

Research Article

Physiochemical, Biochemical, Minerals Content Analysis, and Antioxidant Potential of National and International Honeys in Pakistan

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16 honey samples from Pakistan and two other countries were investigated for their physiochemical, biochemical, minerals, and antioxidant potential. Antioxidant activities of all honey samples were performed by using percentage inhibition of DPPH free radical, AEAC, and FRAP. 5-HMF and mineral contents were determined by HPLC and AAS, respectively. The obtained values of respective parameters, namely, pH, EC, TDS, total acidity, moisture, ash, color intensity, sugars, proline, and protein were in compliance with codex standard and recommendation of council directive by European Union. The total phenolics contents in acacia honey from Germany and jujube honey from Pakistan are similar to monofloral honey from Saudi Arabia and Yemen, respectively. The mineral contents in tested honey samples are comparable with honey from Brazil and Romania. Dark color honeys contained higher phenolic contents than light color ones and attributed to higher oxidation potential and have strong positive correlation with DPPH and FRAP.

1. Introduction

Honey is complex supersaturated sweet natural liquid produced by honeybees from the nectar of plants. The substances collected by bees then after combined with their own specific substances are deposited, dehydrated, stored, and left in honeycombs to ripen and mature. Around 200 substances have been reported in this complex natural liquid but the composition especially its secondary metabolites and quality of honey may be influenced by some external factors such as environmental and seasonal factors, processing, handling, and storage [1–3]. The substances included in honey are sugars (main components: fructose 38%, glucose 31%), proteins, moisture (10–20%), vitamins (ascorbic acid, niacin, etc.), mineral salts (potassium, calcium, sodium, phosphorus, etc.), organic acids (acetic acid and gluconic acid, etc.), 5-hydroxymethylfurfural (HMF), enzymes (phosphatases,

glucose oxidase, invertase, and catalase), flavonoids, phenolic acids, and volatile compounds. EC Directive 2001/110 specified the criteria to ensure the quality of honey by contents analysis, namely, electrical conductivity, moisture, ash, free acidity, sugars (reducing and nonreducing), HMF, and diastase activity [4-6]. As natural antioxidant and having high nutritional values, honey has been consumed over the years by humanity. Many enzymes like glucose oxidase, catalase, and peroxidase and nonenzymes, namely, carotenoids, α tocopherol, vitamin C, proteins, amino acids, flavonoids, polyphenols, Maillard reaction products, and small amount of mineral content, particularly iron and copper, are responsible for the redox properties of honey. Flavonoids present in honey comprising of flavanones, flavones, and flavonols while phenolic acids are substituted cinnamic acids and benzoic acids and these compounds are main contributor for color, taste, and aroma of honey [7-10]. A strong correlation

between phenolic contents especially the total phenolic contents and antioxidant activities of honey from different region of world has been reported. The dark colored honey has higher phenolic contents consequently which indicate the higher antioxidative properties [11-13]. Antioxidant is prominent characteristics of honey but it also possesses various other biological activities like wound healing and antiinflammatory, antimicrobial activities as well as in the treatment of gastrointestinal disorders and skin diseases [14-17]. Since the antioxidant power of honey is strongly correlated with its phenolic contents, its therapeutic potential is associated with its antioxidant capacity against reactive oxygen species produced in physiological and metabolic processes. Antioxidants reduce the risk of chronic disease emerged as a result of oxidative damage such as cancer, heart disease, and neurological degeneration [18]. These degenerative diseases are initiated by the oxidative mechanism of free radicals like peroxides, hydrogen peroxide, and lipid peroxyl so polyphenols, flavonoids, and phenolic acids present in honey scavenge these free radicals to oxidize the proteins, nucleic acids, and lipids and inhibit the degenerative diseases [19]. From different region of world, it has been reported that honey has capacity to prevent inflammatory disease associated with lipid peroxidation and reduce oxidative damage of erythrocytes. Honeys are more potent antioxidant than vitamins A, C, and E and reduce the immune response causing inflammation [20-22]. To the best of our knowledge, this is first study established in Pakistan for physiochemical (pH, electrical conductivity, acidity, total dissolved solids, moisture, color, ash, 5-hydroxymethylfurfural (HMF) concentrations, and color intensity), biochemical (sugars, total phenolics and flavonoids, protein, ascorbic acid, and proline contents), mineral contents (Ca, Cd, Co, Cr, Cu, Fe, K, Mg, Mn, Na, Ni, Pb, and Zn), and antioxidant activities such as AEAC (ascorbic acid equivalent antioxidant content), scavenging activity of DPPH (1,1-diphenyl-2-picrylhydrazyl) free radical, and FRAP (ferric reducing antioxidant power) of honey available in a second largest city (Lahore) of the country.

