Research Article
Synergistic Effect of Nanosilica Aerogel with Phosphorus Flame Retardants on Improving Flame Retardancy and Leaching Resistance of Wood

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Nanosilica (Nano-SiO$_2$) sol fabricated by a sol-gel process was introduced into wood modification with phosphorus flame retardants to improve the flame retardancy and leaching resistance of wood. The obtained materials were characterized by scanning electron microscopy and energy dispersive spectrometer (SEM-EDS), thermogravimetric analysis (TGA), cone calorimetric (CONE), and infrared spectroscopy (FT-IR). The residual rate of flame retardants before and after leaching was determined by a leaching resistance. The results showed that the phosphorus flame retardants and SiO$_2$ sol could reside in the poplar wood and are widely distributed in the vessels, pits, wood timber, and the spaces between wood cells of poplar substrate. TGA and CONE results indicated that the introduction of nano-SiO$_2$ aerogel with phosphorus flame retardants had a significantly synergistic effect on improving the flame retardancy and inhibiting the release of smoke and toxic gases. In addition, the leaching resistance test, combined with infrared analysis and EDS analysis, confirmed that the phosphorus flame retardants were able to be fixed by SiO$_2$ aerogel in the wood.

1. Introduction

Wood is the only truly sustainable resource. Its unique advantages such as good visual characteristics, tactile properties, nontoxicity have made it one of the most important materials in the furniture, building, and construction industries. However, wood is combustible due to its organic constitution, which may not only lead to the spread of fire but also produce a large number of smoke and toxic gases like CO, the leading cause of casualties. Therefore, in order to gain safe and environmentally friendly wood materials, the wood products with fire retardation and smoke suppression are needed urgently.

Inorganic chemicals are attracting more and more attention among common fire retardants due to their high flame retardant efficiency, low cost, and environmental friendliness [1–4]. Specifically, phosphorus flame retardants like ammonium polyphosphate (APP) and ammonium monohydrick phosphate (DAP) are effective fire retardants and have been widely used for fire retardation treatment of wood and other cellulosic combustible materials [5–8]. However, the inorganic salts of the flame retardants are water-soluble, resulting in a gradual loss during using terms, which directly affect the durability of fire-retardant properties, thus limiting its use in contact with water or humidity. What is more, with the use of phosphorus flame retardants, more gases like CO are produced which result in the suffocation of people [9]. Therefore, nontoxic, leach resistance smoke suppression will be a trend in the development of flame retardants for wood.

The inorganic nano-SiO$_2$ aerogel plays a certain role in the adsorption and blocking action, which shows a porous structure when it is resided in interior materials. Meanwhile, the Si element existed in SiO$_2$ aerogel, combined with P or N presented in phosphorus flame retardants, exert a certain synergistic flame retardant and smoke suppression effect [10, 11]. What is more, nano-SiO$_2$ aerogel has large specific surface area and strong adsorbability, which may help to prevent the phosphate flame retardant from running off [12]. Hence, in this work, nano-SiO$_2$ aerogel, obtained by sol-gel method, was used to combine with three different phosphate
flame retardants to treat wood in order to improve the flame retardancy and smoke suppression and leaching resistance of wood.

2. Materials and Methods

2.1. Materials. Poplar wood (Populus deltoides cv. I-69/55) was obtained from Yiyang Forest, Hunan. Wood specimens were cut from untreated poplar sapwood into wood blocks with dimensions of 100 mm × 10 mm × 100 mm and 20 mm × 20 mm × 20 mm (T × R × L), whose moisture content was 8–12%. Tetraethyl orthosilicate (TEOS, C(SiO₂) ≥ 98%), ethanol (ET), and hydrochloric acid (HCl, 36–38%) which were all analytically pure were purchased from XiLong Chemical Co., Ltd., while distilled water (H2O) was prepared in the laboratory. Phosphorus flame retardants including ammonium polyphosphate (APP), ammonium monohydric phosphate (DAP), and ammonium phosphate (AP) were supported by Shanghai Luzhong Chemical Product Co., Ltd. APP, DAP, and AP were dissolved in water, respectively, to prepare three kinds of solutions of 10% mass fraction.

