Research Article

Thermal Diffusion and Mass Transfer Effects on MHD Flow of a Dusty Gas through Porous Medium

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The present problem is concerned with the thermal diffusion mass transfer effects on MHD free convective flow of dusty gas through a porous medium induced by the motion of a semi-infinite flat plate moving with velocity decreasing “exponentially with time”. The effects of various parameters like magnetic parameter M thermal diffusion effect as Soret number S_1 , permeability parameter K_1 , Schmidt number S_c are taken into account. The velocity profile, temperature field, and concentration of incompressible dusty gas and dust particles for several parameters are discussed numerically and explained graphically.

1. Introduction

The thermal diffusion (commonly known as Soret effect) for instance, has been utilized for isotope separation, and in mixtures between gases with very light-molecular-weight [H_2 , He] and the medium-molecular-weight [N_2 , air] the diffusion thermoeffect was found to be of a magnitude such that it cannot be neglected, Eckert and Darke [1]. In view of the importance of this diffusion thermoeffect, recently Basant Kumar and Singh [2] studied the free convection and mass transfer flow in an infinite vertical plate moving impulsively in its own plane, taking into account the Soret effect.

The problems of fluid mechanics involving gas particles mixture arise in many processes of practical importance. One of the earliest problems is that of the heat and mass transfer in the mist-flow region of a boiler tube. The liquid rocket is another example, usually the oxidizer vaporizes much more rapidly than the fuel spray and combustion occurs. Heterogeneously around each droplet, the length of the combustion chamber and the stability of

the flow of acoustic or shock waves are practical two-phase flow problems. The study of the flow of dusty gases, which has gained increased attention recently has wide applications in environmental sciences; one finds in the literature an amazing number of derivations of equations for the flow of a gas particle mixture. The equations have been developed by the several authors for various special problem. Under various assumptions a few derivations primarily for the gas particle mixture are listed here, Saffman [3], Marble [4], and Soo [5]. Using the formulation of Saffman [3] several authors have given exact solutions of various dusty gas problems, Michael and Norey [6], Rao [7] Verma and Mathur [8], Singh [9], Rukman-gadachari [10], and Mitra [11] Studied the problem of circular cylinders under various condition. Gupta [12] considered the unsteady flow of a dusty gas in a channel whose cross-section is an annular sector regarding the plate problems. Liu [13], Michael and Miller [14], Liu [15], and Vimal [16] studied the problems of infinite flat plate under various conditions. Mitra [17] has studied the flow of a dusty gas induced by the motion of a semi-infinite flat plate moving with velocity decreasing exponentially with time. Singh [18] has studied MHD flow of a dusty gas through a porous medium induced by the motion or a semi-infinite flat plate moving with velocity decreasing exponentially with time. Singh and Gupta [19] have discussed MHD free convective flow of a dusty gas through a porous medium induced by the motion of a semi-infinite flat plate moving with velocity decreasing exponentially with time. Recently, Singh and Varshney have seen the mass transfer effects on study of Singh and Gupta. In the present section, we are considering the problem of Singh and Varshney [20] taking thermal diffusion into account under the same conditions taken by Singh and Vaershney.

2. Mathematical Formulation of the Problem and Its Solution

We assume the dusty gas to be confined in the space $y > 0$ and the flow is produced by the motion of the semi-infinite flat plate moving with velocity $ve^{-\lambda^2 t}$ in x direction, x -axis taken along the plate and y -axis to be measured normal to it. Since the plate is semi-infinite, all the physical quantities will be functions of y and t only. According to Saffman [3] the equation of motion of the dusty gas and the dust particles along the x -axis are, respectively, given by

$$\frac{\partial u}{\partial t} = \nu \frac{\partial^2 y}{\partial y^2} + \frac{K_0 N_0}{\rho} (v - u), \quad (2.1)$$

$$\frac{\partial v}{\partial t} = \frac{K_0}{m} (u - v), \quad (2.2)$$

$$\frac{\partial T}{\partial t} = \frac{K_T}{\rho C_P} \frac{\partial^2 T}{\partial y^2}, \quad (2.3)$$

$$\frac{\partial C}{\partial t} = D \frac{\partial^2 C}{\partial y^2} + D_T \frac{\partial^2 T}{\partial y^2}, \quad (2.4)$$

where u and v denote the velocity of gas and dust particles, respectively, ν is the kinematic coefficient of viscosity of the gas, K_0 is the stokes resistance coefficient, N_0 is the number density of the dust particles which is taken to be constant, ρ is the density of the gas, m is the mass of

dust particle, K_T is the thermal conductivity, C_P is the specific heat at the constant pressure, D is the molecular diffusivity, and D_T is the thermal diffusivity.

