

Research Article

Plasma-Based Surface Modification of Polydimethylsiloxane for PDMS-PDMS Molding

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We present and compare two processes for plasma-based surface modification of Polydimethylsiloxane (PDMS) to achieve the antisticking behavior needed for PDMS-PDMS molding. The studied processes were oxygen plasma activation for vapor phase silanization and plasma polymerization with tetrafluoromethane/hydrogen mixtures under different processing conditions. We analyzed topography changes of the treated surfaces by atomic force microscopy and contact angle measurements. Plasma treatment were conducted in a parallel plate reactive ion etching reactor at a pressure of 300 mTorr, 30 Watts of RF power and a total flow rate of 30 sccm of a gas mixture. We found for both processes that short, low power, treatments are better to create long-term modifications of the chemistry of the polymer surface while longer processes or thicker films tend to degrade faster with the use leaving rough surfaces with higher adherence to the molded material.

1. Introduction

As reviewed by Macdonald and Whitesides [1], PDMS (polydimethylsiloxane) has proven to be a suitable material for microfluidic device fabrication. It is optically transparent and can be molded to reproduce structures down to nanoscale resolution [2, 3], usually by direct casting on photoresist features over silicon wafers. SU8 is the most commonly used photoresist for master fabrication; single- or multilevel SU8 lines and structures have been extendedly used for PDMS microchannel definition, however, reutilization of resist masters is limited due to degradation by thermal stress during the curing process and by demolding forces. Some strategies such as coatings with diamond-like carbon (DLC) and DLC-polymer hybrids can be applied to increase the durability of the molds [4].

In this paper we present an improved method to replace the use of conventional photoresist masters with more durable positive copies in PDMS. Due to its elasticity, flexible masters in PDMS are not affected by thermal stress during the heating-cooling cycles of the curing process and it is easier to achieve a conformal contact between mold and counter-mold in closed molding configurations for the production

of parts with through holes. Similar methods have been presented to mold complex 3D structures with through holes: Lucas et al. in 2008 [5] presented the use of flexible molds made with freestanding SU8 layers for double-side molding of PDMS, Jo et al. in 2000 [6] used a plastic sheet as counter-mold for SU8 molds and Janelle et al. in 2000 [7] presented the use of slabs of PDMS treated with thretridecafluoro-tetrahydrooctyl-trichlorosilane as counter-molds.

Here we explore two plasma processes for surface modification of PDMS to achieve the antisticking behavior needed for PDMS-PDMS molding. For this application, three main characteristics are needed from the plasma treatment: the peeling force required to demold the PDMS parts off the treated molds should be low, the surface roughness of the mold should not be considerably increased, and the antisticking effect should be permanent and stable over the time.

2. Materials and Methods

2.1. Sample Preparation. We prepared PDMS samples using Sylgard 184 from Dow Corning, mixed in 1:10 proportion of curing agent to base polymer, degassed in vacuum for

TABLE 1: Recipes for photolithography (SU8-2050).

Thickness	Spinning	Before bake	Exp. dose	After bake	Developing
50 μm	3000 RPM	95° 15 min	600 mJ cm^{-2}	95° 5 min	5 min
300 μm	500 RPM	95° 4 hours	3600 mJ cm^{-2}	95° 15 min	30 min

30 min and oven baked at 100°C for 30 minutes after casting over silicon wafers. We prepared also PDMS samples with rough surfaces, by casting on the rear face of a silicon wafer to increase the contact area for adhesion tests. All PDMS pieces were cast with a thickness of approximately 2 mm, cut in 2 cm \times 2 cm squares and plasma bonded to 1'' \times 3'' glass slides for easy handling. The samples were rinsed with distilled water, sonicated in isopropyl alcohol for 5 minutes, and rinsed again in distilled water to remove particles prior to the treatments.

