

STIMULATED PROCESSES IN A HALF-COLLISION

H.H. TELLE
Department of Physics
University College of Swansea
Swansea SA2 8PP, United Kingdom

Selective photodissociation of diatomic molecules is used to prepare the separating species in a well defined quantum state with narrowly determined final kinetic energy of the particles. During the course of separation to products the repulsive potential is probed by a tunable dye laser, that is by induced absorption in case the dissociation proceeds on a potential leading to ground state products, or by induced emission or absorption for all other cases. The determination of interatomic potentials from the observed spectra is discussed.

1. INTRODUCTION

The probing of atomic and molecular interaction potentials during collisions by means of laser techniques has become more and more a common method. It has been shown that "clean" collision parameters may be essential in the experiments and for the interpretation of their results. Such conditions can be realized in the photodissociation process where initial conditions are in general well known, and the problem is restricted to the treatment of only half a collision. The observation of weak wing emission from an electronically excited dissociation product allows to deduce

information on the interaction potentials of both upper and lower molecular states. This has been discussed in recent years; see e.g. Foth et al.¹ and Telle².

Most experiments undertaken so far have been performed in a vapor cell. This poses a severe drawback in many occasions since evaporation temperatures for various molecules may be quite high, and thus a large number of initial states is populated. A more favorable condition is often found in molecular beam experiments in which by seeding of the investigated molecule its vibrational and rotational temperatures may be lowered significantly. Usually high densities of the species are required since the signals to be observed are extremely feeble; thus a high density beam source³ has to be used.

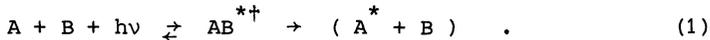
A further increase in sensitivity over pure emission wings during the dissociation process is to be expected when stimulated processes are utilized. Absorption of laser radiation during atom-atom collisions is now well established for probing their interaction potentials. For the dissociation along a potential curve leading to excited state products strong emission may be induced to the ground state when sweeping the wing spectrum with a tunable laser; this process can be viewed as being similar to stimulated Raman scattering, and the induced transition is then evident as a decrease of line intensity in the atomic product. The straight-forward analogy to the absorption during an atom-atom collision is encountered for absorption during the dissociation process, but now only a half-collision has to be treated. Successful absorption is detected as emission from an excited product state.

The scheme is exemplified for the alkali halides. In section 2 the principles are outlined how to derive in-

termolecular potentials, for the case of atom-atom collisions. Induced absorption and emission during the photodissociation process are described in section 3, followed by the discussion of some preliminary results in section 4.

2. PRINCIPLES OF ATOM-ATOM COLLISIONS

One elegant approach for the understanding of what is happening during the course of a collision is the absorption in the (unbound) complex of close encounter in atom-atom scattering. The interaction of the complex with a photon may be described by



In general, the channel indicated parenthetically represents a minor pathway.

The absorption of a photon during the collision, assuming only vertical transitions in the classical Franck-Condon or quasistatic picture, probes the potential difference $\Delta U = U^*(R) - U(R) = h\nu$ at any internuclear separation, and the spectrum resembles a wing to the central atomic transition. Observation of the absorption process is made through some direct or cascading transitions in the collision product (see Figure 1).

In one of the usual approaches for obtaining absorption spectra in atom-atom collisions the beam of a tunable laser is passed through a cell containing a metal atom - rare gas mixture, M - RG. The absorption of a laser photon by a M - RG pair creates a $M^* - RG$ pair, which then may separate to products, $M^* + RG$. The intensity of the M^* fluorescence is proportional to the absorption coefficient $K(\lambda)$; the typical shape of such a spectrum using photon counting techniques⁴ is shown in Figure 2c. The measurement then is repeated for various temperatures of the vapor cell; this is

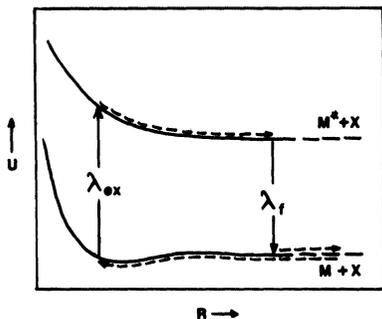


FIGURE 1 Schematics of absorption during atom-atom collisions.

necessary to deduce the two potentials unambiguously, as will be outlined below.

