

Research Article

Microstructure and Microhardness Evolutions of High Fe Containing Near-Eutectic Al-Si Rapidly Solidified Alloy

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Al-11 wt.% Si-11 wt.% Fe (11.29 at.% Si-5.6 at.% Fe) melt was rapidly solidified into ribbons and characterized by X-ray diffraction (XRD), scanning electron microscopy (SEM), energy dispersive X-ray spectroscopy (EDS), and microhardness technique. The Rietveld X-ray diffraction analysis was applied successfully to analyze microstructure and phase precipitations. On the basis of the aluminum peak shifts measured in the XRD scans, a solid solubility extension value of 1 at.% Si in α -Al was determined. SEM investigations confirmed presence of a spherical shape α -phase particles in addition to needle and spherical shape β -phase particles with contents of 1.1 wt.% and 10.1 wt.% as deduced by XRD analysis. During prolonged annealing process at 350°C/25 h, α -phase disappeared, β -phase content increased to 30 wt.%, and Si presence becomes more evident as deduced by XRD analysis. EDS analysis confirmed that these β particles observed in the as-melt spun alloy are of lower Fe content comparing to those usually observed in the as-cast counter-part alloy. Besides, the length distribution of needle shape β -particles has been shortened to be diverse from 1 to 5 μ m. The as-melt spun ribbons exhibited enhancement of hardness to 277 HV and further increased during heat treatment (150°C/12 h) to 450 HV. This improvement of microstructure and hardness are the influence of microstructural refinement and modification obtained during the rapid solidification process.

1. Introduction

Rapid solidification involves cooling of metallic melts at rates $>10^4$ K/s and results in significant microstructural and constitutional changes. The microstructural modifications include grain refinement and reduced segregation effects while the constitutional changes include formation of supersaturated solid solutions and metastable crystalline intermediate and amorphous phases. These effects, either alone or in combination, have improved the mechanical behavior and performance of the rapidly solidified alloys and these results were especially significant for lightweight metals and have been well documented in the literature [1–5]. Eutectic Al-Si alloys are a wide range of useful materials in different fields of industry where high strength to weight ratio, castability, and wear resistance are required [6–10]. The importance of Al-Si and Al-Si-X alloys for automotive applications with some engine parts like connecting rod, cylinder sleeve, piston and

valve retainer, and compressor parts like rotary compressor vane and shoe disc has been well established [11].

The effects of a third element addition to Al-Si alloys such as iron, copper, nickel, and other transition elements have been investigated [12–16]. These transition elements, when alloyed in various combinations with aluminum, form fine dispersions of high-modulus second-phase particles, resulting in increased strength, wear resistance, and thermal stability [17].

Due to the high liquid solubility and the low solid solubility of Fe in Al-Si alloys, Fe has various ways to come into the molten of these alloys and promotes to form various Fe-rich intermetallic phases such as $\text{Al}_8\text{Fe}_2\text{Si}$ (α) and Al_5FeSi (β) [18–20]. In addition, this feature ensures its high chemical homogeneity in the α -Al and its thermal stability. The solubility of Fe in Al can be extended up to 0.6 wt.% by rapid solidification [21]. Therefore, it is expected that the Fe-containing phases may be formed during postsolidification

process including hot forming. However, it is well known that β -phase, formed in as cast aluminum alloys when Fe is present, is a brittle intermetallic phase and regarded as the most harmful element degrading mechanical properties of these alloys [20, 22–27]. Moreover, it is found that the dimension of the β -phase crystals increases with increasing of Fe content and decreasing of cooling rate [28]. Suppression of the β -phase can occur by increasing the cooling rates to typical for rapid solidification processes (10^6 K/s) [28]. The ternary Al-Si-Fe has been extensively studied using rapid solidification techniques [15, 16], including melt spinning [12]. The present investigation aims to study the microstructural characterization of high Fe containing near-eutectic Al-Si rapidly solidified alloy, namely, Al-11 wt.% Si-11 wt.% Fe. For this purpose, X-ray diffraction (XRD), scanning electron microscopy (SEM), and energy dispersive X-ray spectrometer (EDS) techniques were carried out on both alloys. Hardness of the ribbons was also measured.