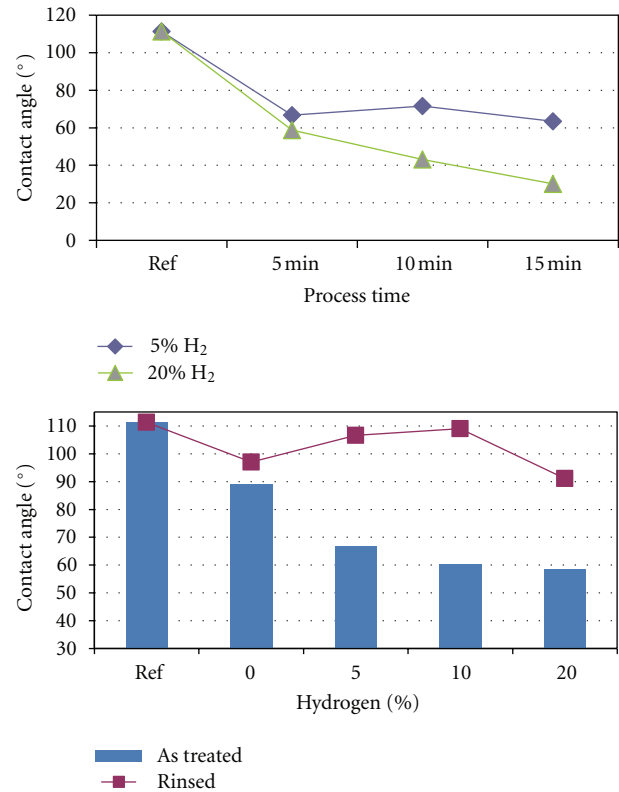
2.2. Fluorinated Plasma Treatments. This treatment consists in a plasma polymerization of fluorinate compounds over the PDMS surface using CF_4/H_2 mixtures. Two different CF_4/H_2 proportions were studied: 5% hydrogen and 20% hydrogen with process times of 1, 5, and 10 minutes. Fluorinated plasma treatments were conducted in a parallel plate RIE reactor at a pressure of 300 mTorr, 30 Watts of RF power (13.56 MHz), and a total flow rate of 30 sccm of the desired gas mixture.

2.3. Plasma-Activated Silanization. The second method consists in an oxygen plasma activation of the PDMS surface to promote de incorporation of Hexamethyldisiloxane molecules. PDMS samples were plasma oxidized in a remote plasma reactor to avoid direct bombardment of the surface. This reactor is composed by a 1'' quartz tube, 20 cm long, with an oxygen inlet at one end connected to a vacuum chamber containing the samples at the other end. High DC voltage is applied between the ends to produce a glow discharge at 300 mTorr 18 W of DC power for 50 seconds and placed on a hotplate at 100°C with two drops of HMDS, covered with a 6'' Petri dish immediately afterward.

2.4. Surface Topography and Adhesion to PDMS. The surface topography of the processed samples was analyzed by atomic force microscopy using a NanoSurf Nanite AFM. We did also contact angle measurements with a simple imaging setup and using the low-bond axisymmetric drop shape analysis (LB-ADSA) script for Image J developed by Stalder et al. [8].

The adhesive forces of the different treated surfaces to the PDMS were qualitatively compared by manual peeling straps of PDMS cured over each sample, ordering from low to high the force required to peel the PDMS apart. The surface interactions with PDMS were also studied by atomic force spectroscopy (AFS), with a contact-mode AFM tip (NanoWorld CNTR-10) coated with PDMS.

2.5. Flexible Molds Fabrication. To illustrate the application of the studied antisticking treatments for the fabrication of flexible PDMS copies of micromolds, we fabricated two level

FIGURE 1: Contact angle of PDMS samples treated with CF_4/H_2 under different conditions.

SU8 masters onto 3'' glass wafers by using high resolution photoplotting sheet as a mask. The first layer was 50 μm thick, to define channels and the second layer 300 μm thick for interconnection posts. Table 1 summarizes the lithographic recipes for each layer.

The flexible molds are obtained in a two steps process, first we use the SU8/silicon mold to cast a PDMS inverted copy that is then treated with the selected antisticking plasma process and used to cast the positive copy in PDMS that is also plasma treated for further use.

3. Results and Discussion

3.1. Contact Angle Measurements. In a preliminary study we had observed that the contact angle of the PDMS samples to water decreases with the plasma processing time and with the hydrogen percentage in the gas mixture, as shown in Figure 1, but the hydrophobicity of the samples is partially recovered after rinsing with water.

