

# Research Article

# Determination of Caffeine, Theobromine, and Theophylline by HPLC-DAD in Beverages Commonly Consumed in Lima, Peru

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The purpose of this study was to determine the content of caffeine, theobromine, and theophylline by high-performance liquid chromatography with diode-array detection (HPLC-DAD) in beverages commonly consumed in Lima, Peru. The samples were divided into 6 groups (herbal teas, coffee powder, chocolate milks, soft drinks, sports drinks, and energy drinks) which included the five most commonly consumed beverages of the different groups. Caffeine was mainly identified in the soft drink and energy drink groups, with the latter having a more significant value  $(10.38 \pm 0.01 \text{ vs}. 95.50 \pm 3.48 \text{ mg/L}$ , respectively). In herbal teas, caffeine showed the highest content  $(0.47 \pm 0.01 \text{ to } 4.91 \pm 0.05 \text{ mg/L})$ , despite theophylline being a characteristic compound of tea leaves. Sports drinks presented very low caffeine levels  $(0.03 \pm 0.01 \text{ to } 0.05 \pm 0.01 \text{ mg/L})$ , and theobromine  $(0.48 \pm 0.01 \text{ to } 6.00 \pm 0.02)$  was also identified. Caffeine  $(4.09 \pm 0.01 \text{ to } 5.70 \pm 0.01 \text{ mg/L})$  and theobromine  $(1.70 \pm 0.01 \text{ to } 12.24 \pm 0.01 \text{ mg/L})$  were found in the five commercial brands of chocolate milk evaluated. Moreover, the group of coffee powder samples had the highest level of caffeine content  $(49.25 \pm 0.24 \text{ to } 964.40 \pm 4.93 \text{ mg/100 g})$ . The results obtained in this study provide reliable information on the composition and quantification of methylxanthines in the beverages most consumed in Lima and impact consumer knowledge.

# 1. Introduction

In the last decades, several studies have focused on the biologically active ingredients, particularly alkaloids in beverages, for their possible beneficial effects on human health [1]. One of those alkaloids is methylxanthine, a compound commonly present in food [2]. Monteiro et al. [3] noted that both animals and plants naturally produce these compounds, being caffeine, theobromine, and theophylline as the most frequently studied. These compounds are mainly found in food such as coffee beans, cocoa beans, and tea leaves [4].

These three methylxanthines are chemically very similar [5], and according to De Sena et al. [6], they share stimulating effects on the central nervous system as well as on other systems such as the gastrointestinal, cardiovascular, renal, and respiratory systems.

Caffeine, for example, is one of the best known and most studied substances, having a stimulating effect on the central nervous system, increasing alertness, and improving long-term memory and concentration and may also improve physical performance in athletes [7-9]. There is even a debate about the neuroprotective effects against certain types of degenerative diseases such as Alzheimer's and Parkinson's [10]. According to a Canadian study [11], moderate daily caffeine intake of up to 400 mg per day by healthy adults is not associated with adverse health effects. Likewise, the Food and Drug Administration (FDA) states that a dose of up to 400 mg of caffeine per day is safe in the healthy adult population, although this may vary according to the sensitivity of the person or vulnerable group (pregnant or breastfeeding women or individuals with any special health condition) [12]. However, excessive consumption may be related to hypertension, anxiety, hyperactivity, and headaches [7].

There are fewer studies on theobromine, which is present in high concentrations in cocoa [13]. According to Martínez-Pinilla et al. [13], the main mechanisms of action are inhibition of phosphodiesterases and blockade of adenosine receptors. Another clinical study suggests that theobromine could be the main active component of cocoa responsible for the effect of increasing HDL cholesterol [14]. However, there is no specific and/or conclusive information regarding their safe or lethal daily dose.

Theophylline has similar pharmacological and toxicological properties to those of caffeine [2]. According to Dolder [15], this compound can mainly be found in tea and in asthma medication used for bronchodilation. Although there is no precise information on a recommended dose, Greene et al. [16] have reported that inadequate consumption may cause toxic effects including tachycardia, hypertension, nausea, vomiting, and diarrhea.

In the last years, access to a huge variety of products [17], such as ready-to-drink beverages, coffee powders, soft drinks, energy drinks, sport drinks, and herbal teas, has become available. These products are commonly marketed in Peru, and most of the nutritional labeling does not clearly indicate the concentrations of certain compounds that they contain, with this information often remaining unknown to the consumer public.

Therefore, the aim of this study was to determine the content of caffeine, theobromine, and theophylline by high-performance liquid chromatography with diode-array detection (HPLC-DAD) and the antioxidant capacity in beverages commonly consumed in Lima, Peru.

#### 2. Material and Methods

2.1. Chemicals. All the solvents and reagents were analytically graded with deionized water (Milli-Q quality) [4]. The standards of caffeine, theobromine, and theophylline were purchased from Sigma-Aldrich (Merck Peruana, Lima). Acetonitrile and methanol (HPLC grade) were also obtained from Sigma-Aldrich (Merck Peruana, Lima).