2. Experimental Details

2.1. Material Preparation. A ternary near-eutectic high Fe containing Al-11 wt.% Si-11 wt.% Fe alloy was prepared from 99.8 wt.% pure Al, 99.75 wt.% pure Fe, and Al-10 wt.% Si master alloy. The ingots were melting in a muffle furnace and poured into a graphite mold after the homogenization process to produce rods of 25 mm in length and 4 mm in diameter. Long uniform ribbons of thickness $50 \mu\text{m}$ and width 2 mm were prepared by melt spinning. A stream of the molten alloy, at 950°C , was ejected by argon gas at a gauge pressure of 1.5 bars from a silica tube with a 0.4 mm orifice diameter. The melt jet fell on a copper disc of 18 cm diameter coated by chromium, rotating at 2950 rpm. The estimated cooling rate was about 10^5 K/s. The produced ribbons were fairly uniform. Deviations of 0.05% mm and $3 \mu\text{m}$ in width and thickness were observed from the whole length of ribbons.

2.2. Material Characterization. XRD patterns were performed using a 1390 Philips diffractometer with filtered $\text{Cu K}\alpha$ radiation at 40 kV and 20 mA. The X-ray samples were prepared from a short length stuck on a glass slide using Vaseline. Microstructure characterizations were performed by scanning electron microscopy. SEM investigations were carried out in a JEOL JSM-T330 scanning electron microscope operated at 25 kV and linked with an energy dispersive spectrometry (EDS) attachment. The ribbons were observed for the wheel side surface using standard metallographic techniques. Optical mount specimens were prepared for SEM investigations followed by chemical etching in a 0.5% HF solution. The annealed samples were heated at $350^\circ\text{C}/25 \text{ h}$. Hardness was measured at room temperature using the Vickers hardness Leitz Wetzlar Germany instrument with loads of 25 g. A total of 10 measurements were performed on the longitudinal section of each ribbon and the average is taken as the microhardness value. In the present study, Rietveld X-ray diffraction analysis [29] was carried out by the X'pert HighScor 2004 program and the Pseudo-Voigt peak shape function. The reliability of the refinement results was

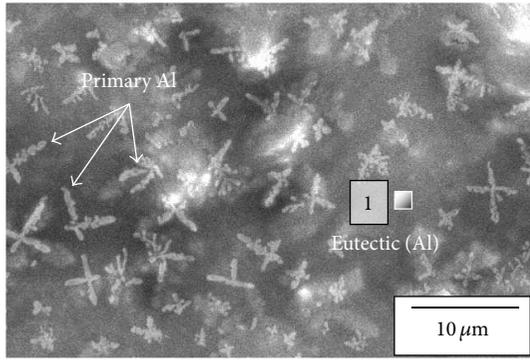
judge by the pattern R factor (R_p), the weighted pattern R factor (R_{wp}), and the “goodness of fit” ($\text{GOF} = (R_{wp}/R_{xp})^2$) [30]. Initial structure parameters of all phases used for Rietveld method in this study were from ICDD (Inorganic Crystal Structure Database) cards. The parameters that had been refined simultaneously include scale factors, zero point shift, lattice parameters, atomic coordinates, atomic sites occupancies, isotropic or anisotropic temperature factors, profile shape parameter, FWHM (full width at half maximum) parameters, asymmetry and preferred orientation parameters. The total parameters to be refined of Al-11Si-11Fe melt spun and annealed ribbons were 26 and 51, respectively. Rietveld method is becoming progressively more popular for microstructure characterization of materials. It is common practice to estimate domain size and strain values from the refined profile width parameters. Moreover, weight fractions of all phases in multiphase sample can be calculated directly by their scale factors which can be obtained by Rietveld fitting. The relationship between the fraction (W_i) for each phase i and its scale factor (S_i) determined is obtained from the following relation [30]:

