

Research Article

Multicolor Emission Tuning and Red/Green Ratio Enhancement of $\text{Yb}^{3+}/\text{Er}^{3+}$ Codoped KGdF_4 Upconversion Nanoparticles

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Herein, a series of $\text{KGdF}_4:\text{Yb}^{3+}/\text{Er}^{3+}$ upconversion nanoparticles (UCNPs) were synthesized through a one-pot hydrothermal method using polyethylene glycol (PEG) as capping ligands. The phase and microstructure studies show all these as-prepared UCNPs are pure cubic phase and uniformed nanoparticle shape by changing the doped Yb^{3+} concentration from 18% to 98%. The as-prepared UCNPs can realize the multicolor emissions from yellow to red and the red-to-green (R/G) ratio can be enhanced from 2.05 to 8.35 when Yb^{3+} varies from 18% to 98%. In addition, the proposed upconversion (UC) mechanisms of these PEGylated UCNPs are investigated in detail. The realization of multicolor tuning and enhanced R/G ratio by only increasing the doping concentration of Yb^{3+} ions in KGdF_4 host indicates that the PEGylated $\text{KGdF}_4:\text{Yb}^{3+}/\text{Er}^{3+}$ UCNPs can find their application on lighting devices, anticounterfeit technology, and even bioimaging field.

1. Introduction

In recent years, lanthanides- (Ln-) doped UCNPs have triggered intense research interest and have been widely applied in lighting devices, magnets, anticounterfeit technology, and biomedical field [1–5]. This is due to the fact that their optical, magnetic, electronic, and chemical nature arises from the 4f shell of Ln ions [6–8]. Besides, Ln-doped UCNPs exhibit large antistocks effects, which can convert a longer wavelength light like near-infrared (NIR) to shorter wavelength luminescence (ultraviolet, visible light, and NIR). Among all the well-developed UCNPs, sodium rare-earth fluorides (NaREF_4), such as NaYF_4 , NaLuF_4 , and NaGdF_4 , are considered as promising and the most efficient host matrix owing to their normally low photon energy, which can decrease the nonradiative radiation probability and increase the UC emission and quantum yield [9–11]. Boyer et al. [12] have reported that $\text{Yb}^{3+}/\text{Er}^{3+}$ and $\text{Yb}^{3+}/\text{Tm}^{3+}$ codoped NaYF_4 colloidal UCNPs as the material with high upconversion efficiency. Wang et al. [13] have shown that NaYF_4 nanocrystals can simultaneously realize size/phase and UC emission color tuning by doping trivalent Ln ions at defined concentrations. Recently,

Tian et al. [14] and Zeng et al. [15] have demonstrated Mn^{2+} doping methods for multicolor tuning and enhanced red UC emission in NaREF_4 systems with fixing the doped Yb^{3+} and Er^{3+} concentration. Nevertheless, how to achieve multicolor UC emission and enhanced red emission by only changing the doped Yb^{3+} concentration is still a huge challenge because $\text{Yb}^{3+}/\text{Er}^{3+}$ codoped UCNPs present green UC emission in general. Recently, Chen et al. [16] have reported that UC NIR emission centered at 800 nm can be enhanced in $\text{Yb}^{3+}/\text{Tm}^{3+}$ codoped NaYF_4 UCNPs along with the increase of Yb^{3+} concentration. This finding inspires us to achieve UC emission color tuning from green to red and to enhance red luminescence in $\text{Yb}^{3+}/\text{Er}^{3+}$ codoped UCNPs.

Apart from the well-established NaREF_4 UCNPs, KREF_4 can also act as effect and ideal UC host lattice. For instance, Wong et al. [17] have synthesized small sized $\text{Yb}^{3+}/\text{Er}^{3+}$ codoped KGdF_4 UCNPs via one-pot solvothermal process utilizing branched polyethyleneimine and 6-aminocaproic acid as capping ligands. However, the authors' emphasis has not focused on multicolor tuning and red emission enhancement but the green emission and paramagnetic nature of

these $\text{KGdF}_4:\text{Yb}^{3+}/\text{Er}^{3+}$ UCNPs. Importantly, enhanced red emission, considered as “optical window” for live beings, is beneficial to bioimaging for deep tissue penetration. In addition, in spite of the perfect UC host, Gd^{3+} -based nanomaterials can also act as promising T_1 -weighted contrast agents for magnetic resonance imaging [9]. However, there are few literatures, up to now, concentrated on KGdF_4 -hosted UCNPs. Therefore, it is of significance to develop multifunctional $\text{Yb}^{3+}/\text{Er}^{3+}$ codoped KGdF_4 UCNPs using a simple method for emission color tuning and enhanced red emission.