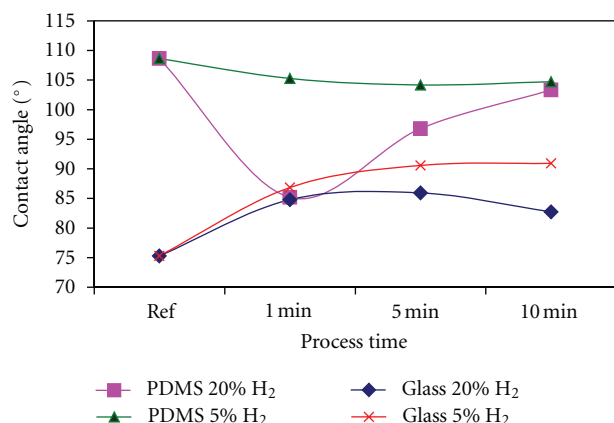


FIGURE 2: Contact angle of PDMS and glass samples processed with CF_4/H_2 plasmas at 300 mTorr 30 W.

The hydrophilic behaviour of the treated samples can be explained by three possible factors: an increment in the surface roughness, the polymerization of a less hydrophobic film on the surface, or the production of polar or charged terminations in the PDMS chains. As the surfaces became again hydrophobic after rinsing with water it is plausible that instable polar groups be primarily responsible for the measured low contact angles and this groups are removed by the dipole interactions with the water.

The curves in Figure 2 correspond to the contact angles measured on PDMS and glass of the treated samples after rinsing for 10 minutes with IPA and water to remove non-bonded compounds. On glass, the fluorinated plasma treatment has the effect of increasing the contact angle while on PDMS the treatment decreases the contact angle what indicates that the deposited polymer has an intermediate contact angle around 90° . The results of both treatment with 5% hydrogen and 20% hydrogen are similar on glass but very dissimilar on the PDMS, this occurs by the depletion of fluorine ions by the etching reactions with the glass that increase locally the relative proportion of hydrogen, getting close to the saturation of polymerization rate observed after 20% of hydrogen.

Comparing the evolution of the contact angle of the “as treated” samples with the process time (Figure 1) and the same for samples after rinsing with IPA (Figure 2), it is evident that the rinsing process degrades somehow the polymerized film, recovering partially the original contact angle value of the original substrate.

From Figure 2 it is also noted that thicker films, deposited with 20% hydrogen and longer times, are more affected by the rinsing process and that the adherence of the fluorinated polymer to the glass substrates seems to be stronger than the adherence to PDMS substrates.

3.2. Antisticking Behavior. All the studied treatments conferred to the PDMS in major or minor degree the desired antisticking effect and it was possible to classify from low to high the peeling force required to manually lift with tweezers a $200\ \mu\text{L}$ drop of PDMS cured on each treated surface. Seven

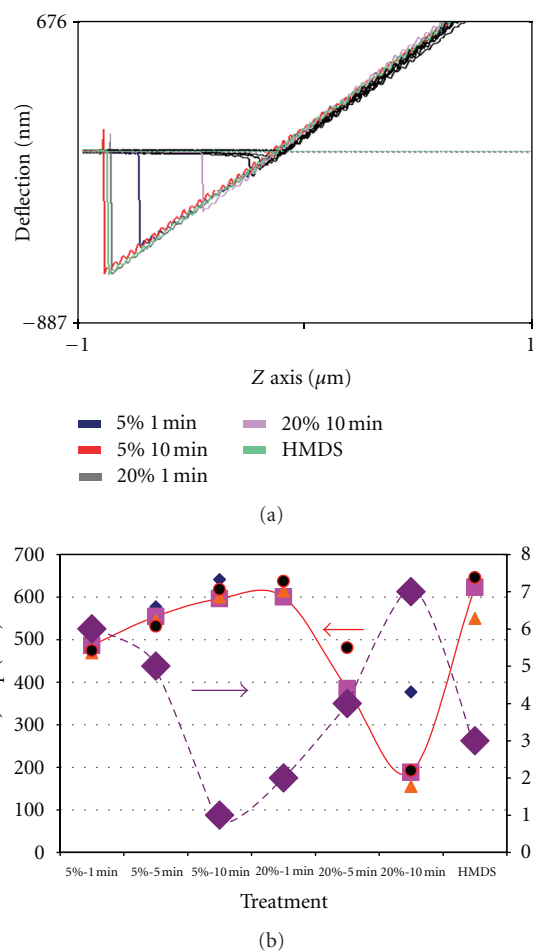


FIGURE 3: Atomic force spectroscopy curves (a) and AFS pull-off jump (b).

samples corresponding to two CF_4/H_2 proportions for three processing times plus the silanized PDMS sample were then indexed from 1 to 7 according to the qualitative comparison made by two experimenters in three independent repetitions.

