places fluid inclusions into assemblages that have been trapped penecontemporaneously. Fluid inclusion data was collected from both primary and pseudosecondary fluid inclusion assemblages, representing fluids trapped during crystal growth. Then, microthermometric data was analyzed from two-three inclusions in each FIA. Phase changes in every FIA generally occurred over 1°–5°C, consistent with petrographic observations that inclusions did not undergo postentrapment modification [25]. Microthermometric analyses were carried out on 258 FIAs, and the results are summarized in Table 2.

After careful examination of their petrography, nine doubly polished-thin sections that have adequate primary fluid inclusions were chosen for Laser Raman spectroscopic analysis (LRSA). LRSA was conducted at the State Key Laboratory of Geological Processes and Mineral Resources (GPMR), China University of Geosciences, Wuhan, China. In order to confirm the volatile species of FIs, representative samples were carried out with a Renishaw 1000 Raman microspectrometer according to the method of Burke [26]. Spectrograph aperture was chosen as 25 µm or 50 µm pinhole. The spectra were recorded with counting time of 20 s to 40 s, and they ranged from 50 to 3,800 cm⁻¹.

Six sphalerite samples from the main economic ore-forming stage II for conventional sulfur isotopic analyses were collected from different localities, including 1030 m in height, 1070 m in height, and 1170 m in height of the Nanmushu zinc-lead deposit (Table 3). Conventional sulfur isotope analyses of the six samples were carried out at Wuhan Institute of Geology and Mineral Resources, China Geological Survey. These samples were first broken into 0.25–0.5 mm fragments, handpicked to >95% purity under a binocular microscope and cleaned ultrasonically, then crushed to <200 mesh, and heated under vacuum condition with CuO. The SO₂ gas was generated by the oxidation reaction and sulfur isotopic compositions are measured on a MAT-251 gas mass spectrometer. The results are reported as δ³⁴S values with an analytical uncertainty of 0.2‰ (2σ). Sulfur isotopic compositions are reported in comparison with V-CDT.

### Table 1: Main characteristics of orebodies in the Nanmushu Zn-Pb deposit.

<table>
<thead>
<tr>
<th>Number</th>
<th>Length (m)</th>
<th>Thickness (m)</th>
<th>Zn grade (wt.%)</th>
<th>Pb grade (wt.%)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Range</td>
<td>Average</td>
<td>Range</td>
<td>Average</td>
</tr>
<tr>
<td>Znl</td>
<td>2560</td>
<td>1.5–32.5</td>
<td>7.6</td>
<td>1.5–11.4</td>
</tr>
<tr>
<td>Zn2</td>
<td>180</td>
<td>1.0–5.3</td>
<td>2.8</td>
<td>1.3–3.4</td>
</tr>
<tr>
<td>Zn3</td>
<td>260</td>
<td>0.5–6.4</td>
<td>2.1</td>
<td>2.7–13.1</td>
</tr>
<tr>
<td>Zn4</td>
<td>106</td>
<td>1.3</td>
<td>1.3</td>
<td>1.4</td>
</tr>
<tr>
<td>Zn5</td>
<td>100</td>
<td>1.7–2.5</td>
<td>2.1</td>
<td>1.7–1.9</td>
</tr>
<tr>
<td>PbZn1</td>
<td>86</td>
<td>1.6</td>
<td>1.6</td>
<td>1.9</td>
</tr>
<tr>
<td>PbZn2</td>
<td>160</td>
<td>0.8–8.4</td>
<td>4.6</td>
<td>1.1–3.2</td>
</tr>
<tr>
<td>PbZn3</td>
<td>160</td>
<td>5.8</td>
<td>5.8</td>
<td>1.1–6.4</td>
</tr>
</tbody>
</table>

Geofluids
Figure 3: (a) Representative simplified geologic map showing the occurrence and morphology of major orebody Zn1 at 1030 m above sea level in the Nanmushu Zn-Pb deposit (after Li et al., 2007 [9]). (b) Cross-section of the Nanmushu Zn-Pb deposit (modified from Hou et al., 2007 [23]).

4. Results

4.1. Paragenetic Mineralization Stages. According to the mineral assemblages and the cross-cutting relationships of the orebodies, three mineralization stages are identified, including the sedimentary diagenetic stage, the hydrothermal ore-forming stage, and the post-ore supergene stage. In the sedimentary diagenetic stage, vein-type bitumen occurs parallel to the country rock beddings or fractures and far away from the orebodies (Figure 4(a)). The fine-grained euhedral pyrite Py(s) of this stage is 10 to 50 μm in size and is disseminated within dolomite with the hydrothermal minerals filling interspace (Figure 4(b)). The hydrothermal ore-forming stage can be subdivided into three substages.

Stage I. Stage I characterized by the assemblage of dolomite + quartz + pyrite + sphalerite + pyrrhotite without galena (Figure 4(c)). This type mineral assemblage usually fills breccias and open spaces with sphalerite impregnation. In this stage, medium-grained pyrite Py(a) and medium-grained sphalerite Sp(a) vary in size ranging from 30 to 100 μm and 30 to 200 μm in diameter, respectively. Minor inclusions of pyrrhotite are common within the euhedral Py(a).
Table 2: A summary of fluid inclusion data for the Nanmushu Zn-Pb deposit.