For the interpretation of the observed line profiles various models have been utilized; for example, profiles can be described, at least partially, by the quasi-static theory. According to this interpretation, the absorption coefficient is given by

$$K(\lambda) \sim 4\pi \cdot R^2 n_M n_{RG} \cdot f(R) \cdot \exp(- (U^*(R) - U(R)) / kT) \cdot dR/d\lambda \quad (2)$$

Here n_M and n_{RG} are the concentrations of perturbed and perturbing atoms, $f(R)$ is the oscillator strength of the transition, T is the temperature and U^* and U are the interaction potentials between the perturber and the atom in its initial or final state, respectively. However, when there is an extremum in the difference between upper and lower state potentials the spectrum generally exhibits so-called "satellites". In this region the quasi-static approximation is no longer valid, and a more refined interpretation has to be used, for example, the unified Franck-Condon (UFC) treatment⁵ (see Figure 2b).

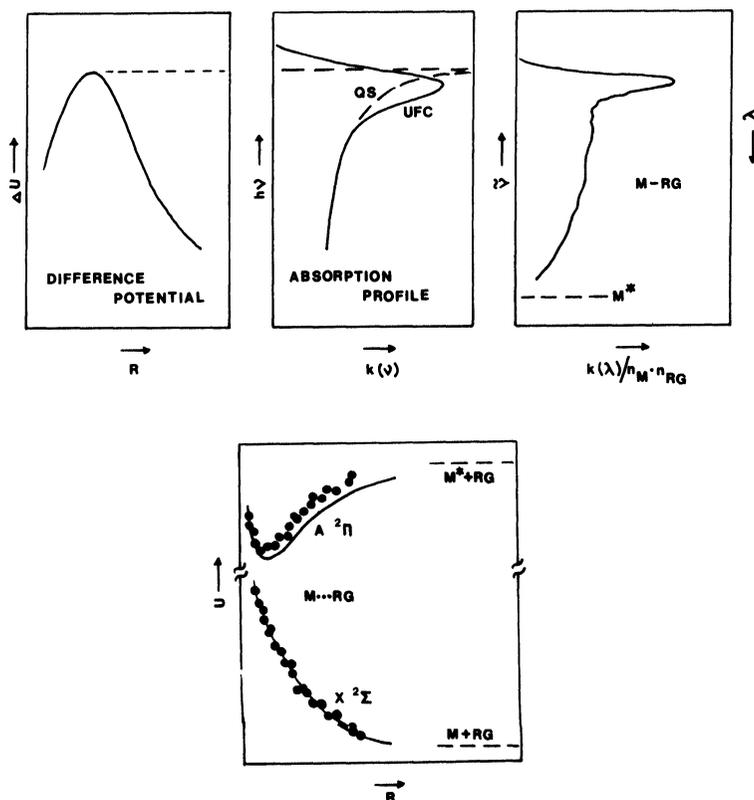


FIGURE 2 Wing spectra in atom-perturber collisions; (a) difference potential, (b) quasi-static (QS) and unified Franck-Condon (UFC) profiles, (c) experimental absorption spectrum (d) potentials deduced from spectra like in (c).

According to relation (2), the temperature dependence of the absorption spectrum can be interpreted as follows: at a given λ the plot $\ln K(\lambda) = -(1/T)$ is a straight line, and its slope yields $U(R_\lambda)$. The upper state potential, $U^*(R_\lambda)$, is then obtained by adding the difference potential (proportional to the photon energy) to this potential (see Figure

2d). Such experimental potentials can be compared with those from theoretical calculations (full lines in Figure 2d). Moreover, the absolute measurement of the intensity of the absorption spectrum leads to the determination of the oscillator strength of the transition as a function of R .