The atomic force spectrograms of the samples exhibited different pull-off jumps between treatments and also different point to point variability within each sample. In Figure 3 we plotted the different AFS spectrograms. It was expected that the pull-off jump, which is the maximum negative deflection of the AFM cantilever before it loses contact with the sample in the backward part of the AFS cycle, would be proportional to the adhesive force of the PDMS to the treated surface, however when we plotted together the 1–7 index of the peeling force and the AFS data of the seven samples the results were completely inverse; the graph in Figure 4 at the right shows this superimposed data plots where the lower pull off jump corresponded to the higher peeling force and vice versa.

This apparent inconsistency could be related to the surface roughness that can affect the effective contact area between the coated tip and the sample; if the surface of a sample is so rough that only a few peaks of its topography

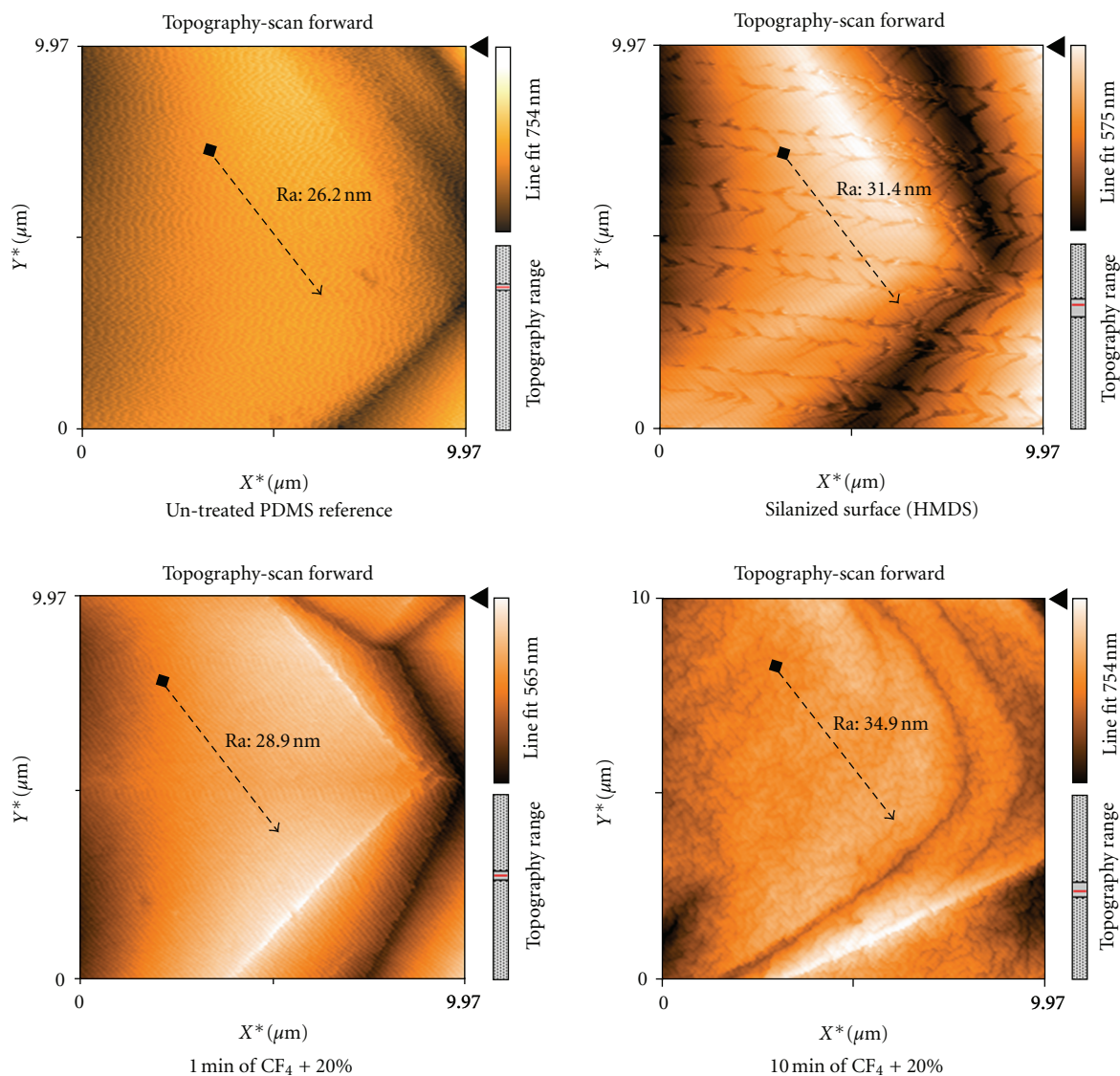


FIGURE 4: AFM images of the PDMS samples: untreated PDMS reference, silanized surface, and $\text{CF}_4 + 20\% \text{H}_2$ for 1 and 10 minutes.

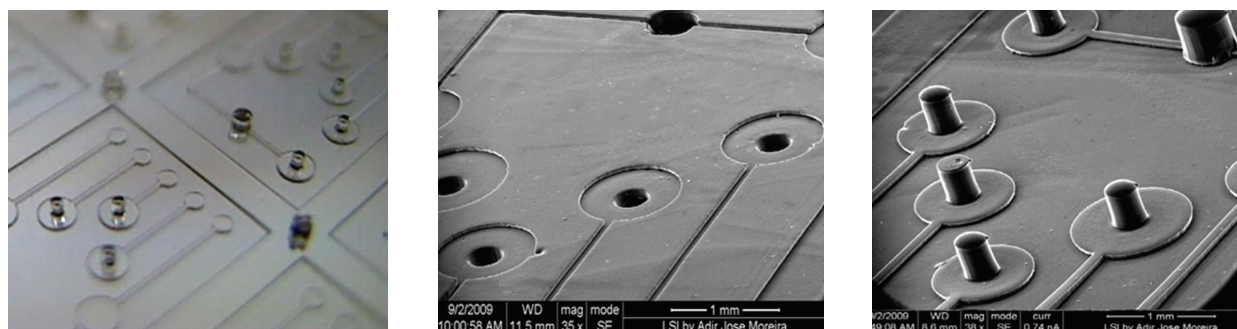


FIGURE 5: SU8 mold (left), negative copy in PDMS (center), and positive copy (right).

touch the coated AFM tip then the measured pull of force will be low, but on the contrary the effective contact area with a PDMS drop cured on it will be higher requiring a bigger force to peel it off.

The samples with higher variability that correspond to the lower pull off forces are also the samples with thicker polymerized films (20% hydrogen, 5 and 10 min) that appeared to be degraded during the rinsing, this reaffirms

that in some places the film could be away giving different measurements from point to point. The AFM images of the surface topography can give more information on this topic.