<table>
<thead>
<tr>
<th>Host mineral</th>
<th>Stage I</th>
<th>Stage II</th>
<th>Stage III</th>
</tr>
</thead>
<tbody>
<tr>
<td>FIs size (µm)</td>
<td>Q</td>
<td>Sp(a)</td>
<td>Q</td>
</tr>
<tr>
<td>FI size (µm)</td>
<td>3–18</td>
<td>6–11</td>
<td>6–40</td>
</tr>
<tr>
<td>Counts of FIs</td>
<td>61</td>
<td>7</td>
<td>34</td>
</tr>
<tr>
<td>$T_m$ (°C) of type I inclusions</td>
<td>−31.3–−20.7</td>
<td>−34.7–−21.3</td>
<td>−51.6–51.3</td>
</tr>
<tr>
<td>$T_{m,HH}$ (°C) of type I inclusions</td>
<td>−27.2–−21.3</td>
<td>−11.9–−2.8</td>
<td>−15.4–−1.8</td>
</tr>
<tr>
<td>$T_{ic}$ (°C) of type I inclusions</td>
<td>−9.4–−2.4</td>
<td>−10.3–−2.8</td>
<td>−9.4–−3.6</td>
</tr>
<tr>
<td>Mean of $T_{ic}$ (°C)</td>
<td>171–340</td>
<td>186–264</td>
<td>183–276</td>
</tr>
<tr>
<td>Salinity (wt.% NaCl equivalent)</td>
<td>4.3–13.3</td>
<td>6.2–11.2</td>
<td>5.9–13.3</td>
</tr>
<tr>
<td>Salinity (wt.% NaCl + CaCl$_2$ equivalent)</td>
<td>6.6–11.6</td>
<td>11.6–14.8</td>
<td></td>
</tr>
<tr>
<td>Bulk density (g/cm$^3$)</td>
<td>0.73–0.96</td>
<td>0.75–0.92</td>
<td>0.84–0.98</td>
</tr>
<tr>
<td>Homogenization style</td>
<td>L + V $\rightarrow$ L</td>
<td>L + V $\rightarrow$ L</td>
<td>L + V $\rightarrow$ L</td>
</tr>
<tr>
<td>Laser Raman spectroscopic analysis of composition</td>
<td>$H_2O$, CH$_4$</td>
<td>$H_2O$, CH$_4$, Bitumen</td>
<td>$H_2O$, CH$_4$, C$_2$H$_2$, Bitumen</td>
</tr>
<tr>
<td>Dominant aqueous-salt species</td>
<td>$H_2O$-NaCl</td>
<td>$H_2O$-NaCl-CaCl$_2$ $\pm$ MgCl$_2$</td>
<td>$H_2O$-NaCl-CaCl$_2$ $\pm$ MgCl$_2$</td>
</tr>
</tbody>
</table>

$T_m$, first melting temperature; $T_{m,HH}$, melting temperature of hydrohalite; $T_{ic}$, melting temperature of final ice; $T_{gh}$, total homogenization temperature; Brt, barite; Cal, calcite; Dol, dolomite; Q, quartz; Sp(a), fine-grained sphalerite from hydrothermal stage I; Sp(b), coarse-grained sphalerite from hydrothermal stage II; Notes: Salinity (wt.% NaCl equivalent) estimated by final ice melting temperature ($T_{ic}$) of type I inclusions (Bodnar, 1993) [13]; Salinity (wt.% NaCl + CaCl$_2$ equivalent) estimated by $T_m$, $T_{m,HH}$, and $T_{ic}$ of type I inclusions by using the Software Package Fluids (Steele-MacInnis et al., 2010) [16]. Bulk density of type I inclusions estimated using the equations of Bodnar (1998) [13]. Dominant aqueous-salt species concluded on the basis of $T_m$ and the microthermometry behavior of fluid inclusions.
Table 3: Sulfur isotopic data of sulfides and sulfate from the Mayuand district.

<table>
<thead>
<tr>
<th>Sample ID</th>
<th>Location</th>
<th>Mineral</th>
<th>$\delta^{34}$S (‰)</th>
<th>Data source</th>
</tr>
</thead>
<tbody>
<tr>
<td>NMS-1170-14</td>
<td>Nanmushu</td>
<td>Sphalerite</td>
<td>+18.3</td>
<td>This study</td>
</tr>
<tr>
<td>NMS-1030-2</td>
<td>Nanmushu</td>
<td>Sphalerite</td>
<td>+17.4</td>
<td>This study</td>
</tr>
<tr>
<td>NMS-1170-13</td>
<td>Nanmushu</td>
<td>Sphalerite</td>
<td>+18.0</td>
<td>This study</td>
</tr>
<tr>
<td>NMS-1070-13</td>
<td>Nanmushu</td>
<td>Sphalerite</td>
<td>+17.6</td>
<td>This study</td>
</tr>
<tr>
<td>NMS-1030-3</td>
<td>Nanmushu</td>
<td>Sphalerite</td>
<td>+17.4</td>
<td>This study</td>
</tr>
<tr>
<td>NMS-1070-2</td>
<td>Nanmushu</td>
<td>Sphalerite</td>
<td>+17.8</td>
<td>This study</td>
</tr>
<tr>
<td>MAYN-1</td>
<td>Nanmushu</td>
<td>Sphalerite</td>
<td>+18.9</td>
<td>Li et al., 2007 [9]</td>
</tr>
<tr>
<td>MAYN-2</td>
<td>Nanmushu</td>
<td>Sphalerite</td>
<td>+17.7</td>
<td>Li et al., 2007 [9]</td>
</tr>
<tr>
<td>MAYN-3</td>
<td>Nanmushu</td>
<td>Sphalerite</td>
<td>+17.7</td>
<td>Li et al., 2007 [9]</td>
</tr>
<tr>
<td>MAYN-4</td>
<td>Nanmushu</td>
<td>Sphalerite</td>
<td>+18.5</td>
<td>Li et al., 2007 [9]</td>
</tr>
<tr>
<td>MAYN-5</td>
<td>Nanmushu</td>
<td>Sphalerite</td>
<td>+18.1</td>
<td>Li et al., 2007 [9]</td>
</tr>
<tr>
<td>MAYN-6</td>
<td>Nanmushu</td>
<td>Sphalerite</td>
<td>+18.5</td>
<td>Li et al., 2007 [9]</td>
</tr>
<tr>
<td>MAYN-7</td>
<td>Nanmushu</td>
<td>Sphalerite</td>
<td>+19.4</td>
<td>Li et al., 2007 [9]</td>
</tr>
<tr>
<td>MY-8</td>
<td>Nanmushu</td>
<td>Sphalerite</td>
<td>+17.0</td>
<td>Wang et al., 2008 [7]</td>
</tr>
<tr>
<td>S-1</td>
<td>Nanmushu</td>
<td>Sphalerite</td>
<td>+17.8</td>
<td>Wang et al., 2008 [7]</td>
</tr>
<tr>
<td>S-2</td>
<td>Nanmushu</td>
<td>Sphalerite</td>
<td>+18.0</td>
<td>Wang et al., 2008 [7]</td>
</tr>
<tr>
<td>MWT1</td>
<td>Nanmushu</td>
<td>Sphalerite</td>
<td>+18.2</td>
<td>Hou et al., 2007 [17]</td>
</tr>
<tr>
<td>MWT3</td>
<td>Nanmushu</td>
<td>Sphalerite</td>
<td>+17.9</td>
<td>Hou et al., 2007 [17]</td>
</tr>
<tr>
<td>H-1</td>
<td>Nanmushu</td>
<td>Pyrite</td>
<td>+18.8</td>
<td>Wang et al., 2008 [7]</td>
</tr>
<tr>
<td>H-2</td>
<td>Nanmushu</td>
<td>Pyrite</td>
<td>+18.6</td>
<td>Wang et al., 2008 [7]</td>
</tr>
<tr>
<td>F-1</td>
<td>Nanmushu</td>
<td>Galena</td>
<td>+16.8</td>
<td>Wang et al., 2008 [7]</td>
</tr>
<tr>
<td>F-2</td>
<td>Nanmushu</td>
<td>Galena</td>
<td>+16.6</td>
<td>Wang et al., 2008 [7]</td>
</tr>
<tr>
<td>F-3</td>
<td>Nanmushu</td>
<td>Galena</td>
<td>+15.6</td>
<td>Wang et al., 2008 [7]</td>
</tr>
<tr>
<td>MWT2</td>
<td>Nanmushu</td>
<td>Barite</td>
<td>+32.2</td>
<td>Hou et al., 2007 [17]</td>
</tr>
<tr>
<td>MWT3-1</td>
<td>Zhujiahe</td>
<td>Barite</td>
<td>+33.3</td>
<td>Hou et al., 2007 [17]</td>
</tr>
<tr>
<td>MWT3-2</td>
<td>Zhujiahe</td>
<td>Barite</td>
<td>+33.5</td>
<td>Hou et al., 2007 [17]</td>
</tr>
<tr>
<td>KXG-9</td>
<td>Kongxigou</td>
<td>Sphalerite</td>
<td>+16.0</td>
<td>Wang et al., 2008 [7]</td>
</tr>
<tr>
<td>MWT4</td>
<td>Nan’anshan</td>
<td>Galena</td>
<td>+12.9</td>
<td>Hou et al., 2007 [17]</td>
</tr>
</tbody>
</table>

