

***Program to calculate standard free energy change,
equilibrium constant and equilibrium ozone pressure of
SiC oxidation reactions with ozone (O3) at temperatures
from 2000 to 3200 K***

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*** Constants***

R = 8.3143 / 1000;

co2d = 0;

cod = 0;

siod = 0;

- *** Input data from thermodynamic tables : Temperatures and corresponding standard free energies of formation (dg) of the compounds***

(* SiC *)

temp3 = {2000, 2100, 2200, 2300, 2400, 2500, 2600, 2700, 2800, 2900, 3000, 3100, 3200};

dg_{SiC3} = {-48.450, -44.738, -41.034, -37.338, -33.649,
-29.967, -26.294, -22.627, -18.967, -15.315, -11.669, -8.030, -4.397};

dg_{SiO23} = {-551.112, -531.950, -512.900, -493.93, -475.047, -456.253,
-437.529, -418.894, -400.321, -381.807, -363.376, -344.996, -326.678};

dg_{CO23} = {-396.442, -396.417, -396.409, -396.371, -396.325, -396.283,
-396.216, -396.154, -396.066, -395.97, -395.869, -395.752, -395.639};

dg_{SiO3} = {-256.651, -261.228, -265.776, -270.278, -274.726, -279.161,
-283.541, -287.888, -292.206, -296.482, -300.738, -304.955, -309.152};

dg_{CO3} = {-285.989, -294.328, -302.65, -310.917, -319.164, -327.385,
-335.565, -343.728, -351.845, -359.941, -368.012, -376.05, -384.07};

```

dgcgr3 = 0;

dgsi3 = 0;

caldgo33 = {66.680, 68.272, 69.861, 71.449, 73.039,
  74.623, 76.211, 77.797, 79.383, 80.971, 82.557, 84.145, 85.731};

dgo33 = 4.184 * caldgo33;

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(* SiC+ 4/3 O3-----> SiO₂ (liq)+CO₂ *)

■ * Standard free energy change of reaction 13 (G13)

```
G13 = (dgsio23 + dgco23 - dgsic3 - 4 / 3 * dgo33) ;
```

■ * Temperature - Equilibrium oxygen activity list for reaction '13' (eqbs13)

```

a13 = (3 * G13 / (4 * R * temp3 * 2.303)) + 3 / 4 * co2d;

ozlw013 = {{temp3[[1]], a13[[1]]}, {temp3[[2]], a13[[2]]},
  {temp3[[3]], a13[[3]]}, {temp3[[4]], a13[[4]]}, {temp3[[5]], a13[[5]]},
  {temp3[[6]], a13[[6]]}, {temp3[[7]], a13[[7]]}, {temp3[[8]], a13[[8]]},
  {temp3[[9]], a13[[9]]}, {temp3[[10]], a13[[10]]}, {temp3[[11]], a13[[11]]},
  {temp3[[12]], a13[[12]]}, {temp3[[13]], a13[[13]]}};

```

■ * Plot of Temperature - Equilibrium oxygen activity list for reaction 13*

```
ListPlot[ozlw013]
```

■ * Export data*

```
Export["ozlw013", ozlw013, "Table"];
```

■ * List of Temperature-Standard free energy change of reaction 13

```

omn13 = {{temp3[[1]], G13[[1]]}, {temp3[[2]], G13[[2]]},
  {temp3[[3]], G13[[3]]}, {temp3[[4]], G13[[4]]}, {temp3[[5]], G13[[5]]},
  {temp3[[6]], G13[[6]]}, {temp3[[7]], G13[[7]]}, {temp3[[8]], G13[[8]]},
  {temp3[[9]], G13[[9]]}, {temp3[[10]], G13[[10]]}, {temp3[[11]], G13[[11]]},
  {temp3[[12]], G13[[12]]}, {temp3[[13]], G13[[13]]}};

```

■ * Plot of Temperature-Standard free energy change of reaction 13*

```
ListPlot[omn13]
```

■ * Export list of Temperature-Standard free energy change of reaction 13*

```
Export["omn13", omn13, "Table"];
```

[illegible]

■ * Standard free energy change of reaction 23 (G23)

$$G_{23} = (dg_{\text{SiO}_3} + dg_{\text{CO}_3} - dg_{\text{SiC}_3} - 2 / 3 * dg_{\text{O}_{33}}) ;$$

■ *** Equilibrium oxygen activity for reaction 23 (a23)**

