

# SUPPORTING INFORMATION

associated with the paper

## Indole-3-carbaldehyde semicarbazone derivatives: synthesis, characterization and antibacterial activities

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**Table S1.** Computational results, at B3LYP/6-311++G(d,p) for the synthesized compounds **1-4**, indicating the relative enthalpies ( $\Delta H$ ) of stable conformers and their equilibrium molar fractions (populations).

Compound	$E_0^a$	$G_{298}^a$	$H_{298}^a$	$\Delta H^b$	$\mu^c$	Population (%)	
<b>Gas phase</b>							
<b>1</b>	<i>cisE</i>	-3255.050809	-3254.911800	-3254.851474	0.0	6.6	98.4
	<i>transE</i>	-3255.040251	-3254.900459	-3254.840973	27.6	1.3	< 0.1
	<i>cisZ</i>	-3255.049470	-3254.907928	-3254.849459	5.3	3.2	1.6
	<i>transZ</i>	-3255.034754	-3254.894224	-3254.835092	43.0	8.6	< 0.1
<b>2</b>	<i>cisE</i>	-1141.131072	-1140.990312	-1140.931390	0.0	6.6	98.3
	<i>transE</i>	-1141.120482	-1140.979122	-1140.920872	27.6	1.3	< 0.1
	<i>cisZ</i>	-1141.129728	-1140.986502	-1140.929365	5.3	3.2	1.7
	<i>transZ</i>	-1141.115249	-1140.972923	-1140.915190	42.5	7.8	< 0.1
<b>3</b>	<i>cisE</i>	-796.063614	-795.882397	-795.820764	0.0	5.0	97.1
	<i>transE</i>	-796.052321	-795.870550	-795.809559	29.4	3.5	< 0.1
	<i>cisZ</i>	-796.062800	-795.879095	-795.819238	4.0	3.7	2.9
	<i>transZ</i>	-796.048584	-795.865752	-795.805313	40.6	6.4	< 0.1
<b>4</b>	<i>cisE</i>	-886.067000	-885.914670	-885.853751	0.0	9.8	99.3
	<i>transE</i>	-886.057790	-885.905898	-885.844689	23.8	3.5	< 0.1
	<i>cisZ</i>	-886.063201	-885.909912	-885.849531	11.1	6.7	0.6

	<i>transZ</i>	-886.048547	-885.896563	-885.835265	48.5	10.6	< 0.1
<b>DMSO</b>							
1	<i>cisE</i>	-3255.071545	-3254.932256	-3254.872302	0.0	8.9	99.0
	<i>transE</i>	-3255.065972	-3254.926233	-3254.866717	14.7	2.5	0.2
	<i>cisZ</i>	-3255.068655	-3254.927791	-3254.868932	8.8	4.4	0.9
	<i>transZ</i>	-3255.062547	-3254.921449	-3254.862782	25.0	12.4	< 0.1
2	<i>cisE</i>	-1141.151749	-1141.010544	-1140.952116	0.0	9.0	99.1
	<i>transE</i>	-1141.146194	-1141.004357	-1140.946454	14.9	2.4	0.1
	<i>cisZ</i>	-1141.148867	-1141.005914	-1140.948695	9.0	4.4	0.7
	<i>transZ</i>	-1141.142547	-1141.000303	-1140.942509	25.2	11.8	< 0.1
3	<i>cisE</i>	-796.085256	-795.903600	-795.842489	0.0	6.6	98.8
	<i>transE</i>	-796.079589	-795.897464	-795.836751	15.1	5.4	0.1
	<i>cisZ</i>	-796.082821	-795.899337	-795.839470	7.9	4.6	1.1
	<i>transZ</i>	-796.076475	-795.893690	-795.833279	24.2	9.4	< 0.1
4	<i>cisE</i>	-886.091918	-885.941255	-885.878999	0.0	13.9	99.5
	<i>transE</i>	-886.086453	-885.935397	-885.873509	14.4	5.2	0.2
	<i>cisZ</i>	-886.088371	-885.935785	-885.875106	10.2	9.7	0.3
	<i>transZ</i>	-886.082251	-885.929722	-885.868982	26.3	16.2	< 0.1
<b>Acetone</b>							
1	<i>cisE</i>	-3255.070572	-3254.931303	-3254.871323	0.0	8.8	99.0
	<i>transE</i>	-3255.064633	-3254.924875	-3254.865411	15.5	2.4	0.1
	<i>cisZ</i>	-3255.067789	-3254.926894	-3254.868050	8.6	4.4	0.9
	<i>transZ</i>	-3255.061059	-3254.919791	-3254.861288	26.3	12.2	< 0.1
2	<i>cisE</i>	-1141.150780	-1141.009575	-1140.951127	0.0	8.9	99.1
	<i>transE</i>	-1141.144868	-1141.002958	-1140.945101	15.8	2.3	0.1
	<i>cisZ</i>	-1141.148005	-1141.005007	-1140.947811	8.7	4.4	0.8
	<i>transZ</i>	-1141.141101	-1140.998529	-1140.941015	26.5	11.6	< 0.1
3	<i>cisE</i>	-796.084220	-795.902582	-795.841433	0.0	6.6	98.8
	<i>transE</i>	-796.078141	-795.895937	-795.835275	16.2	5.3	0.1
	<i>cisZ</i>	-796.081904	-795.898369	-795.838530	7.6	4.5	1.1
	<i>transZ</i>	-796.075004	-795.892006	-795.831779	25.3	9.2	< 0.1
4	<i>cisE</i>	-886.090707	-885.939664	-885.877745	0.0	13.8	99.4
	<i>transE</i>	-886.085030	-885.933621	-885.872072	14.9	5.1	0.2
	<i>cisZ</i>	-886.087206	-885.934617	-885.873907	10.1	9.5	0.5
	<i>transZ</i>	-886.080458	-885.927848	-885.867158	27.8	15.8	< 0.1

<sup>a</sup>All values in units of Hartree/particle. <sup>b</sup> Values in in kJ·mol<sup>-1</sup>. <sup>c</sup>Values in Debye.  $E_0$  = Total Energies at 0 K.  $H_{298}$  = Enthalpies and  $G_{298}$  = Gibbs Free Energies at 298.15 K.  $\mu$  = Dipole Moment.

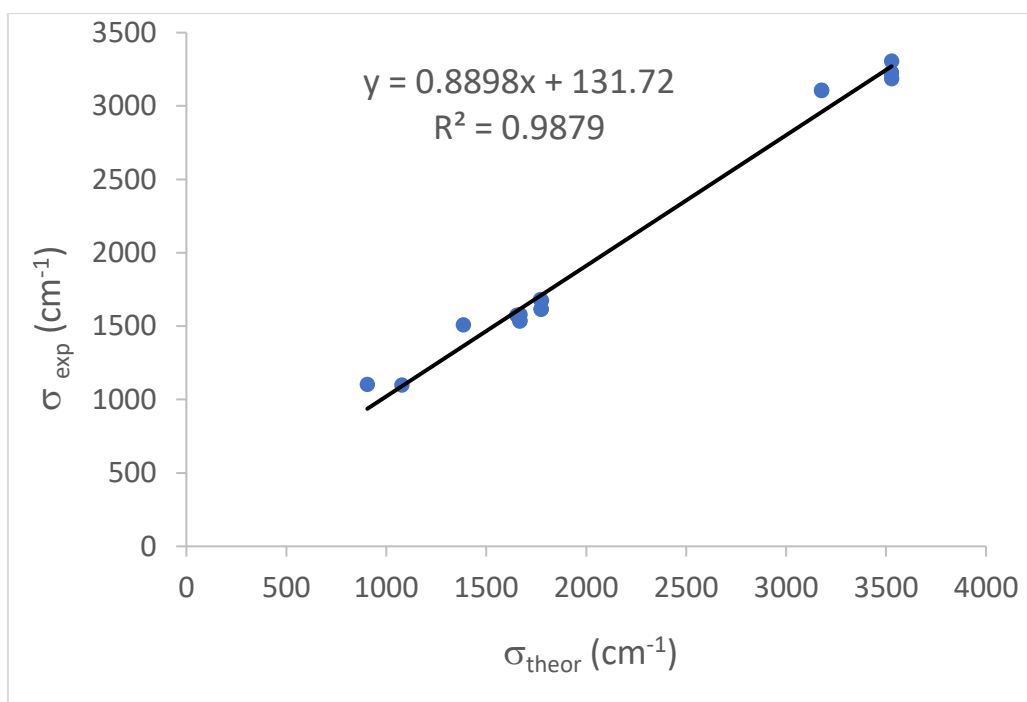


Figure 1. Linear Correlation of vibrational frequencies (given as wave numbers  $\sigma$ , in  $\text{cm}^{-1}$ ) between experimental IR data and the corresponding to B3LYP/6-311++G(d,p) theoretical values for the synthesized compounds **1-4**.

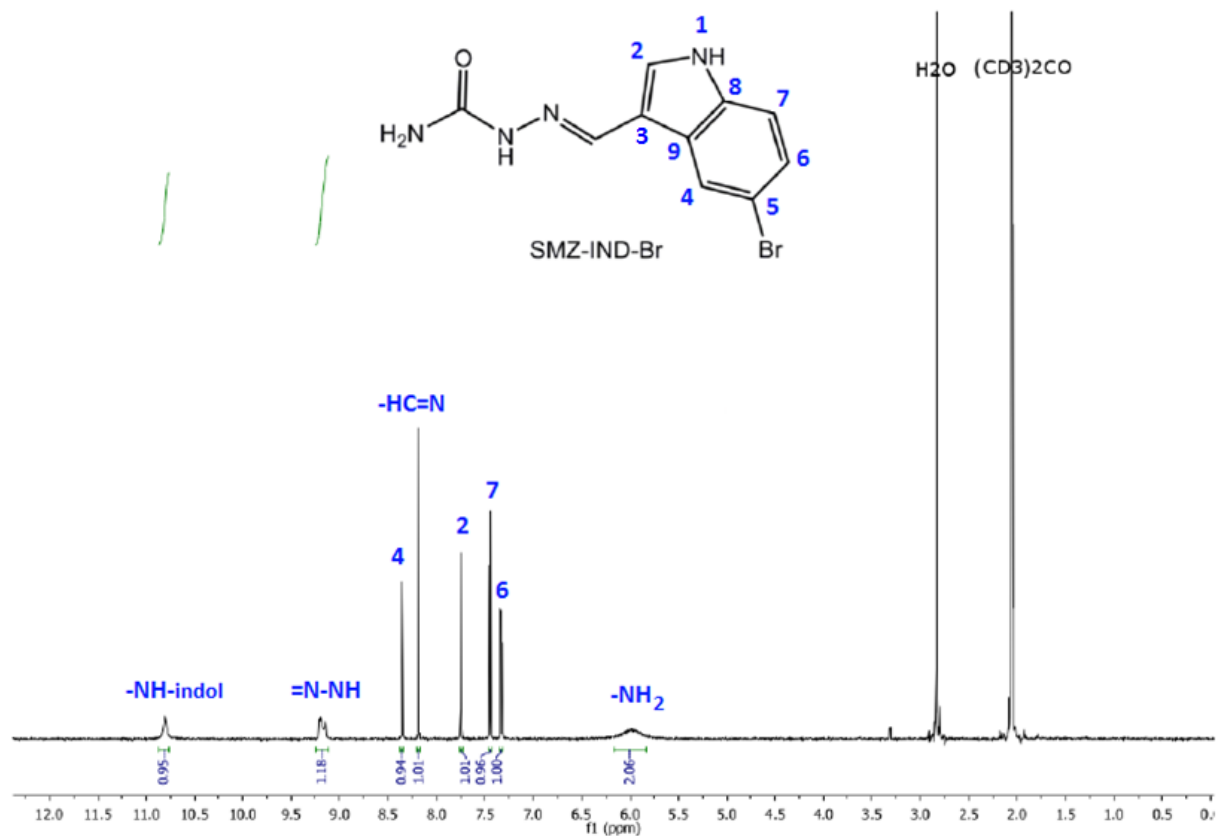


Figure 2.  $^1\text{H}$  NMR spectrum of compound 1 (500 MHz, acetone- $d_6$ )

