

Research Article

Effects of High-Density Pulse Currents on the Solidification Structures of Cu-SiC_p/AZ91D Composites

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In this study, Cu-SiC_p/AZ91D composites were prepared with high-density pulse currents. The wettability between SiC_p and matrix during solidification was improved by coating 0.095- μ m thick copper film on the surface of SiC_p. By comparing the composites prepared with/without pulse currents, the solidification structure and its formation mechanism of Cu-SiC_p/AZ91D composites were analyzed under different conditions. The Cu-SiC_p/AZ91D composites prepared without high-density pulse currents were mainly composed of α -Mg, β -Mg₁₇Al₁₂, and a small amount of Mg₂Si phases, with coarse grains and uneven structures. Under the action of high-density pulse currents, the structures of Cu-SiC_p/AZ91D composites were transformed into α -Mg and Mg₂Si phases with refined grain, and the homogeneity of the structures was improved significantly.

1. Introduction

Magnesium matrix composites are widely used in aerospace, construction, marine, and mineral processing industries due to their low density, good mechanical properties, and good corrosion resistance [1–3]. However, the performance of traditional magnesium alloys can no longer meet the needs of social development, and people are committed to the preparation of high-performance magnesium alloys. At present, the commonly used preparation methods of magnesium matrix composites include powder metallurgy, stirring casting, in situ synthesis, and melt infiltration methods [4–6]. After treatment, the reinforced particles are evenly distributed, and the properties of alloys are improved. However, these preparation methods require strict pretreatment procedures, high manufacturing costs, and complex operation. As a new technology, the treatment of alloy melt by using

high-density pulsed current has attracted the attention of researchers, and some research results have been achieved. Especially under the action of high-density pulsed current, it can effectively inhibit the segregation of the second phase and refine the solidification structure. At the same time, this technology has the characteristics of in situ synthesis technology for some composite materials [7–9].

SiC particles are often used as reinforcing phase for magnesium matrix composites. However, the surface activity of micron and nano-SiC particles is easy to agglomerate, and the wettability of SiC particles is poor [10]. The wettability can generally be improved by increasing the surface energy of the reinforcing phase to lower the surface tension of the melt. Preheat treatment of reinforcing particles, addition of appropriate elements to melt, surface coating, and ultrasonic dispersion can be adopted. Therefore, the modification of the SiC surface is beneficial to

promote the wettability of interface and the uniform distribution of reinforcing particles in the process of liquid casting. At present, the volume fraction of SiC reinforcing phase added by the full liquid stirring casting method is 5 vol.%~10 vol.% [11]. In this paper, a new type of full liquid treatment device under the condition of electric pulse is used. SiC particles coated with copper film are added into the AZ91D matrix as reinforcing phase to improve the wettability of the matrix and reinforcing phase. The Cu-SiC_p/AZ91D composites prepared under different conditions are studied. The microstructures of AZ91D composites and the strengthening mechanism are discussed, which provides a new design concept and method for future research.

2. Experimental

Cu-SiC_p/AZ91D composites contain 10 vol.% Cu-SiC_p, in which a layer of 0.095 μm copper film is deposited on the surface of SiC_p. The mass of Cu is 17.3% of that of Cu-SiC_p. The sample size of AZ91D magnesium alloy is 16 mm × 16 mm × 30 mm. The chemical composition of the AZ91D alloy is shown in Table 1 [12]. The magnesium alloy was used as the matrix in the experiment, and 10 μm SiC particles were chosen as the reinforcer to prepare Cu-SiC_p/AZ91D composites.

Figure 1 shows the schematic diagram of the electric pulse melting device used in this study. The preparation process is described as follows. Firstly, magnesium alloy samples were placed in the boron nitride crucible, and then, SiC_p was placed on the surface of magnesium alloys. The pressure in the vacuum box was pumped to 2×10^{-4} Pa, and then, argon gas was filled to the pressure of 50 kPa. Then turning on the high-frequency induction power supply, the samples were heated to 700°C in the vacuum box, and all the samples were melted in the crucible. Then, the metal melt was heated for 10 min. Finally, the electric pulse was applied to the metal melt for 5 min. The electric pulse treatment process of the metal melt was provided as follows. First, set the pulse width 10 μs, the frequency is 30 Hz, and start the power, then read the required current peak on the oscilloscope by adjusting the voltage, and finally the heating device was closed to start the cooling process in the furnace. When the body was completely solidified, the pulse power supply was turned off.