$$W_i = \frac{S_i (ZMV)}{\sum_{i=1}^n S_i (ZMV)_i}, \quad (1)$$

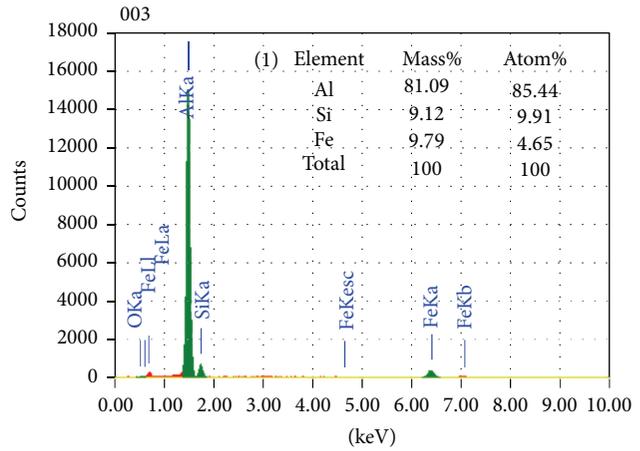
where Z , M , and V are the numbers of formula unite cell, unit molecular weight of the formula, and unit cell volume of phase i in a mixture of n phases, respectively. Weight fraction of all phases observed in this work, crystallite size of and microstrain% of α -Al have been estimated using Rietveld X-ray diffraction analysis.

3. Results and Discussion

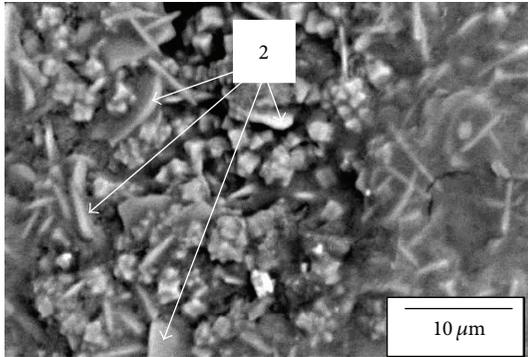
3.1. Scanning Electron Microscopy. It was reported that the intermetallic phases α and β observed in Fe-containing Al-Si alloys follow another solidification path under nonequilibrium solidification condition which leads to the formation of metastable forms of α and β with different compositions [31–35]. Rapid solidification is considered as a nonequilibrium process; therefore it is expected to obtain a metastable α and β phases. The representative microstructures of the investigated melt spun alloy are shown in Figure 1. An ultra-fine dendritic primary (Al) phase, together with a very fine eutectic (Al), is observed as shown in Figure 1(a). This eutectic Al has a large amount of Si and Fe as deduced by EDS shown in Figure 1(b). Moreover, a large number of tiny needle-shaped β -phase particles (about $0.5 \mu\text{m}$ thickness and $5 \mu\text{m}$ width) are present and distributed in the eutectic region as shown in Figure 1(c). Chemical composition of these needle shape particles has been deduced by EDS and shown in Figure 1(d). Besides, a large number of ultra-fine cubic α -phase particles are present as shown in Figure 1(e). Chemical composition of these cubic particles has been deduced by EDS and shown in Figure 1(f). The chemical compositions of intermetallic compounds (α and β phases) observed in the rapidly solidified Al-11Si-11Fe and measured by EDS are summarized in Table 1. This fine microstructure of α and β -phase can be related to the high cooling rate obtained in the rapid solidification process



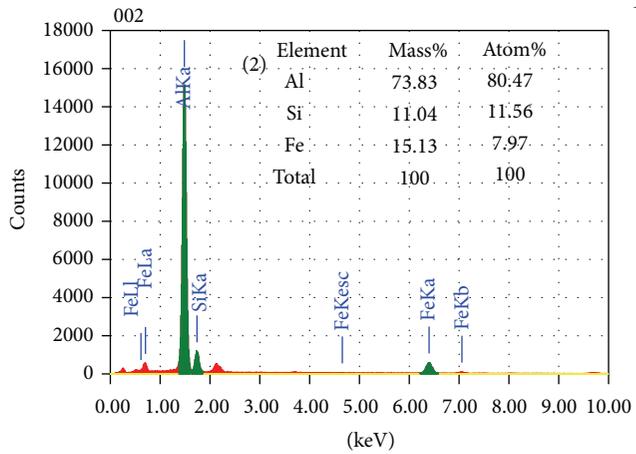
(a)