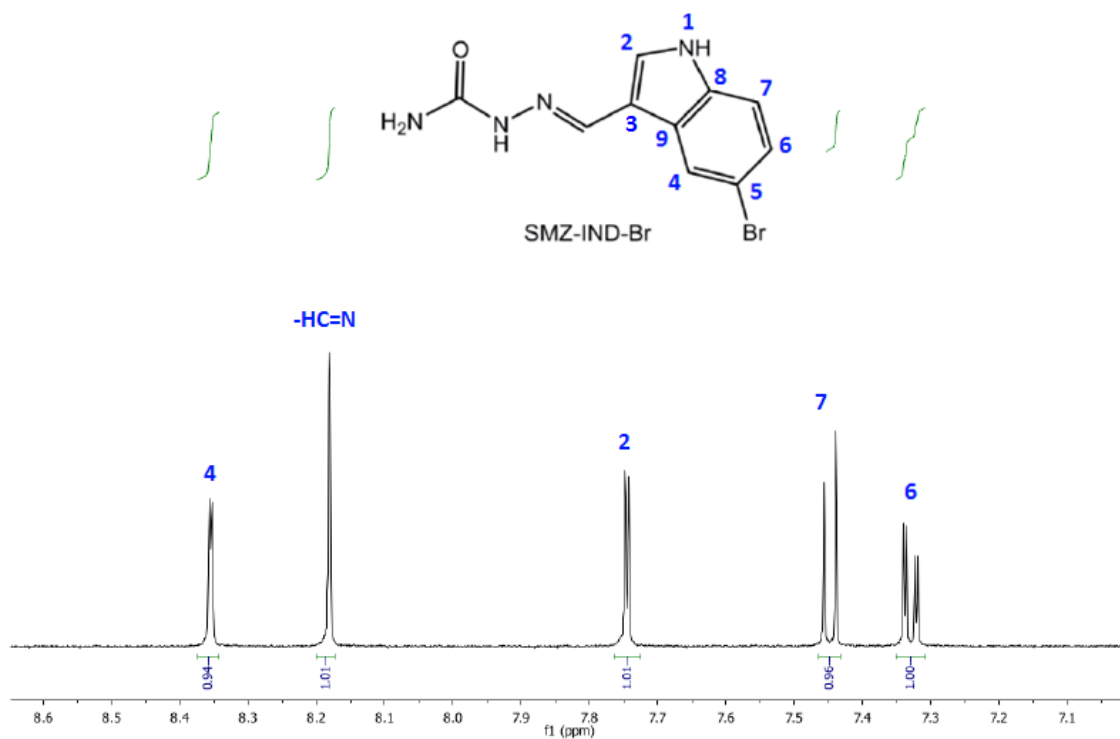


Figure 3.  $^1\text{H}$  NMR spectrum showing the aromatic zone of compound 1 (500 MHz, acetone- $d_6$ )

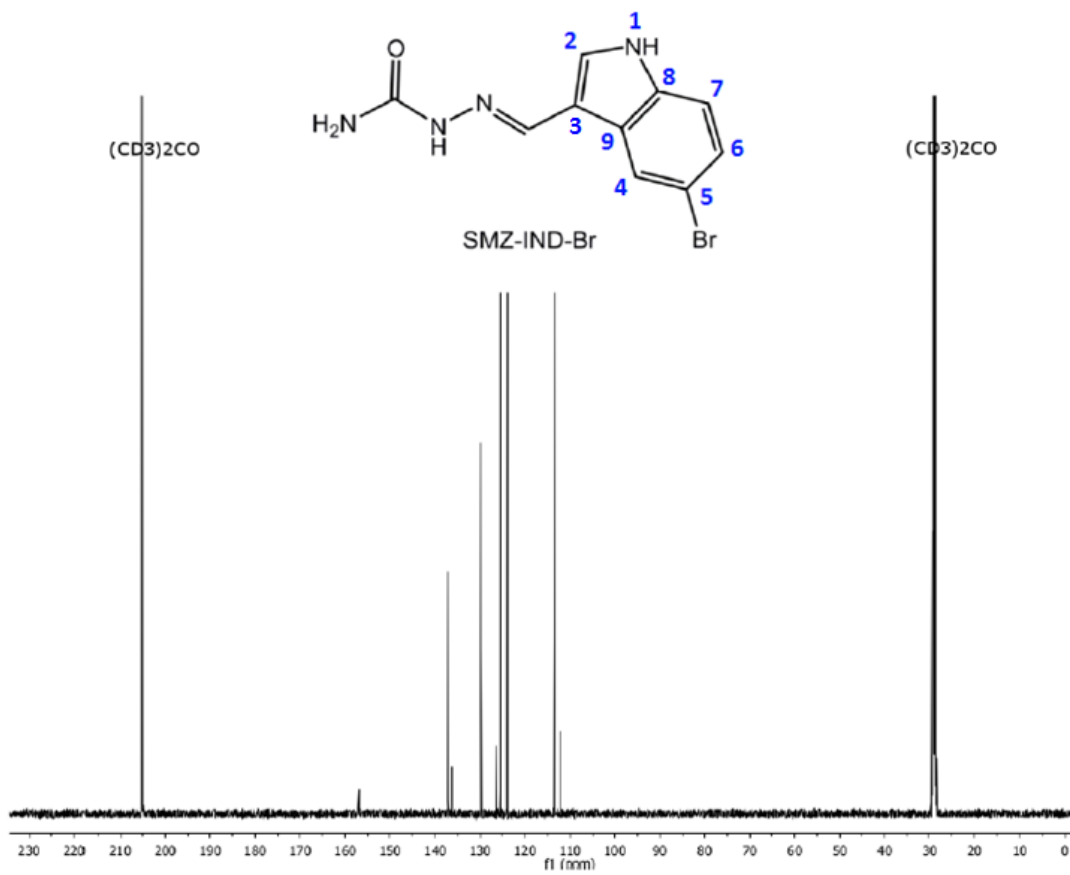


Figure 4.  $^{13}\text{C}\{^1\text{H}\}$  NMR spectrum of compound 1 (126 MHz, acetone- $d_6$ )

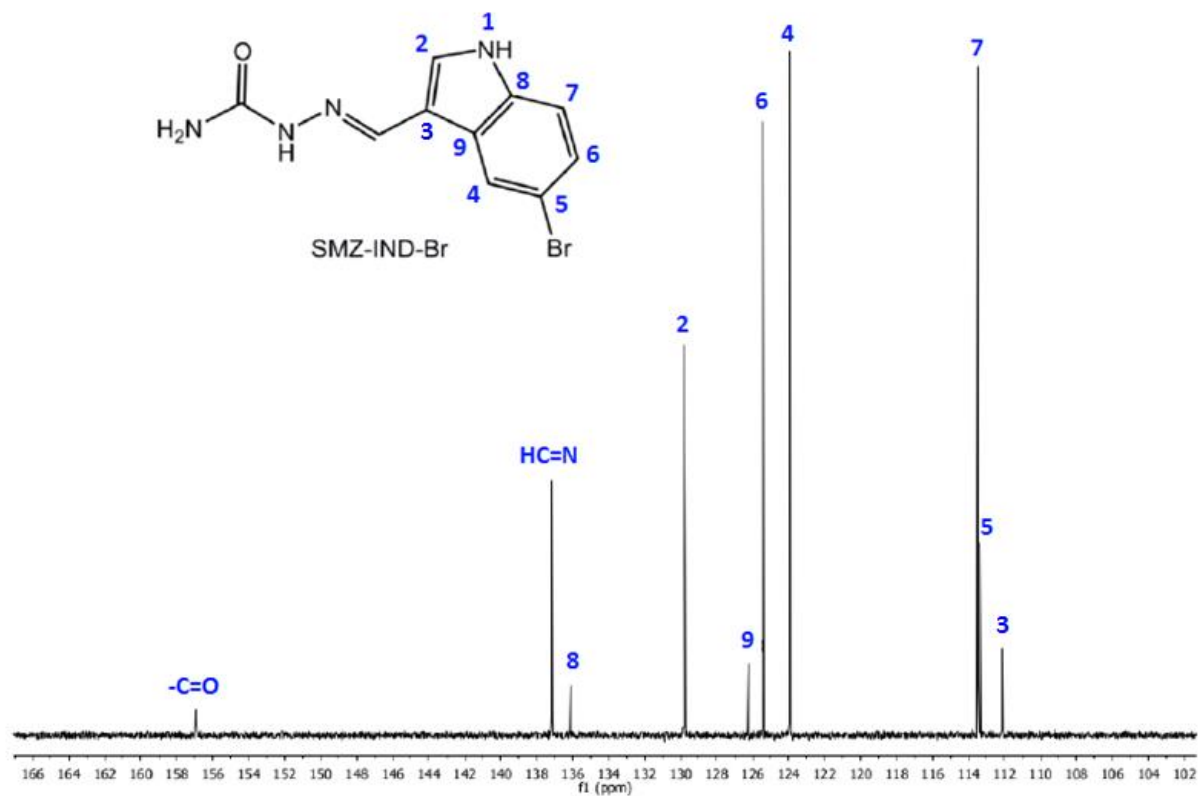


Figure 5.  $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum showing the aromatic zone of compound **1** (126 MHz, acetone- $d_6$ )

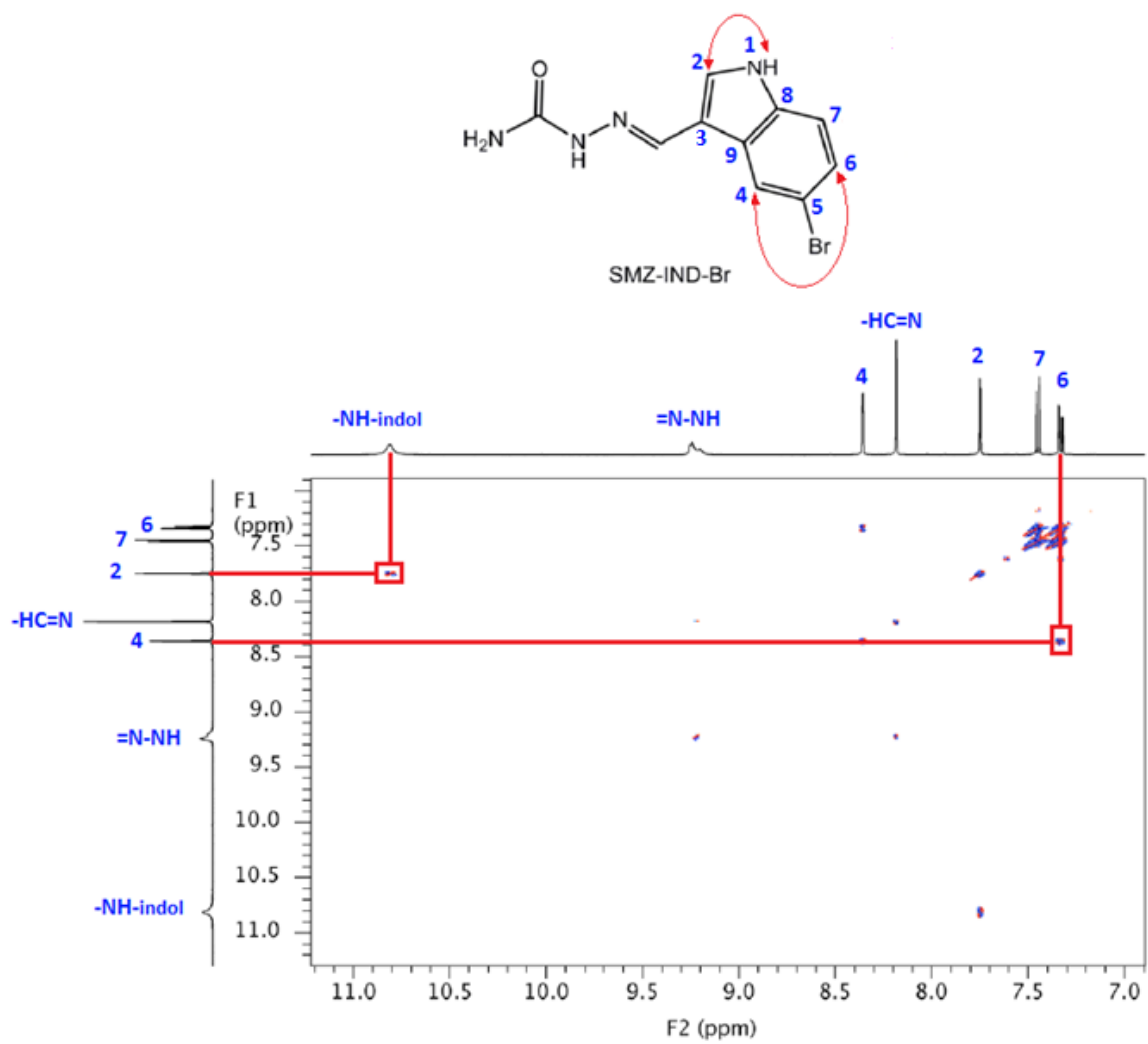


Figure 6. Two-dimensional  $^1\text{H}$ - $^1\text{H}$  DQFCOSY NMR spectrum recorded in acetone- $d_6$  for the compound **1**