Applying the magnetic field, porous medium, free convection, mass transfer, and thermal diffusion along the x -axis the equation of motion (2.1) reduces to

$$\frac{\partial u}{\partial t} = \nu \frac{\partial^2 u}{\partial y^2} + \frac{K_0 N_0}{\rho} (v - u) - \frac{\sigma B_0^2 u}{\rho} - \frac{\nu}{K} u + g\beta\theta + g\beta'\phi, \quad (2.5)$$

where

$$\theta = T - T_\infty, \quad \phi = C - C_\infty \quad (2.6)$$

The boundary conditions are

$$\begin{aligned} \theta &= \nu e^{-\lambda^2 t}, & \phi &= \nu e^{-\lambda^2 t}, & u &= \nu e^{-\lambda^2 t}, \\ \text{At } y &= 0, & \theta &= 0, & \phi &= 0 \quad \text{as } y \rightarrow \infty. \end{aligned} \quad (2.7)$$

Let the nondimensionals introduced be

$$\begin{aligned} y^* &= \frac{y}{(\gamma t)^{1/2}}, & u^* &= \frac{u}{\nu}, & v^* &= \frac{v}{\nu}, & t^* &= \frac{t}{\tau}, \\ \tau &= \frac{m}{K_0}, & \theta^* &= \frac{\theta}{\nu}, & \phi^* &= \frac{\phi}{\nu}. \end{aligned} \quad (2.8)$$

On applying nondimensionals the dimensionless forms of (2.5), (2.2), (2.3), and (2.4) are, respectively,

$$\frac{\partial u}{\partial t} = \frac{\partial^2 u}{\partial y^2} + f(v - u) - Mu - \frac{1}{K_1} u + \beta_1 \theta + \beta_2 \phi, \quad (2.9)$$

$$\frac{\partial v}{\partial t} = (u - v), \quad (2.10)$$

$$\frac{\partial \theta}{\partial t} = \frac{1}{P_r} \frac{\partial^2 \theta}{\partial y^2}, \quad (2.11)$$

$$\frac{\partial \phi}{\partial t} = \frac{1}{S_c} \frac{\partial^2 \phi}{\partial y^2} + \frac{1}{S_1} \frac{\partial^2 \theta}{\partial y^2}, \quad (2.12)$$

where f is the mass of concentration of dust particles, M is the magnetic parameter, β_1 volumetric expansion parameter, β_2 is the mass expansion parameter, S_c is the Schmidt number, P_r

is the prandtl number, K_1 is the permeability parameter, and S_1 is the thermal diffusion parameter as soret number

$$\begin{aligned} f &= \frac{mN_0}{\rho}, & M &= \frac{m\sigma B_0^2}{K_0\rho}, & \beta_1 &= g\beta\tau, & \beta_2 &= g\beta'\tau, \\ S_c &= \frac{\nu}{D}, & P_r &= \frac{\rho\nu C_p}{K\tau}, & \frac{1}{K_1} &= \frac{\nu T}{K}, & S_1 &= \frac{\nu}{D_T}. \end{aligned} \quad (2.13)$$

The boundary condition (2.7) is reduced to

$$\begin{aligned} \theta &= e^{-\lambda^2 t}, & \phi &= e^{-\lambda^2 t}, & u &= e^{-\lambda^2 t} & \text{at } y = 0, \\ \theta &\longrightarrow 0, & \phi &\longrightarrow 0, & u &\longrightarrow 0 & \text{as } y \longrightarrow \infty. \end{aligned} \quad (2.14)$$

Let us choose the solutions of (2.9), (2.10), (2.11), and (2.12), respectively, as

$$u = F(y)e^{-\lambda^2 t}, \quad (2.15)$$

$$v = G(y)e^{-\lambda^2 t}, \quad (2.16)$$

$$\theta = H(y)e^{-\lambda^2 t}, \quad (2.17)$$

$$\phi = I(y)e^{-\lambda^2 t}. \quad (2.18)$$

Using the solution the boundary condition (2.14),

$$\begin{aligned} H &= 1, & I &= 1, & F &= 1 & \text{at } y = 0, \\ H &\longrightarrow 0, & I &\longrightarrow 0, & F &\longrightarrow 0 & \text{at } y \longrightarrow \infty. \end{aligned} \quad (2.19)$$

