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**Abstract** High Gradient Magnetic Separation of small (5-38  $\mu\text{m}$ ) weakly magnetic copper mineral particles from a copper concentrate and ore has been performed. In previous work coarser fractions of these minerals, bornite and chalcopyrite, were separated successfully. The recovery of the smaller particles in the magnetic fraction decreases but their grade increases compared to the results obtained on the larger particles. At a magnetic background field of 1.3 T the concentrate was upgraded from 72% bornite and chalcopyrite to 86% with a recovery of 82% and the ore from 16% magnetic minerals to 44% with a recovery of 72%.

### INTRODUCTION

In Zambia there is a growing interest in the search of new concentration techniques of copper minerals from poor ores and tailings. The most important copper minerals, bornite and chalcopyrite, are both weakly magnetic. Magnetic separation has already been performed on two particle size fractions i.e. 53-75  $\mu\text{m}$  and 38-53  $\mu\text{m}$  of various copper samples<sup>1</sup>. Presently magnetic separation was applied to a smaller particle size fraction i.e. 5-38  $\mu\text{m}$  of a copper concentrate and ore sample. The minerals that are present in these samples are listed in Table 1. In addition to their chemical composition and average magnetic mass susceptibility their weight distribution, determined by ore microscopy, in both samples is also given.

TABLE 1

Minerals	Chemical Composition	$\chi \cdot 10^6$ $\text{m}^3 \cdot \text{kg}$	Concentrate w%	Ore w%
Chalcopyrite	$\text{CuFeS}_2$	0.08	55	5
Bornite	$\text{Cu}_5\text{FeS}_4$	0.15	17	11
Pyrite	$\text{FeS}_2$	0.02	< 3	1
Carrollite	$\text{Co}_2\text{CuS}_4$	0.03	2	1
Chalcocite	$\text{Cu}_2\text{S}$	-0.002		2
Strongly magnetic particles <sup>1</sup>			< 1	3
Gangue (mainly quartz)	$\text{SiO}_2$	-0.006	23	77
Copper			18	4
Iron			13	
Cobalt			0.5	1.3
$\chi \cdot 10^6$ $\text{m}^3/\text{kg}$			0.8	0.7

The second part of Table 1 contains the copper, iron and cobalt content measured by atomic absorption spectroscopy and the magnetic susceptibility measured with a mutual inductance bridge. The major contribution to the susceptibility of the samples arises from the strongly magnetic particles, which contain mainly pyrrhotite and some hematite and magnetite.

#### MAGNETIC SEPARATION

The magnetic separation was performed with an in-house built 1.3 T transverse High Gradient Magnetic Separator. An oblong canister 23x32x104mm filled with 5% steelwool of filament diameter 50-100  $\mu\text{m}$  (supplied by Frantz Co.) was used. Only 55% of the filter volume was within the field of the circular poles of a horse-shoe magnet, so that the effective volume was 42 ml.

The samples were prepared from wet sieved fractions with particles smaller than 38  $\mu\text{m}$ . Particles smaller than 5-10  $\mu\text{m}$ , depending on their specific gravity, were removed by elutriation.

In the magnetic separation process the slurry was fed downwards into the filter with an average flow velocity of 70 mm/s. Trapped nonmagnetic particles were flushed out with a flow velocity twice as high with the magnetic field still on. Magnetic particles were flushed out with much higher velocities at zero field. In both cases the flushing was done in both upward and downward directions until the effluent was clear.

Nonmagnetic particles that are recovered in the magnetic fraction arise from clogging and mechanical capture (together called mechanical trapping later on) of these particles in the matrix. This mechanical trapping was determined at zero field for different particle size fractions of the concentrate by measuring the amount of material that remained in the filter after the first flushing step. The results given in Table 2 show that the amount of mechanical trapping decreases for smaller particles. As the estimated size of the voids in the matrix is 250-500  $\mu\text{m}$  it can therefore be assumed that the mechanical trapping of coarser particles must be mainly due to clogging.

TABLE 2

Particle size fraction in $\mu\text{m}$	Mechanical trapping in %
53-75	8
38-53	6
5-38	3

The filter was saturated when about 3 g of material was trapped under the used separation conditions. Therefore magnetic separation was applied to batches of 5 g, which were mixed with 250 ml water.