In this work, a series of $\text{KGdF}_4:\text{Yb}^{3+}/\text{Er}^{3+}$ UCNPs were fabricated by a one-pot hydrothermal method using PEG as capping ligands. The phase compositions and microstructure of these as-prepared products were characterized by X-ray diffraction (XRD) and field-emission scanning electron microscope (FE-SEM), respectively. The UC properties of these UCNPs were detected using emission spectra and the UC emission mechanisms were analyzed in detail. As expected, multicolor tuning and enhanced red UC emission were achieved by only tuning the doped Yb^{3+} concentration in the host, which indicated that these as-prepared $\text{KGdF}_4:\text{Yb}^{3+}/\text{Er}^{3+}$ UCNPs could expand their application from optics to biomedicine.

2. Experimental Section

2.1. Materials and Regents. $\text{RE}(\text{NO}_3)_3 \cdot 6\text{H}_2\text{O}$ ($\text{RE} = \text{Gd}^{3+}$, Yb^{3+} , and Er^{3+} , 99.99%) were purchased from Sigma-Aldrich. Ethylene glycol (EG), PEG, KF, and other reagents were purchased from Sinopharm Chemical Reagent Co., China. All of these chemicals were used as received without further purification.

2.2. One-Pot Method for Fabrication of $\text{KGdF}_4:\text{Yb}^{3+}/\text{Er}^{3+}$ UCNPs. $\text{KGdF}_4:x\%\text{Yb}^{3+}/2\%\text{Er}^{3+}$ ($x = 18, 38, 58, 78, \text{ and } 98$) UCNPs were synthesized through a simple one-pot hydrothermal method using PEG as a surface modifier [10]. In a typical process, 1.0 g of PEG was dissolved into 15 mL of EG solution for the formation of aqueous solution. And then, a total amount (1 mmol) of $\text{RE}(\text{NO}_3)_3$ water solutions with as-designed molar ratio was added. After 10 min agitation, 6 mL of KF aqueous solution (1 mM) was added into the former mixture. After another 30 min agitation, the mixture was poured in a 50 mL Teflon-lined stainless steel autoclave. The system was sealed and kept at 200°C for 24 h. After reaction, the system was cooled to room temperature and the product was washed with absolute ethyl alcohol and deionized water and dried at 60°C for 24 h.

2.3. Characterizations. XRD analysis was measured by a D/max- γ A System X-ray diffractometer at 40 kV and 250 mA with $\text{Cu-K}\alpha$ radiation ($\lambda = 1.54056 \text{ \AA}$). Morphologies of these UCNPs were characterized by FE-SEM (FEI NanoSEM 450). UC photoluminescence spectra of these UCNPs were recorded by a Zolix analytical instrument (fluoroSENS 9000A) equipped with a 980 nm laser.

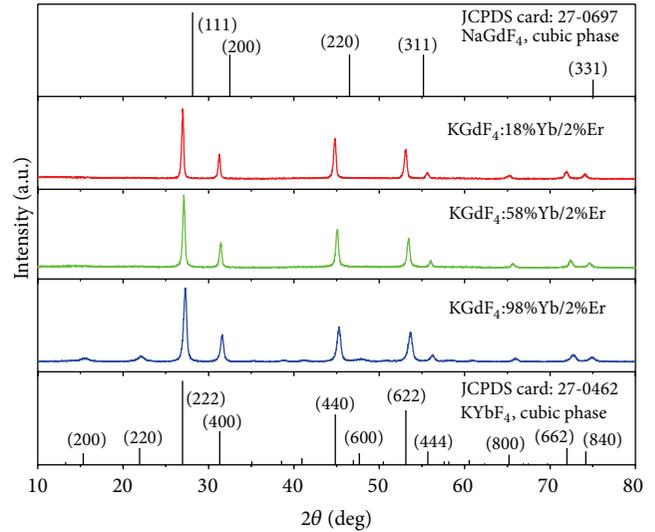


FIGURE 1: XRD patterns of $\text{KGdF}_4:18\%\text{Yb}^{3+}/2\%\text{Er}^{3+}$, $\text{KGdF}_4:58\%\text{Yb}^{3+}/2\%\text{Er}^{3+}$, $\text{KGdF}_4:98\%\text{Yb}^{3+}/2\%\text{Er}^{3+}$ UCNPs, and standard JCPDS Card of cubic phase NaGdF_4 and KYbF_4 .