2.2. Sample Collection. The samples were divided into six groups: herbal teas (HT), sports drinks (SD), chocolate milk (CM), soft drinks (SOD), energy drinks (ED), and coffee powder (CP). Each group included five of the most popular beverages of the group, obtained by nonprobabilistic convenience sampling, based on a survey of consumers in Lima. Samples were purchased in three supermarkets in Metropolitan Lima during 2020. The nutritional labels of each sample analyzed were reviewed to corroborate the declared composition.

HTs are ready-to-drink cold drinks marketed as herbal infusions with an added value (e.g., fruit flavors or various properties), which are highly appreciated by a certain sector of the national and international markets [18]. SDs provide carbohydrates, electrolytes, and liquids to the body, helping the body hydrate before, during, and after physical activity [19].

CM is a mixture based on milk, sucrose, cocoa and/or cocoa powder, and some hydrocolloids, which are added to improve the consistency and avoid sedimentation of the cocoa particles [20]. On the other hand, SODs are flavored drinks, which are produced by adding  $CO_2$  gas directly to the drink, diluted with sucrose, and may or may not contain caffeine depending on the drink [21].

EDs are used to provide an extra burst of energy as well as promote wakefulness, increase attention span, maintain alertness, and improve athletic performance [22].

Lastly, CP is a drink prepared with coffee beans in powder form [23], with caffeine as its main compound. All coffee powder samples were prepared.

None of the samples in this study were sweetened with noncaloric sweeteners.

2.3. Total Polyphenol Measurement. The content of total phenolic was determined according to Singleton and Rossi [24]. Briefly, 1 mL of 10% Folin-Ciocalteu's reagent was mixed with 0.1 mL of sample for 5 min at room temperature; then, 1 mL of 5% sodium carbonate was added, and the mixture was placed in a water bath at 45°C for 15 min. All samples were analyzed in triplicate. Absorbance was read on a spectrophotometer at 725 nm. The results were expressed in mg gallic acid equivalent/100 mL (mg GAE/100 mL).

2.4. Total Flavonoid Measurement. Total flavonoid measurement was performed according to Wolfe and Liu [25]. In a 0.250 mL sample, 0.075 mL of 5% sodium nitrite was added and allowed to react for 5 min. Then, 0.150 mL of aluminum chloride was added, and the mixture was left to stand for 5 min. Finally, 0.275 mL of sodium hydroxide (1 M) was added to the mixture, and it was left to react for 15 min. All samples were analyzed in triplicate. The reading was performed at 510 nm in a spectrophotometer. The results were expressed in  $\mu$ g catechin equivalent/100 mL ( $\mu$ g CE/100 mL).

2.5. Antioxidant Capacity by Ferric Reducing Antioxidant Power. Antioxidant activity evaluated by the ferric reducing antioxidant power (FRAP) assay was performed according to the methodology proposed by Benzie and Strain [26]. Briefly, 1 mL of distilled water and 1 mL of the FRAP reagent were added to 0.1 mL of sample. The mixture was then placed in a 37°C water bath and allowed to react for 10 min. The reading was made in a spectrophotometer (Pharo 300, Spectroquant, USA) at 593 nm. A standard curve was prepared using different concentrations of Fe<sup>2+</sup> ranging from 15 to 75 mM. All samples were analyzed in triplicate. The results were expressed in  $\mu$ M Fe<sup>2+</sup>/100 mL ( $\mu$ M Fe<sup>2+</sup>/100 mL).

2.6. Chromatographic Conditions. HPLC analyses of caffeine, theobromine, and theophylline were performed according to Srdjenovic et al. [4], with slight modifications. Briefly, a VWR HITACHI Chromaster 600 HPLC with a diode-array detector (DAD CM 5430), autosampler, and a reversed phase purospher STAR RP-8 column (5  $\mu$ m particle size, i.d. 4.6 × 150 mm) was used in an isocratic elution mode with the mobile phase water-THF(A) (0.1% THF in water, pH 8)-acetonitrile (B) (90:10,  $\nu/\nu$ ). The pH was adjusted with 0.1 M NaOH. The mobile phase and all the solutions were filtered (0.45  $\mu$ m × 47 mm Millipore nylon filter), and the run time was 5 min, with a flow rate of 0.8 mL/min. The column temperature was 25°C, and the analytes were detected at 273 nm.

The quantification of individual compounds was calculated with a calibration curve of the standard compound [27]. All samples were run in triplicate for each analysis.

2.7. Sample Preparation. The HTs and SDs were filtered through a  $0.22 \,\mu\text{m}$  nylon filter. Ten milliliters of the filtrate was adjusted to pH 8 with 0.1 M NaOH [4].

The CMs (25 mL) containing suspended particles were filled up to 200 mL with water in a separate container and extracted for 30 minutes at 60°C in an ultrasonic bath. The extracted sample was filtered through filter paper to remove solids. Then, 10 mL of the filtrate was adjusted to pH 8 with a 0.1 M NaOH solution [4].

The SOD and ED samples were degassed for 15 min to release CO<sub>2</sub>. Before analysis, the samples were adjusted to pH 8 with a 0.1 M NaOH solution and filtered through the 0.22  $\mu$ m × 33 mm Millipore Millex-GN nylon filter.

