

Supplementary Information

Derivation of equations (1) and (2)

For the process: $P + H^+ \rightarrow PH^+$, the proton affinity (PA) is defined as: $PA(P) = \Delta H_f(P) + \Delta H_f(H^+) - \Delta H_f(PH^+)$

Similarly, for the process: $P + Ca^{2+} \rightarrow (P \cdots Ca^{2+})$, the calcium affinity (Ca_{aff}) is: $Ca_{aff} = \Delta H_f(P) + \Delta H_f(Ca^{2+}) - \Delta H_f(P \cdots Ca^{2+})$

1. $P_1 \cdots Ca^{2+} \cdots P_2$ dissociating to $P_1 + (P_2 \cdots Ca^{2+})$ or to $P_2 + (P_1 \cdots Ca^{2+})$

$$\Delta H_f(P_1 \cdots Ca^{2+} \cdots P_2) \rightarrow \Delta H_f P_1 + \Delta H_f P_2 + \Delta H_f Ca^{2+} - Ca_{aff}(P_2)$$

$$\Delta H_f(P_1 \cdots Ca^{2+} \cdots P_2) \rightarrow \Delta H_f P_2 + \Delta H_f P_1 + \Delta H_f Ca^{2+} - Ca_{aff}(P_1)$$

Subtract:

$$\Delta = \Delta Ca_{aff}(P)$$

2. $P_1 \cdots Ca^{2+} \cdots P_2$ dissociating to $P_1 H^+ + [(P_2-H)^- \cdots Ca^{2+}]$ or to $P_2 H^+ + [(P_2-H)^- \cdots Ca^{2+}]$

$$\Delta H_f [P_1 H^+ \cdots Ca^{2+} \cdots (P_2-H)^-] \rightarrow \Delta H_f(P_1 H^+) + \Delta H_f[(P_2-H)^- \cdots Ca^{2+}]$$

$$\Delta H_f [P_1 H^+ \cdots Ca^{2+} \cdots (P_2-H)^-] \rightarrow \Delta H_f(P_2 H^+) + \Delta H_f[(P_1-H)^- \cdots Ca^{2+}]$$

or

$$\Delta H_f [P_1 H^+ \cdots Ca^{2+} \cdots (P_2-H)^-] \rightarrow \Delta H_f P_1 + \Delta H_f H^+ - PA(P_1) + \Delta H_f (P_2-H)^- + \Delta H_f Ca^{2+} - Ca_{aff}[P_2-H]^-$$

$$\Delta H_f [P_2 H^+ \cdots Ca^{2+} \cdots (P_1-H)^-] \rightarrow \Delta H_f P_2 + \Delta H_f H^+ - PA(P_2) + \Delta H_f (P_1-H)^- + \Delta H_f Ca^{2+} - Ca_{aff}[P_1-H]^-$$

Subtract:

$$\Delta = \Delta H_f P_1 - \Delta H_f P_2 + \Delta H_f (P_2-H)^- - \Delta H_f (P_1-H)^- - PA(P_1) + PA(P_2) - Ca_{aff}[P_2-H]^- + Ca_{aff}[P_1-H]^-$$

$$\Delta = Ca_{aff}[P_1-H]^- - Ca_{aff}[P_2-H]^- - PA(P_1) - PA(P_2) + \Delta H_f P_1 - \Delta H_f P_2 + \Delta H_f [P_2-H]^- - \Delta H_f [P_1-H]^- \dots \dots \dots \text{(a)}$$

Also:

$$\Delta H_f P_1 = \Delta H_f (P_1-H)^- + \Delta H_f H^+ - PA [P_1-H]^-$$

$$\Delta H_f P_2 = \Delta H_f (P_2-H)^- + \Delta H_f H^+ - PA (P_2-H)^-$$

Subtract:

$$\Delta H_f P_1 - \Delta H_f P_2 = - (PA (P_1-H)^- - PA (P_2-H)^-) + \Delta H_f (P_1-H)^- - \Delta H_f (P_2-H)^- \quad \text{(b)}$$

Put (b) in (a):

$$\Delta = Ca_{aff}(P_1-H)^- - Ca_{aff}(P_2-H)^- - (PA(P_1) - PA(P_2)) - (PA(P_1-H)^- - PA(P_2-H)^-) + \Delta H_f [P_2-H]^- - \Delta H_f [P_1-H]^- - \Delta H_f (P_1-H)^- + \Delta H_f (P_2-H)^-$$

$$\Delta = \Delta C_{\text{aff}} [\text{P-H}]^- - \Delta \text{PA}(\text{P}) - \Delta \text{PA} (\text{P-H})^- \quad (1)$$