Note. Nanmushu and Kongxigou deposits are from South ore zone of Mayuan district, and Zhujiahe and Nan’anshan are ore deposits (spots) from Middle ore zone of Mayuan district.

Stage II. Stage II is dominated by the widespread coarse-grained sphalerite Sp(b) and galena, as well as minor amount of coarse-grained pyrite Py(b). Zn-Pb mineralization starts at stage I and continues with sphalerite impregnation, followed by galena + sphalerite impregnation which subordinate stage II. The cross-cutting relationships of stage I and stage II are observed and shown in Figure 4(d). Stage II is the major mineralization stage for zinc and lead. This mineralization stage is most important in terms of tonnage and mining. Major types of ores include breccia-type, vein-type (Figure 4(d)), and high-grade massive sulfide type (Figure 4(e)), with a simple mineralogy dominated by sphalerite, galena, and pyrite in this stage. Bitumen Bit(a) can also be observed in this stage (Figure 4(f)), occurring in vein-type and breccia-type ores and usually associating with contemporary sphalerite and galena and gangue minerals, such as dolomite, calcite, and quartz. The Py(b) in this stage surrounded the Sp(b) or filled the fractures of Sp(b) and galena grains (Figure 4(g)), indicating a relatively late timing of formation.

Stage III. Stage III marked the end of hydrothermal ore-forming epoch and was characterized by quartz + barite + calcite + bitumen. Abundant bitumens Bit(b) in this stage are emulsion drops filling the open interspace between other gangue minerals and concentrated along crystal interface of gangue minerals as well as microfissuring or pores (Figures 4(h) and 4(i)), implying the bitumen might have formed a little later than the gangue minerals.

In the supergene stage, oxide minerals include smithsonite, cerussite, and limonite. A paragenetic sequence for minerals in the Nanmushu Zn-Pb deposit is summarized in Figure 5.

4.2. Fluid Inclusions Petrography and Laser Raman Spectroscopic Analysis. Samples of sphalerite (Sp(a) and Sp(b)), dolomite, quartz, calcite, and barite were selected from three hydrothermal stages in the Nanmushu deposit to determine the nature of the fluids associated with mineralization. Abundant primary, pseudosecondary, and secondary FIs in these minerals are found. In order to avoid erroneous interpretations, secondary FIs and/or FIs, which partially decrepitate during the microthermometric processes, were not documented during experiment. FIs were studied in samples that are characterized with absence of plentiful
Figure 4: Photographs of mineralization characteristics of the Nanmushu Zn-Pb deposit. (a) Vein-type bitumen (Bit(s)) occurs parallel to the country rock. (b) Breccia-type ore, dolomite breccia with Py(s) cemented by later hydrothermal mineral. (c) Breccia-type ore, yellow sphalerite + quartz (stage I) and brown sphalerite + quartz (stage II) are cemented by the dolomite breccia. (d) Vein-type ore, assemblages of coarse-grained galena with minor sphalerite as a vein (stage II) crosscut the early sphalerite vein (stage I). (e) High-grade massive ore, assemblages of coarse-grained galena and sphalerite (stage II). (f) Bit(a) associating with contemporary sphalerite occurs in vein (stage II). (g) Later Py(b) surround the early Sp(b) in stage II. (h) Sphalerite and galena as breccia (stage II), cemented by late barite + calcite + quartz + bitumen (stage III). (i) Abundant Bit(b) filled in the interspace of quartz in stage III. Bit, bitumen; Brt, barite; Cal, calcite; Dol, dolomite; Gn, galena; Py, pyrite; Qz, quartz; Sp, sphalerite.

Figure 5: Paragenetic sequence for minerals in the Nanmushu Zn-Pb deposit. Stage II is the main economic stage with abundant sphalerite and galena.
Figure 6: Photomicrographs of typical FIs hosted in different minerals of the Nanmushu Zn-Pb deposit, under transmitted light, cross-nicols. (a) A brown grain sphalerite contains abundant FIs, including one relatively large type I FI. (b) In an automorphic quartz grain, there are abundant type I and type II FIs, which contain detectable bitumen. (c) One type I FI with liquid + vapor phase in calcite. (d) One two-phase CH$_4$-rich inclusion in quartz is dominated by water with detectable methane. (e) CH$_4$-rich inclusions with dark bubbles are dominated by detectable methane, coexisting with type I FIs in barite. Brt, barite; Cal, calcite; Dol, dolomite; Qz, quartz; Sp, sphalerite.

secondary inclusions trails and exhibit no evidence of post-entrapment modification (i.e., stretching, leaking, and necking down; [25]).