The CP samples were weighed (5 g) and extracted with boiling water (200 mL) and then mixed in a thermal flask for 5 min on the magnetic stirrer [4]. The extracts were then filtered, and 10 mL of the filtrate was adjusted to pH 8 with 0.1 M NaOH.

2.8. Cleanup Procedure. According to Srdjenovic et al. [4], the Supelclean LC-18 SPE cartridges were conditioned with  $2 \times 6 \text{ mL}$  of methanol, followed by  $2 \times 6 \text{ mL}$  of ultrapure deionized water. The extracts from each sample were then passed through the solid phase extraction (SPE) cartridges, followed by washing with 6 mL of ultrapure deionized water. They were air dried under vacuum for 10 min, and the fluids were discarded. Lastly, caffeine, theobromine, and theophylline were extracted from the SPE cartridges with 10 mL of chloroform. The solution was evaporated until dryness. The residue of all samples was reconstituted in 1 mL of pH 8 water, with the exception of CM, which was reconstituted in 2 mL. Samples were filtered through a 0.22  $\mu$ m nylon filter and injected into the HPLC [4].

2.9. Statistical Analysis. The results of each analysis were presented in tables, and the statistical analyses were performed using JASP version 0.14.1 (2020). Preliminarily, it was tested whether the variables met the assumptions of normality by means of the Shapiro-Wilk test, with a significance level of p > 0.05. In the comparison tests between beverage groups, the Kruskal-Wallis test was used, with a p value < 0.05 being considered significant.

#### 3. Results and Discussion

3.1. Bioactive Compounds and Antioxidant Capacity. The total polyphenol content, total flavonoid content, and FRAP of the six groups of beverages are presented in Table 1. As can be seen, the total polyphenol content in the different beverages varied according to the group, with reductions in the following order: CP > ED > HT > CM > SOD > SD (Table 1). These differences are due to the numerous phenolic compounds found in coffee ranging from  $3.57 \pm 0.17$  to  $19.97 \pm 0.17$  mg GAE/ 100 mL compared to the other samples studied. In fact, chlorogenic acid attributes to more potential benefits of coffee

TABLE 1: Total polyphenols, flavonoids, and FRAP results of the beverages most consumed in Lima, Peru<sup>\*</sup>.

Beverage Samples	Total polyphenols (mg GAE/ 100 mL)	Total flavonoids (μg CE(+)/100 mL)	FRAP (µmol Fe <sup>2+</sup> /100 mL)	
SOD-1	$1.12\pm0.04$	$2718.31 \pm 70.42$	$3.59\pm0.10$	
SOD-2	$0.19\pm0.01$	$915.49\pm70.42$	<lod< td=""></lod<>	
SOD-3	$0.08\pm0.01$	$887.32 \pm 42.25$	<lod< td=""></lod<>	
SOD-4	$0.46\pm0.04$	$971.83 \pm 42.25$	<lod< td=""></lod<>	
SOD-5	$1.85\pm0.01$	$2647.89 \pm 28.17$	$6.17\pm0.15$	
ED-1	$1.58\pm0.02$	$3732.39 \pm 42.25$	$15.63\pm0.10$	
ED-2	$5.19\pm0.10$	$4126.76 \pm 70.42$	$17.04\pm0.05$	
ED-3	$0.57\pm0.10$	$2723.00 \pm 70.89$	$6.21\pm0.29$	
ED-4	$2.16\pm0.04$	$3859.15 \pm 28.17$	$8.58\pm0.15$	
ED-5	$0.49\pm0.02$	$2971.83 \pm 79.42$	$3.06\pm0.05$	
HT-1	$4.74\pm0.18$	$4.00\pm0.40$	$6.55\pm0.05$	
HT-2	$4.08\pm0.27$	$7.33 \pm 0.61$	$38.98\pm0.05$	
HT-3	$0.40\pm0.01$	$9.20\pm0.40$	$7.91\pm0.05$	
HT-4	$0.06\pm0.01$	<lod< td=""><td><math display="block">5.08\pm0.06</math></td></lod<>	$5.08\pm0.06$	
HT-5	$0.12\pm0.03$	$37.40\pm0.60$	$4.85\pm0.10$	
SD-1	$0.53\pm0.03$	$2873.24\pm56.34$	$0.74\pm0.06$	
SD-2	$0.11\pm0.02$	$3070.42 \pm 84.51$	<lod< td=""></lod<>	
SD-3	$0.58\pm0.06$	$4225.35 \pm 56.34$	$2.23\pm0.10$	
SD-4	$0.38\pm0.01$	$3436.62\pm56.34$	$7.43\pm0.05$	
SD-5	$1.76\pm0.05$	$2323.94\pm79.42$	$9.66\pm0.05$	
CM-1	$0.38\pm0.01$	$21647.89 \pm 211.27$	$86.60\pm0.42$	
CM-2	$0.74\pm0.02$	$14859.15 \pm 14.08$	$47.91 \pm 0.34$	
CM-3	$0.76\pm0.28$	$14774.65 \pm 211.27$	$4.95\pm0.10$	
CM-4	$1.80\pm0.03$	$32676.06 \pm 281.69$	$62.77\pm0.53$	
CM-5	$1.69\pm0.03$	$19971.83 \pm 169.01$	$1.41\pm0.05$	
CP-1	$3.57\pm0.17$	$51154.93 \pm 225.35$	$29.51 \pm 0.29$	
CP-2	$5.09\pm0.16$	$78971.83 \pm 1478.87$	$54.08\pm0.39$	
CP-3	$15.92\pm0.24$	$89478.87 \pm 70.42$	$167.86\pm0.49$	
CP-4	$19.97\pm0.17$	$38873.24 \pm 647.89$	$228.88 \pm 1.60$	
CP-5	$9.23\pm0.28$	28746.48 ± 1450.70	$103.20\pm0.97$	