3. $(\text{P}_1\text{-2H})^{2-} \cdots \text{Ca}^{2+} \cdots (\text{P}_2\text{-2H})^{2-}$ dissociating to $[(\text{P}_1\text{-3H})^{3-} + \text{Ca}^{2+}] + (\text{P}_2\text{-H})^-$ or to $[(\text{P}_2\text{-3H})^{3-} + \text{Ca}^{2+}] + (\text{P}_1\text{-H})^-$

$$\Delta H_f [(\text{P}_1\text{-H})^- \cdots \text{Ca}^{2+} \cdots (\text{P}_2\text{-3H})^{3-}] \rightarrow \Delta H_f(\text{P}_1\text{-2H})^{2-} + \Delta H_f \text{H}^+ - \text{PA}(\text{P}_1\text{-2H})^{2-} + \Delta H_f(\text{P}_2\text{-3H})^{3-} + \Delta H_f \text{Ca}^{2+} - \text{Ca}_{\text{aff}} (\text{P}_2\text{-3H})^{3-}$$

$$\Delta H_f [(\text{P}_2\text{-H})^- \cdots \text{Ca}^{2+} \cdots (\text{P}_1\text{-3H})^{3-}] \rightarrow \Delta H_f (\text{P}_2\text{-2H})^{2-} + \Delta H_f \text{H}^+ - \text{PA}(\text{P}_2\text{-2H})^{2-} + \Delta H_f (\text{P}_1\text{-3H})^{3-} + \Delta H_f \text{Ca}^{2+} - \text{Ca}_{\text{aff}} (\text{P}_1\text{-3H})^{3-}$$

Subtract

$$\Delta = \Delta H_f (\text{P}_1\text{-2H})^{2-} - \Delta H_f (\text{P}_2\text{-2H})^{2-} - \text{PA}(\text{P}_1\text{-2H})^{2-} + \text{PA} (\text{P}_2\text{-2H})^{2-} + \Delta H_f (\text{P}_2\text{-3H})^{3-} - \Delta H_f(\text{P}_1\text{-3H})^{3-} - \text{Ca}_{\text{aff}} [\text{P}_2\text{-3H}]^{3-} + \text{Ca}_{\text{aff}} [\text{P}_1\text{-3H}]^{3-}$$

$$\Delta = \text{Ca}_{\text{aff}}[\text{P}_1\text{-3H}]^{3-} - \text{Ca}_{\text{aff}}[\text{P}_2\text{-3H}]^{3-} + \Delta H_f (\text{P}_1\text{-2H})^{2-} - \Delta H_f (\text{P}_2\text{-2H})^{2-} + \Delta H_f (\text{P}_2\text{-3H})^{3-} - \Delta H_f (\text{P}_1\text{-3H})^{3-} \quad (c)$$

Also:

$$\Delta H_f (\text{P}_1\text{-2H})^{2-} = \Delta H_f (\text{P}_1\text{-3H})^{3-} + \Delta H_f \text{H}^+ - \text{PA} [\text{P}_1\text{-3H}]^{3-}$$

$$\Delta H_f (\text{P}_2\text{-2H})^{2-} = \Delta H_f (\text{P}_2\text{-3H})^{3-} + \Delta H_f \text{H}^+ - \text{PA}(\text{P}_2\text{-3H})^{3-}$$

Subtract:

$$\Delta H_f (\text{P}_1\text{-2H})^{2-} - \Delta H_f (\text{P}_2\text{-2H})^{2-} = \Delta H_f (\text{P}_1\text{-3H})^{3-} - \Delta H_f (\text{P}_2\text{-3H})^{3-} - \text{PA}[\text{P}_1\text{-3H}]^{3-} + \text{PA}[\text{P}_2\text{-3H}]^{3-} \quad (d)$$

Put (d) in (c):

$$\Delta = \text{Ca}_{\text{aff}}[\text{P}_1\text{-3H}]^{3-} - \text{Ca}_{\text{aff}}[\text{P}_2\text{-3H}]^{3-} - (\text{PA}(\text{P}_1\text{-2H})^{2-} - \text{PA} (\text{P}_2\text{-2H})^{2-}) + \Delta H_f (\text{P}_1\text{-3H})^{3-} - \Delta H_f (\text{P}_2\text{-3H})^{3-} - \text{PA}[\text{P}_1\text{-3H}]^{3-} + \text{PA}[\text{P}_2\text{-3H}]^{3-} + \Delta H_f (\text{P}_2\text{-3H})^{3-} - \Delta H_f (\text{P}_1\text{-3H})^{3-}$$

$$\Delta = \text{Ca}_{\text{aff}}[\text{P}_1\text{-3H}]^{3-} - \text{Ca}_{\text{aff}}[\text{P}_2\text{-3H}]^{3-} - (\text{PA}(\text{P}_1\text{-2H})^{2-} - \text{PA}(\text{P}_2\text{-2H})^{2-}) - (\text{PA}(\text{P}_1\text{-3H})^{3-} - \text{PA}(\text{P}_2\text{-3H})^{3-})$$

$$\Delta = \Delta C_{\text{aff}}(\text{P-3H})^{3-} - \Delta \text{PA}(\text{P-2H})^{2-} - \Delta \text{PA}(\text{P-3H})^{3-} \quad (2)$$