3. PRINCIPLES OF A HALF-COLLISION

3.1. Photodissociation and emission during the course of separation

Molecules may be dissociated by photoexcitation; if this dissociation proceeds on a potential surface which correlates to an excited product state then fluorescence from that excited state is expected. Accompanying the strong atomic emission for infinitely separated particles weak wings may also be observed originating from the molecular complex in the course of falling apart. The situation is shown schematically in Figure 3; the dashed lines for the wing emission indicate that the intensity observed is very small

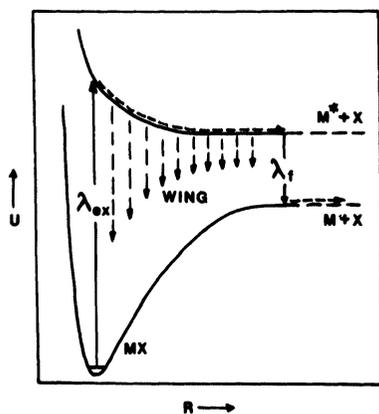


FIGURE 3 Schematics of emission during dissociation following the excitation with a photon.

in comparison to the fluorescence at wavelength λ_f . The spectra obtained from the emission of the transient dissociating state can be linked to the interaction potentials. In general, this poses a simplification over the case where a full atom-atom collision has to be considered, since initial conditions for the photodissociation process (a half-collision) are usually quite distinct. Wing spectra originating from the described process have been observed in the photodissociation of NaI^2 . In the corresponding experiments NaI contained in a vapor cell is dissociated by the radiation of a KrCl laser; strong emission of the sodium D-line is observed. The accompanying wings are 5 to 7 orders of magnitude weaker than the atomic emission but nevertheless show interesting structure; however, interpretation of the spectra in terms of molecular potentials is still not complete.

3.2. Induced absorption and emission phenomena

The experimental scheme utilized for the NaI photodissociation can be improved significantly. One major restriction in a vapor cell experiment is the relatively high temperature needed to produce sufficient particle densities; this in turn results in a rather broad population distribution, and the advantage of the half-collision scheme is diminished. The situation improves dramatically when molecular beam techniques are applied. In a seeded beam the effective rotational and vibrational temperatures can be lowered significantly, so that only a few very low lying energetic states are populated. One may also wish to increase the sensitivity in the detection channel. As it has been pointed out above the spontaneous emission during the separation of the molecule to its atomic products is very weak; the signal strength can be largely enhanced when using laser-induced

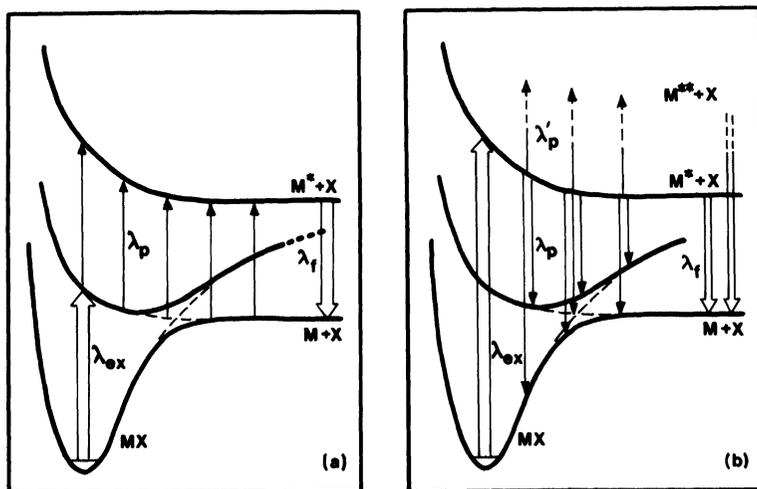


FIGURE 4 Schematics of laser-induced processes during photodissociation; (a) stimulated absorption from a potential leading to ground state products, and (b) stimulated emission from a potential leading to excited state products, where various lower potentials may be accessed; alternatively, also stimulated absorption to higher potentials is feasible.

processes. This has the further advantage that also potentials are accessible which lead to nonradiative product states.

In Figure 4a the schematics for induced absorption are shown. A strong laser at wavelength λ_{ex} excites the molecule to the dissociative branch of a potential correlating to ground state products; this laser may also be tunable in order to access various parts of the repulsive potential. While the molecule is en route to products a second laser probes the corresponding potential. The successful absorption of a photon can be observed via the emission from the then excited atomic product. In this way the complete po-

tentials from the internuclear separation of dissociative excitation out to infinite separation of the product atoms can be explored. The procedure closely resembles the atom-atom collision described in section 2, but only for a half-collision.