Note\*: LOD: level of detail; SOD: soft drinks; ED: energy drinks; HT: herbal teas; SD: sport drinks; CM: chocolate milks; CP: coffee powder; CE: catechin equivalent.

and has shown to contribute to the prevention of Alzheimer's disease-induced cognitive dysfunction [28].

However, in the group of HTs, there was a notable difference in polyphenol content between the infusions based on black tea (HT-3, HT-4, and HT-5:  $0.06 \pm 0.01$  to  $0.40 \pm 0.01$  mg GAE/100 mL) and green tea (HT-1 and HT-2:  $4.08 \pm 0.27$  to  $4.74 \pm 0.18$  mg GAE/100 mL). In agreement with Horžić et al. [1], green tea was found to be the richest source of total phenols.

Regarding the total flavonoid content, CP and CMs presented the highest concentrations (57445.07  $\pm$  5969.47  $\mu$ g

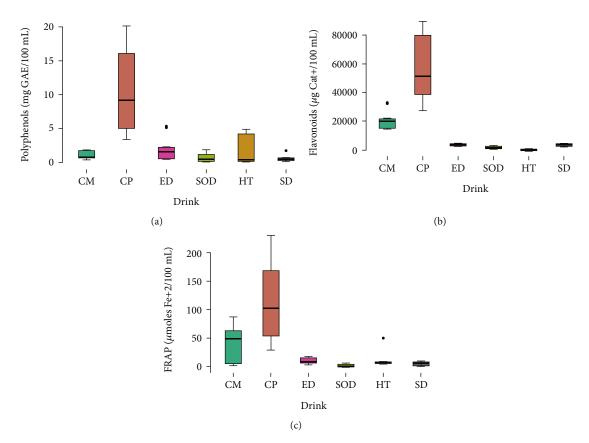


FIGURE 1: Box plot displaying the distribution of the concentrations of (a) polyphenols (mg GAE/100 mL), (b) flavonoids ( $\mu$ g Cat+/100 mL), and (c) FRAP ( $\mu$ moles Fe+2/100 mL) detected in the samples analyzed for the different brands of chocolate milks (CM), coffee powder (CP), energy drinks (ED), soft drinks (SOD), herbal teas (HT), and sport drinks (SD).

CE(+)/100 mL and  $20785.92 \pm 7315.11 \,\mu\text{g}$  CE(+)/100 mL, respectively) compared to the other groups (Figure 1(a)).

According to the results obtained, the group with the lowest flavonoid content was the HTs (Figure 1(b)). No previous study has compared these groups of beverages to establish why teas have lower flavonoid content. On the other hand, among the tea beverages, it is conceivable that the catechin content may vary according to tea variety, origin, time of harvest, and sun exposure [1], and these factors should be considered in future research. The FRAP assay showed that CP had a much higher antioxidant capacity (29.51 ± 0.29 to 228.88 ± 1.60  $\mu$ mol Fe(2+)/100 mL) compared to the other samples, such as SOD, which showed a very low or almost no antioxidant capacity (Figure 1(c)).

Based on these results, the group of CP drinks predominated in most of the tests carried out, presenting the greatest antioxidant capacity (Figure 1(c)) due to its bioactive compounds such as chlorogenic acid. The other groups varied in relation to the compounds of each sample.

3.2. Results of Methylxanthine (CF, TB, and TP) Measurement. The content of caffeine, theobromine, and theophylline obtained by the HPLC-DAD method in the different beverage samples is shown in Table 2 and Figure 2. The declared and determined values of these methylxanthines found by HPLC-DAD in the analyses of the beverages are also shown. 3.2.1. Soft Drinks and Energy Drinks. The SOD and ED groups had no detectable amounts of theobromine or theophylline. These groups were characterized by caffeine as the most predominant methylxanthine. The content of caffeine in SOD ranged from  $0.54 \pm 0.01$  mg/L for SOD-3 to  $16.68 \pm 0.01$  mg/L for SOD-2 (Figure 3(a)). The caffeine content of EDs ranged from  $10.38 \pm 0.01$  for ED-4 to  $95.5 \pm 3.48$  mg/L for ED-2 (Figure 4(b)), being almost 8-fold greater than the SODs analyzed.

In comparison with Serbia [4], one study reported that caffeine content in EDs ranged from 237 mg/L to 348.7 mg/L, with a 2- to 3-fold greater caffeine content than SOD with a minimum value of 106.7 mg/L and a maximum of 119.1 mg/L. Therefore, while the ratio of caffeine content between EDs and SODs seems to be higher in this study, their values were lower.