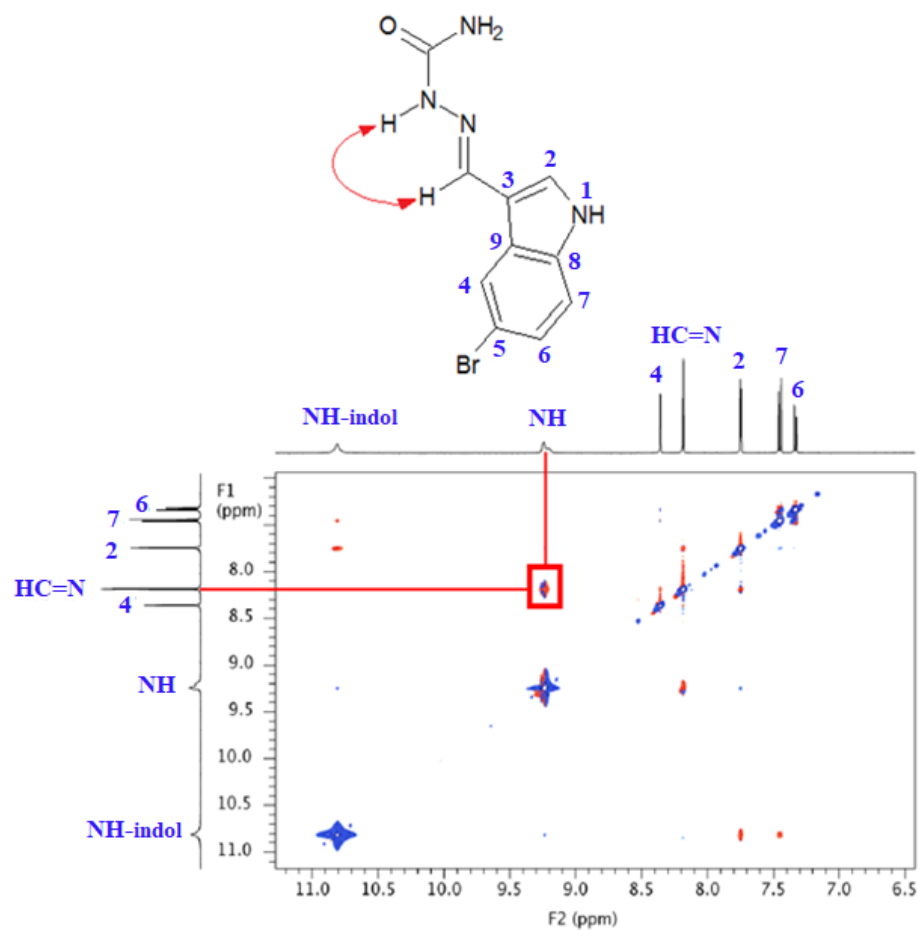


Figure 7. Two-dimensional  $^1\text{H}$ - $^1\text{H}$  NOESY NMR spectrum, recorded in acetone- $d_6$  for the compound **1**



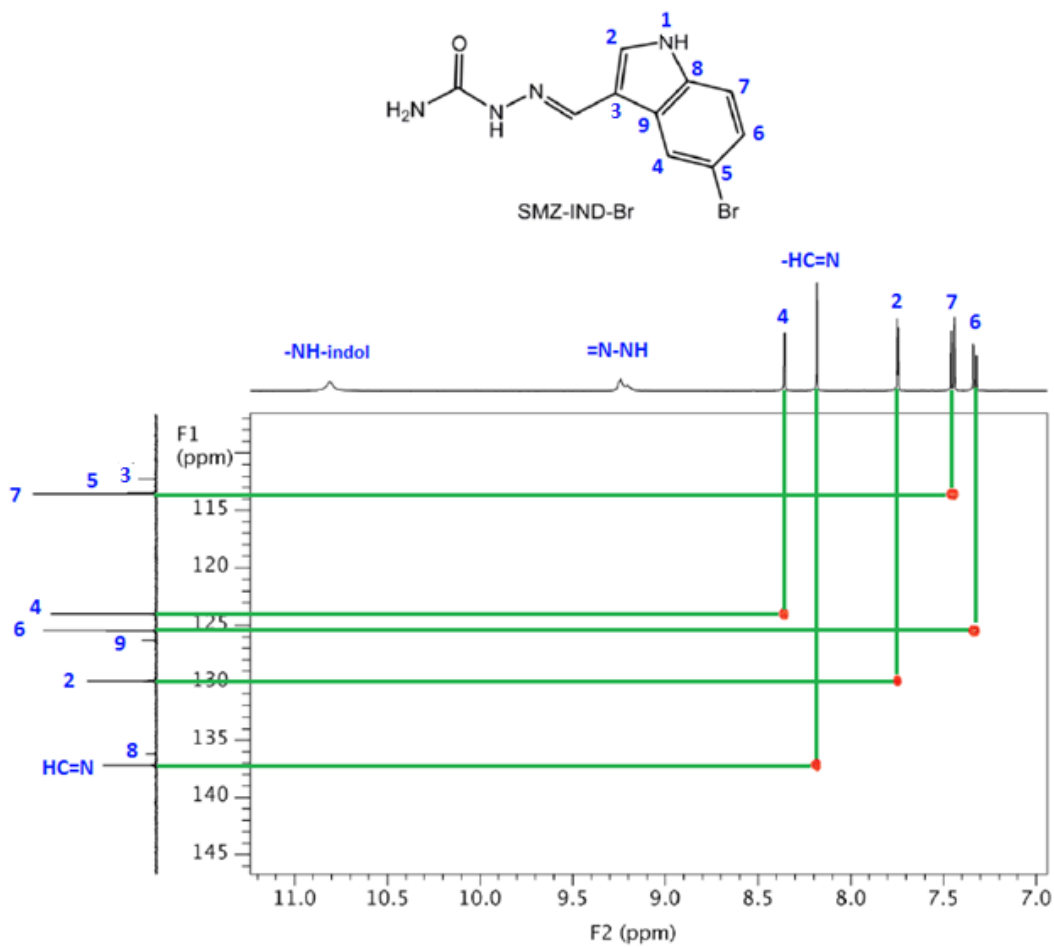


Figure 8. Two-dimensional  $^1\text{H}$ - $^{13}\text{C}$  HSQC NMR spectrum showing the aromatic zone for the compound **1** recorded in acetone- $d_6$

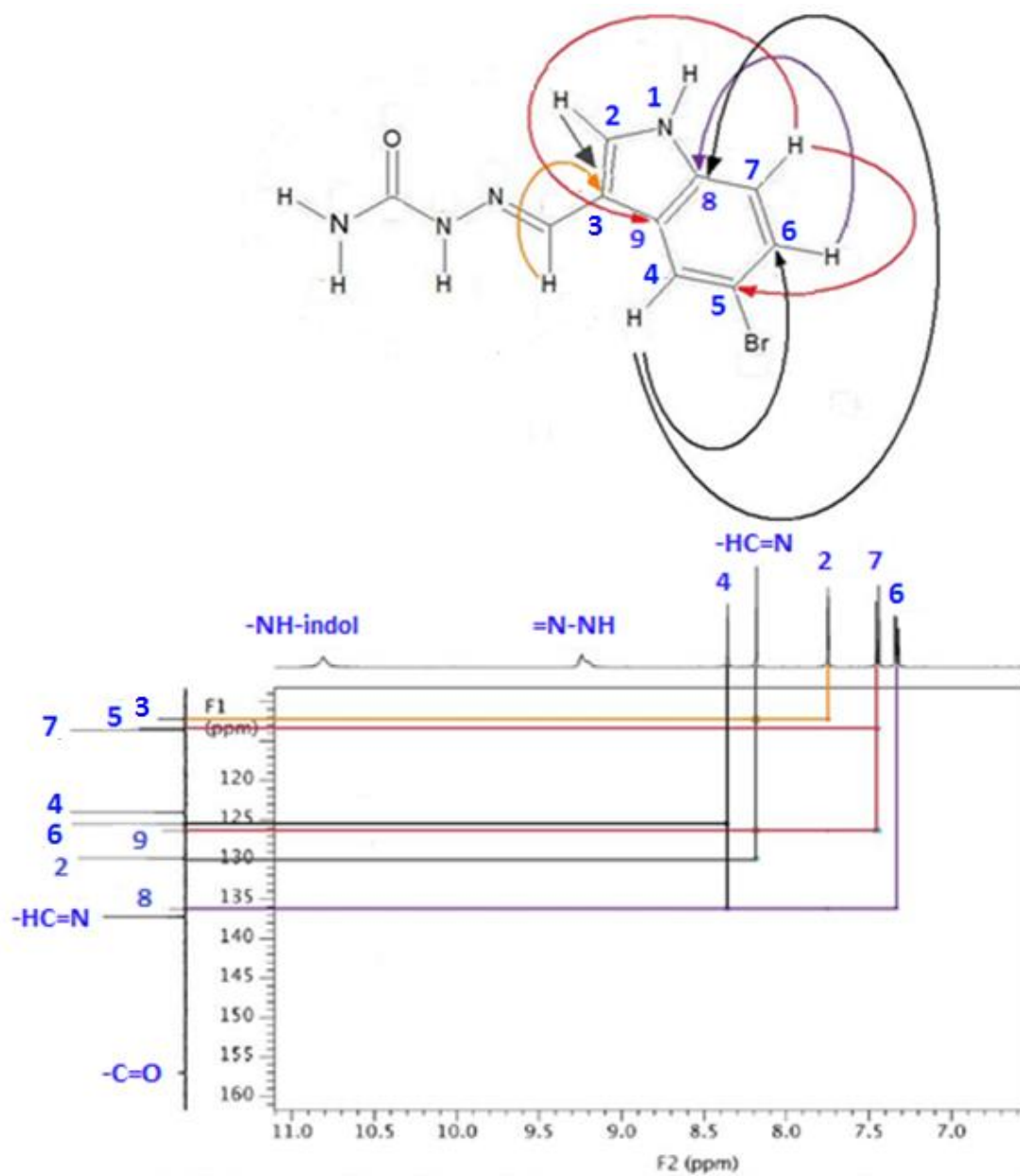


Figure 9. Two-dimensional  $^1\text{H}$ - $^{13}\text{C}$  HMBC NMR spectrum showing the aromatic zone for the compound **1** recorded in acetone- $d_6$