The majority of samples are dominated by primary inclusions in terms of volume, although small secondary inclusions are easily observed. Fluid inclusion morphology in the samples studied varies from polyhedral to round, and their sizes are generally ranging from 4 to 20 $\mu$m. At room temperature ($20^\circ$C), most of FIs are two-phase (liquid + vapor) (Figure 6). In the order of decreasing abundance, two compositional types of FIs are distinguished.

1. The first type is aqueous-salt dominant inclusions (type I), which are found in all examined samples. Type I inclusions are about 6–18 $\mu$m on average but probably up to 40 $\mu$m in length, and they are negative crystal, ellipsoidal, near-spherical, or irregular in shape (Figures 6(a) and 6(c)). The inclusions typically consist of liquid water and a water-dominated vapor phase which occupies 5 to 40 vol. %.

2. The second type is hydrocarbon-bearing inclusions (type II), which include three-phase hydrocarbon-bearing inclusions (liquid + vapor + solid phase with bitumen, e.g., Figure 6(b)), two phases' CH$_4$-rich inclusions (Figure 6(d)), one-phase CH$_4$-rich inclusions (Figure 6(e)), and two phases' oil-bearing inclusions (Figure 7). They are of ellipsoidal, near-spherical, or irregular shapes and on average about 6–25 $\mu$m in size. The inclusions typically have very consistent aqueous to carbonic phase ratios, and fluorescent oil phase was concentrated around gas bubbles (Figures 7(a) and 7(c)), displaying consistent fluorescence colors within a particular cluster (Figure 7(f)).

At stage I, we observed abundant type I inclusions in dolomite, Sp(a), and quartz with size ranging from 3 to 18 $\mu$m.
A smaller number of fluid inclusions contain hydrocarbon (type II) in this stage. Not only CH₄-bearing inclusions are observed (determined by Laser Raman spectroscopic analysis) but also oil-bearing inclusions with oil exhibiting a yellow to cream fluorescing rim around the hydrocarbon phase (Figures 7, 8(a), and 8(b)).

At stage II, primary fluid inclusion assemblages occur as clusters of inclusions in gangue minerals, including dolomite, quartz, and coarse-grained Sp(b) with growth zones. They show regular to negative crystal shape with the size varying from 4 to 40 μm. Ninety percent of these inclusions are type I. A small number of type II inclusions in sphalerite, containing considerable amounts of CH₄ and bitumen, were observed in our study through Laser Raman spectroscopic analysis (Figure 8(c)).

At stage III, type I inclusion occurs in quartz, barite, and calcite with the size varying from 3 to 16 μm. This stage has more type II inclusions than the two early stages (I and II). There are three-phase hydrocarbon-bearing inclusions (liquid + vapor + solid phase with bitumen), two phases’ CH₄-rich inclusions, and one-phase CH₄-rich inclusions. The compositions of H₂O, CH₄, C₂H₆, and bitumen were detected by Laser Raman spectroscopic analysis (Figures 8(d)–8(f)).

4.3. Homogenization Temperature and Salinity. The fluid inclusion microthermometry was based on the concept of FIA [24], and microthermometric data were analyzed from two-three inclusions in each FIA. There are aqueous-salt dominant inclusions (type I) and hydrocarbon-bearing inclusions (type II) in the Nanmushu deposit. Figure 9 shows typical FIA in stages I, II, and III. In our study, we attempted measurements of type II inclusions in different stages; however, only a few type II inclusions were measured as a result of difficulties in observing the temperature (T_h) of phase transition. A limited number of CH₄-rich inclusions in quartz were measured for T_h. The CH₄-rich inclusions in quartz show T_h values from −139°C to −90°C (to liquid). Within individual FIA, the ranges of T_h are relatively small, for example, −139°C to −135°C, −91°C to −90°C, and −118°C to −120°C. The coexistence of the CH₄-rich inclusions with two-phase inclusions in barite suggests fluid immiscibility (Figure 6(e)). The coexistence of gas, gas-aqueous, and liquid-only inclusions in quartz indicates immiscibility and heterogeneous trapping (Figure 9(f)). As for a few two-phase inclusions with minor CH₄, their homogenization from liquid + vapor phase to liquid phase took place between 142° and 165°C.
Type I inclusion as the predominant type was performed on various minerals from different stages, and microthermometric results are summarized in Table 2 and Figure 10 (stage I, Figures 10(a) and 10(b); stage II, Figures 10(c) and 10(d); stage III, Figures 10(e) and 10(f)). The first melting temperatures ($T_m$), melting temperature of hydrohalite ($T_{m,HH}$), last ice-melting temperature ($T_{ice}$), and the temperature of total homogenization ($T_h$) have been measured in FIs hosted in dolomite, calcite, barite, sphalerite, and quartz. Salinities are calculated as wt.% NaCl equivalent using $T_{ice}$ and the equation of Roedder [12] and wt.% NaCl + CaCl$_2$ equivalent using $T_m$, $T_{m,HH}$, and $T_{ice}$ by using the Software Package Fluids [16]. At stage I, we observed abundant type I inclusions with measurements of homogenization (to liquid phase) temperatures of 171° to 340°C for primary inclusions in quartz and 225° to 314°C for inclusions in Sp(a), respectively (Table 2). Freezing experiments on type I inclusions in quartz yielded $T_m$ of −20.7°C, corresponding to the stable eutectic melting temperature (−20.8°C) of the H$_2$O-NaCl system (e.g., [13, 15]). However, one type I inclusion in quartz yielded $T_m$
Figure 9: Photomicrographs showing fluid inclusions assemblages (FIAs) of the Nanmushu Zn-Pb deposit, under transmitted light, crossed-nicols. (a) In stage I, a trail of two-phase type I FIs (a-1, a-2, and a-3) in the quartz grain, they have consistent vapor/liquid ratios. Microthermometry measurement shows that the ranges of histograms of total homogenization temperatures ($T_h$) of this FIA are relatively small, for example, 311°C for (a-1), 302°C for (a-2), and 308°C for (a-3), respectively. They all homogenized to liquid phase. Average temperature 307°C as $T_h$ of this FIA recorded in the result. (b) In stage II, a brown grain sphalerite contains two-phase FIs (b-1 and b-2) with consistent vapor/liquid ratios and similar $T_h$ (255°C for (b-1) and 251°C for (b-2), resp.). Both of them homogenized to liquid phase. Average temperature 253°C of this FIA recorded in the result. ((c), (d), (e)) A trail of CH$_4$-rich inclusions in quartz. When the temperature decreases to −190°C, most of one-phase CH$_4$-rich inclusions were frozen. When heated gradually, this type inclusions appeared two-phase ((c), at −170°C), followed by the bubbles reduced ((d), at −145°C), and then homogenized to one liquid phase ((e), −90°C). (f) Coexistence of gas, gas-aqueous, and liquid-only inclusions in quartz indicating immiscibility and heterogeneous trapping; Qz, quartz; Sp, sphalerite.
Figure 10: Histograms of total homogenization temperatures ($T_h$) and salinities at different hydrothermal stages in the Nanmushu Zn-Pb deposit. Brt, barite; Cal, calcite; Dol, dolomite; Qz, quartz; Sp(a), fine-grained sphalerite of hydrothermal stage I; Sp(b), coarse-grained sphalerite of hydrothermal stage II.
of −31.3°C, which means it may contain other composition (e.g., CH₄). T_{ice} occurred between −9.4° and −2.4°C in quartz and −28° and −3.8°C in Sp(a), corresponding to salinities of 4.0–13.3 and 6.2–11.2 wt.% NaCl equivalent, respectively (Table 2; Figure 10(b)).