According to several studies [29], caffeine is one of the most widely used psychoactive substances in the world and is often used as an additive in SOD [30] and ED [31].

On the other hand, most of the commercial brands analyzed in these categories do not declare the total caffeine content of their product. However, it is included in the list of ingredients on the nutrition label.

In the SOD and ED groups, the amount of total caffeine content was below the lower limit of consumption of 400 mg per day in healthy adults [11, 12]. However, it was of note that ED had the second highest caffeine content (Figure 5(c)). Only

TABLE 2: HPLC-DAD determination of theobromine, theophylline, and caffeine concentrations in the beverage most frequently consumed
in Lima*.

Beverage samples	Values found				Declared values		
Deverage samples	ТВ	TP	CF	TB	TP	CF	
SOD-1 (mg/L)	<lod< td=""><td><lod< td=""><td><math display="block">9.34\pm0.02</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<></td></lod<>	<lod< td=""><td><math display="block">9.34\pm0.02</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$9.34\pm0.02$	n.d	n.d	n.d	
SOD-2 (mg/L)	<lod< td=""><td><lod< td=""><td><math display="block">16.68\pm0.01</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<></td></lod<>	<lod< td=""><td><math display="block">16.68\pm0.01</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$16.68\pm0.01$	n.d	n.d	n.d	
SOD-3 (mg/L)	<lod< td=""><td><lod< td=""><td><math display="block">0.54\pm0.01</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<></td></lod<>	<lod< td=""><td><math display="block">0.54\pm0.01</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$0.54\pm0.01$	n.d	n.d	n.d	
SOD-4 (mg/L)	<lod< td=""><td><lod< td=""><td><math display="block">4.63\pm0.02</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<></td></lod<>	<lod< td=""><td><math display="block">4.63\pm0.02</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$4.63\pm0.02$	n.d	n.d	n.d	
SOD-5 (mg/L)	<lod< td=""><td><lod< td=""><td><math display="block">4.82\pm0.06</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<></td></lod<>	<lod< td=""><td><math display="block">4.82\pm0.06</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$4.82\pm0.06$	n.d	n.d	n.d	
ED-1 (mg/L)	$1.17\pm0.09$	<lod< td=""><td><math display="block">20.38\pm0.12</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$20.38\pm0.12$	n.d	n.d	n.d	
ED-2 (mg/L)	<lod< td=""><td><lod< td=""><td><math display="block">95.50 \pm 3.48</math></td><td>n.d</td><td>n.d</td><td>≤320</td></lod<></td></lod<>	<lod< td=""><td><math display="block">95.50 \pm 3.48</math></td><td>n.d</td><td>n.d</td><td>≤320</td></lod<>	$95.50 \pm 3.48$	n.d	n.d	≤320	
ED-3 (mg/L)	<lod< td=""><td><lod< td=""><td><math display="block">73.45\pm0.02</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<></td></lod<>	<lod< td=""><td><math display="block">73.45\pm0.02</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$73.45\pm0.02$	n.d	n.d	n.d	
ED-4 (mg/L)	<lod< td=""><td><lod< td=""><td><math display="block">10.38\pm0.01</math></td><td>n.d</td><td>n.d</td><td>≤200</td></lod<></td></lod<>	<lod< td=""><td><math display="block">10.38\pm0.01</math></td><td>n.d</td><td>n.d</td><td>≤200</td></lod<>	$10.38\pm0.01$	n.d	n.d	≤200	
ED-5 (mg/L)	<lod< td=""><td><lod< td=""><td><math display="block">86.58\pm0.04</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<></td></lod<>	<lod< td=""><td><math display="block">86.58\pm0.04</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$86.58\pm0.04$	n.d	n.d	n.d	
HT-1 (mg/L)	$0.31\pm0.01$	$0.17\pm0.01$	$4.91\pm0.05$	n.d	n.d	n.d	
HT-2 (mg/L)	$3.64\pm0.58$	<lod< td=""><td><math display="block">2.21\pm0.08</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$2.21\pm0.08$	n.d	n.d	n.d	
HT-3 (mg/L)	$0.07\pm0.01$	<lod< td=""><td><math display="block">1.70\pm0.01</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$1.70\pm0.01$	n.d	n.d	n.d	
HT-4 (mg/L)	$0.03\pm0.01$	<lod< td=""><td><math display="block">0.47\pm0.01</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$0.47\pm0.01$	n.d	n.d	n.d	
HT-5 (mg/L)	$0.09\pm0.01$	<lod< td=""><td><math display="block">2.96\pm0.01</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$2.96\pm0.01$	n.d	n.d	n.d	
SD-1 (mg/L)	$0.05\pm0.01$	<lod< td=""><td><math display="block">6.00\pm0.02</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$6.00\pm0.02$	n.d	n.d	n.d	
SD-2 (mg/L)	$0.03\pm0.01$	<lod< td=""><td><math display="block">0.75\pm0.01</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$0.75\pm0.01$	n.d	n.d	n.d	
SD-3 (mg/L)	$0.04\pm0.01$	<lod< td=""><td><math display="block">0.84\pm0.01</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$0.84\pm0.01$	n.d	n.d	n.d	
SD-4 (mg/L)	<lod< td=""><td><lod< td=""><td><lod< td=""><td>n.d</td><td>n.d</td><td>n.d</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>n.d</td><td>n.d</td><td>n.d</td></lod<></td></lod<>	<lod< td=""><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	n.d	n.d	n.d	
SD-5 (mg/L)	<lod< td=""><td><lod< td=""><td><math display="block">0.48\pm0.01</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<></td></lod<>	<lod< td=""><td><math display="block">0.48\pm0.01</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$0.48\pm0.01$	n.d	n.d	n.d	
CM-1 (mg/L)	$2.92\pm0.01$	<lod< td=""><td><math display="block">4.71\pm0.01</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$4.71\pm0.01$	n.d	n.d	n.d	
CM-2 (mg/L)	$6.43\pm0.01$	<lod< td=""><td><math display="block">5.70\pm0.01</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$5.70\pm0.01$	n.d	n.d	n.d	
CM-3 (mg/L)	$3.96\pm0.01$	<lod< td=""><td><math display="block">5.37 \pm 0.01</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$5.37 \pm 0.01$	n.d	n.d	n.d	
CM-4 (mg/L)	$1.70\pm0.01$	<lod< td=""><td><math display="block">4.09\pm0.01</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$4.09\pm0.01$	n.d	n.d	n.d	
CM-5 (mg/L)	$12.24\pm0.01$	<lod< td=""><td><math display="block">5.69\pm0.01</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$5.69\pm0.01$	n.d	n.d	n.d	
CP-1 (mg/100 g sample)	$0.45\pm0.01$	$0.79\pm0.01$	$49.25\pm0.24$	n.d	n.d	n.d	
CP-2 (mg/100 g sample)	$0.30\pm0.01$	$0.95\pm0.01$	$133.00\pm0.07$	n.d	n.d	n.d	
CP-3 (mg/100 g sample)	$0.83\pm0.07$	$0.87\pm0.01$	$964.40 \pm 4.93$	n.d	n.d	n.d	
CP-4 (mg/100 g sample)	$0.80\pm0.01$	$0.87\pm0.03$	$778.19\pm0.61$	n.d	n.d	n.d	
CP-5 (mg/100 g sample)	$0.58\pm0.02$	<lod< td=""><td><math display="block">389.79\pm0.13</math></td><td>n.d</td><td>n.d</td><td>n.d</td></lod<>	$389.79\pm0.13$	n.d	n.d	n.d	