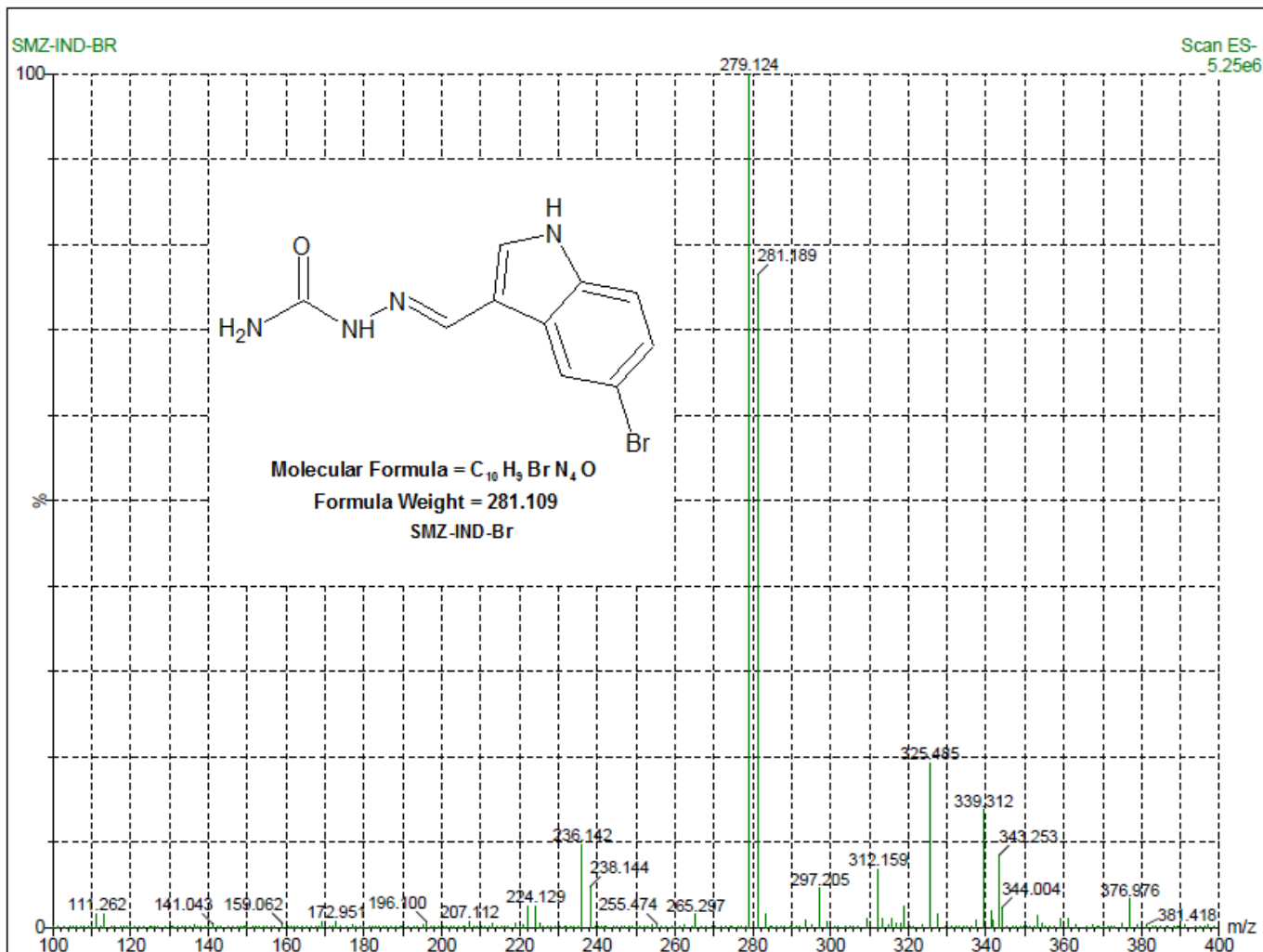


Figure 10. ESI-mass spectrum of the compound **1**

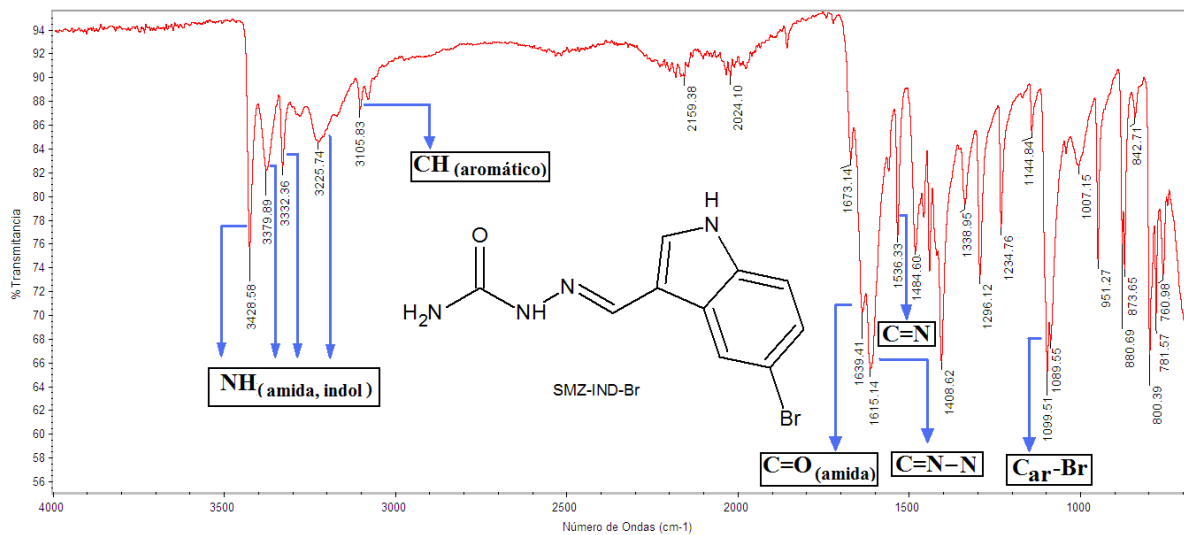


Figure 11. FT-IR spectrum of the compound 1

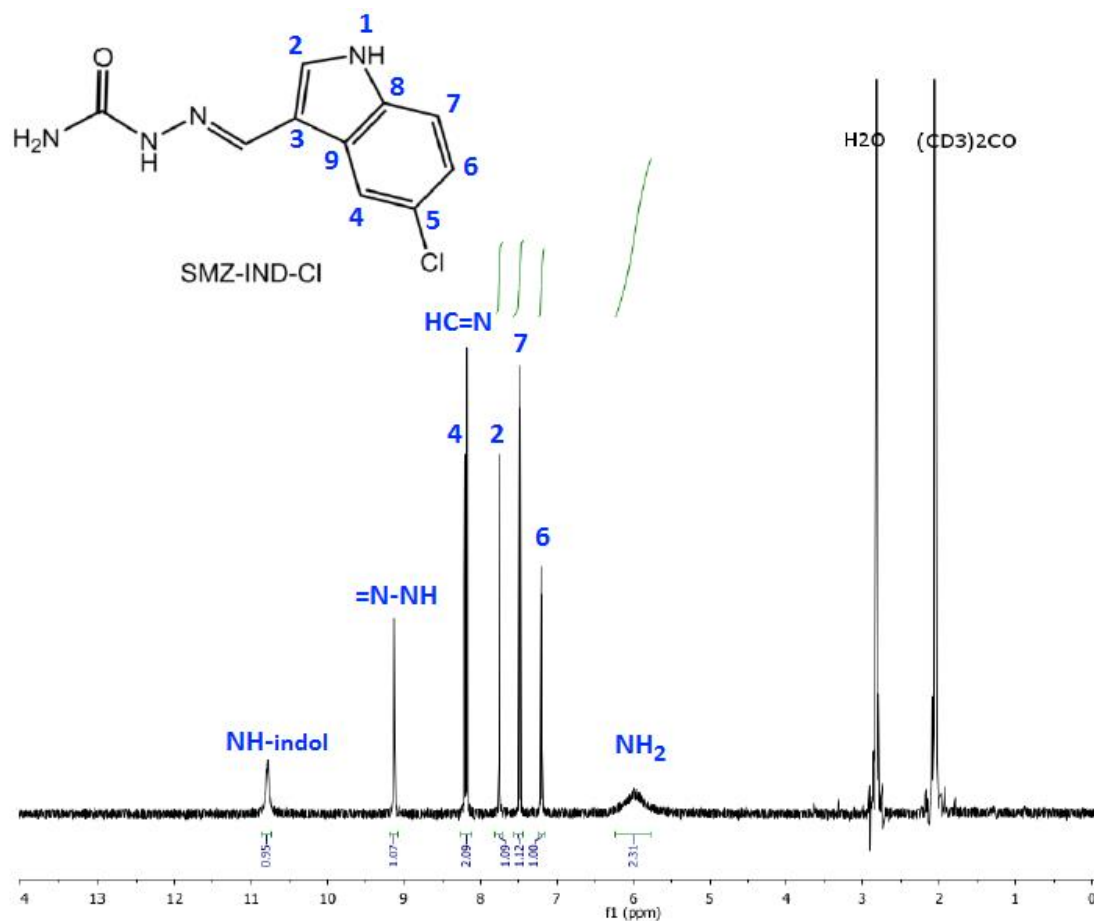


Figure 12. <sup>1</sup>H NMR spectrum of compound 2 (500 MHz, acetone-*d*<sub>6</sub>)

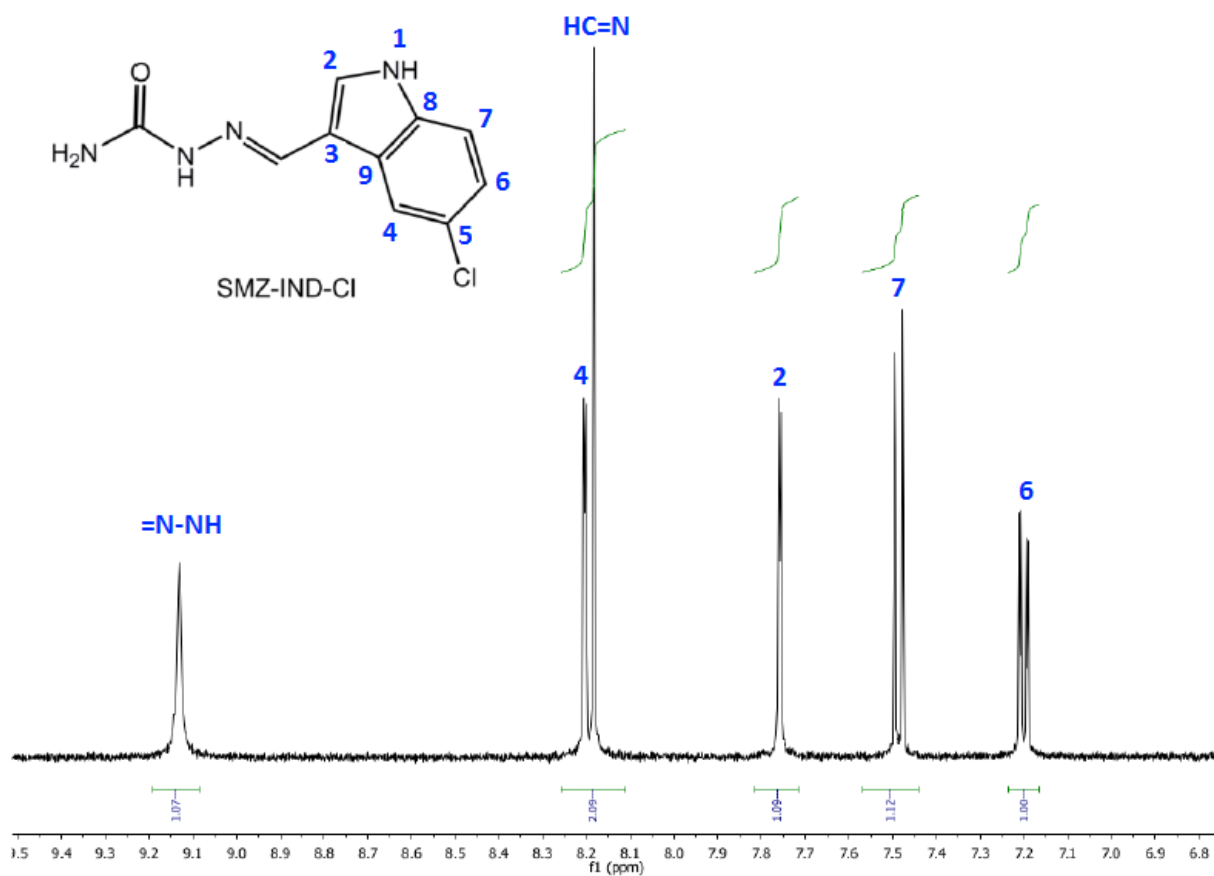


Figure 13. <sup>1</sup>H NMR spectrum showing the aromatic zone of compound 2 (500 MHz, acetone-*d*<sub>6</sub>)