2.2. Preparation of Nano-SiO₂ Aerogel. The SiO₂ aerogel with nanoporous network structure was made by sol-gel method. The preparation was shown in Figure 1. The TEOS/ET/H₂O with a molar ratio of 1:5:8 were poured into the reaction system. To promote the hydrolysis process, HCl was added until the pH value was 3. The mixture solution was stirred at 50°C for 60 min until the solution was clear.

2.3. Preparation of Nanosilica Aerogel with Phosphorus Flame Retardants Treated Wood. The poplar wood specimens were treated with flame retardants by impregnation method. The prepared solutions of APP, DAP, and AP were impregnated into poplar wood specimens at room temperature under a high pressure (0.5 MPa) for 40 min, respectively, while H₂O was impregnated into wood at the same condition as control group. The impregnated specimens were then placed in an oven controlled to be 60°C for 24 h. In addition, half of the impregnated samples again with the silica sol solution were treated under the same process condition as mentioned above. After that, they were placed in an oven controlled at 60°C for 24 h and at 105°C for another 24 h to age the gels until SiO₂ aerogel formed in the cell of wood. The specific formula of impregnation experiment of every group could be seen in Table 1.

2.4. Evaluation and Characterization

2.4.1. Weight Percent Gain (WPG). The WPG of the samples (20 mm × 20 mm × 20 mm (T × R × L)) was determined on an oven-dried weight basis as shown in (1), by measuring the oven-dried weights of untreated specimens (M₀) and the composite specimens (M₁):

\[ \text{WPG} = \frac{M₁ - M₀}{M₀} \times 100\% \]  

2.4.2. Leaching Resistance. The prepared wood blocks including untreated and treated ones (20 mm × 20 mm× 20 mm (T × R × L)) were subject to a leaching test in a certain flow rate of water for 24 h. After the test, the specimens were removed out to dry in an air atmosphere for 24 h. After which, they should be placed in an oven controlled to be 60°C for another 24 h and then set to be 103°C until the weight was almost the same.

2.4.3. Microstructure Observation. Small samples with dimensions of 10 mm × 20 mm × 3 mm (T × R × L) were cut from the untreated and treated poplar wood, which were located 3 mm from each radial, tangential, and transverse surface of the wood. Before observing the samples with a scanning electron microscope (SEM, FEI Quanta 450) at an accelerating voltage of 20 kV, each sample surface should be sputter-coated with a thin layer of gold. The element composition was determined by regional analysis using an
Table 2: Mass variety of different stage for untreated and treated poplar wood.

<table>
<thead>
<tr>
<th>Samples</th>
<th>$W_0^a$ (g)</th>
<th>$W_1^a$ (g)</th>
<th>$W_2^a$ (g)</th>
<th>WPG$^b$ (%)</th>
<th>RL$^c$ (%)</th>
<th>ERL$^d$ (%)</th>
<th>RSFR$^e$ (%)</th>
<th>LRV$^f$ (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>S0</td>
<td>2.8811</td>
<td>2.8811</td>
<td>2.8673</td>
<td>0</td>
<td>0.48</td>
<td></td>
<td>4.22</td>
<td>31.46</td>
</tr>
<tr>
<td>S1</td>
<td>3.3466</td>
<td>3.795</td>
<td>3.4283</td>
<td>13.40</td>
<td>9.66</td>
<td>10.45</td>
<td>2.98</td>
<td>22.19</td>
</tr>
<tr>
<td>S2</td>
<td>3.2395</td>
<td>3.6746</td>
<td>3.2906</td>
<td>13.43</td>
<td>10.45</td>
<td>8.26</td>
<td>12.48</td>
<td>60.18</td>
</tr>
<tr>
<td>S3</td>
<td>3.3678</td>
<td>3.8283</td>
<td>3.4149</td>
<td>13.67</td>
<td>10.80</td>
<td>10.56</td>
<td>10.56</td>
<td>57.87</td>
</tr>
<tr>
<td>S4</td>
<td>3.2961</td>
<td>3.9796</td>
<td>3.6510</td>
<td>20.73</td>
<td>8.26</td>
<td>12.48</td>
<td>9.99</td>
<td>55.03</td>
</tr>
<tr>
<td>S5</td>
<td>3.4955</td>
<td>4.1337</td>
<td>3.8157</td>
<td>18.26</td>
<td>7.69</td>
<td>10.56</td>
<td>10.56</td>
<td></td>
</tr>
<tr>
<td>S6</td>
<td>3.1912</td>
<td>3.7708</td>
<td>3.4628</td>
<td>18.16</td>
<td>8.17</td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