The structure and composition of the samples were analyzed by SS-550 Shimadzu scanning electron microscope, Phoenix EDAX-2000 energy dispersive spectrometer, and X-ray diffractometer (6000×).

3. Results and Discussion

3.1. XRD Analysis of Solidification Structures. XRD analysis was carried out with the prepared magnesium matrix composites (Figure 2). As shown in Figure 2, the Cu-SiC_p/AZ91D magnesium matrix composites prepared under conventional conditions mainly consist of three phases: α-Mg, β-Mg₁₇Al₁₂, and Mg₂Si. The composites prepared under pulse currents mainly consist of two phases: α-Mg and Mg₂Si, and the diffraction peaks of Mg₂Si are enhanced.

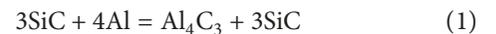
TABLE 1: Chemical composition of AZ91D alloy.

ω (%)					
Cu	Al	Zn	Mn	Si	Mg
0.015	9	0.67	0.25	0.05	Bal.

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3.2. SEM Analysis of Solidification Structures. In order to further analyze the tissue changes, the samples were photographed at a high solution for elemental scanning. The scanning results are listed in Table 2. The XRD spectrum of Figure 2 and Table 2 indicated that the material was composed of α-Mg phase (Spectrogram 1 and Spectrogram 4). The black skeletal dendrite is α-Mg₁₇Al₁₂ phase (Spectrogram 2). The new phase Mg₂Si is formed on the dendrite of β-Mg₁₇Al₁₂ (Spectrogram 3). The gray-white structure near Mg₂Si is composed of α-Mg and Al₄C₃ as well as less attached Cu (Spectrogram 5).

Figure 3 shows high-resolution SEM photographs of AZ91D magnesium alloy, Cu-SiC_p/AZ91D magnesium matrix composite, and Cu-SiC_p/AZ91D composite in pulsed electric fields. The AZ91D magnesium alloy is mainly composed of gray-black α-Mg and dark-black skeletal eutectic β-Mg₁₇Al₁₂ (Figure 3(a)). The phase of β-Mg₁₇Al₁₂ grows along the grain boundary. Figures 3(b) and 3(c) show that the hard strengthening phase Mg₂Si grows along the β-phase after adding Cu-SiC_p, and the Mg₂Si phase replaces the β-Mg₁₇Al₁₂ phase after applying pulsed electric fields. This is consistent with the bright white structure shown in Figures 4(b) and 4(c), the formation of Mg₂Si diffraction peaks in Figure 2, and the disappearance of Mg₁₇Al₁₂ diffraction peaks. The results show that Mg₂Si precipitates as a heterogeneous nucleation point at the grain boundary. SiC and Mg₁₇Al₁₂ peaks were not observed in the XRD spectra (Figure 2). It was inferred that the formation of Mg₂Si was ascribed to the reaction of SiC_p and Al [6]:



SiC_p in the copper-SiC_p was depleted, and Cu was distributed among the dendrites. Formed Al₄C₃ was dispersed in the vicinity of Mg₂Si [2]. Al₄C₃ was mixed with the matrix α-Mg, thus changing the color from gray to gray-white (red circle in Figure 3(c)), which was consistent with Spectrogram 5 in Table 2. The color change in Figure 3(b) was not obvious because SiC_p was not mixed evenly, and the content of Al₄C₃ was too small. As a heterogeneous nucleation point, Al₄C₃ increased the nucleation rate and refined grains [13, 14].

Copper-SiC_p used in this study could effectively change the wettability between particles and melt, but segregation of copper-SiC_p still occurred, and a large number of copper-SiC_p had not yet reacted with Al in the melt. Dendrites in the structure were mainly composed of β-Mg₁₇Al₁₂ and Mg₂Si phases. Al₄C₃ had fewer heterogeneous nucleation points and poor homogeneity, and the grain refinement effect was