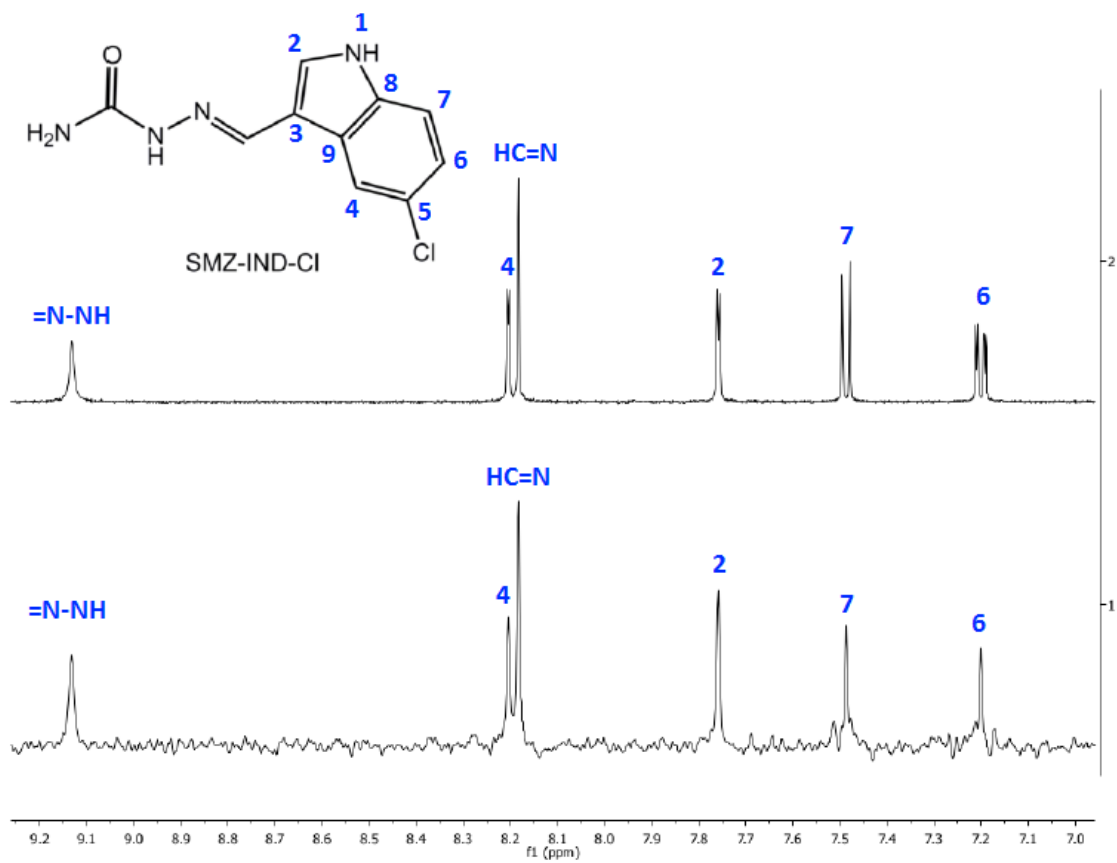


Figure 14. comparison between the experiment "pure shift" of  $^1\text{H}$  (lower) with the NMR spectrum of  $^1\text{H}$  (upper) (500 MHz, acetone- $d_6$ )

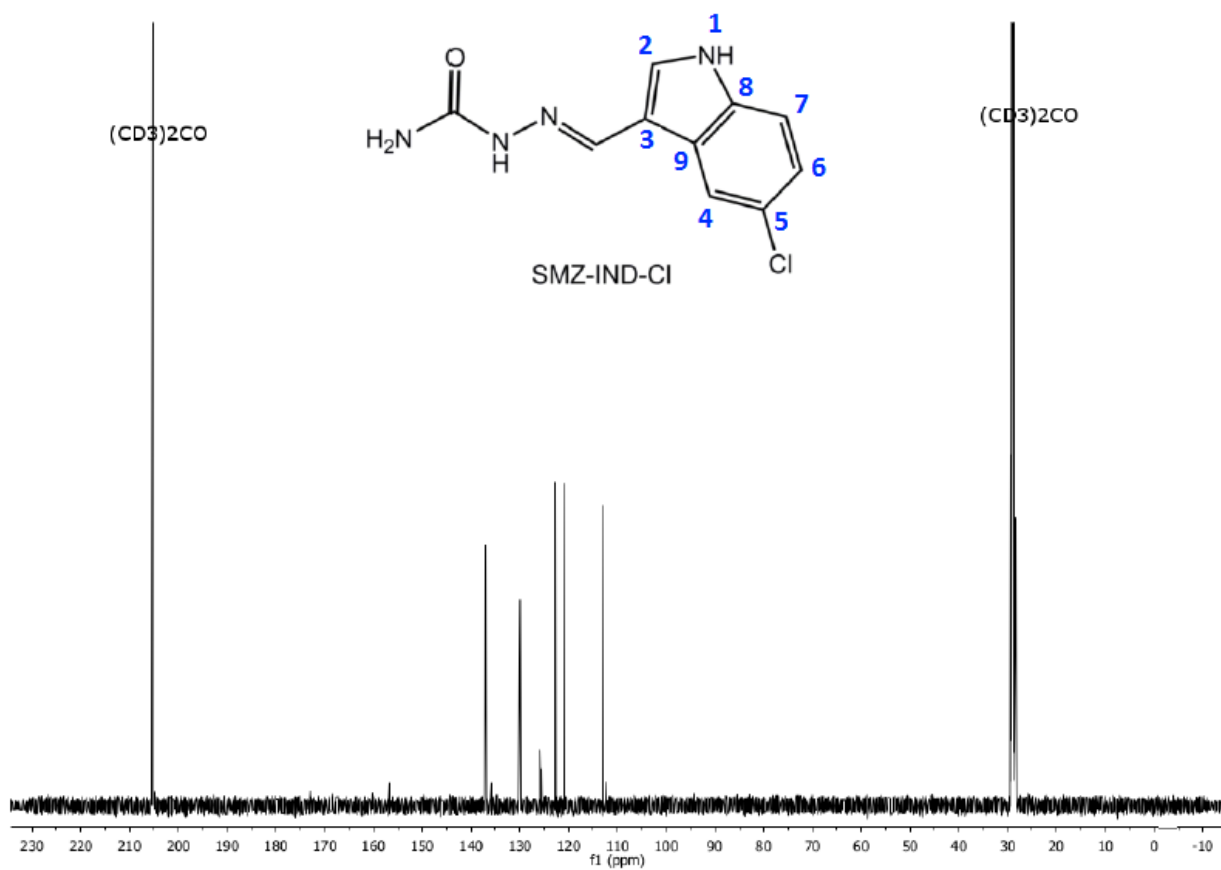


Figure 15.  $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of compound 2 (126 MHz, acetone- $d_6$ )



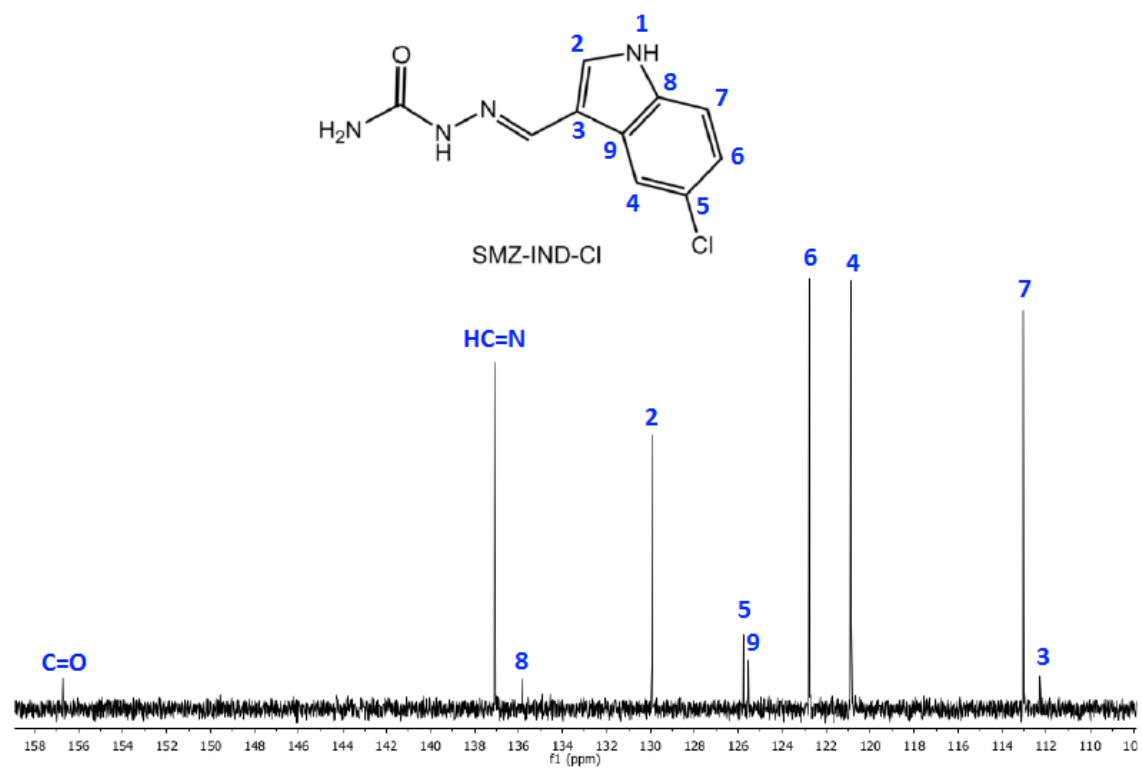


Figure 16.  $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum showing the aromatic zone of compound **2** (126 MHz, acetone- $d_6$ )

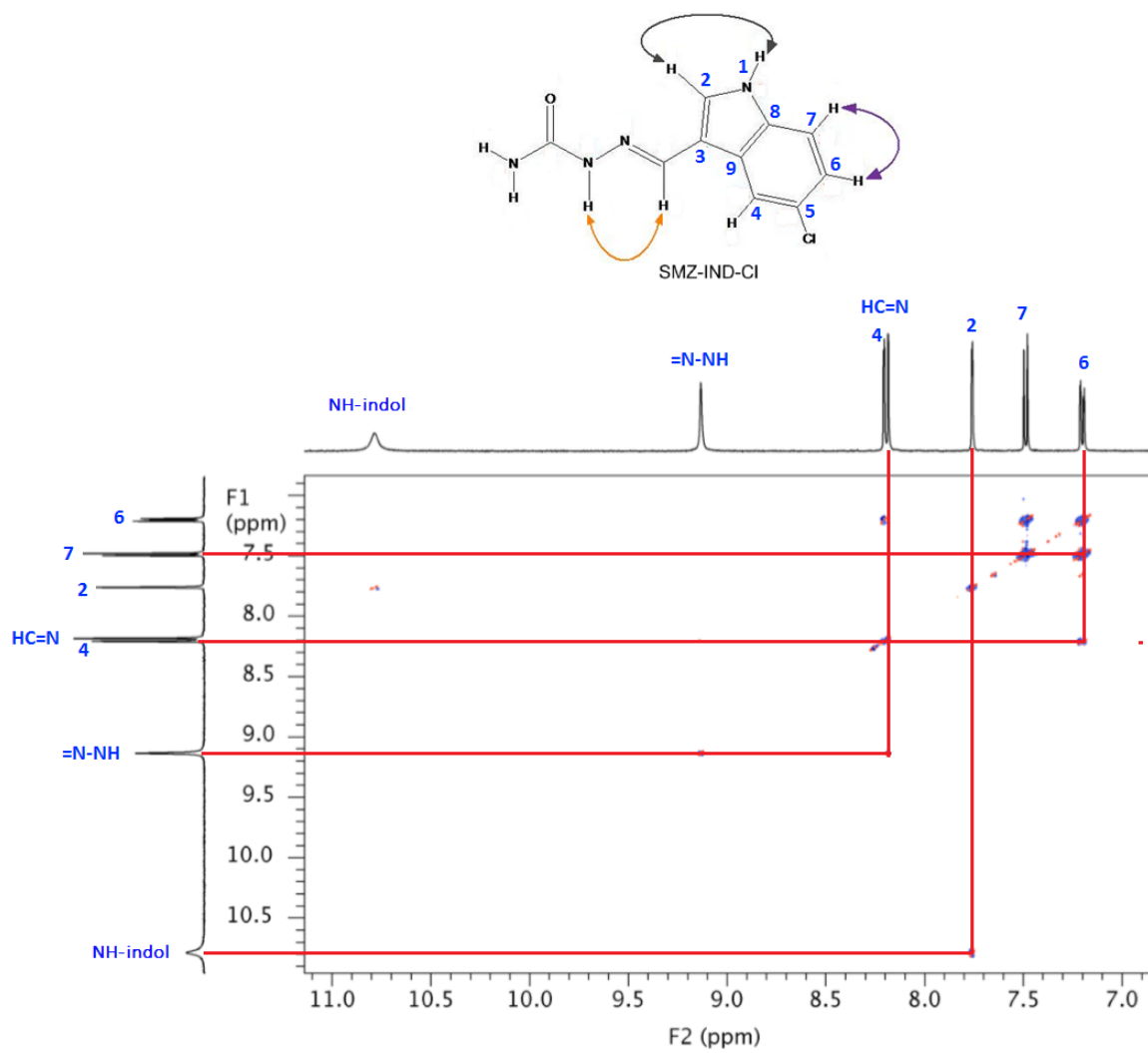
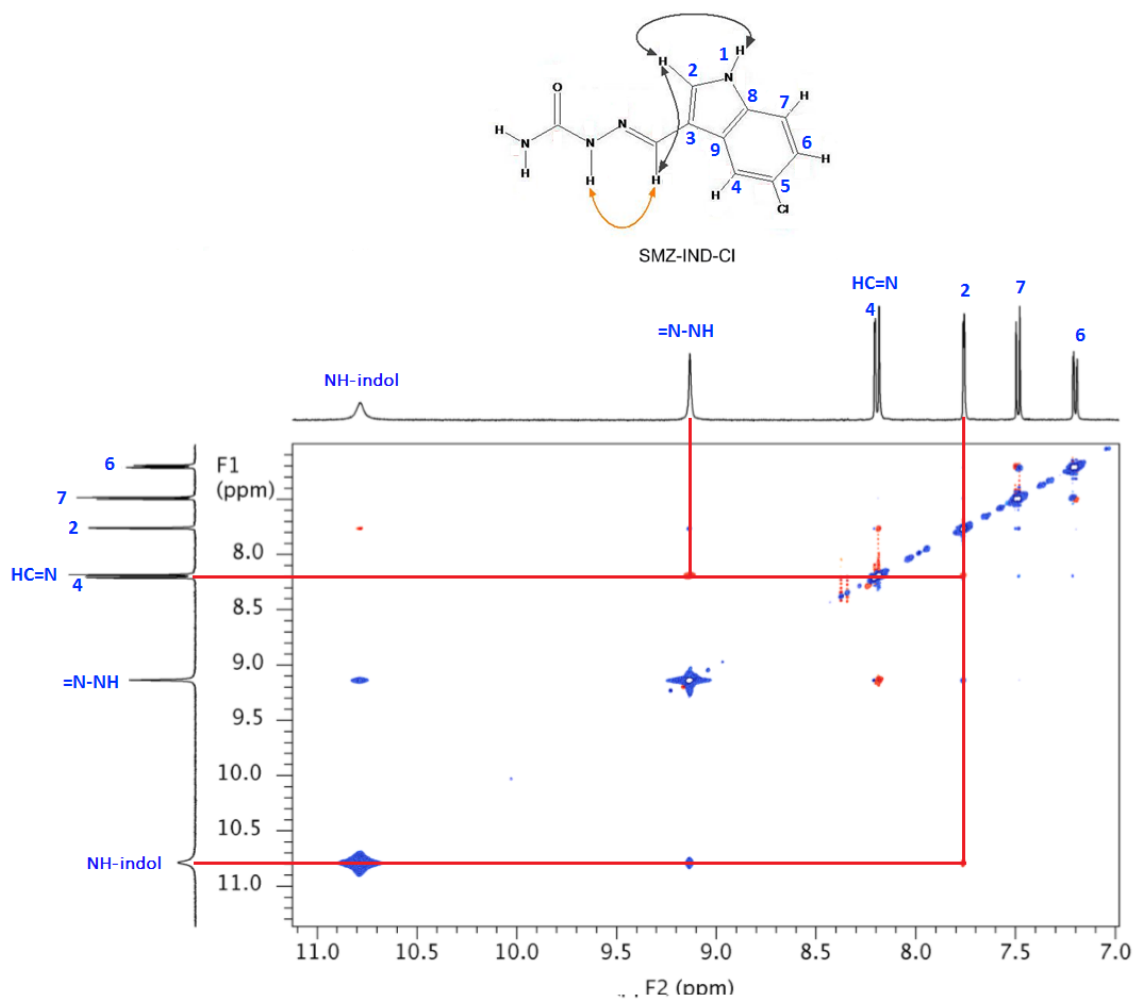