In order to remove strongly magnetic particles before the maximum magnetic field was applied, magnetic separation was first performed at a lower fieldstrength of 0.5 T. Thereafter each bath was treated twice at 1.3 T.

### RESULTS

The first magnetic separation was applied to a 5-38  $\mu\text{m}$  fraction of the copper concentrate. Table 3 shows the obtained results. The reported grades and recoveries were determined from incident light microscopic analysis. This technique is less accurate for the size fraction considered due to the wide range of particle sizes. Further, the amount of gangue is undervalued. The grade and recovery of the copper and iron were obtained by atomic absorption spectroscopy. The grade and recovery of the weakly magnetic minerals, bornite and chalcopyrite, in the mags are 86% and 82% respectively, which is good. Also the recovery of the copper and iron in the mags is good in spite of its presence in the weaker ('non') magnetic minerals in the non-mags.

TABLE 3 Results Concentration

Minerals	MAGS 58% of feed		NON MAGS 42% of feed	
	Grade%	Recovery %	Grade %	Recovery %
Chalcopyrite	63	80	22	20
Bornite	23	90	4	10
Pyrite+Carrollite	4	40	8	60
Strong mags	<1	100		
Gangue	10	17	66	83
Copper	27	88	5	12
Iron	19	86	4	14

A comparison of these results with those obtained for coarser particles<sup>1</sup> is given in Table 4. The stated improvement (reduction) of the grade is the relative increase (decrease) of the grade in the (non)-mags compared to the feed. In case of the coarser particles the nonmags were retreated five times starting from batches of 15 g per 500 ml water. Here the nonmags were re-

treated only twice starting from batches of 5 g per 250 ml. It can be seen that, although the improvement of the grade of the magnetic minerals remains the same, their recovery decreases slightly for smaller particles. Nevertheless there is a considerable decrease of their grade in the nonmags. This leads to a higher grade and recovery of the copper in the mags and a lower grade in the nonmags. The reason for this is the reduction of mechanical trapping, which is clearly demonstrated by the increase of the grade and recovery of the non-magnetic minerals and especially gangue in the nonmags.

TABLE 4

	Improvement		Recovery in %			
	Grade in %					
	Particle Size Fractions in $\mu\text{m}$					
	53-75	38-53	5-38	53-75	38-53	5-38
Magn. Mineral in Mags	20	20	20	90	87	82
Copper in Mags	5	15	50	85	85	88
Non Magn. Min. in Non Mags	25	0	60	27	40	60
Gangue in Non Mags	65	80	185	44	57	83
	Reduction Grade in %					
Magn. Min. in Non Mags	25	40	180			
Copper in Non Mags	35	40	70			

TABLE 5 Results Copper Ore

Minerals	MAGS 25% of feed		NON MAGS 75% of feed	
	Grade %	Recovery %	Grade %	Recovery %
Chalcopyrite	12	58	3	42
Bornite	32	79	3	21
Pyrite + Carrollite + Calcocite	3	17	5	83
Strong Mags	5	100		
Gangue	49	16	89	84
Copper	9	58	2	42
Iron	7	59	2	41

The results obtained on the copper ore are presented in Table 5. The recovery of bornite and chalcopyrite in the mags (together 72%) is reasonable. Although their grade in the mags is not very high, the increase of the grade (175%) is quite good. The copper recovery is low because about half the copper in the non-mags is present in the high copper bearing (80% Cu) and diamagnetic chalcocite (4% in the nonmags). No comparison is made with the coarser particle size fraction, because there is too much inter-growth in those fractions.

### CONCLUSIONS

Even at a magnetic background field of 1.3T an effective magnetic separation of small bornite and chalcopyrite particles (5-38  $\mu\text{m}$ ) can be achieved. For larger particles the performance of the magnetic separation can be improved by the use of less dense or coarser matrices. This was also demonstrated by El Tawil et al<sup>2</sup>. The number of times the nonmags have to be retreated in order to increase the recovery of especially chalcopyrite can be reduced by avoiding the feeding of a saturated filter. This will also lead to a higher grade in the mags. The concentration of bornite and chalcopyrite from low grade feeds by High Gradient Magnetic Separation (HGMS) will improve by applying HGMS at higher magnetic fields ( $\approx 2$  T) and retreatment of the mags at lower fields.

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