$^aW_0$: weight of untreated wood. $W_1$: weight of treated wood after impregnation experiment. $W_2$: weight of treated wood after leaching test. $^b$WPG (weight percent gain) = ($W_1 - W_0$)/$W_0$. $^c$RL (average turnover rate) = ($W_1 - W_2$)/$W_1$. $^d$ERL was the average turnover rate of untreated wood. $^e$RSFR (rest loading efficiency) = WPG – RL + ERL. $^f$LRV (leaching resistance value) = RSFR/WPG.

energy dispersive spectrometer (EDS) combined with the SEM.

2.4.4. Thermogravimetric Analysis (TGA). TG data of the untreated and treated wood powder were obtained using a Pyris-6 thermal analyzer. Each sample powder of about 8 mg was placed in a platinum crucible and heated at the rate of 10°C/min from ambient temperature to 700°C. The flow of dynamic carrier nitrogen gas was set at a rate of 40 mL/min.

2.4.5. Cone Calorimetric (CONE) Test. The CONE tests of the untreated and treated wood were performed on Stanton Redcroft Inc. (Fire Testing Technology) following ISO5660-1. For each test, the specimen (100 mm × 10 mm × 10 mm (T × R × L)) covered with aluminum foil except the upper surface was placed into a specific steel frame, and the steel frame was mounted horizontally on the loader and exposed to the heat radiation of 50 kW/m2. This heat radiation corresponded to a temperature of 780°C on the upper surface of test sample.

2.4.6. Fourier Transform Infrared Spectrometer (FTIR) Analysis. For testing the change of chemical bonds and the combination mode between functional components and poplar wood, the method of tabletting with KBr was used with FTIR (IRAffinity-1). The powder of untreated and treated specimens was oven-dried, and the KBr was dried at 200°C in a muffle oven for 24 h. Subsequently, the testing sample was obtained by mixing about 1-2 mg oven dried samples with 100–200 mg KBr. The FTIR curves of samples were obtained at the test scanning range of 4000-400 cm$^{-1}$.

3. Results and Discussion

3.1. Weight Percent Gain (WPG) and Leaching Resistance. The impregnation experiment and leaching resistance test are evaluated by weight percent gain (WPG) and leaching resistance value (LRV), respectively. Every group has been prepared with ten samples to calculate the average weight. As we know, the mass of fire-retardant treated wood decreases after the extraction with water, which means the fire retardant is lost during the process of extraction. The LRV is the percentage of residue loading efficiency after extraction (RSFR) to the WPG. The lager the LRV is, the better the performance of leach resistance will be [13]. The results of mass variety of different stages for untreated and treated wood are shown in Table 2.