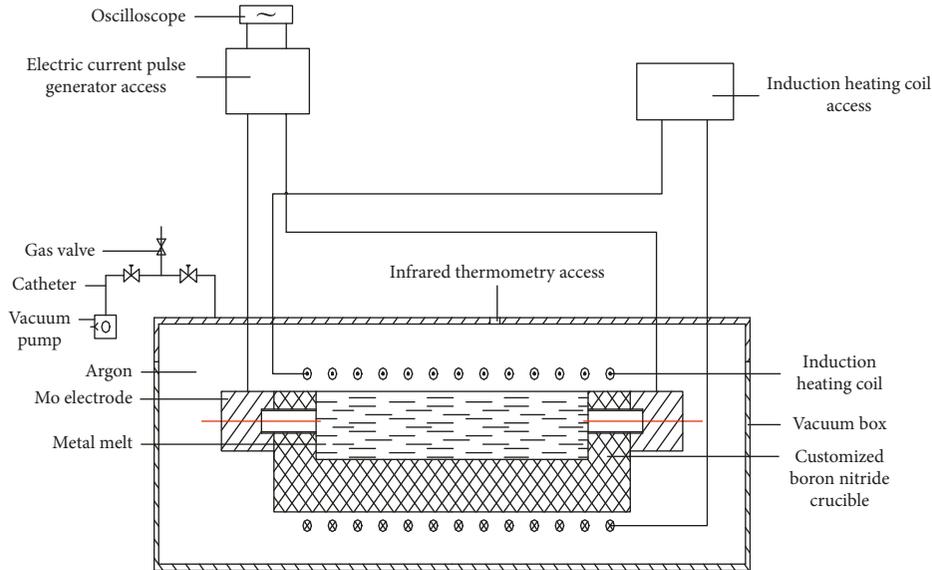


FIGURE 1: Schematic diagram of the electric pulse melting device.

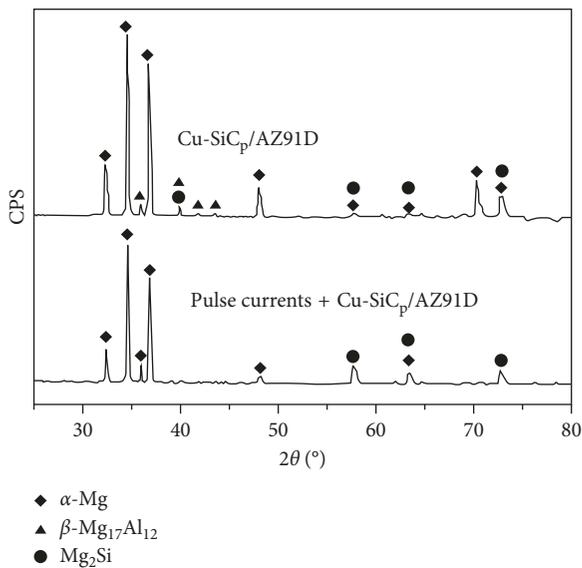


FIGURE 2: XRD spectrum of Cu-SiC_p/AZ91D magnesium matrix composites.

TABLE 2: AZ91D and Cu-SiC_p/AZ91D magnesium matrix composite element scanning atomic content table.

Elemental atomic percentage	C	Mg	Al	Si	Zn	Cu
Spectrogram 1	—	90.97	8.27	—	0.76	—
Spectrogram 2	—	60.48	39.52	—	—	—
Spectrogram 3	—	66.71	—	33.29	—	—
Spectrogram 4	—	93.78	5.68	—	0.54	—
Spectrogram 5	21.94	51.23	23.84	—	—	2.99

not observed (Figure 3(b)). When high-density pulse currents were applied, the Lorentz force was produced by the metal melt under the action of electric fields [15], and the

first cyclotron force of the Lorentz force formed strong convection together with the melt, thus resulting in the decreased temperature gradient, widened two-phase zone in the melt. Therefore, the segregation of Cu-SiC_p was effectively restrained, and Cu-SiC_p was allowed to join the melt and maintained in the uniformly mixing state. Due to the wetting effect and the increase in the contact area, Cu-SiC_p reacted with Mg₁₇Al₁₂ sufficiently, thus resulting in phase replacement of β-Mg₁₇Al₁₂ phase by the vermicular Mg₂Si (Figure 3(c)). In addition, the formation of a large number of heterogeneous nucleation points Al₄C₃ increased the nucleation rate, promoted heterogeneous nucleation, obtained uniform structures, and further refined the grains [16] (Figure 4(c)).