At stage II, first melting temperature (T_m) was observed in a single type I inclusion in quartz at −58.7°C, indicating the presence of CaCl₂ in addition to NaCl (e.g., [12, 27]). T_m of the inclusions in Sp(b) ranges from −34.7° to −21.3°C, which are typical characteristics for the H₂O-NaCl ± CaCl₂ ± MgCl₂ system [12, 15, 27, 28]. When heated gradually, T_m was followed by the melting of hydrohalite (T_{m,HH}) and then last ice-melting temperature (T_{ice}). We attempted measurements of the temperatures of phase transition of hydrohalite (T_{m,HH}) in each FIA, but it was difficult to observe. Only a few inclusions in quartz observed T_{m,HH} values range of −27.2° to −21.3°C and T_{ice} values range of −7.5° to −3.6°C. Assuming a system of H₂O-NaCl-CaCl₂, NaCl/(NaCl + CaCl₂) ratios and compositions in wt.% NaCl + CaCl₂ equivalent were calculated from T_{m,HH} and T_{m,ice} using the Software Package Fluids of Steele-MacInnis et al. [16]. The values of NaCl/(NaCl + CaCl₂) ratios range from 0.4 to 1.0. The salinities are 6.1 to 11.5 wt.% NaCl + CaCl₂ equivalent, including 4.9–10.4 wt.% NaCl equivalent and 0.2–6.5 wt.% CaCl₂ equivalent.

Fluid inclusions in quartz have T_h values of 166° to 264°C and T_{ice} values of −10.3° to −2.8°C, corresponding to salinities of 4.7 to 14.3 wt.% NaCl equivalent. Dolomite-hosted fluid inclusions share similar T_h values of 152°–265°C and T_{ice} values from −8.3° to −1.2°C, corresponding to salinities of 5.9 to 13.3 wt.% NaCl equivalent. Sp(b) that is associated with quartz and dolomite has inclusions with T_h values of 183° to 276°C and T_{ice} values of −9.4° to −3.6°C, corresponding to salinities of 2.0 to 12.2 wt.% NaCl equivalent (Figures 10(c) and 10(d)).

At stage III, fluid inclusions have T_h values of 207° to 202°C and T_{ice} values of −15.4° to −1.2°C, corresponding to salinities of 2.1 to 19.0 wt.% NaCl equivalent (Figures 10(e) and 10(f)). When the temperature decreases to −100°C, the liquid phase of some type I inclusion was yellowish in color, which is characteristic for the eutectic composition of the H₂O-NaCl-CaCl₂ system (e.g., [4, 17]). The T_m (−51.6° to −51.3°C) and T_{m,HH} (−37.0°C) were observed in barite, supporting the H₂O-NaCl-CaCl₂ ± MgCl₂ system. Assuming a system of H₂O-NaCl-CaCl₂, salinities of 11.6–14.8 wt.% NaCl + CaCl₂ equivalent were calculated using the Software Package Fluids [16] based on the values of T_m, T_{m,HH}, and T_{m,ice}.

4.4. Sulfur Isotopic Composition. The δ³⁴S values of the analyzed sulfides are summarized in Table 3. Sulfur isotopic data in this study were combined with previous published data from the Nanmushu deposit (Figure II; [7, 9, 17]), and other deposits (e.g., Kongxigou deposit: Wang et al. [7]; Nananshan deposit: Hou et al. [17]) in the Mayuan district. Sulfur isotopic compositions of six sphalerite samples from the same stage II but different locality that was stated above have a narrow δ³⁴S range of +17.4 to +18.3‰, which are similar to those previous data at Nanmushu deposit, between +11.9 and +23.9‰ [7, 9, 17] (Table 3). The sphalerite in the Kongxigou deposit shows δ³⁴S value of +16.0‰ [7] and galena in the Nananshan deposit at Mayuan shows δ³⁴S value of +12.9‰ [7]. In addition, barites from the Nanmushu deposit and the Zhujiawei ore prospect have nearly the same δ³⁴S values of +32.2‰, and +33.3 to +33.5‰, respectively [17].

5. Discussion

5.1. Nature of the Ore-Forming Fluids. Previous published fluid inclusion microthermometric data from the Mayuan district are limited to the work of Hou et al. [17] and Liu et al. [8], Hou et al. [17] analyzed five samples from Mayuan district, with result of homogenization temperatures ranging from 130° to 210°C but without detailed sample description and fluid inclusion petrography. Liu et al. [8] reported microthermometric measurements only from quartz and calcite, and no data on other minerals, for example, dolomite, barite, and sphalerite. Their results show homogenization temperatures of 98° to 337°C (typically range from 150° to 300°C) and salinities of 7.7 to 22.2 wt.% NaCl equivalent.