3. Results and Discussion

Crystal phases and compositions of these as-prepared $\text{KGdF}_4:\text{Yb}^{3+}/\text{Er}^{3+}$ UCNPs were synthesized by a simple one-pot hydrothermal process utilizing PEG as capping ligands. These UCNPs were recorded by powder XRD patterns. As presented by the red line in Figure 1, the measured XRD pattern of the PEGylated $\text{KGdF}_4:18\%\text{Yb}^{3+}/2\%\text{Er}^{3+}$ UCNPs is in agreement with that reported in previous literatures [16], which is similar to that of the cubic phase NaGdF_4 (JCPDS Card: 27-0697) and space group $Fm\bar{3}m$. It therefore can be seen that the $\text{KGdF}_4:18\%\text{Yb}^{3+}/2\%\text{Er}^{3+}$ UCNPs crystallize in the fluorite structure known for cubic phase NaGdF_4 . When the doped Yb^{3+} concentration increases by 58%, the diffraction peaks shift towards high angel sides owing to the replacement of Gd^{3+} (1.193 \AA) by relatively smaller Yb^{3+} ions (1.125 \AA) in the host [17]. After the total substitution of Gd^{3+} through Yb^{3+} , the UCNPs show the pure cubic KYbF_4 phase (JCPDS Card: 27-0462). As observed, all these as-prepared UCNPs are of pure and single cubic phase.

To further study the morphology of these PEGylated UCNPs, FE-SEM images were carried out. As shown in Figures 2(a)–2(e), all these as-prepared PEGylated UCNPs are uniformed nanoparticle shape and the high doping has not changed the crystal structures and morphology of the host materials. The average diameter of these UCNPs is measured to 86.7, 64.8, 60.8, 46.6, and 43.1 nm while the doping ions of Yb^{3+} vary from 18% to 98% based on FE-SEM images. In addition, the EDS of $\text{KGdF}_4:18\%\text{Yb}^{3+}/2\%\text{Er}^{3+}$ UCNPs (Figure 2(f)) shows the major elements composition of K, Gd, and Yb. The detected C and Si come from the silicon pellet of FE-SEM. In addition, the absence signal of Er element is due to the low doping concentration and the accuracy of the detector. Figure 2(g) shows the EDS of