3.3. Surface Topography. In Figure 4 we present the AFM images of some treated surfaces and the untreated PDMS reference. The three samples processed with 5% hydrogen and the sample for one minute process with 20% hydrogen showed a surface quality similar to the untreated reference, all of them remained optically clear with a slight increase in roughness; only the samples with longer treatments for higher hydrogen concentration produced thicker brown films that led to a rough texture after rinsing.

These results are consistent with the previous observations of AFS and contact angle measurements, indicating that thinner films are more stable than thicker films. The HMDS silanization formed a cracked-like texture on the samples but with a low main roughness.

Comparing all the studied treatments we found very similar results for 1 minute CF_4 + 20% hydrogen and 10 minutes CF_4 + 5% hydrogen what could be associated with a similar film thickness, both having low peeling forces, low roughness and the higher measured AFS pull-off jump. The silanization process showed also similar AFS values and a low peeling force, however, with a characteristic texture that could be problematic for some applications.

3.4. PDMS-PMDS Molding. In Figure 5 we present optical and electronic micrographs of a micromold and its negative and positive copies obtained in PDMS with the proposed replication technique. Demolding PDMS parts from this flexible copies was found to be even easier than from the original SU8 Mold.

4. Conclusion

All studied treatments achieved the desired antisticking effect on PDMS surfaces, however long plasma treatments increase the surface roughness and create thicker films that are delaminated by rinsing or demolding forces losing its effect. The best results were obtained with 1 min treatment CF_4 + 20% H_2 , 10 min treatment CF_4 + 5% H_2 , and plasma-activated silanization, all of them requiring low peeling forces and having a good surface quality.

We demonstrated the applicability of the developed method for the fabrication of flexible copies of SU8 masters for microfluidics.

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