Note\*: TB: theobromine; TP: theophylline; CF: caffeine; n.d: not declared; LOD: level of detail; SOD: soft drinks; ED: energy drinks; HT: herbal teas; SD: sport drinks; CM: chocolate milks; CP: coffee powder.

two brands of the five EDs indicated the amount of caffeine contained by their product in the nutritional labeling. It is important for this information to be included on the nutritional labeling of the product in order for consumers to control their consumption.

3.2.2. Herbal Teas and Sport Drinks. The groups of HT and SD showed the presence of only two methylxanthines, theobromine and caffeine. In the HT, the highest content identified was caffeine  $(0.47 \pm 0.01$  to  $4.91 \pm 0.05$  mg/L), with the lowest value in HT-4 and the highest value in HT-1, respectively (Figure 3(c)). One would expect to find theophylline in the HTs as it is a characteristic and predominant compound

in tea leaves [16], and some are marketed precisely because of the beneficial health properties of tea leaves.

However, according to De Paula and Farah [32], there is no perceptible trace of this methylxanthine in these HTs possibly due to the fact that these infusion drinks are sweetened, contributing to the dilution of the tea solids and producing a higher content of soluble solids. In addition, many other factors could intervene in the variation of the properties of the product.

The results obtained are consistent with other studies [4, 33, 34], which evaluated different beverages and described that ready-to-drink HTs have a high content of caffeine and also of theobromine in some samples.

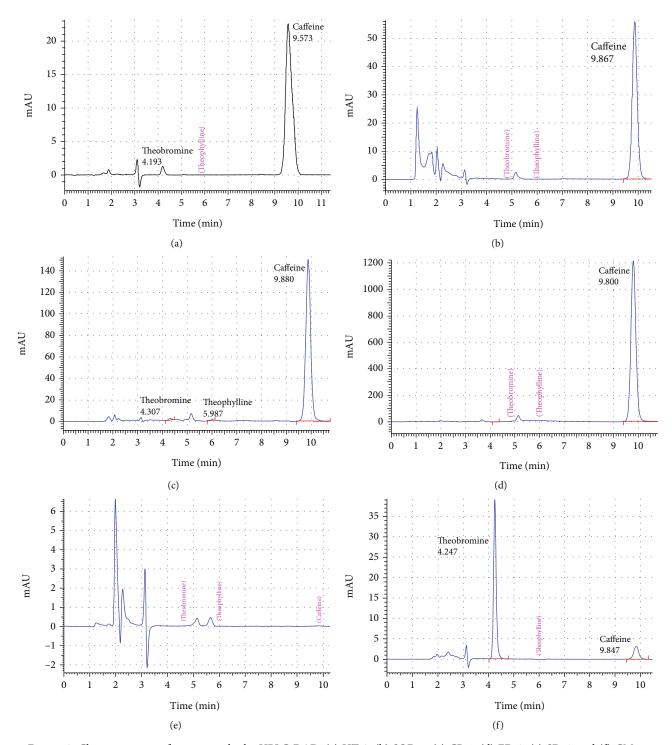


FIGURE 2: Chromatograms of some samples by HPLC-DAD: (a) HT-3, (b) SOD-5, (c) CP-1, (d) ED-2, (e) SD-4, and (f) CM-5.