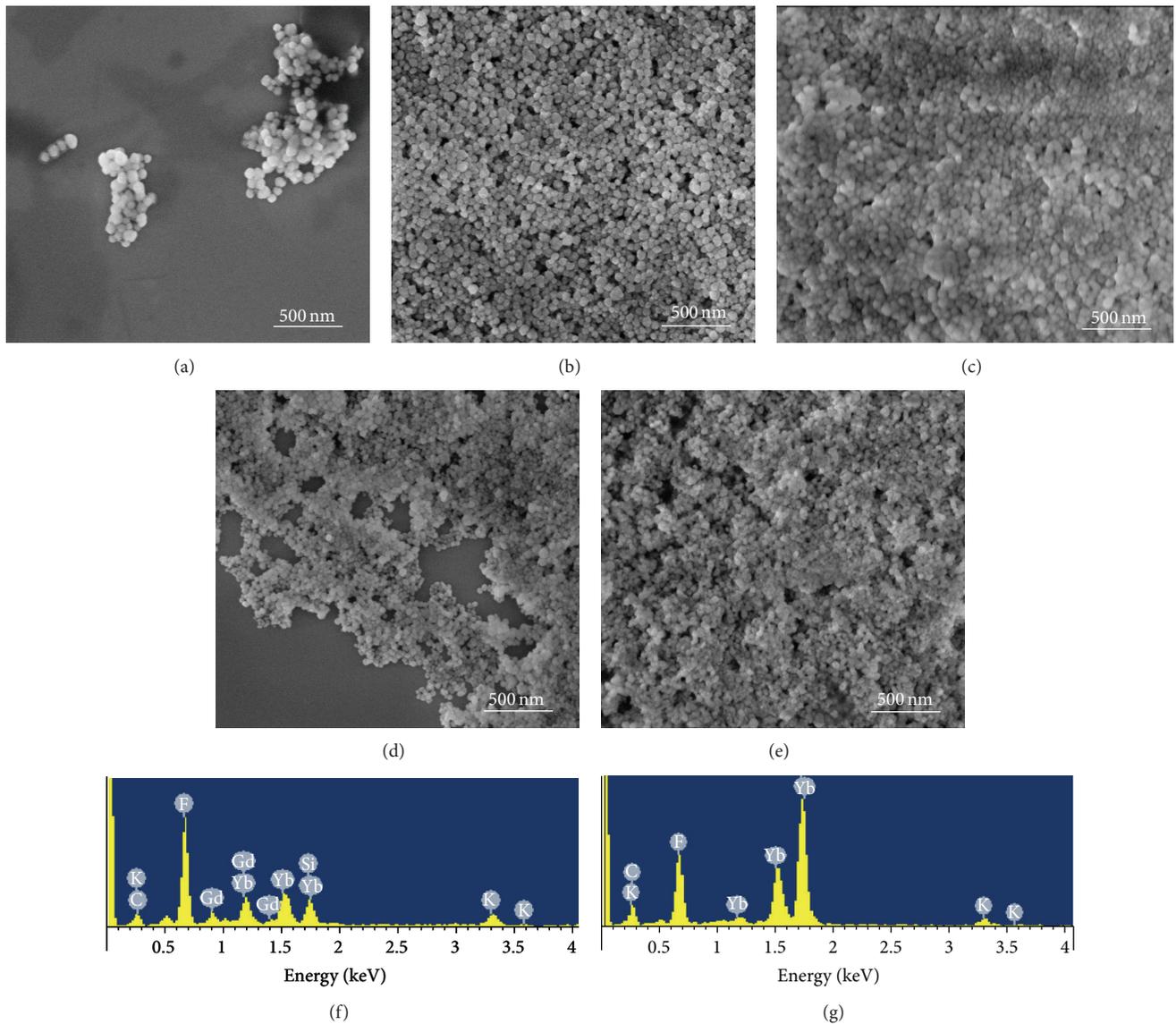


FIGURE 2: FE-SEM images of $\text{KGdF}_4:\text{Yb}^{3+}/\text{Er}^{3+}$ UCNPs with doping different Yb^{3+} ions: (a) 18%, (b) 38%, (c) 58%, (d) 78%, and (e) 98%; the EDS of (f) $\text{KGdF}_4:18\%\text{Yb}^{3+}/2\%\text{Er}^{3+}$ and (g) $\text{KGdF}_4:98\%\text{Yb}^{3+}/2\%\text{Er}^{3+}$ UCNPs.

$\text{KGdF}_4:98\%\text{Yb}^{3+}/2\%\text{Er}^{3+}$ UCNPs, indicating the absolute substitution of Gd by Yb elements in the nanocrystals.

Figure 3 gives the UC emission spectra of $\text{KGdF}_4:x\%\text{Yb}^{3+}/2\%\text{Er}^{3+}$ ($x = 18, 38, 58, 78, \text{ and } 98$) UCNPs with different doping concentration of Yb^{3+} . The emission bands can be attributed to the ${}^2\text{H}_{11/2} \rightarrow {}^4\text{I}_{15/2}$, ${}^4\text{S}_{3/2} \rightarrow {}^4\text{I}_{15/2}$, and ${}^4\text{F}_{9/2} \rightarrow {}^4\text{I}_{15/2}$ transitions of Er^{3+} , respectively [18, 19]. As observed in Figure 3, $\text{KGdF}_4:\text{Yb}^{3+}/\text{Er}^{3+}$ UCNPs present the strong green and red emission when doping 18% Yb^{3+} . However, the green emission has been inhibited and the R/G ratio (Figure 4) can be promoted from 2.05 to 2.28, 3.12, 3.48, and finally 8.35 by ranging from the doping concentration of Yb^{3+} from 18% to 98%. The corresponding CIE chromaticity of $\text{KGdF}_4:x\%\text{Yb}^{3+}/2\%\text{Er}^{3+}$ ($x = 18, 38, 58, 78, \text{ and } 98$) UCNPs is shown in Figure 5. The output emission colors

tune obviously from the yellow region to the red region with the increase of the doped Yb^{3+} concentration owing to the intensities variation of the green and red emissions.