In the magnesium alloy of AZ91D, α-Mg formed the matrix, and the β-Mg₁₇Al₁₂ phase was distributed along the crystal boundary with the large crystal grain. In the Cu-SiC_p/AZ91D composite material obtained under conventional conditions, the tissue was mainly composed of three phases: β-Mg₁₇Al₁₂, Mg₂Si, and a small amount of Al₄C₃ phase. Under pulsed electric fields, the composite material showed the uniform structure, and the vermicular Mg₂Si phase replaced the β-Mg₁₇Al₁₂ phase. In the vicinity of the Mg₂Si phase, α-Mg was mixed with Al₄C₃ to form the gray-white zone. When a high-density pulse current is applied, the Lorentz force is generated in the melt under the electric field [16], and the melt is strongly convected by the Lorentz force, resulting in a decrease in the temperature gradient inside the melt and a widening of the two-phase region. In this way, the nucleation rate of Cu-SiC_p/AZ91D increased, and crystal grains were refined. After adding Cu-SiC_p particles, the Al₄C₃ phase was not detected because the following hydrolysis reaction occurred in the sampling preparation process [6]:



3.3. Refinement Mechanism of Solidification Structures. Figure 3 shows the metallographic pictures of AZ91D magnesium alloy, Cu-SiC_p/AZ91D magnesium matrix

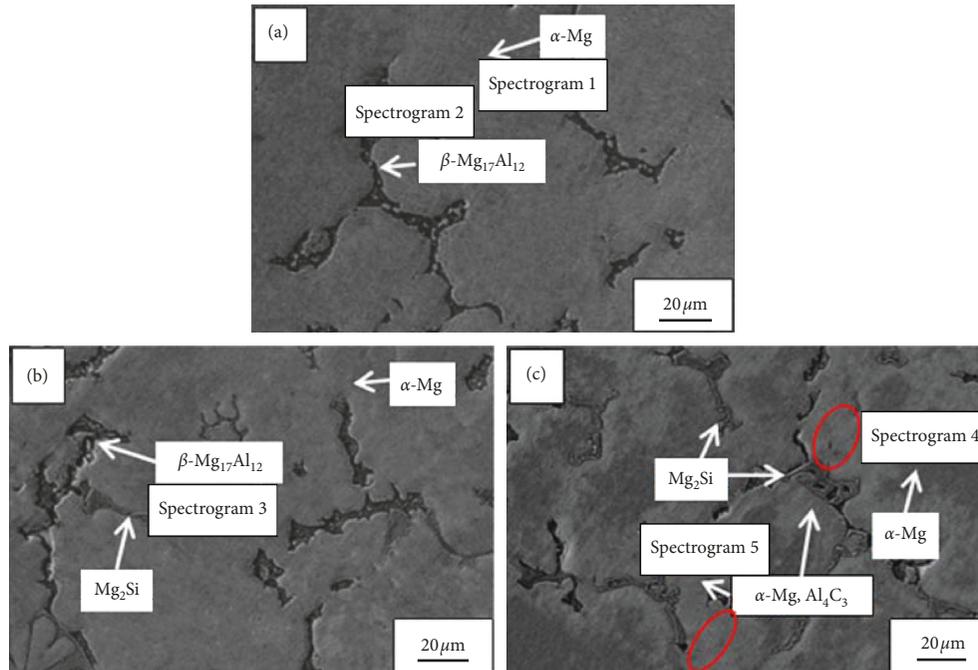


FIGURE 3: High times SEM of AZ91D magnesium alloy and Cu-SiC_p/AZ91D magnesium matrix composites: (a) AZ91D, (b) Cu-SiC_p/AZ91D, and (c) pulsed electric field+Cu-SiC_p/AZ91D).