The composition of SDs is basically carbohydrates and electrolytes [19], and their nutritional labeling does not indicate or allude to any ingredient containing caffeine, theobromine, or theophylline, and therefore, they should not present significant values of these methylxanthines. According to the results obtained, almost undetectable values of only two methylxanthines, caffeine  $(0.03 \pm 0.01 \text{ to } 0.05 \pm 0.01 \text{ mg/L})$  and theobromine  $(0.48 \pm 0.01 \text{ to } 6.00 \pm 0.02 \text{ mg/L})$ , were

found in the SDs analyzed (Table 2 and Figure 3(b)), thereby confirming and validating the expected content.

*3.2.3. Chocolate Milks (CM).* The CM drinks are mixture based on milk, sucrose, and cocoa and/or cocoa powder [21], cocoa being the ingredient leading to the denomination of a CM. According to some studies [35], cocoa is rich in two methyl-xanthines, predominantly theobromine followed by caffeine.

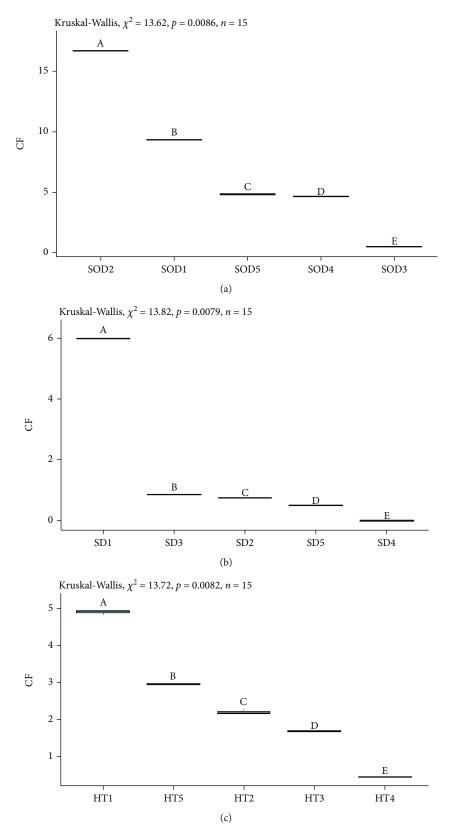


FIGURE 3: Box plot showing the distribution of the concentrations of the most representative methylxanthine in each group: (a) soft drink-caffeine (mg/100 g sample), (b) sport drink-caffeine (mg/100 g sample), and (c) herbal tea-caffeine (mg/100 g sample).

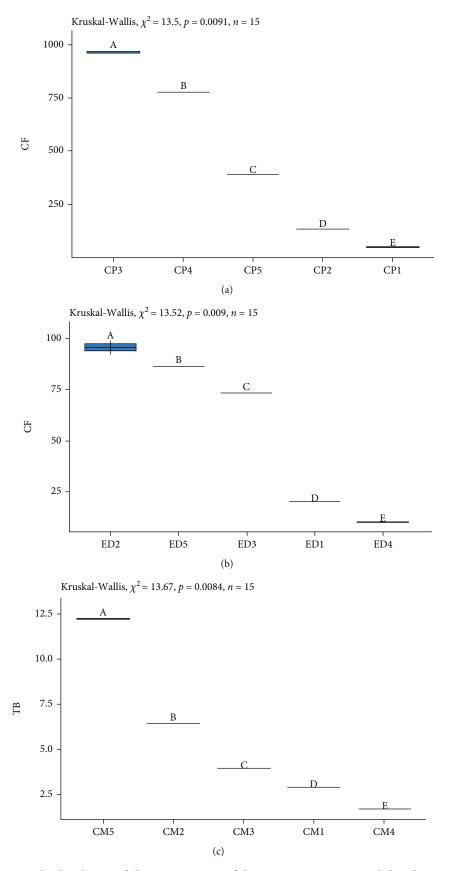


FIGURE 4: Box plot showing the distribution of the concentrations of the most representative methylxanthine in each group: (a) coffee powder-caffeine, (b) energy drink-caffeine, and (c) chocolate milk-theobromine.