To further understand the UC emission properties of the PEGylated $\text{KGdF}_4:\text{Yb}^{3+}/\text{Er}^{3+}$ UCNPs by tuning the doped Yb^{3+} concentration, the proposed UC mechanisms should be investigated in detail. As illustrated in Figure 6, the first energy transfer from Yb^{3+} to Er^{3+} excites the ${}^4\text{I}_{15/2}$ to ${}^4\text{I}_{11/2}$ state followed by the redundant energy dissipated by phonons. As a consequence, Er^{3+} relaxes nonradiatively to the lower ${}^4\text{I}_{13/2}$ state and then suffers a second energy transfer process to ${}^4\text{F}_{9/2}$ state. The red emission is associated with the radiative transition from ${}^4\text{F}_{9/2}$ state to ground state. In addition, another energy transfer from ${}^4\text{I}_{11/2}$ to ${}^4\text{F}_{7/2}$ state results in ${}^2\text{H}_{11/2}/{}^4\text{S}_{3/2}$ to the ground state with the green emissions centered at

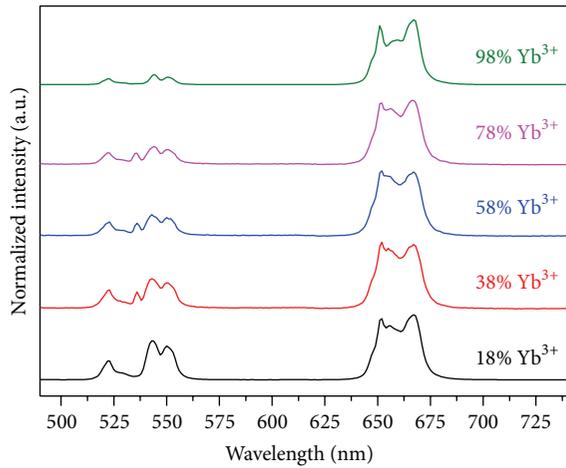


FIGURE 3: UC emission spectra of $\text{KGdF}_4:x\% \text{Yb}^{3+}/2\% \text{Er}^{3+}$ ($x = 18, 38, 58, 78, \text{ and } 98$) UCNPs.

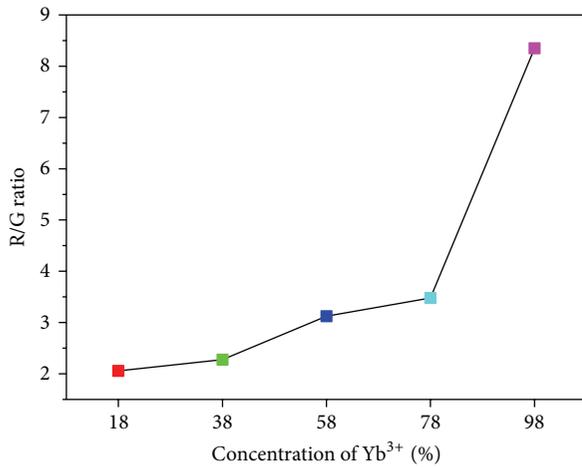


FIGURE 4: The variation of R/G ratio of $\text{KGdF}_4:\text{Yb}^{3+}/\text{Er}^{3+}$ UCNPs as a function of Yb^{3+} concentration.