```
a23 = (3 * G23 / (2 * R * temp3 * 2.303)) + 3 / 2 * cod + 3 / 2 * siod;
```

■ * Temperature - Equilibrium oxygen activity list for reaction 23 (eqbs23)

```
ozlw023 = {{temp3[[1]], a23[[1]]}, {temp3[[2]], a23[[2]]},
  {temp3[[3]], a23[[3]]}, {temp3[[4]], a23[[4]]}, {temp3[[5]], a23[[5]]},
  {temp3[[6]], a23[[6]]}, {temp3[[7]], a23[[7]]}, {temp3[[8]], a23[[8]]},
  {temp3[[9]], a23[[9]]}, {temp3[[10]], a23[[10]]}, {temp3[[11]], a23[[11]]},
  {temp3[[12]], a23[[12]]}, {temp3[[13]], a23[[13]]}};
```

■ * Plot of Temperature-Standard free energy change of reaction 23*

```
ListPlot[ozlw023]
```

■ *** Export data***

```
Export["ozlw023", ozlw023, "Table"];
```

■ * List of Temperature-Standard free energy change of reaction 23

```
omn23 = {{temp3[[1]], G23[[1]]}, {temp3[[2]], G23[[2]]},
  {temp3[[3]], G23[[3]]}, {temp3[[4]], G23[[4]]}, {temp3[[5]], G23[[5]]},
  {temp3[[6]], G23[[6]]}, {temp3[[7]], G23[[7]]}, {temp3[[8]], G23[[8]]},
  {temp3[[9]], G23[[9]]}, {temp3[[10]], G23[[10]]}, {temp3[[11]], G23[[11]]},
  {temp3[[12]], G23[[12]]}, {temp3[[13]], G23[[13]]}};
```

■ * Plot of Temperature-Standard free energy change of reaction 2*

```
ListPlot[omn23]
```


■ *** Export list of Temperature-Standard free energy change of reaction 43***

```
Export["omn43", omn43, "Table"];
```

```

XXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXX
XXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXX

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```
53 (* SiC+ O3-----> SiO + CO2 *) ;
```

■ *** Standard free energy change of reaction 53 (G53)**

```
g53 = (dgsio3 + dgco23 - dgsic3 - dgo33) ;
```

■ *** Equilibrium oxygen activity for reaction 53 (a53)**

```
a53 = (G53) / (R * temp3 * 2.303) + co2d + siod;
```

■ *** Temperature - Equilibrium oxygen activity list for reaction 53 (ozlw53)**

```
ozlw53 = {{temp3[[1]], a53[[1]]}, {temp3[[2]], a53[[2]]},
  {temp3[[3]], a53[[3]]}, {temp3[[4]], a53[[4]]}, {temp3[[5]], a53[[5]]},
  {temp3[[6]], a53[[6]]}, {temp3[[7]], a53[[7]]}, {temp3[[8]], a53[[8]]},
  {temp3[[9]], a53[[9]]}, {temp3[[10]], a53[[10]]}, {temp3[[11]], a53[[11]]},
  {temp3[[12]], a53[[12]]}, {temp3[[13]], a53[[13]]}};
```

■ *** Plot of Temperature - Equilibrium oxygen activity list for reaction 53***

```
ListPlot[ozlw53]
```

■ *** Export data***

```
Export["ozlw53", ozlw53, "Table"];
```

■ *** List of Temperature-Standard free energy change of reaction 53**

```
omn53 = {{temp3[[1]], G53[[1]]}, {temp3[[2]], G53[[2]]},
  {temp3[[3]], G53[[3]]}, {temp3[[4]], G53[[4]]}, {temp3[[5]], G53[[5]]},
  {temp3[[6]], G53[[6]]}, {temp3[[7]], G53[[7]]}, {temp3[[8]], G53[[8]]},
  {temp3[[9]], G53[[9]]}, {temp3[[10]], G53[[10]]}, {temp3[[11]], G53[[11]]},
  {temp3[[12]], G53[[12]]}, {temp3[[13]], G53[[13]]}};
```

■ *** Plot of Temperature-Standard free energy change of reaction 53***

```
ListPlot[omn53]
```