Figure 17. Two-dimensional  $^1\text{H}$ - $^1\text{H}$  DQFCOSY NMR spectrum recorded in acetone- $d_6$  for the compound **2**



Expansión de la zona aromática del experimento de RMN bidimensional  $^1\text{H}$ - $^1\text{H}$  NOESY

Figure 18. Two-dimensional  $^1\text{H}$ - $^1\text{H}$  NOESY NMR spectrum, recorded in acetone- $d_6$  for the compound **2**

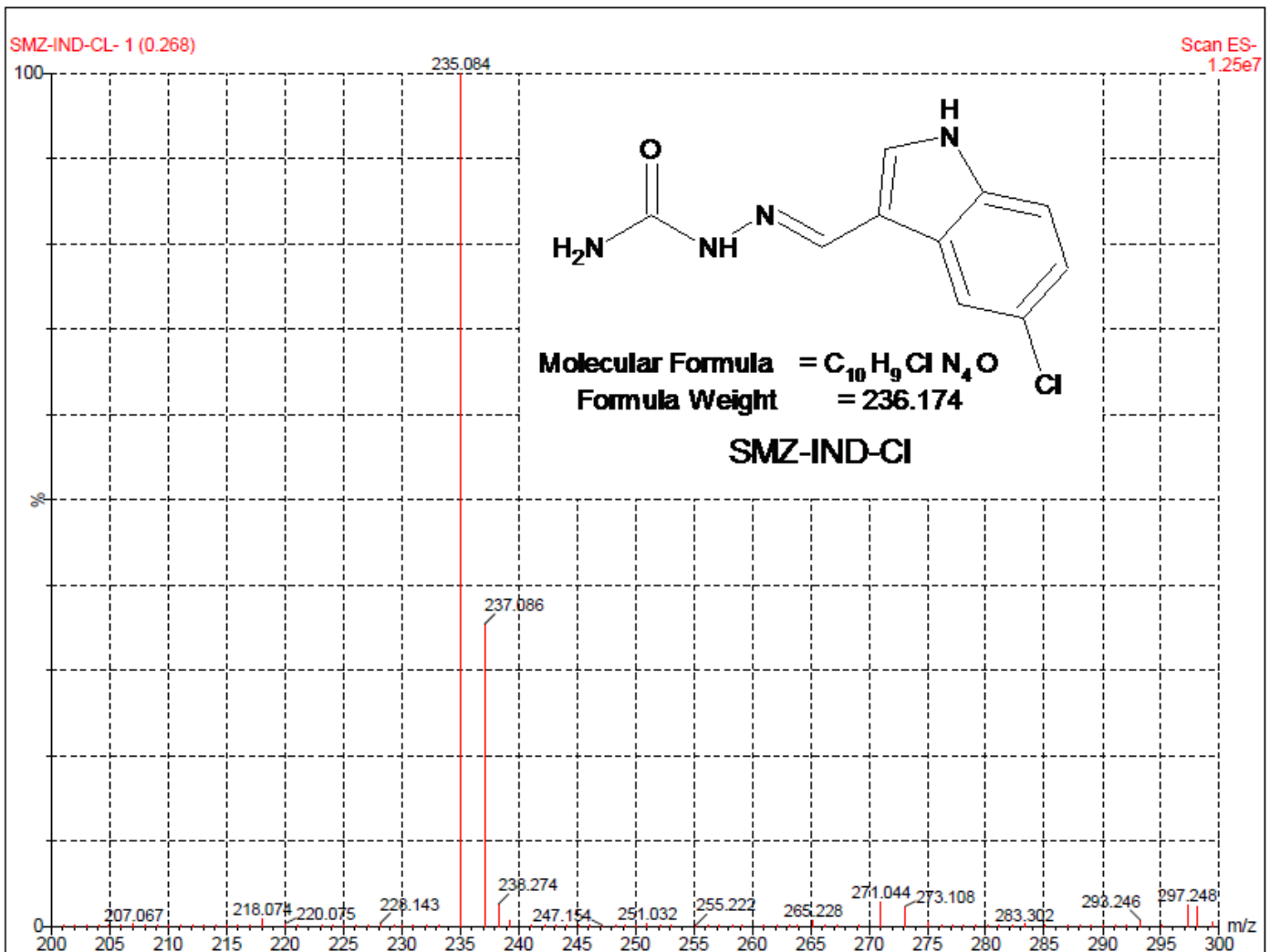


Figure 19. ESI-mass spectrum of the compound 2

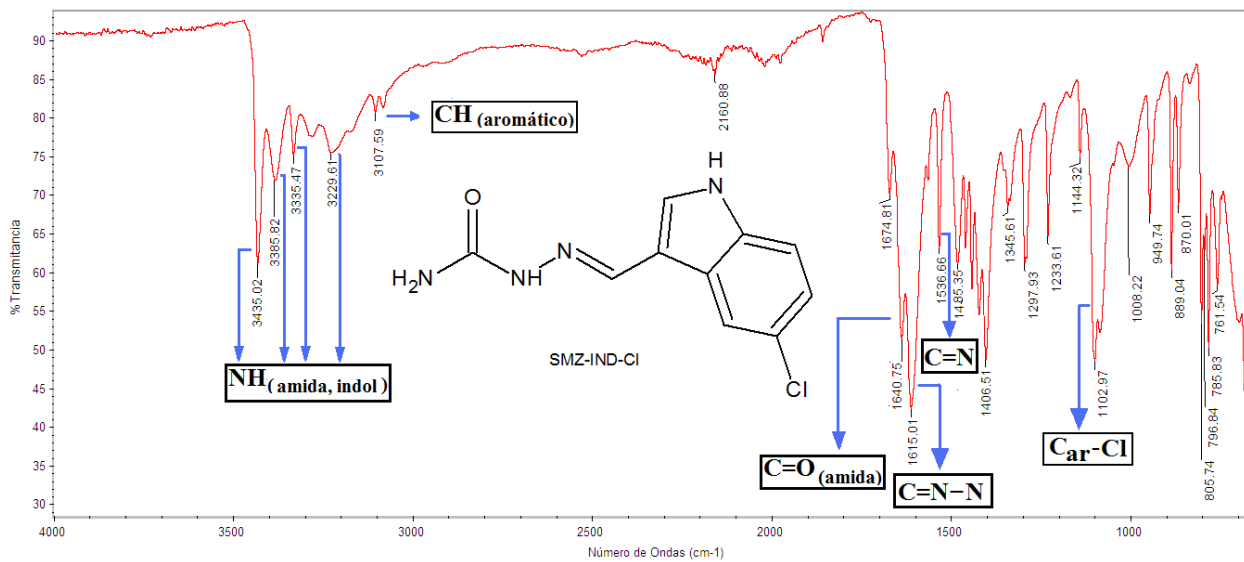


Figure 20. FT-IR spectrum of the compound **2**

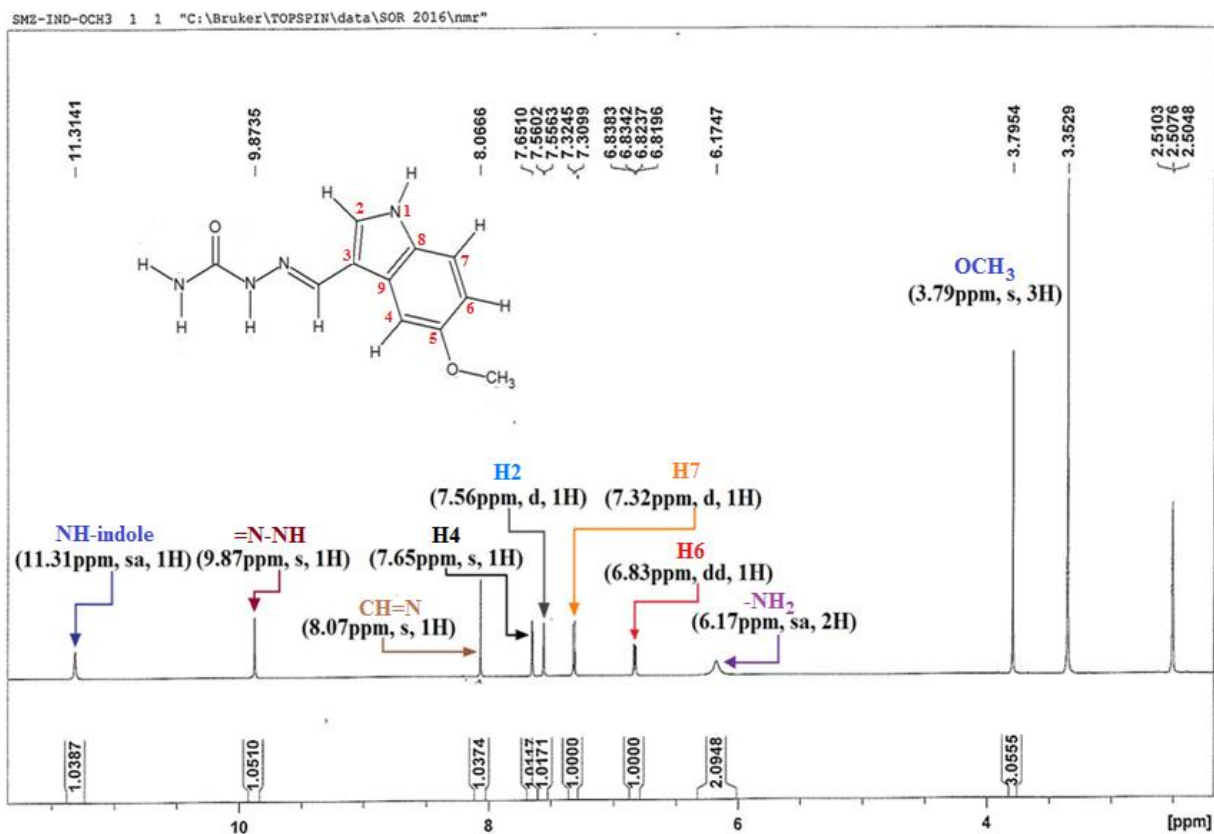


Figure 21. <sup>1</sup>H NMR spectrum of compound 3 (600 MHz, DMSO-*d*<sub>6</sub>)