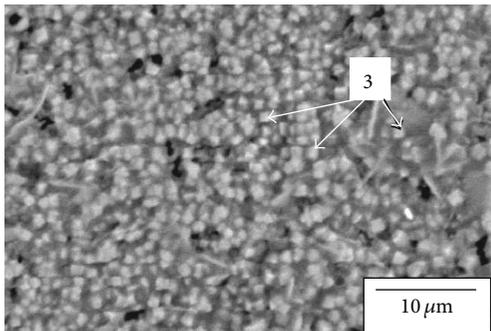
(b)



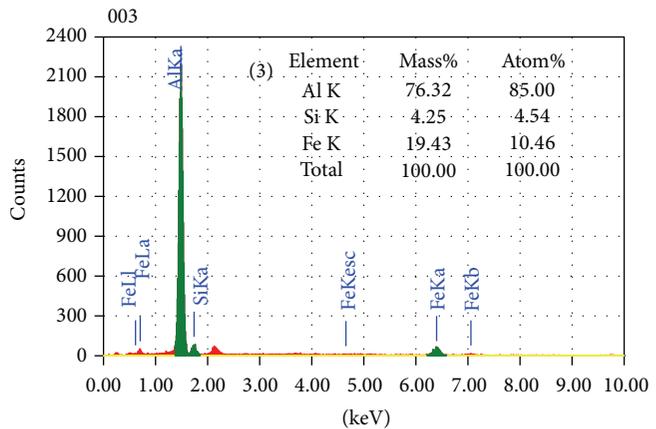
(c)



(d)



(e)



(f)

FIGURE 1: SEM micrographs of Al-11Si-11Fe as-spun samples (a), (c), and (e) and the corresponding EDS-spectra (b), (d), and (f); additional red signals are corresponding to oxygen and carbon.

which inhibits the formation of coarse phases. Moreover, β -particles grow easily straight along the lengthwise direction. It was found that the length distribution of the existing β -particles were very diverse from $1\mu\text{m}$ to $5\mu\text{m}$. All phases observed in the rapid solidified alloy at room temperature

are expected except α -phase. For the chemical composition range of this alloy, under equilibrium solidification condition, the first solid phase γ begins to crystallize at 829°C as shown in Figure 2. As the temperature drops, the primary γ phase continuous to precipitate out of the liquid. As the temperature