It can be seen from Table 2 that the WPG of phosphate flame retardant treated wood blocks is 13-14%, while that of wood treated with phosphate flame retardant and SiO2 sol reaches 18–21%. This illustrates that the effect of impregnation results is prominent. The mass of untreated wood after extraction is lost with average turnover rate (RL) of 0.48%, which means there is almost nothing lost of wood itself. The mass loss of wood treated with phosphate flame retardants is obvious. It is reflected in that the RL reaches by 9–11% while WPG is 13-14%, meaning a big loss of phosphate flame retardants. Besides, the LRV is only 20–32%. That is to say, its performance of leach resistance is poor. The LRV of wood treated with the combination of phosphate flame retardants and SiO2 sol reaches 55–61%, which is much bigger than that of wood treated with phosphate flame retardants. This means the wood treated with the combination of two materials have the better performance of leaching resistance. It can be explained that SiO2 sol has large specific surface area and strong absorbability, which helps to prevent the phosphate flame retardant from running off.

3.2. SEM-EDS. Figure 2 shows the SEM images of the surface of both transverse section and tangential section for untreated wood (S0) and treated wood samples (S1, S3, S4, and S6). It is clear that the vessel, wood fiber, and pit of poplar wood are interconnected. The porous structure of poplar wood is the main channels for phosphorus flame retardants and SiO2 sol into the poplar wood. It is obvious that no compound was detected in any part of the untreated wood blocks. When the samples were treated with phosphorus flame retardant (S1, S3), there are just white particles in the vessel, wood fiber, and wood ray. While the samples were treated with the mixture of phosphorus flame retardant and SiO2 sol (S4, S6), many agglomerates can be found in the vessel, wood fiber, wood ray, and pit. This is because the fluid can diffuse, resulting in the impregnation of agents into the wood.

To confirm the presence of flame retardants and SiO2 sol inside the cell of poplar wood, SEM-EDS mapping was carried out. Figure 3 shows the SEM-EDS map for the
Figure 2: SEM images of the surface of transverse section of (a) S0, (b) S1, and (c) S4 and the tangential section of (d) S0, (e) S3, and (f) S6, respectively.

Figure 3: EDS spectral profiles of untreated (S0) and treated wood (S2, S5).
Table 3: Values of combustion parameters for untreated and treated poplar wood.

<table>
<thead>
<tr>
<th>Samples</th>
<th>THR (MJ/m²)</th>
<th>TSR (m²/m²)</th>
<th>Mean COY (kg/kg)</th>
<th>Mean CO₂Y (kg/kg)</th>
</tr>
</thead>
<tbody>
<tr>
<td>S0</td>
<td>56.24</td>
<td>142.76</td>
<td>0.032</td>
<td>1.08</td>
</tr>
<tr>
<td>S1</td>
<td>36.72</td>
<td>55.63</td>
<td>0.065</td>
<td>0.69</td>
</tr>
<tr>
<td>S2</td>
<td>34.46</td>
<td>56.52</td>
<td>0.057</td>
<td>0.75</td>
</tr>
<tr>
<td>S3</td>
<td>38.19</td>
<td>68.53</td>
<td>0.062</td>
<td>0.72</td>
</tr>
<tr>
<td>S4</td>
<td>29.25</td>
<td>21.42</td>
<td>0.035</td>
<td>0.79</td>
</tr>
<tr>
<td>S5</td>
<td>29.03</td>
<td>23.83</td>
<td>0.037</td>
<td>0.78</td>
</tr>
<tr>
<td>S6</td>
<td>31.35</td>
<td>22.48</td>
<td>0.044</td>
<td>0.76</td>
</tr>
</tbody>
</table>

untreated and treated blocks and indicates that there is no functional element except the basic elements of wood such as C and O. When samples were treated with only phosphorus based flame retardant (S2), it was verified that the functional element P distribute inside porous structure of wood from the SEM-EDS analysis, although there was no obvious agglomerate block found in the vessel, wood fiber, and pit of wood. Additionally, the SEM-EDS images of the samples treated with both phosphorus flame retardants and SiO₂ sol (S5) indicate that the P and Si elements exist concurrently in the wood cell. This means that phosphorus based flame retardants and SiO₂ sol can reside in the poplar wood since poplar wood itself does not have P and Si elements. As the samples were obtained 3 mm from each radial, tangential, and transverse surface of the wood, it can be concluded that phosphorus flame retardants and SiO₂ sol are easy to be brought by water into the poplar wood. Similarly, because the phosphorus flame retardants are water-soluble, this is why the phosphorus flame retardants are easy to be washed away by water.