The induced emission scheme to replace the spontaneous emission case described above is shown in Figure 4b. The exciting laser now transfers the molecule to a dissociative potential correlating to excited products. The tunable laser then induces downward transitions from the potential on which the molecule separates to lower lying potentials; this is indicated in the figure by multiple arrows. The successful induced emission may be detected via a decrease in emission intensity in the atomic product. However, this could be rather difficult; the change in atomic emission intensity may be minute because only rather small fractions of all particles may suffer induced downward transitions. To overcome this problem induced absorption to higher lying potentials could be helpful since this creates atomic emission lines differing from the original $M^* \rightarrow M$ line; this process is indicated by the dashed lines in Figure 4b.

4. PRELIMINARY RESULTS AND DISCUSSION

The procedure outlined in the previous section is now being tested in its initial stages with NaI in order to have data from the spontaneous emission scheme for comparison.

The experimental set-up is shown in Figure 5. Since the processes to be observed are rather feeble sufficient particle densities are required. This can be achieved with a high-flow seeded beam source for molecules developed earlier³. NaI densities comparable to those in a vapor cell can be produced. The molecular beam is irradiated perpen-

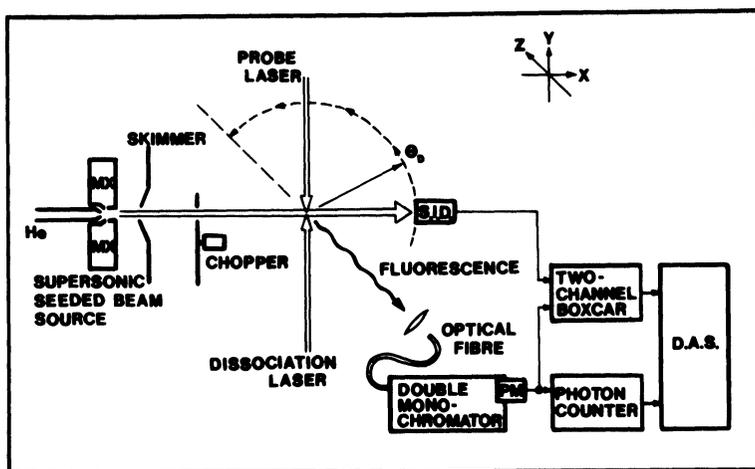


FIGURE 5 Experimental set-up.

dicular by two counter-propagating laser beams. One is the dissociating laser at wavelength λ_{ex} , whereas the second laser induces transitions during the separation of the molecule at the tunable wavelength λ_p . Two detectors are available in the experiment. The dissociation to ground state products may be detected by a surface ionization (S.I.) detector, which can be moved around the interaction volume. As a bonus the angular distribution of the photodissociation products can be traced. The fluorescence signal from the sodium D-line is filtered by a double-monochromator; in this way straylight from the lasers and parasitic fluorescence can be suppressed. Alternatively the monochromator can select short-wavelength atomic transitions when induced absorption is used as in the case depicted in Figure 4b. The signals from the photomultiplier and the S.I. detector are sampled with a photon-counting / boxcar-integrator arrangement and handled by a data acquisition system.

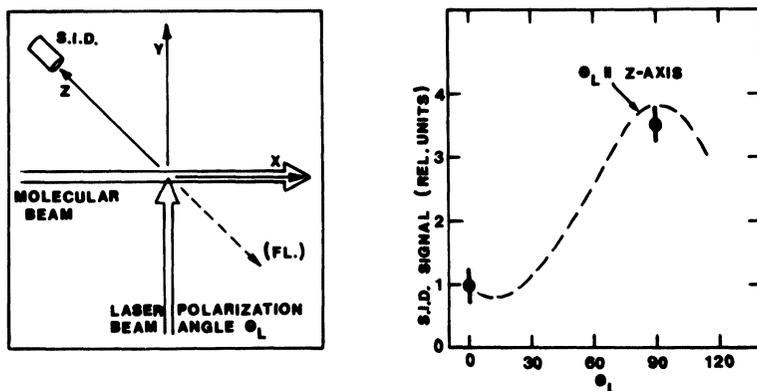


FIGURE 6 Test data; dependence of the photodissociation probability on the polarization of the exciting laser.

In a preliminary study the S.I. detector was set at 90° to both the molecular and laser beam. In Figure 6 the measured relative intensity from the S.I. detector is plotted as a function of the polarization of the dissociation laser, in this study the 3rd harmonic of a Nd:YAG laser; its polarization is with respect to the axis of the S.I. detector. The results are in good agreement with other measurements⁶.