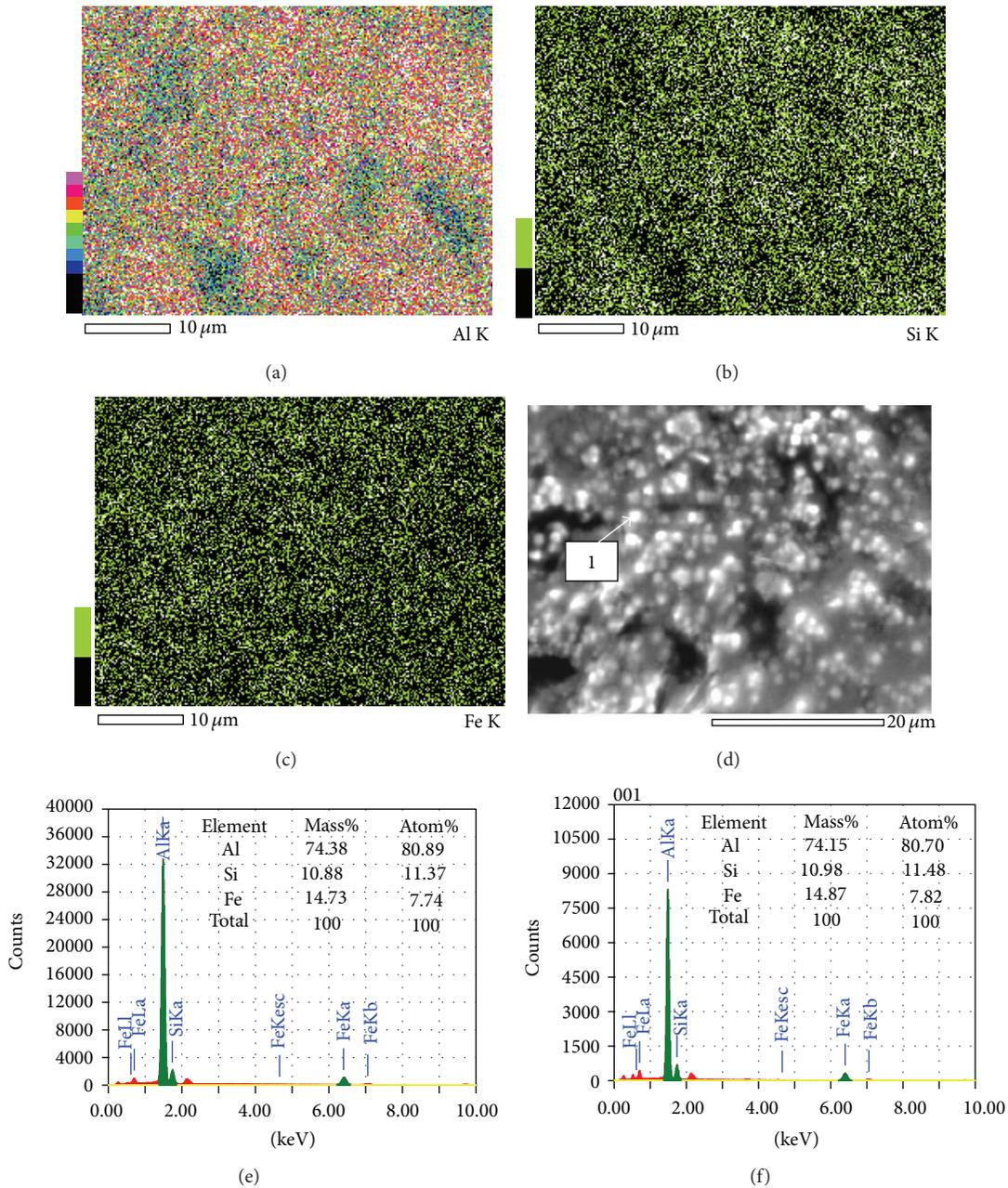


FIGURE 3: SEM micrograph of Al-11Si-11Fe as-spun alloy (d); corresponding elemental maps of Al (a), Si (b), and Fe (c); corresponding EDS-spectra (e) and EDS-spectra of point 1 (f); additional signals are corresponding to oxygen and carbon.

for 25 hours on Al-11Si-11Fe as-spun sample appeared in the increasing of the $Al_{4.5}FeSi$ and Si contents to 35 wt.% and 5.1 wt.%, respectively. No XRD lines corresponding to Fe could be detected. The estimated crystallite size and microstrain % of α -Al for the as-melt spun alloy are 27 nm and 0.126, respectively, as shown in Table 2. After prolonged annealing (350°C/25 h) the estimated crystallite size and microstrain % of α -Al become 29 nm and 0.072, respectively.

Figure 7 shows the respective DSC data for the as-melt spun Al-11Si-11Fe alloy scanned at a rate of 10°C/min between 50 and 500°C. The DSC scan exhibits a small shallow exothermic peak indicating the existence of large amount

of Si nanoparticles in as-spun state and coarsening of the Si nanoparticles occurs at 374.3°C. Similar exothermic peak corresponding to Si coarsening has been reported at 348°C for Al-7.7Si-3.3Fe rapidly solidified alloy [13].

4. Microhardness

Comparing to hardness of as cast counterpart alloy (hardness = 115 HV), a significant improvement of hardness for as-spun Al-11Si-11Fe alloy has been obtained. The as-spun alloy exhibits microhardness as high as 270 HV. Figure 8 represents

TABLE 2: Rietveld refinement results for Al-11Si-11Fe melt spun and annealed (350°C/25 h) ribbons.