3.3. Thermal Degradation. The TGA curves of untreated and treated poplar wood are shown in Figure 4. All the materials show three stages of decomposition. The first decomposition stage of all materials starts at 30°C and ends at 120°C, which mainly is the evaporation of moisture resulting in about 1.5–3% of weight loss. In the range of 200°C–420°C, the weight of untreated wood decreases significantly up to 76.01% due to the decomposition or dehydration of hemicellulose and the decomposition of cellulose [7]. Whereas the second decomposition stage of treated wood starts at 200°C and ends at 330°C, the range of decomposition temperature is shorter than untreated wood, and the weight loss is only 42%–45%, which is less than untreated wood. And the rate of weight loss of treated wood is faster than untreated wood during this temperature range. After that there is a gradual loss in both cases.

The thermal degradation of treated wood ends earlier than that of untreated wood, while the qualities of carbon residue of treated wood are higher than those of untreated group. All these indicate that flame retardants catalyze the decomposition of treated wood and advance the decomposition of wood. Simultaneously, flame retardants change the reaction mechanism and direction of decomposition of treated wood which can generate more carbon and water when the treated wood decompose [7, 8]. What is more, the weight loss of the compound wood material treated with both phosphorus-containing flame retardants and SiO₂ sol is even less than the wood treated with only phosphorus-containing flame retardants. The weight loss of S4, S5, and S6 is 60.28%, 61.72%, and 63.36%, while the weight loss of S1, S2, and S3 is 62.59%, 63.65%, and 64.63%, respectively, indicating that SiO₂ sol can further improve the flame retardation of wood, because the presence of SiO₂ sol can enhance char formation of the matrix material and form a protective surface barrier to prevent the immediate damage of substrate materials [14].

3.4. Flame-Retardation and Smoke-Suppression Characteristics. The results of cone calorimeter tests for untreated and treated poplar wood are displayed in Figure 5 and Table 3. It is observed from Figure 5(a) and Table 3 that untreated wood burns very fast after ignition and a sharp heat release rate (HRR) peak appearing with a pk-HRR as high as 309.40 kW/m², whose total heat release (THR) values are 56.24 MJ/m². After treatment with phosphorus flame retardant, the HRR values all decrease by over 45%. The THR values of treated wood are lowered by 30% to 40%, which suggested that the addition of phosphorus flame retardant alone can improve the flame retardancy of wood. Besides, when the combination of phosphorus flame
retardant and SiO2 aerogel is applied for wood, the HRR, pk-
HRR, and THR values show an obvious reduction compared
with that of untreated wood, while present more reduction
compared with that of wood treated with phosphorus flame
retardant alone. For example, the poplar wood treated with
APP and SiO2 aerogel whose THR values are 29.25 MJ/m2
reduces by 48% and 20%, respectively, compared to that of
untreated wood whose THR values are 56.24 MJ/m2 and that
of APP treated wood whose THR values are 36.72 MJ/m2.
These results demonstrated a synergistic effect of phosphorus
flame retardant and SiO2 aerogel on improving the flame
erdancy of wood. It can be illustrated that phosphorus
flame retardants decompose at low temperature to form char
residue layer on the surface of the sample, which acts as a
barrier to prevent heat to underlying materials and flammable
gases into flame zone. On the other hand, SiO2 aerogel
significantly enhances the thermal stability of carbon, thus
forming a defending charring layer of Si–O–C and Si-C
bonds, which protect the poplar wood from burning.