523 and 543 nm, respectively, followed by the corresponding nonradiative relaxations. However, the increase of the doped Yb^{3+} concentration into the host results in the inhibition of green emission and the enhancement of R/G ratio when ranging Yb^{3+} concentration from 18% to 98%. This is due to the decrease in average distance between Yb^{3+} and Er^{3+} with the increase in number of Yb^{3+} per Er^{3+} , which can increase the energy transfer efficiency between Yb^{3+} and Er^{3+} . However, the cross relaxation process ${}^4\text{F}_{7/2} (\text{Er}^{3+}) + {}^2\text{F}_{7/2} (\text{Yb}^{3+}) \rightarrow {}^4\text{I}_{11/2} (\text{Er}^{3+}) + {}^2\text{F}_{5/2} (\text{Yb}^{3+})$ from Er^{3+} to Yb^{3+} is favoured at the same time, which results in the decrease of ${}^4\text{F}_{7/2} (\text{Er}^{3+})$ population and green emissions [20]. As a consequence, through energy transfer process ${}^2\text{F}_{5/2} (\text{Yb}^{3+}) + {}^4\text{I}_{13/2} (\text{Er}^{3+}) \rightarrow {}^2\text{F}_{7/2} (\text{Yb}^{3+}) + {}^4\text{F}_{9/2} (\text{Er}^{3+})$, the excited Yb^{3+} can transfer its energy to Er^{3+} , which contributes to the more population of Er^{3+} at ${}^4\text{F}_{9/2}$ state and results in the enhancement of red emission.

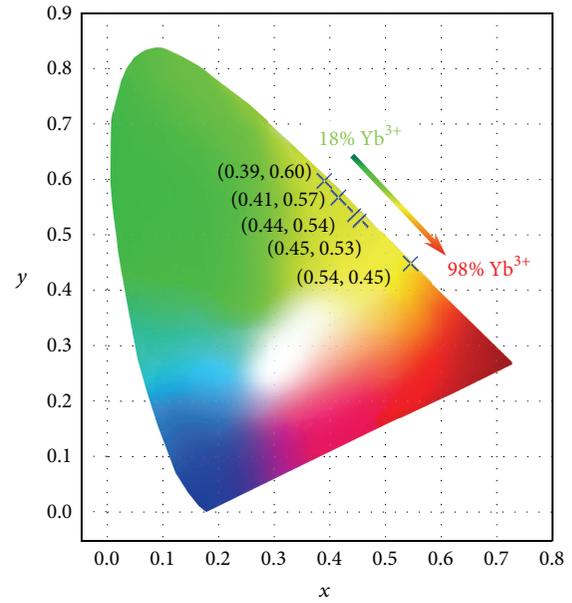


FIGURE 5: The CIE-1931 chromaticity diagram and calculated chromaticity coordinates of these UCNPs.

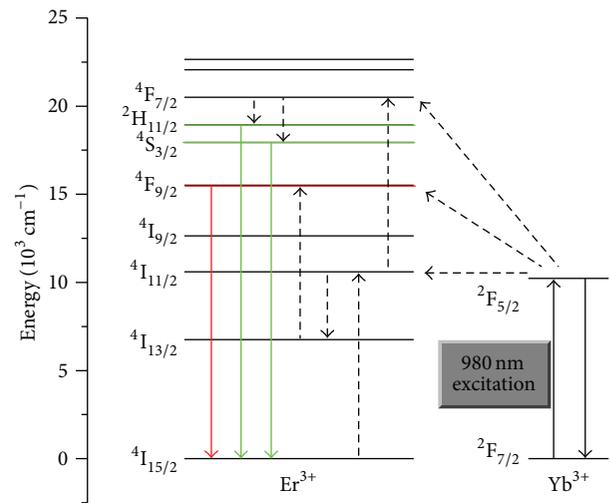


FIGURE 6: Schematic energy level of Yb^{3+} and Er^{3+} and their proposed energy transfer mechanisms in KGdF_4 UCNPs.

4. Conclusions

In conclusion, $\text{KGdF}_4:x\% \text{Yb}^{3+}/2\% \text{Er}^{3+}$ ($x = 18, 38, 58, 78, \text{ and } 98$) UCNPs were fabricated via a one-pot hydrothermal method using PEG as capping ligands. All these PEGylated UCNPs were cubic phase and uniformed nanoparticle shape. These UCNPs realized yellow to red color emission by tuning the Yb^{3+} concentration from 18% to 98%, as well as the R/G ratio enhancement from 2.05 to 8.35. The multicolor emission and enhanced R/G ratio revealed that these PEGylated $\text{KGdF}_4:\text{Yb}^{3+}/\text{Er}^{3+}$ UCNPs can act as nanophosphors and expand their application on lighting devices, anticounterfeit technology, and even bioimaging field.

Conflict of Interests

The authors do not have any conflict of interests in their submitted paper.

Acknowledgment

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