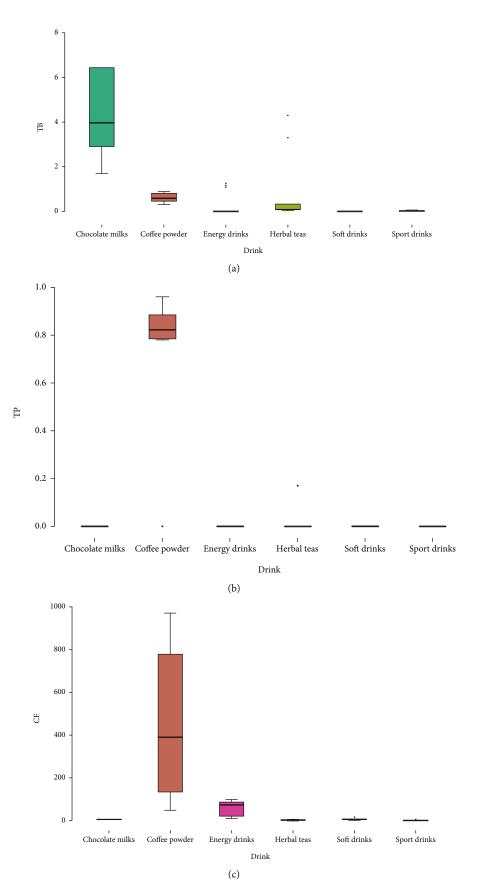


FIGURE 5: Box plot showing the distribution of the concentrations of methylxanthines: (a) theobromine (mg/L), (b) theophylline (mg/L), and (c) caffeine (mg/100 g sample).

Among the five commercial brands of chocolate milk evaluated (Figure 5(a)), theobromine was identified in greater quantity  $(1.7 \pm 0.01 \text{ mg/L} \text{ to } 12.24 \pm 0.01 \text{ mg/L})$  followed by caffeine  $(4.09 \pm 0.01 \text{ mg/L} \text{ to } 5.69 \pm 0.01 \text{ mg/L})$ . According to two studies, one carried out in Serbia [4] and the other in Spain [36], the mean theobromine content of the CM evaluated was 225.8 mg/L and 205.7 mg/L, while the mean caffeine content was 5.6 mg/L and 18.4 mg/L, respectively. However, the values found in the Serbian and Spanish studies were much higher than those found in the beverages analyzed in the present study.

On the other hand, Table 2 shows that theobromine levels varied greatly between one of the brands of CM (CM-5) compared to the others analyzed (Figure 4(c)). The CM-5 sample had two to four times more theobromine than the other brands of CM (Figure 4(c)). According to its nutritional labelling, the CM-5 brand of CM included cocoa in second place in its list of ingredients, while in the other brands, cocoa appears as the third ingredient. This would explain the difference in the concentration of theobromine in the CM-5 drink (Table 2).

Taking into account these differences in theobromine concentrations in the different CM products, it is important to understand and know the content of each compound included in the products consumed.

3.3. Coffee Powder (CP). Among the six beverage groups evaluated, CP had the highest level of caffeine in CP-3  $(49.25 \pm 0.24 \text{ mg}/100 \text{ g})$  and the lowest value in CP-1  $(964.40 \pm 4.93 \text{ mg}/100 \text{ g})$  (Figure 4(a)).

The coffee bean is the main source of caffeine [22] and the main ingredient of the samples evaluated, which corroborates the results found. However, the level of caffeine in each of the five samples studied varied, likely due to the variety of coffee bean used by each brand [36], among other factors, for the processing treatment of this product.

As can be seen, compared to the other CP samples, two (CP-3 and CP-4) presented the highest level of caffeine (964.40  $\pm$  4.93 and 778.19  $\pm$  0.61 mg/100 g sample, respectively) (Figure 4(a)). This means that, in the case of CP-3, a person would be consuming approximately 19.29 mg of caffeine per cup, provided that only 2g of coffee powder had been added, as suggested in the nutritional labels of the product. This does not exceed the limit described in several studies [11, 12], reporting that an intake of up to 400 mg/day is not associated with adverse health effects.

While some people choose to consume this drink for its characteristics, benefits, or simply its taste, it must be remembered that excess intake is related to negative effects such as hypertension, insomnia, constant agitation, headaches, and anxiety [7].

Furthermore, people sensitive to caffeine or nonregular consumers or those with a health problem in which caffeine interferes or aggravates their symptoms [32] should take greater care and should be taken to limit or restrict their consumption.

# 4. Conclusions

The results obtained in this study provide reliable information on the composition and quantification of methylxanthines in the beverages most consumed in Lima, using the HPLC-DAD method.

The group of CPs predominated as having the highest antioxidant activity according to the tests carried out. Similarly, this group presented the highest level of caffeine, which was to be expected since its main active ingredient is the coffee bean.

Theobromine was predominantly found in the group of CMs analyzed.

Although theophylline is mainly found in the tea leaf, there was no detectable or significant value within the group of ready-to-drink HTs, and little antioxidant activity was identified. This should be analyzed in future research.

In general, it was observed that caffeine was one of the bioactive compounds present in each of the groups of beverages analyzed.

Inclusion of the concentrations of the different compounds found in the nutritional labels of food products is essential. This information should be stated in a clear and understandable way to consumers, improving the nutritional education of the population and favoring purchasing decisions.

# **Data Availability**

The datasets generated during and/or analyzed during the current study are available from the corresponding author on reasonable request.

## **Conflicts of Interest**

The authors declare no conflict of interest.

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