In this study, systematic petrographic and microthermometric data have been carried out, and measurements of fluid inclusions in different stages and minerals in the Nannmusu Zn-Pb deposit have been presented in Table 2 and Figure 10. Aqueous-salt dominant inclusions (type I) in the Nanmushu deposit illustrate that they contain medium-low temperature (107° to 340°C) and medium-high salinity (up to 19 wt.% NaCl equivalent) brines (Table 2). Such fluids with this nature are universal in sedimentary basins and are frequently associated with sedimentary hosted Zn-Pb deposits (e.g., [14, 29]). In addition, hydrocarbon-bearing inclusions (type II) are easily observed on three hydrothermal stages, occurring as three-phase hydrocarbon-bearing inclusions (liquid + vapor + solid phase with bitumen, e.g., Figure 6(b)), two phases' CH₄-rich inclusions (Figure 6(d)), one-phase CH₄-rich inclusions (Figure 6(e)), or two phases' oil-bearing
inclusions (Figure 7). The similar characteristics are also found in some sedimentary rock hosted Zn-Pb deposits which are located in around sedimentary basins worldwide [29–32]. According to the discussion above and geological location (e.g., near Sichuan Basin) of the Nanmushu Zn-Pb deposit, we propose that the ore-forming fluids are basinal brines. As the features of ore-forming fluids (medium-low temperature, medium-high salinity with hydrocarbon), brines of sedimentary basins were thought to supply ore-forming materials and contribute to the mineralization.

The presence of hydrocarbon-bearing inclusions and bitumen, especially widely distributed CH$_4$-rich inclusions, demonstrates that there were extensive hydrocarbon activities in the Nanmushu deposit. In general, the CH$_4$-rich hydrothermal fluids may come from three ultimate sources: (1) abiogenic origin derived from the mantle [33, 34]; (2) abiogenic origin derived from postmagmatic alteration by Fischer–Tropsch type synthesis [35–37]; (3) incorporation of thermally decomposed organic material (thermogenesis) or/and products from microbial processes (bacteriogenesis) [38–41]. The deposit characteristics (without postmagmatic alteration by thermally decomposed organic material (thermogenesis) or/and products from microbial processes (bacteriogenesis) [38–41]. The deposit characteristics (without postmagmatic alteration by thermally decomposed organic material (thermogenesis) or/and products from microbial processes (bacteriogenesis) [38–41]. The deposit characteristics (without postmagmatic alteration by thermally decomposed organic material (thermogenesis) or/and products from microbial processes (bacteriogenesis) [38–41]. The deposit characteristics (without postmagmatic alteration by thermally decomposed organic material (thermogenesis) or/and products from microbial processes (bacteriogenesis) [38–41]. The deposit characteristics (without postmagmatic alteration by thermally decomposed organic material (thermogenesis) or/and products from microbial processes (bacteriogenesis) [38–41]. The deposit characteristics (without postmagmatic alteration by thermally decomposed organic material (thermogenesis) or/and products from microbial processes (bacteriogenesis) [38–41]. The deposit characteristics (without postmagmatic alteration by thermally decomposed organic material (thermogenesis) or/and products from microbial processes (bacteriogenesis) [38–41].

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5.2. Origin of the Sulfur. As many important sulfur reservoirs have distinct sulfur isotopic signature, the δ$^{34}$S values can be used as representative of the sulfur sources [42]. The δ$^{34}$S values of the sulfides show a relatively narrow range (Figure 11, Table 3) but significantly different from those of mantle-derived magmatic sulfur (around 0‰; [43]). The δ$^{34}$S values of sulfides in the Nanmushu deposit are slightly lower than those of evaporites in the Cambrian to Triassic strata (+15 to +35‰; [44]) and the seawater sulfates of the Cambrian to Triassic strata (+15 to +35‰; [44]) and the seawater sulfates of the Cambrian to Triassic strata (+15 to +35‰; [44]) and the seawater sulfates of the Cambrian to Triassic strata (+15 to +35‰; [44]) and the seawater sulfates of the Cambrian to Triassic strata (+15 to +35‰; [44]) and the seawater sulfates of the Cambrian to Triassic strata (+15 to +35‰; [44]) and the seawater sulfates of the Cambrian to Triassic strata (+15 to +35‰; [44]) and the seawater sulfates of the Cambrian to Triassic strata (+15 to +35‰; [44]) and the seawater sulfates of the Cambrian to Triassic strata (+15 to +35‰; [44]). However, the δ$^{34}$S value of barite in the Nanmushu deposit is 32.2‰ [17], similar to those of evaporites in the Cambrian to Triassic strata (Figures 6, 7, and 9). TSR-related chemical reactions may explain the abundant CH$_4$-rich inclusions.

The δ$^{34}$S values of sulfides from the Nanmushu deposit suggest that the sulfur was likely to be derived from SO$_4^{2−}$ reduction from seawater and/or evaporitic sulfates that have possibly been leached by basinal brines during mineralization. In consideration of temperatures of FIs in the Sp(a) and Sp(b) at the range of 183−314 °C, the possible mechanism for reduced sulfur is thermochemical sulfate reduction (TSR), because TSR can occur under high temperatures while bacterial sulfate reduction (BSR) can only take place under low temperatures (<127 °C) [38, 46]. In addition, TSR would like to produce a series of organic matter (e.g., bitumen, C$_2$H$_6$, C$_3$H$_8$, and CH$_4$) [38, 40, 46, 47]. Ore-forming fluids show an increase in carbonaceous contents over time. In stage I, quartz-hosted fluid inclusions with CH$_4$ in volatile and some oil-bearing inclusions occur. In stage II, CH$_4$-rich inclusions are relatively widespread, which occur not only in gangue minerals but also in sphalerite. Moreover, carbonic phase in sphalerite also includes bitumen. In stage III, the increasing amount and species of hydrocarbon-bearing inclusions (e.g., CH$_4$, C$_2$H$_6$, and bitumen) are consistent with changing fluid composition, which is likely caused by TSR during mineralization.