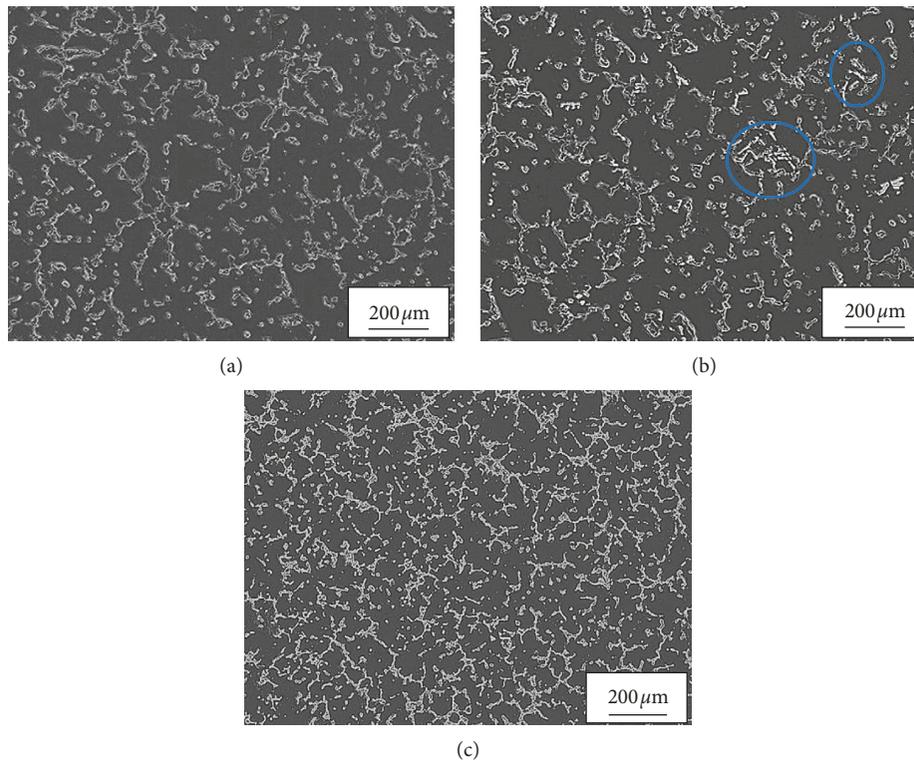


FIGURE 4: Photos of AZ91D magnesium alloy and Cu-SiC_p/AZ91D magnesium matrix composites: (a) AZ91D, (b) Cu-SiC_p/AZ91D, and (c) pulsed electric field +Cu-SiC_p/AZ91D.

composite, and Cu-SiC_p/AZ91D composite under pulsed electric fields. As shown in Figure 4(a), AZ91D magnesium alloy is composed of black matrix and gray reticulated

structure, and its grain size is relatively large. As shown in Figure 3(b), after adding Cu-SiC_p, a little bright white fine structure occurs in the gray-white reticulated structure

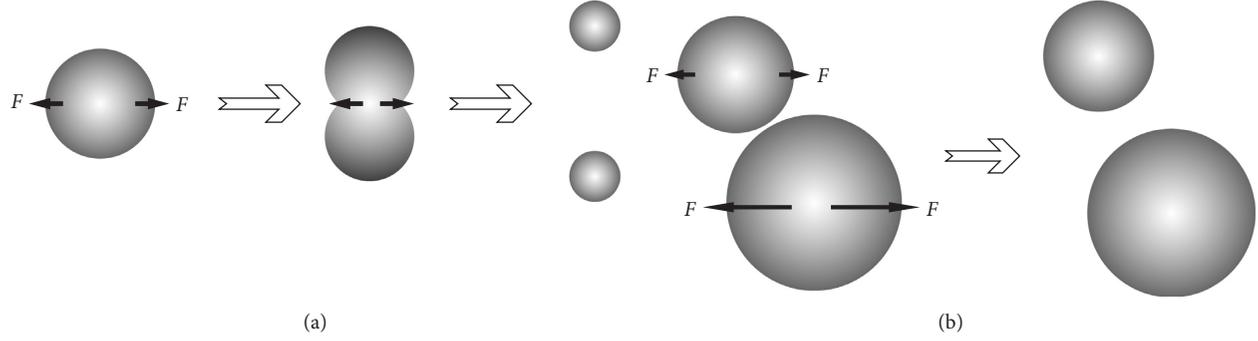


FIGURE 5: Particle condensation process under the influence of electromagnetic force.

(marked by the blue ellipse), and the grains near the bright white structure are refined Cu-SiC_p that can provide potential nucleation points for the melt, thus increasing the number of fine grains in the composites. The copper-SiC_p particles have the pinning effect on grain boundaries and inhibit the grain growth [17]. As shown in Figure 4(c), after applying pulsed electric fields and adding Cu-SiC_p, all the gray-white reticulated structures become the finer bright white structures with dense dendrites and fine grains.

According to the theory of electromagnetic field dynamics, under the action of pulse currents, the particles and the whole melt are affected by the changing electromagnetic force. This vibration will produce the following effects on the aggregation of particles in the melt. Firstly, the vibration of the electromagnetic force can break coagulated Cu-SiC_p into smaller particles, as shown in Figure 5(a). Secondly, under the action of pulse currents, the SiC particles of different sizes are also affected by inertial force, which results in the relative motion between them. The effect of relative motion also weakens the coagulation effect of Cu-SiC_p, as shown in Figure 5(b). Furthermore, under the action of pulsed electric fields, the undercooling of the alloy melt increases, thus leading to the increase in the viscosity of the alloy melt and weakening the coagulation effect of Cu-SiC_p particles [10, 18].