For the step of stimulated absorption the 2nd harmonic of the same Nd:YAG laser has been used. This wavelength is sufficiently far from the centre D-line to be outside the Lorentz core. Strong emission of the sodium D-line is observed.

The two fixed wavelengths in these test measurements will now be replaced by the output from two tunable dye lasers to probe the complete range of internuclear separations.

Measurement of the wing spectra, both for the induced absorption and/or emission, for various excitation wave-

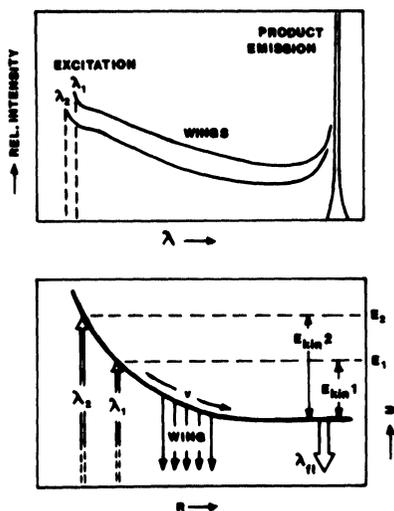


FIGURE 7 Conversion procedure from experimental wing spectra to molecular potentials, in dependence from the excitation wavelength.

lengths resembles an equivalence to the temperature dependence in the absorption during atom-atom collisions. The respective dissociation energies result in different kinetic energies with which the atoms recess from their transient configuration on the repulsive potential. The situation is schematically shown in Figure 7.

For any frequency ν_p of the probe laser the emitted fluorescence intensity in the D-line is correlated to the time the atoms linger at the corresponding internuclear distance,

$$I_f(\nu_p, E_i) \sim |D(R_p)|^2 \cdot v_i(R_p)^{-1} \\ = |D(R_p)|^2 \cdot (2(E_i - U^*(R_p))/\mu)^{-1/2} ; \quad (3)$$

E_i stands for the dissociation energy at the i -th excitation wavelength, $v_i(R)$ is the recoil velocity at distance R , and

μ is the reduced mass of the molecule. The intensity ratio, Q , for different energies, say E_1 and E_2 (corresponding to λ_1 and λ_2), is

$$Q = (I_f(\nu_p, E_1) / I_f(\nu_p, E_2))^{1/2} \\ = (E_1 - U^*(R_p)) / (E_2 - U^*(R_p)) \quad , \quad (4)$$

and one deduces for the potential of the excited state

$$U^*(R_p) = (E_1 - Q \cdot E_2) / (1 - Q) \quad , \quad (5)$$

while the difference potential ΔU is related to the transition frequency ν_p by $h\nu_p = \Delta U(R_p) = U^*(R_p) - U(R_p)$. Hence, both potentials can be derived from the wing spectra. The well known position of the minimum of the ground state, from which the dissociation step starts, serves as a reference point in the scale of internuclear separations. Care should be taken that the fluorescence intensity in the product state follows the strength of the probe laser radiation linearly, otherwise a conversion from the observed spectra to molecular potentials is not possible.

5. CONCLUSION

We have shown that the proposed scheme of induced absorption and emission for the investigation of molecules in the process of falling apart is feasible. The procedure will be of superior sensitivity to the previously used method of wings produced by spontaneous emission, and a precise tracking of repulsive potentials should now be possible.

REFERENCES

1. H. J. FOTH, J. C. POLANYI, H. H. TELLE, J. Phys. Chem., 86, 5027-5043 (1982)
2. H. H. TELLE, J. Physique Colloq. C1, 46, 287-301 (1985)

3. U. BRINKMANN, H. BUSSE, K. PIPPERT, H. H. TELLE,
Appl. Phys. , 18, 249-255 (1979)
4. B. SAYER, M. FERRAY, J. P. VISTICOT, J. LOZINGOT,
J. Phys. B: Atom. Molec. Phys., 13, 177-191 (1980)
5. J. SZUDY, W. E. BAYLIS, J. Quant. Spectrosc. Radiat.
Transfer, 15, 64-73 (1975)
6. W. R. ANDERSON, B. M. WILSON, R. C. ORMEROD, T. L.
ROSE, J. Chem. Phys., 74, 3295-3306 (1981)