Therefore, it is reasonable to suggest that the positive δ$^{34}$S values (+15.6 to +19.4‰) in the Nanmushu Zn-Pb deposit were resulted from TSR. This phenomenon agrees well with geologic and FIs characteristics in the Nanmushu deposit, which show occurrence of organic matter including bitumen in mineralization stages (Figure 4) and abundant carbonaceous contents in hydrocarbon-bearing inclusions (Figures 6, 7, and 9). TSR-related chemical reactions may happen [38, 46]:

$$\text{CH}_4 + \text{SO}_4^{2−} \rightarrow \text{CO}_2^{2−} + \text{H}_2\text{S} + 2\text{H}_2\text{O}. \quad (1)$$

Hydrocarbons + CaSO$_4$ → altered hydrocarbons + solid bitumen + CaCO$_3$ + H$_2$S + H$_2$O. TSR-related reduced sulfur (e.g., H$_2$S) could have been involved in the ore-forming process with chemical reactions:

$$\text{Zn}^{2+} + \text{H}_2\text{S} \rightarrow \text{ZnS} \downarrow \text{ (sphalerite)} + 2\text{H}^+; \quad (2)$$

$$\text{Pb}^{2+} + \text{H}_2\text{S} \rightarrow \text{PbS} \downarrow \text{ (galena)} + 2\text{H}^+. \quad (2)$$

They are responsible for the deposition of sphalerite and galena in the ore-forming processes.

5.3. Evolution of the Ore-Forming Fluids. In the Nanmushu Zn-Pb deposit, microthermometry and Laser Raman spectroscopic analyses for fluid inclusions have identified amounts of salts (mainly NaCl mixed with CaCl$_2$ and MgCl$_2$) and carbonaceous contents (mainly CH$_4$, C$_2$H$_6$, and bitumen, Figure 8) in fluids. Hydrocarbon-bearing inclusions have changeable liquid/vapor ratios. Oil-bearing inclusions exhibit continuous different filling degrees with yellow fluorcesces (Figure 6). They typically occur together with only vapor hydrocarbon phase CH$_4$ inclusions and aqueous-salt inclusions (Figures 6 and 7). The coexistence of gas (CH$_4$), gas-aqueous, and liquid-only inclusions was also observed (Figure 9). Such phenomenon provides evidence that the fluid system of the Nanmushu Zn-Pb deposit consisted of an aqueous-salt solution and an immiscible hydrocarbon phase (e.g., [3, 4, 48]).

Temperatures of FIs exhibit a decreasing trend from stage I to stage III (Table 2, Figures 10 and 12). Salinities of stage I (4.3 to 13.3 wt.% NaCl equivalent), stage II (2.0 to 14.3 wt.% NaCl equivalent), and stage III (2.1 to 19.0 wt.% NaCl equivalent) are similar (Figures 10 and 12). Notably, FIs
hosted in the barite which were only found at stage III have the relatively high salinities (up to 19.0 wt.% NaCl equivalent, Table 2). On the basis of ore-forming fluids characteristics in the Nanmushu Zn-Pb deposit (Figures 10 and 12), the decoupling between the temperature and salinity in fluids may suggest that there was an addition of external fluid with higher salinity (with CaCl₂ ± MgCl₂-enriched) flow into the relatively lower salinity primary fluid system during mineralization of this deposit.

In hydrothermal stage I, the original H₂O–NaCl fluid with methane enrichment is predominately in the fluid system, with a small quantity of later fluid into the connatural fluid system. Temperature of FIs shows a distinct bimodal with maxima at around 180° and 300°C (Figure 10(a)), probably because of the original fluid and later external fluid with difference temperatures and/or heterogeneous trapping. FIs with wide ranges of Th values in stage I imply heterogeneous trapping, accompanied by immiscibility of hydrocarbon phase and aqueous-salt solution.

In stage II, with adding an increasing amount of higher salinity CaCl₂ ± MgCl₂-rich brine, the temperatures of FIs continuously decreased, and the salinities of the fluid system can reach up to 14.3 wt.% NaCl equivalent (Figure 12 and Table 2). CH₄-rich inclusions appear not only in gangue minerals but also in sphalerite. Hydrocarbon-bearing inclusions in sphalerite with CH₄ and bitumen in the carbonic phase were detected by Laser Raman spectroscopic analysis (Figure 8).

In stage III, owing to the continuous addition of the extraneous fluid, temperature of fluid became lower with its salinities rising inversely (Table 2, Figure 12). Higher number species of CH₄, C₅H₁₀, and bitumen were detected in the carbonic phase of hydrocarbon-bearing inclusions (Figure 8).

The presence of CH₄-rich inclusions in each hydrothermal stage (Table 2 and Figure 8) suggests that the original fluids contain abundant hydrocarbons. The hydrocarbons may react with sulfates to produce a series of organic matter through TSR. Thus, ore-forming fluids show an increase in carbonaceous contents and species. It implies that the hydrocarbons were entrained with the hydrothermal fluids throughout the paragenetic sequence during mineralization. Bitumen associated with contemporary sphalerite vein occurs in stage II (Figure 4). In the main stage sphalerite and quartz, primary hydrocarbon-bearing inclusions containing bitumen, also were observed and detected (Figures 4 and 8). These characteristics strongly suggest an intimate relationship between bitumen and mineralization.

Although PVTX modelling and equations of fluid inclusion from simultaneously and homogeneously entrapped FIs can be used to estimate fluid pressures, when irreversible physical and chemical changes happened, the PVTX modelling and equations are invalid. Based on the foregoing discussion, the fluid system of the Nanmushu Zn-Pb deposit is a heterogenous fluid system, which consisted of an aqueous-salt solution and an immiscible hydrocarbon phase. However, TSR is an irreversible process [38]. When solid bitumen as hydrocarbon phase appears in the hydrocarbon-bearing inclusions, it implies that irreversible physical and chemical changes happened in the closed system of the hydrocarbon inclusions [15]. Actually, TSR and solid bitumen as hydrocarbon phase in the hydrocarbon-bearing inclusions both exist in the Nanmushu Zn-Pb deposit. Taking these into account, it seems there are no appropriate FIs to calculating pressure on account of complex fluid system in the Nanmushu Zn-Pb deposit.

5.4. Genetic Type of the Deposit. Ore genesis of the Nanmushu Zn-Pb deposit has been discussed since it was first discovered in 2004. In spite of the orebodies generally controlled by stratum and structure, the occurrence of abundant vein-type and breccia-type ores shows that hydrothermal activity is critical for the mineralization. Ore-forming fluids in the Nanmushu deposit are medium-low temperature (107–340°C) and medium-high salinity (up to 19 wt.% NaCl equivalent) brines with carbonaceous components of CH₄, C₅H₁₀, and bitumen concentrations, similar to those of basinal brine systems with hydrocarbons worldwide [49–51]. Lower Cambrian Guojiaba Formation carbonaceous slate may have acted as a capping aquitard for mineralizing fluids flowing through the more permeable sandstone and dolomite.