By virtue of (2.15), (2.16), (2.17), and (2.18), (2.9), (2.10), (2.11), and (2.12) are, respectively, reduced to

$$\frac{d^2 F}{dy^2} + f \cdot G + F \left(\lambda^2 - f - M - \frac{1}{K_1} \right) = -\beta_1 H - \beta_2 I, \quad (2.20)$$

$$G(1 - \lambda^2) = F, \quad (2.21)$$

$$\frac{d^2 H}{dy^2} + \lambda^2 H P_r = 0, \quad (2.22)$$

$$\frac{d^2 I}{dy^2} + \frac{d^2 H}{dy^2} + \lambda^2 I S_1 S_c = 0. \quad (2.23)$$

Eliminating G from (2.20) and (2.21) we get

$$\frac{d^2 F}{dy^2} + \frac{fF}{1-\lambda^2} + F \left(\lambda^2 - f - M - \frac{1}{K} \right) = -\beta_1 H - \beta_2 I, \quad (2.24)$$

$$\frac{d^2 F}{dy^2} + n^2 F = -\beta_1 H - \beta_2 I, \quad (2.25)$$

where $n = [(\lambda^4 - \lambda^2(1 + f + M + K_1^{-1}) + M + K_1^{-1})/(\lambda^2 - 1)]^{1/2}$.

From (2.22), we get

$$H = e^{-isy}, \quad (2.26)$$

where $s = \lambda\sqrt{P_r}$.

From (2.23)

$$I = \left(1 - \frac{s^2}{m^2 - s^2} \right) e^{-imy} + \frac{s^2}{m^2 - s^2} e^{-is_1 y}, \quad (2.27)$$

where $s_1 = \lambda\sqrt{S_c}$.

By the boundary condition (2.19) the solution of (2.25) is obtained as

$$F = e^{-iny} + \frac{\beta_1}{n^2 - s^2} (e^{-iny} - e^{-isy}) + \frac{\beta_2}{n^2 - s_1^2} (e^{-iny} - e^{-is_1 y}). \quad (2.28)$$

Then from (2.15) we get the velocity of dusty gas as

$$u = \left[e^{-iny} + \frac{\beta_1}{n^2 - s^2} (e^{-iny} - e^{-isy}) + \frac{\beta_2}{n^2 - s_1^2} (e^{-iny} - e^{-is_1 y}) \right] e^{-\lambda^2 t}. \quad (2.29)$$

Real part of u is given by

$$u = e^{-\lambda^2 t} \cos ny + \frac{\beta_1}{n^2 - s^2} e^{-\lambda^2 t} (\cos ny - \cos sy) + \frac{\beta_2}{n^2 - s_1^2} e^{-\lambda^2 t} (\cos ny - \cos s_1 y). \quad (2.30)$$

Using (2.16), (2.21), and (2.28) the real part of velocity of dust particle v is obtained as

$$v = \frac{1}{1 - \lambda^2} \left[e^{-\lambda^2 t} \cos ny + \frac{\beta_1}{n^2 - s^2} e^{-\lambda^2 t} (\cos ny - \cos sy) + \frac{\beta_2}{n^2 - s_1^2} e^{-\lambda^2 t} (\cos ny - \cos s_1 y) \right]. \quad (2.31)$$

Using (2.26) temperature distribution is given by

$$\theta = e^{-isy} e^{-\lambda^2 t}. \quad (2.32)$$

Figure 1: Velocity profile for $\lambda = .5$, $f = .2$, $P_r = .71$, $\beta_1 = 5$, $\beta_2 = 2$, $S_1 = .38$, $S_c = .4$ at the different values of M , K_1 , S_c , and t .

Real part of θ

$$\theta = e^{-\lambda^2 t} \cos sy. \quad (2.33)$$

Using (2.27), concentration is given by

$$\phi = \left[\left(1 - \frac{s^2}{m^2 - s^2} \right) e^{-imy} + \frac{s^2}{m^2 - s^2} e^{-is_1 y} \right] e^{-\lambda^2 t}. \quad (2.34)$$

Real part of Φ is given by

$$\phi = \left[\left(1 - \frac{s^2}{m^2 - s^2} \right) \cos my + \frac{s^2}{m^2 - s^2} \cos s_1 y \right] e^{-\lambda^2 t}. \quad (2.35)$$

3. Results and Discussion

From the solid and dotted graphs of Figure 1 it is clear that velocity for dusty gas decrease with the increasing values of y and increases with the increasing values of t . And for the increasing values of λ keeping y , t , M , S_1 constant the velocity of dusty gas increases as well as the velocity of dust particles decreases. Increasing values of t increases the velocity of dusty gas as well as the velocity of dust particles. From the solid and dotted graphs of Figure 2 it is noted that the temperature profile θ decreases and concentration profile ϕ increases when time t is increases. From the solid and dotted graphs of Figure 3 it is noted that the temperature profile decreases and the concentration profile ϕ increases as thermal diffusion S_1 parameter as soret number increases at $\lambda = .5$, $f = .2$, $P_r = .71$, $\beta_1 = 5$, $\beta_2 = 2$.