Sample	Lattice parameter α -Al (nm)	Crystallite size (nm) α -Al	Strain, $\epsilon\%$	Agreement factors			
				R_p	R_{WP}	R_{ex}	GOF
Melt spun	0.404753	27	0.126	5.7	7.8	7.4	1.1
Annealed	0.40501	29	0.072	5.3	7.1	6.5	1.2

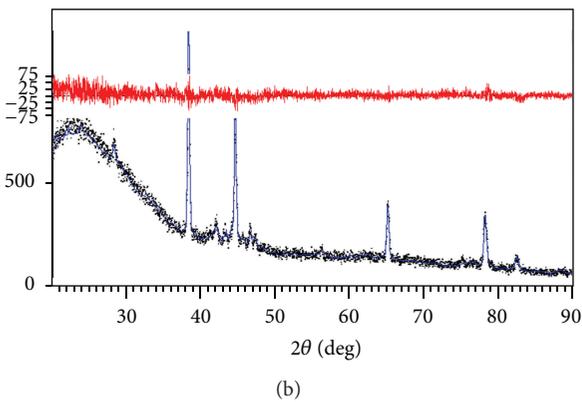
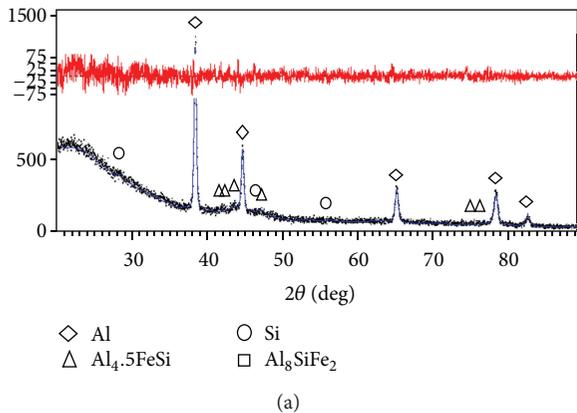


FIGURE 4: Observed profile (points), calculated profile (the solid line), and the profile difference above for the Rietveld patterns of Al-11Si-11Fe melt spun (a) and annealed at 350°C/25 h (b) alloys.

development of the microhardness for the Al-11Si-11Fe as-melt spun alloy during annealing at elevated temperature of 150°C. Further increase of microhardness up to 450 HV after 12 hours at 150°C has been obtained. After prolonged annealing (at 350°C/25 h) hardness decreases to 190 HV. This improvement of microhardness for the as-melt spun alloys can be related to the supersaturated solid solution of α -Al with Si, fine structural low Fe-containing needle and spherical shape β -phase, and ultra-fine hard Si particles. Microhardness for as-spun Al-Si base alloys is expected to decrease by increasing of annealing time even at low elevated temperature as 150°C due to the precipitation solutes from the supersaturated solid solution α -Al [40]. Instead, it increases by increasing of the annealing time at the first stage and then decreases. This increase of hardness with time during the first stage may be related to precipitation of much ultra-fine β -phase particles before Si precipitation from the supersaturated solid solution α -Al becomes effective and

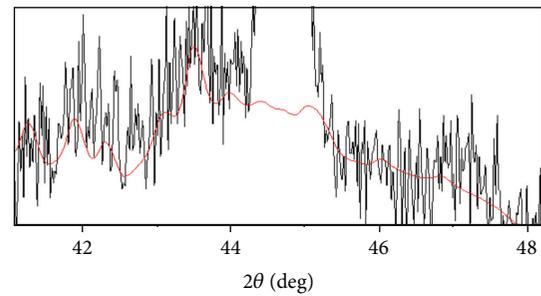


FIGURE 5: Expanded XRD pattern for as-melt spun Al-11Si-11Fe alloy showing part of the calculated profile (red line) of $Al_{4.5}SiFe$ β -phase in as-quenched state.