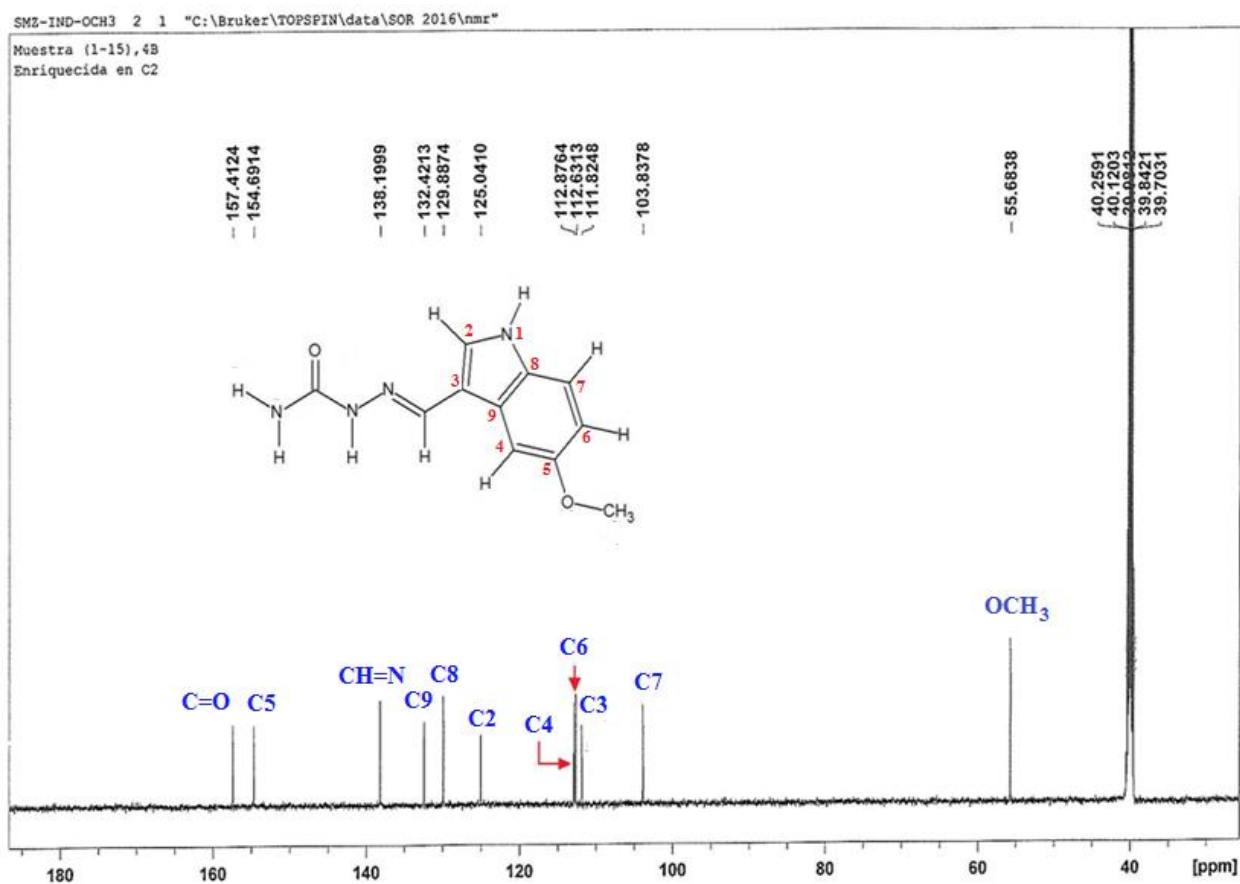


Figure 22.  $^{13}\text{C}$  NMR spectrum of compound 3 (150 MHz, DMSO- $d_6$ )