Figure 2: Concentration profile $\lambda = .5$ and $S_c = .6$ and the different values of S_1 and t .

Figure 3: Concentration and temperature profile $\lambda = .5$, $S_c = .6$ and different values of S_1 and t are $S_1 = .3, .4, .5, .6$ (see Table 3).

The velocity profile for dusty gas is tabulated in Table 1 and plotted in Figure 1 having solid graphs 1 to 3 for $t = 1$ and dotted graphs 4 to 6 at $t = 3$ and $\lambda = .5$, $f = .2$, $P_r = .71$, $\beta_1 = 5$, $\beta_2 = 2$ and different values of M , K_1 , S_c , and t are as in Table 4.

4. Conclusion

We conclude our study on thermal diffusion as well as mass transfer effect on MHD free convective flow of a dusty gas through a porous medium induced by the motion of a semi-infinite flat plate moving with velocity decreasing exponentially with time as follows.

Table 1: Velocity profile of dusty fluid and dust particle at $\lambda = .5$, $f = .2$, $P_r = .71$, $\beta_1 = 5$, $\beta_2 = 2$, $S_1 = .38$, and $S_c = .4$ at the different values of M , K_1 , S_c , and t .

Y	Graph 1	Graph 2	Graph 3	Graph 4	Graph 5	Graph 6
0	0.7788	0.7788	0.7788	1.0384	1.0384	1.0384
1	-1.93407	-1.90368	-1.91017	-2.57876	-2.53823	-2.5469
2	-9.39053	-9.37132	-8.81633	-12.5207	-12.4951	-11.7551
3	-19.6764	-20.0147	-16.8816	-26.2352	-26.6863	-22.5088
4	-30.0193	-31.554	-25.2733	-40.0257	-42.072	-35.6977

Table 2: The concentration profile is tabulated in Table 2 plotted in Figure 2 having graphs 1 to 3 at $\lambda = .5$ and different values of S_1 and t are taken for velocity values of concentration at $\lambda = .5$, $S_c = .6$ and the different values of S_1 and t .

Y	Graph 1	Graph 2	Graph 3	Graph 4	Graph 5	Graph 6
0	0.7788	0.7788	0.7788	0.7788	0.7788	0.7788
1	0.54152	0.63284	0.72612	0.71732	0.68346	0.64277
2	0.00017	0.26157	0.5473	0.5426	0.42079	0.2822
3	-0.46131	-0.17084	0.20561	0.2822	0.05509	0.17695
4	-0.74208	-0.48685	-0.29205	-0.02274	-0.3241	0.57428

Table 3: As taken for velocity, values of concentration profile $\lambda = .5$, $S_c = .6$ and different values of S_1 and t are $S_1 = .3, .4, .5, .6$.

y	Graph 1	Graph 2	Graph 3	Graph 4	Graph 5	Graph 6
0	0.7788	0.7788	0.7788	0.7788	0.7788	0.7788
1	0.71732	0.68346	0.64277	0.83886	0.8716	0.91158
2	0.5426	0.42079	0.2822	1.00911	1.12606	1.26137
3	0.2822	0.05509	-0.17695	1.26137	1.47629	1.69947
4	-0.02274	-0.3241	-0.57428	1.55363	1.83056	2.06203

Table 4: Different values of M , K_1 , S_c , and t taken for various graphs.

	Graph 1	Graph 2	Graph 3	Graph 4	Graph 5	Graph 6
M	0.1	0.2	0.1	0.1	0.1	0.2
K_1	10	10	20	10	10	10
S_c	0.4	0.4	0.4	0.8	0.4	0.4
t	1	1	1	1	3	3

- (1) Increasing the viscoelastic parameter (λ) increases the velocity of dusty gas and decreases the velocity of dust particles as well as decreases the temperature profile and increases the concentration profile.
- (2) Increasing values of y decreases the velocity of dusty gas while it increases the velocity of dust particles.
- (3) Increasing values of thermal diffusion parameters as sorlet number (S_1) decreases the temperature profile (θ) and increases the concentration profile ϕ .

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