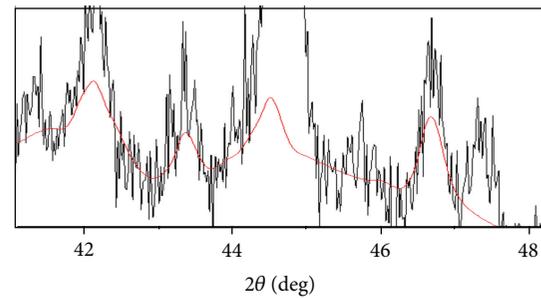


FIGURE 6: Expanded XRD pattern for annealed (350°C/25 h) Al-11Si-11Fe alloy showing part of the calculated profile (red line) of $Al_{4.5}SiFe$ β -phase.

before β -phase becomes coarser at the second stage during the heat treatment process.

5. Conclusions

Microstructure and microhardness of Al-11Si-11Fe rapidly solidified alloy have been improved as deduced by SEM, EDS, XRD, and microhardness measurements. Rapid solidification with estimated cooling rate of 1.2×10^5 K/s has lowered Fe content of β -phase particles as deduced by EDS analysis and suppressed the formation of β -phase incompletely. Besides, rapid solidification has a great influence on the morphology of β phase since these phase particles become very fine with straight needle shape (with 0.5 μ m thickness and 5 μ m width) not with curved shape as compared with its classically solidified counterpart. The estimated content of the as-melt spun β -phase is 10.1 wt.%. Solid solubility of Si in α -Al has been extended to 1 wt.%. Si nanoparticles in as-melt spun state coarsened at temperature of 374.3°C as deduced by DSC analysis with estimated content of 2.9 wt.% as deduced by

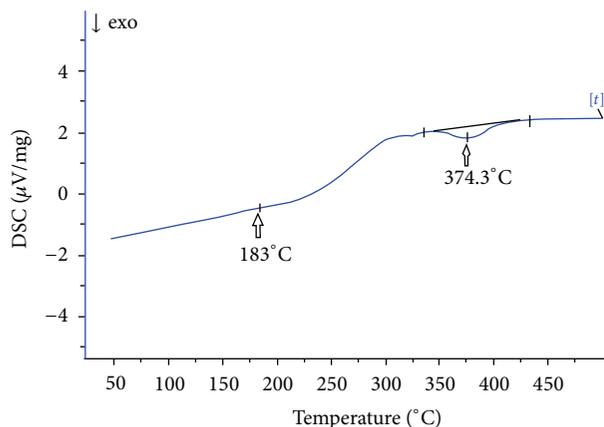


FIGURE 7: DSC curve of the melt-spun Al-11Si-11Fe alloy ribbon scanned at the heating rate of 10°C/min between 50 and 500°C.

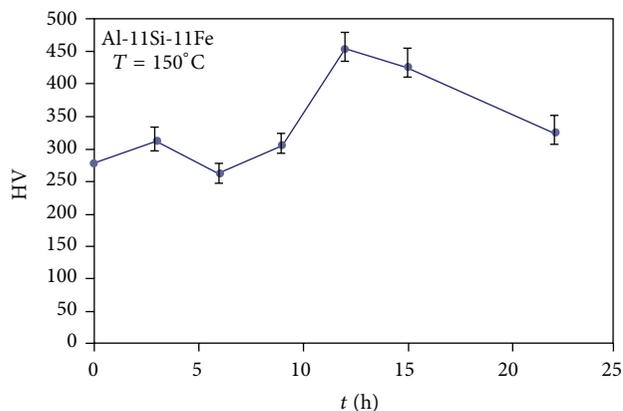


FIGURE 8: Variation of hardness (HV) with time (h) at 150°C.

XRD analysis. The average crystallite size and microstrain of α -Al in as-melt spun state were 27 nm and 0.125, respectively. During annealing process (350°C/25 h), the estimated contents of both β -phase and Si phases increased to 35 wt.% and 5.1 wt.%, respectively. Microhardness value of the melt spun ribbons was as high as 277 HV and further improved up to 450 HV during heat treatment at 150°C/12 h. This improvement of microhardness can be related to the influence of rapid solidification for microstructure refinement and modification occurring for the Al-11Si-11Fe as-melt spun alloy.

Conflict of Interests

The authors declare that there is no conflict of interests regarding the publication of this paper.

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