The sphalerites from the sediment-hosted Zn-Pb ores at the Mayu district have been dated with a Rb-Sr isochron age of 486 ± 12 Ma (Ordovician, Li et al. [9]). Recently, Han et al. [10] collected the later hydrothermal bitumen and obtained a bitumen Re-Os isochron age of 128 ± 8 Ma. In this deposit, three generations of bitumen have been observed, including Bit(s), Bit(a), and Bit(b) that have distinctive association of minerals and textures. The Bit(s) formed during sedimentary diagenetic stage, far from the orebody, and they occur parallel to the country rock beddings and fractures. The Bit(a) formed during mineralization of the main economic stage II. Post-ore Bit(b) formed later than the gangue minerals of stage III or even later than the whole mineralization stages. Based on the sample descriptions of Li et al. [9] and Han et al. [10] and the petrography observations of our study, we regard that ca.486 Ma for sphalerite (Sp(b)) represents the major mineralization age of stage II and ca.128 Ma for bitumen (Bit(b)) formed during post-ore stage. Both of them
are much younger than the sedimentation age of the host Neoproterozoic Dengying Formation (~540 Ma, [52]). The mineralization of the Nanmushu deposit, located in the northern Yangtze Block, may have lasted for a prolonged period. The obtained ages fall in the range between Caledonian orogeny and Indosinian-Yanshanian movement, which both have been regarded as the two main mineralization events in the Yangtze Block. Compaction and topographically driven fluid flows are the likely mechanisms in the Nanmushu Zn-Pb deposit. Sediment compaction probably occurred during the Neoproterozoic to Permian that caused thick cover sequence (over 9 km) and major sedimentation in the basin, while the basin compression and topographic relief occurred during the Indosinian-Yanshanian movement. The foreland basin system of the northern Yangtze plate records the intracontinental tectonic deformation and oblique amalgamation of the Yangtze and North China-Qinling-Dabie complex plates during the late Triassic (late Indosinian) to probably early Jurassic (early Yanshanian) [53]. The northern Sichuan basins were filled by the depositional system and the Micianshan fold-and-thrust belt belonged to the foreland belt. It is noted that the Nanmushu Zn-Pb deposit occurs near northern Sichuan Basin and the Micianshan fold-and-thrust belt. Therefore, the aforementioned tectonic movement could have driven the large-scale flow of basal brines carrying ore-forming metals to migrate a long distance, which was in response to the Nanmushu Zn-Pb deposit mineralization.

In the Sichuan basins, the Pb-Zn deposits and paleo-oil-gas reservoirs show a closed spatial distribution and they are possibly controlled by the basal fluid flow [47, 54]. The compressional tectonic event may influence the Chipu Pb-Zn deposit and the gas reservoir hosted in the Dengying Formation may contribute to mineralization [54]. It is noted that the Chipu Pb-Zn deposit has abundant organic matter and is hosted within the Dengying Formation, similar to the Namushu Zn-Pb deposit. We speculate organic matter in the Nanmushu Zn-Pb deposit possibly came from a paleo-oil-gas reservoir on and/or near the sites of the Nanmushu Zn-Pb deposit. Given geological, geochemical, and fluid inclusion characteristics mentioned above for the Nanmushu Zn-Pb deposit, it is consistent with typical MVT deposits. They are epigenetic deposits, which are hosted in carbonate-dominated sedimentary rocks, generally stratabound and stratiform with sulfide minerals replacing carbonates and filling open spaces in the host rocks [29, 55]. Most of the world’s MVT Zn-Pb deposits occur in orogenic forelands and mineralizations are accompanied by tectonic movement [29, 56]. In addition, many of typical MVT deposits have low-moderate temperatures and medium salinities of basinal brines with abundant hydrocarbons [29], which are consistent with the Nanmushu deposit studied here. Therefore, we suggest that the Nanmushu deposit is a MVT Zn-Pb deposit.

6. Conclusions

The Nanmushu deposit is a MVT Zn-Pb deposit, and the mineralization contains three stages: (1) sedimentary diagenetic stage, (2) hydrothermal ore-forming stage with three substages (I, II, and III), and (3) supergene stage. The hydrothermal stage II is the main economic mineralization stage.

The $\delta^{34}$S values of sulfides in the Nanmushu deposit are characterized by positive values with a peak around +18%, and the reduced sulfur may have derived from reduction of $SO_4^{2-}$ from seawater and/or evaporitic sulfates. Thermochemical sulfate reduction (TSR) is the likely key factor for sulfur isotope fractionation during the ore-forming processes.

Ore-forming fluids are basinal brines, trapped in a heterogeneous fluid system, with medium-low temperature ($107^\circ$ to $340^\circ$C) and medium-high salinity (up to 19 wt.% NaCl equivalent). The majority of fluid inclusions are aqueous-salt dominant inclusions (type I) with minor hydrocarbon-bearing inclusions (type II) with amount of CH$_4$, C$_4$H$_8$, and bitumen. Microthermometric data exhibit temperature continuously decreased (171 to 340°C in stage I, 152 to 276°C in stage II, and 107 to 202°C in stage III); however, the salinities maintain approximately the same at around 6 to 12 wt.% NaCl equivalent in the two major ore-forming stages (I and II), and the salinities of barite-hosted FIs in stage III show a slight increase up to 19.0 wt.% NaCl equivalent. These features suggest a mixing of higher salinity CaCl$_2$ + MgCl$_2$-rich fluids into the original H$_2$O-NaCl with methane-rich fluid system. Different temperature-salinity fluids mixing and TSR could be the predominant factors to control the Zn-Pb mineralization in the Nanmushu deposit.

Conflicts of Interest

The authors declare that there are no conflicts of interest regarding the publication of this paper.

Acknowledgments

This research was supported by projects from the National Natural Science Foundation of China (41603042 and 41672140), China Geological Survey (12120103094200), and the Fundamental Research Funds for the Central Universities, China University of Geosciences, Wuhan (CUGQYZX1733). The authors are grateful to Mr. Ying Ma and Qi-Zhi Chen and Dr. Man-Rong Jiang for helping with sample preparation and Professor Mou-Chun He for helping with Laser Raman spectroscopy analysis.

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