Figure 5(b) shows the total smoke production (TSP)
curves of wood at a flux of 50 kW/m2 in the cone calorimeter
test. The total smoke release (TSR), mean CO yield (COY),
and CO2 yield (CO2Y) parameters can be seen in Table 3.
As is portrayed in Figure 5(b), the distinction between
pure wood and the flame retarded samples is apparent.
Furthermore, the TSP values of wood significantly decrease
with the addition of SiO2 aerogel. According to Table 3,
the TSR of wood that was treated with phosphorus flame
retardants alone and the combination of phosphorus flame
retardants and SiO2 aerogel was reduced by 51–62% and 83–
85%, respectively, compared to untreated wood. However,
the mean COY of wood treated with phosphorus flame
retardants is much higher than that of untreated wood. While
SiO2 aerogel is incorporated into the system, the mean COY
can reduce almost the same to that of untreated wood.
Besides, the CO2Y of treated wood is all lower than that of
untreated wood, while the SiO2 aerogel can inhibit the release
of smoke and toxic gases, which are the leading cause of
fire casualties. The phosphorus flame retardants act by the
condensed phase mechanism, during which dehydration of
the substrates leads to form more char, and protective coating
of polyphosphoric acid, phosphorus oxynitride structures or
polyphosphoramicid structures on the surface of burning
material is formed, resulting in the fact that less smoke and
gas were generated from wood [15]. The reason for good
smoke suppression of nano-SiO2 aerogel is attributed to the
SiO2 with highly adsorptive capacity and catalytic conversion
effect on the release of smoke and gas.

3.5. FT-IR Spectroscopy. The FTIR spectra of wood treated
with phosphorus and silicon-phosphorus flame retardant
are shown in Figure 6. It can be seen that most of the
characteristic peaks of treated wood are similar to those of
untreated one. The FTIR spectrum of wood treated with
phosphorus flame retardant (S1, S2, S3) shows an absorption
band at 1250–1253 cm−1, attributed to the stretching vibration
of P=O. The FTIR spectrum of wood treated with silicon-
phosphorus flame retardant (S4, S5, S6) indicated that the
composite not only has the peaks of phosphorus-oxygen
bond which can prove that the phosphorus flame retardant
exists in the poplar wood, but also shows new intensities
of peak which is proof that silica aerogel is present in the
wood. The FT-IR peaks of 1080–1110 cm−1 and 450–460 cm−1
corresponded to the asymmetric stretching of linear Si-O-
Si and bending vibration Si–O–Si [16, 17]. The absorption
band located between 1020 and 1040 cm−1 ascribes to Si-
OH groups [18]. What is more, the peak of 3000–3500 cm−1
which correspond to the stretching vibration of OH of treated
wood is different than that of untreated group; that is to say, the phosphorus flame retardants or SiO$_2$ aerogel may bond with wood cellulose by hydrogen bonds.

4. Conclusion

The nano-SiO$_2$ sol, obtained by sol-gel method, combined with three different phosphate flame retardants to treat wood in order to improve the flame retardancy and leaching resistance of wood. The SEM and SEM-EDS results showed that phosphorus flame retardants and SiO$_2$ sol resided in the poplar wood and widely distributed in the vessels, pits, wood timber, and the spaces between wood cells of poplar substrate. TGA indicated that phosphorus flame retardants could significantly improve the thermal stability of wood, while the SiO$_2$ sol could further enhance the thermal stability of the system. It was found that phosphorus flame retardant and nano-SiO$_2$ aerogel have a synergistic effect on improving the flame retardation of wood by CONE test. Besides, the combination of phosphorus flame retardants and nano-SiO$_2$ aerogel exerts a significant influence on inhibiting the release of smoke and toxic gases in the combustion processes, while the phosphorus flame retardants alone would lead to more smoke and toxic gases than untreated wood. In addition, the average LRV of composite could reach 55–61%, while that of untreated wood was 20–32% by a leaching resistance test, which means the phosphorus flame retardants were able to be fixed in the wood by nano-SiO$_2$ aerogel.

Conflict of Interests

The authors declare that there is no conflict of interests regarding the publication of this paper.

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