In order to study the effect of pulse currents on the nucleation rate of alloy melt, the nucleation rate of classical nucleation theory [15] is expressed as follows:

$$N = \frac{nKT}{h} \exp\left(-\frac{16\pi\sigma_{sl}^3 T_m^2}{3L_m^2 \Delta T^2 KT}\right) \cdot \exp\left(-\frac{\Delta G_A}{KT}\right), \quad (4)$$

where h is the Planck constant; n is the number of atoms per unit volume; K is the Boltzmann constant; T is the absolute temperature; σ_{sl} is the surface free energy; T_m is the melting point; $\Delta T = (\Delta T_m - T)$ is the undercooling of the alloy melt; L_m is the latent heat of melting; ΔG_A is the liquid atom nucleation barrier. When the axisymmetric current $\vec{j} = j(r)\vec{e}_z$ passes through a cylindrical conductive melt, the magnetic field $\vec{B} = B(r)\vec{e}_\theta$ is formed. Pulse currents usually affect the undercooling of the alloy melt through the generated Joule heat and electromagnetic force. In addition, when pulse currents are applied during the melt solidification process, more solute atoms are stimulated to break the energy barrier and enter the matrix due to the effect of

instantaneous discharge. At the same time, pulse currents enhance the vibration of atoms deviating from the equilibrium position, reduce the energy barrier, and change the nucleation barrier. Therefore, in the original nuclear rate equation (1), $\Delta G_A = (\Delta G_0 + \Delta G_E)$, where G_0 is the thermodynamic barrier for the nucleation without applying an external field and G_E is the thermodynamic barrier for the nucleation after applying an external field. Then, we get

$$\Delta G_A = (\Delta G_0 + \Delta G_E) = \Delta G_0 + K_1 \cdot j_2 \cdot \xi_2 \cdot V, \quad (5)$$

$$\xi = (\sigma_0 - \sigma_n) \cdot (\sigma_n - 2\sigma_0)^{-1}.$$

Among them, K_1 is a parameter related to materials; J is the pulsed current density; σ_0 is the conductivity of disordered dielectrics; σ_n is the conductivity of nuclei; V is the volume of nuclei; and K is the Boltzmann constant. For the crystalline melt, if $\sigma_n > \sigma_0$, then $\xi > 0$. Therefore, it can be concluded that the pulse currents reduce the nucleation barrier in the alloy melt. The effect of pulse currents increases the nucleation rate in the alloy melt, and the increase in the nucleation rate leads to the grain refinement in the alloy melt. When EPT is applied after heat preservation, the pulse currents contact directly with the melt and the nucleation growth stops, thus forming an equiaxed region, which can effectively improve the nucleation rate of liquid metal and semisolid metal and trigger the heterogeneous nucleation mechanism [19]. Fine structures were obtained during the rapid solidification because the increase in the undercooling promoted the nucleation rate. The mechanism of dendrite breakage induced by Loren magnetic force under electric pulse treatment allowed the grain refinement [16, 20].

4. Conclusion

Cu-SiC_p/AZ91D composites prepared without high-density pulse currents mainly consisted of three phases: α -Mg, β -Mg₁₇Al₁₂, and Mg₂Si. By applying high-density pulse currents, the structures of Cu-SiC_p/AZ91D composites were transformed into the phases of α -Mg and Mg₂Si.

Cu-SiC_p/AZ91D composites were prepared by different testing methods. The results showed that the Cu-SiC_p/AZ91D composites under high-density pulse currents had uniform structures, and the grains were significantly refined. The nucleation barrier was reduced, and the nucleation rate

was effectively increased by applying high-density pulse currents. Therefore, the fine structure was obtained.

The microstructures of Cu-SiC_p/AZ91D composites were transformed into α -Mg, Al₄C₃ and Mg₂Si phases under the action of high-density pulse currents. Al₄C₃ and Mg₂Si phases as heterogeneous nucleation points increased the nucleation rate of the composites. The Al₄C₃ phase was not detected in the obtained tissues due to the hydrolysis reaction.

Data Availability

The data used to support the findings of this study are available from the corresponding author upon request.

Conflicts of Interest

The authors declare that there are no conflicts of interest regarding the publication of this paper.

Acknowledgments

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