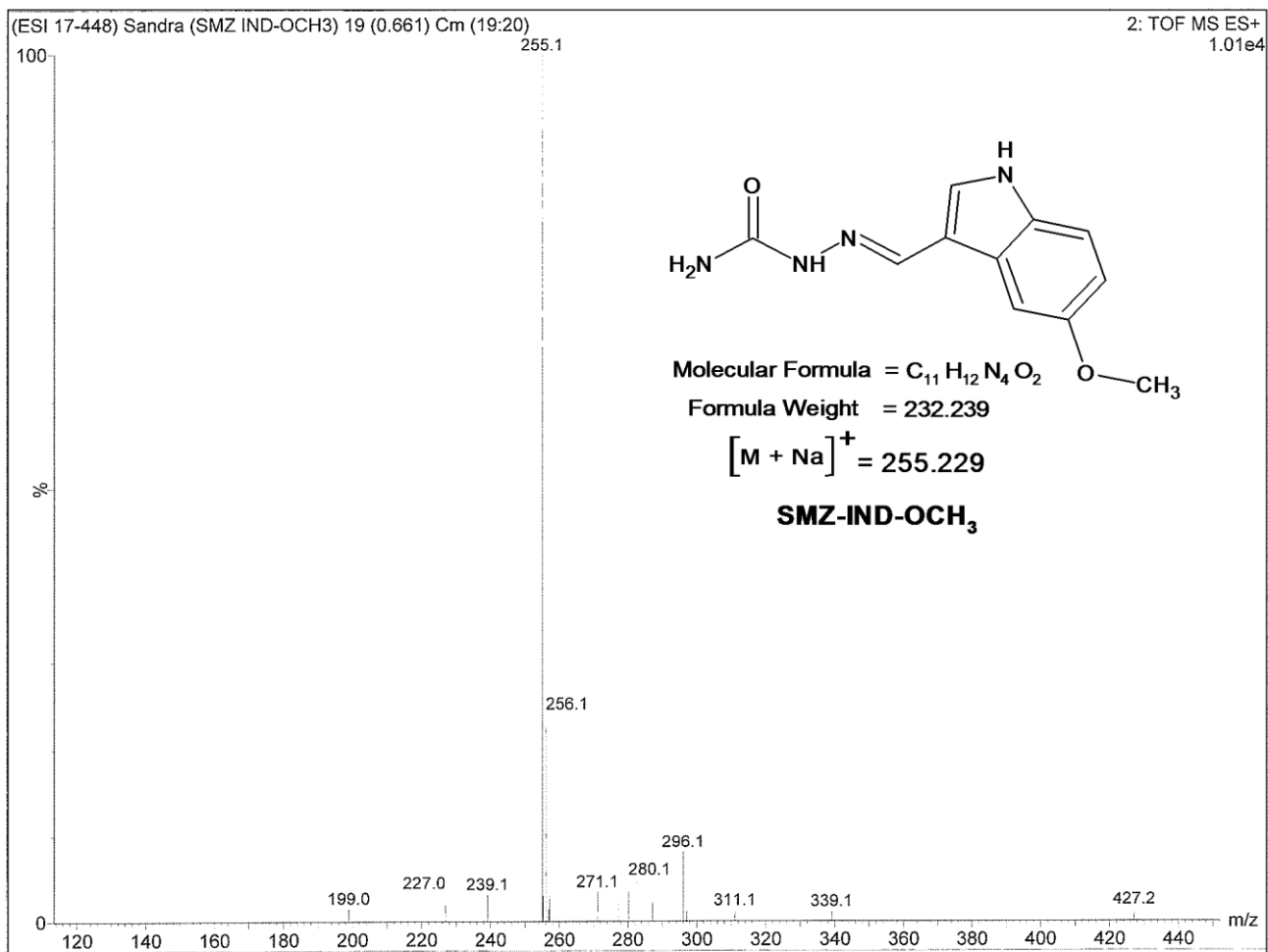


Figure 23. ESI-mass spectrum of the compound **3**



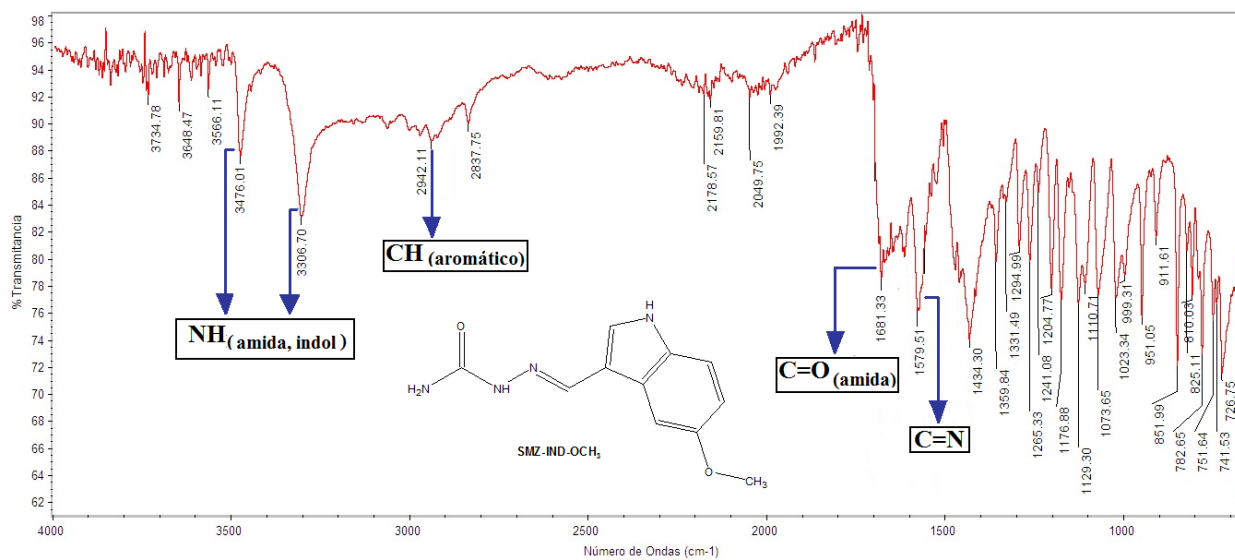


Figure 24. FT-IR spectrum of the compound 3

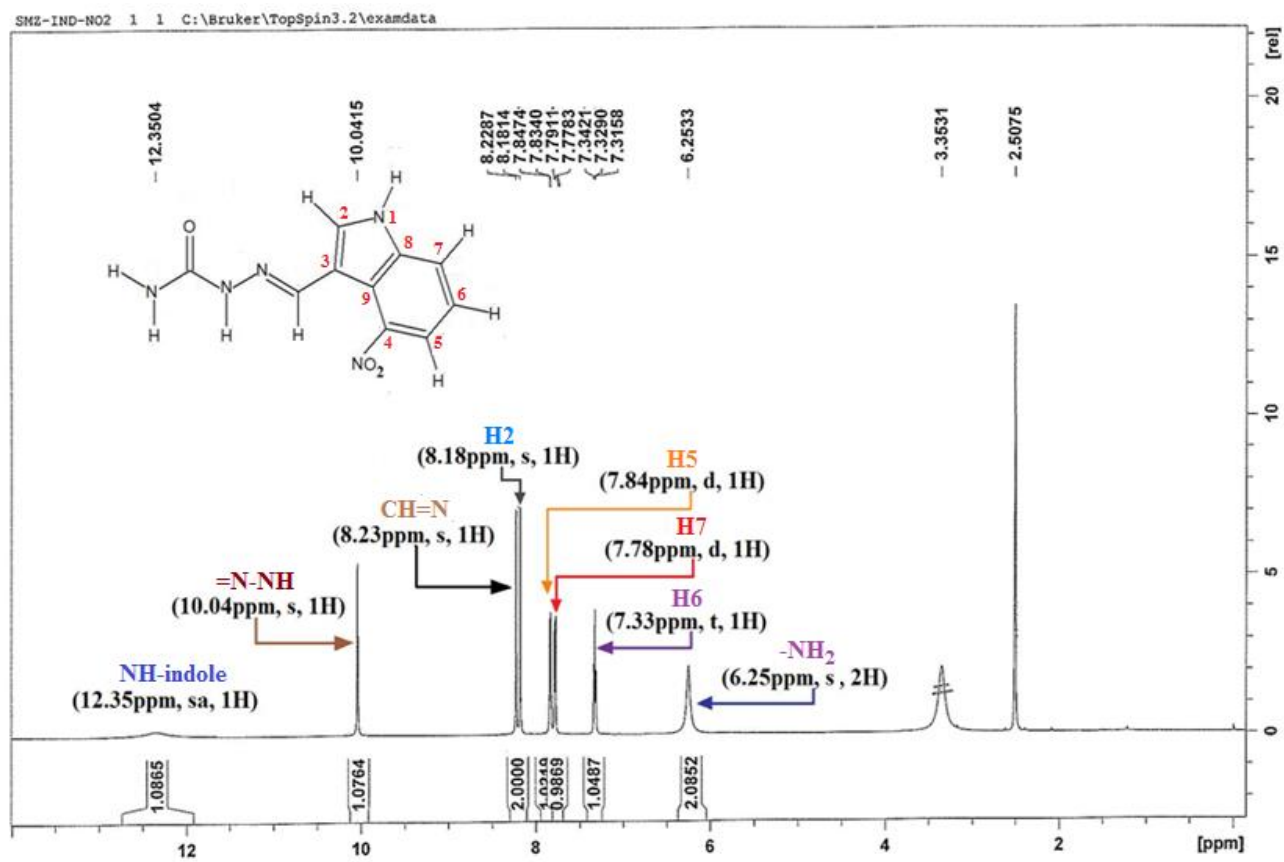


Figure 25. <sup>1</sup>H NMR spectrum of compound 4 (600 MHz, DMSO-*d*<sub>6</sub>)

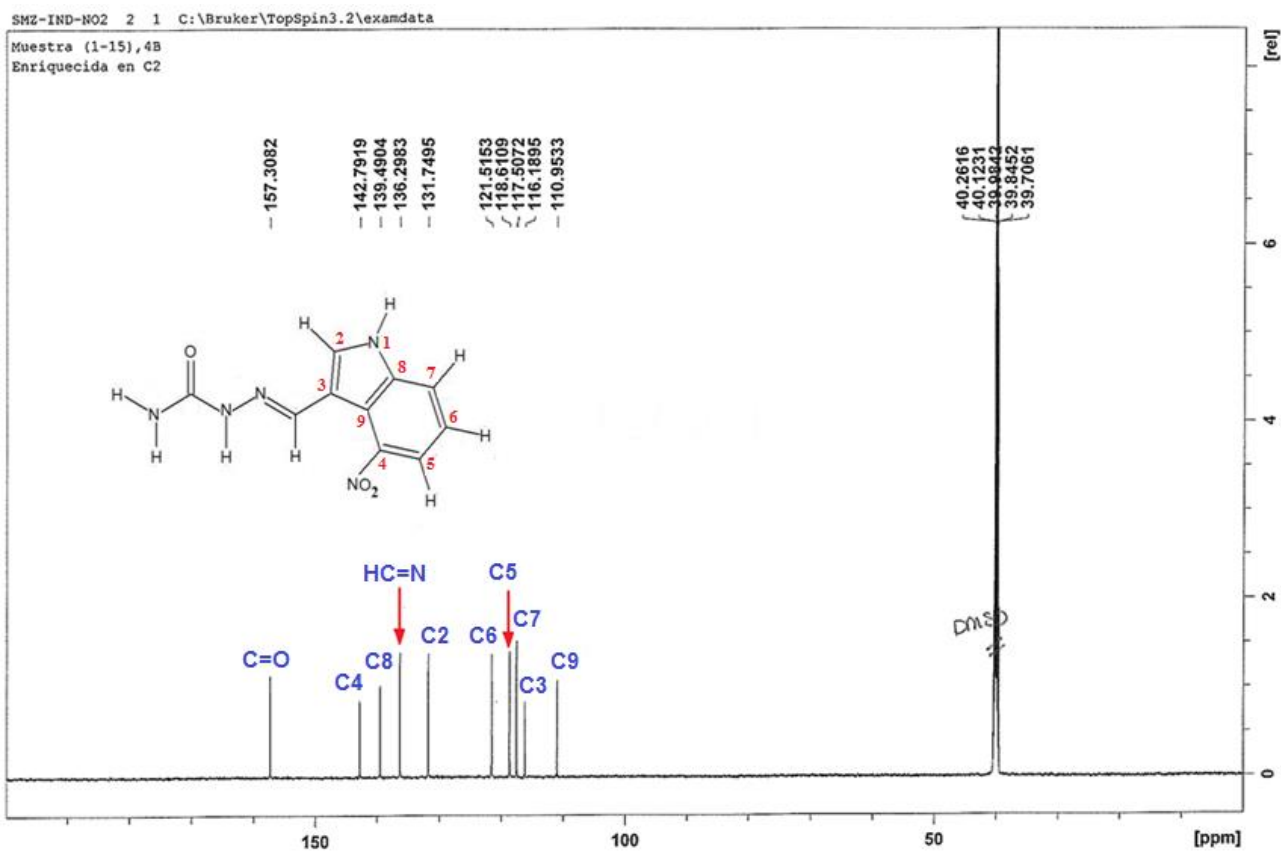


Figure 26. <sup>13</sup>C NMR spectrum of compound 4 (150 MHz, DMSO-*d*<sub>6</sub>)

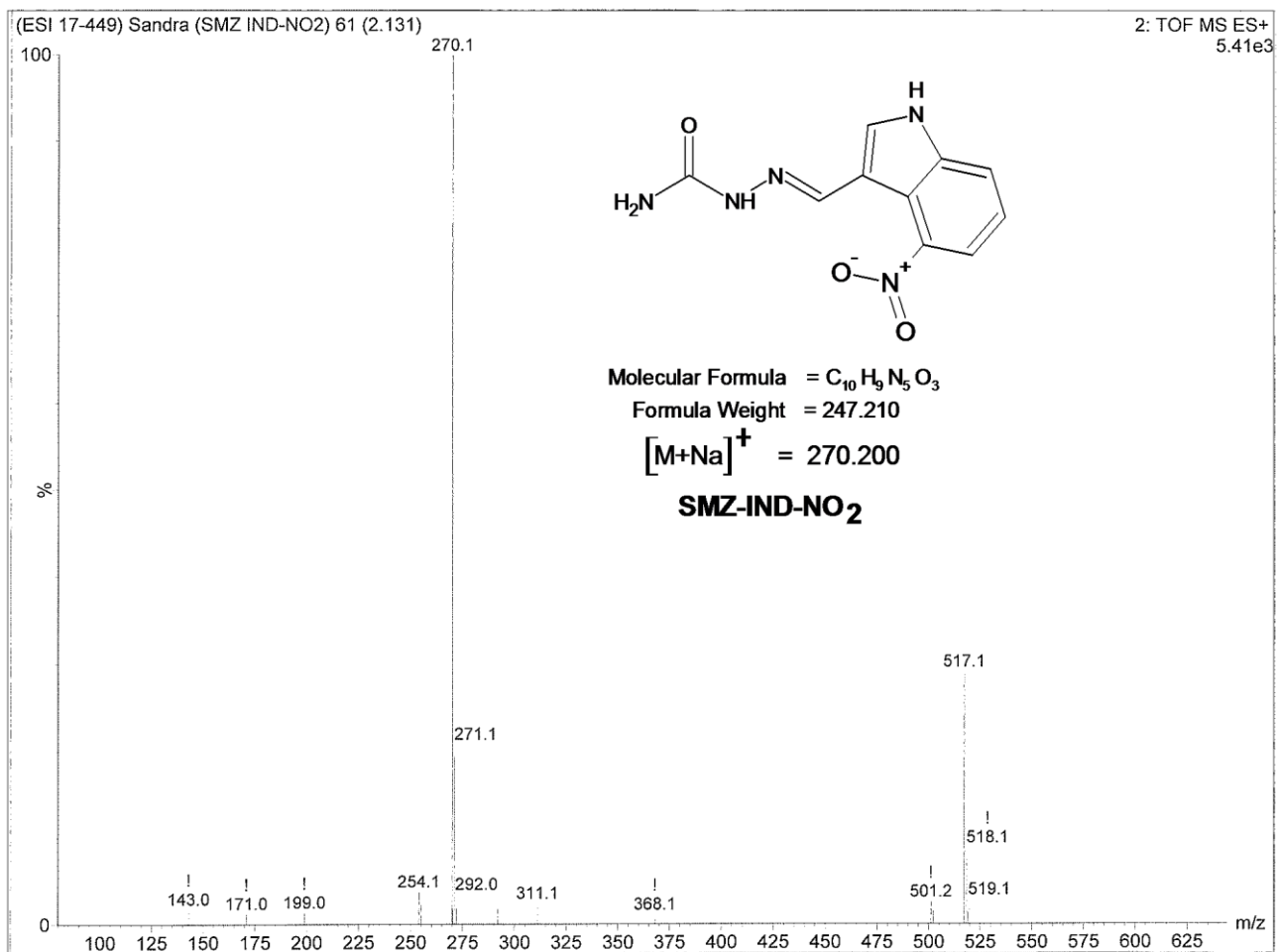


Figure 27. ESI-mass spectrum of the compound **4**

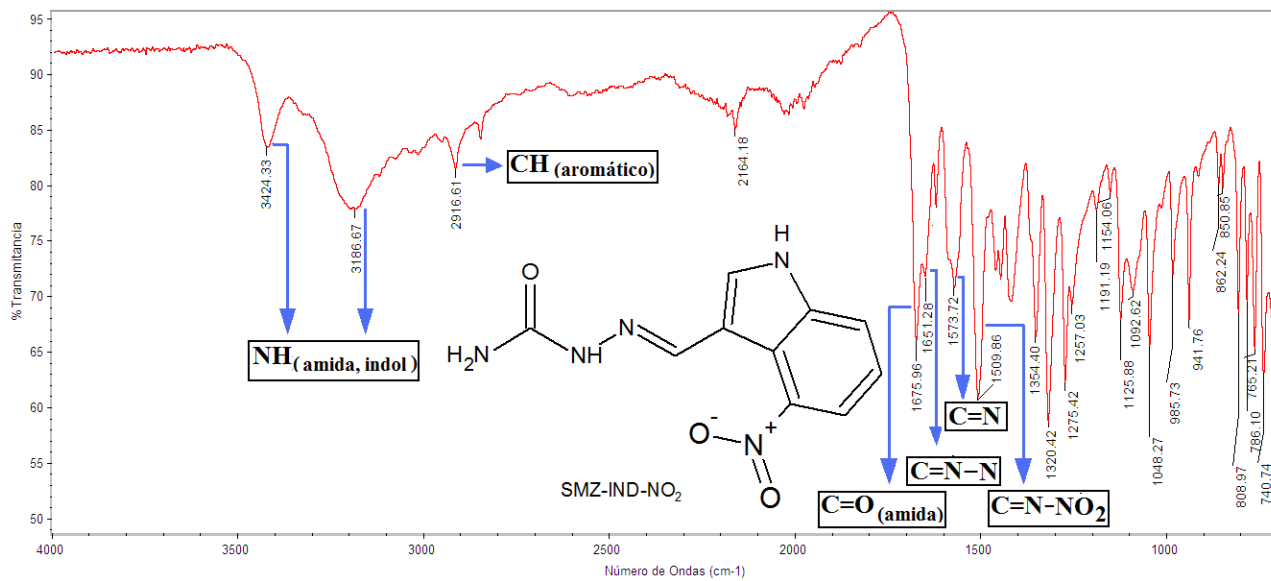


Figure 28. FT-IR spectrum of the compound **4**