2. Materials and Methods

2.1. Honey Samples. Sixteen honey samples were collected from local market shelves in the capital city Lahore, Punjab-Pakistan. Each sample was of 250 g packed in glass bottle, and none of samples exceeded the storage period of three months. All the samples were stored at 4° C and before analysis kept overnight at $25 \pm 2^{\circ}$ C. Among the tested honey samples twelve samples were of Pakistani honey coded as PAKH-1-PAKH-12, two samples each of German honey coded GERH-13, GERH-14, and French honey coded FRNH-15, FRNH-16. The PAKH-11 and GERH-13 are monofloral as jujube and acacia, respectively, while remaining were multifloral.

2.2. Physiochemical Analysis

2.2.1. pH, Acidity, Electrical Conductivity (EC), and Total Dissolved Solids (TDS). A 10% (w/v) honey solution was

prepared in high purity water ($0.01 \,\mu$ S/cm) using purification system (Milli-Q, USA) and pH was determined by multimeter (Orion 5 star, Thermo Scientific, UK). 0.05 M sodium hydroxide (Merck, Germany) was used for free acidity by plotting the neutralization curve while for lactone acidity, excess sodium hydroxide was added to honey solution and back titrated with 0.025 M sulphuric acid (Merck, Germany). Summation of both free and lactone acidity is equivalent to total acidity. TDS in ppm and EC in mS/cm were measured by Orion 5 star multimeter (Thermo Scientific, UK). 20 g honeys were suspended in 75 mL of high purity water in volumetric flask and make the volume up to 100 mL with same solvent.

2.2.2. Moisture, Ash Contents, Color, and Its Intensity. Refractive index measurement method by Wedmore's formula [23] was used and the moisture was expressed in percentage. Ash dish was heated at the ashing temperature and then 5.0 g of honey is placed along with two drops of ash-free olive oil. We placed the ash dish in furnace (Thermolyne[™], Thermo Scientific UK) and commenced ashing without loss at a low heat rising to 350-400°C. The ash dish is cooled at room temperature and ash content was calculated as g ash per 100 g of honey [24]. The color of honey was categorized by Pfund scale after changing an absorbance value. Color grades were expressed in millimeter (mm) Pfund grades using glycerol (Sigma Aldrich, USA) as reference standard for comparison. PG-T80⁺ UV-Vis spectrophotometer, UK, was used to measure the absorption at 635 nm. In warm high pure water 50% (w/v) of honey solution was made and then filtered using a nylon 0.45 μ m filter. The absorption was measured by using spectrophotometer at 450 and 720 nm. Differences of absorbance were stated is mAU [25].

2.2.3. 5-Hydroxymethylfurfural (HMF) Concentration. 5-Hydroxymethylfurfural (Sigma Aldrich, USA) concentration was determined by slight modification in reported method [26]. For stock solution, 1.0 g 5-hydroxymethylfurfural was dissolved in water and make volume up to 1000 mL; 0-80 mg/kg standard solutions for linearity were prepared by dissolving the respective volumes in mobile phase water (Milli-Q) and methanol (Thermo Scientific, UK) (90:10) from stock solution. The chromatography was performed on MediterraneanTM sea 18 (5 μ m, 4.6 × 250 mm) column at room temperature ($25 \pm 2^{\circ}$ C) under optimized experimental conditions. 1.0 mL/min flow rate was adjusted by injecting the volume of $20 \,\mu L$ using Rheodyne sample injection port. The separation was performed on HPLC system (Shimadzu, Kyoto-Japan) equipped with degasser (DGU-4A), pump (LC-20AD), column oven (CTO-20A), and UV-Visible detector (SPD20A) and detection was performed at 285 nm. Each test required just 5 min. and before injection all the standards, sample solutions, and mobile phase were filtered through membrane filter of $0.45 \,\mu\text{m}$ and then sonicated. Peak areas integration was done by Shimadzu LC solution software (version 1.227). All the reagents used were of HPLC grade. The results of 5-HMF are expressed in mg/kg of honey.

2.3. Biochemical Analysis

2.3.1. Sucrose, Reducing Sugar, and Total Sugar. 1.0 mL of 1.0 mg/mL of honey solution in water was mixed with 1.0 mL of 3,5-dinitrosalicylic acid (DNSA) (Sigma Aldrich, USA), the solution was warmed in water bath for ten minutes, and the absorption of resulting reddish-orange color was measured spectrophotometrically at 540 nm. The glucose (Merck, Germany) was used as reference standard. Sucrose content in each honey sample was measured by refractometric method using 20% (w/v) solution. Sum of concentration of reducing sugar and sucrose results in the total sugar contents in g/100 g of honey [2].

2.3.2. Total Phenolics and Flavonoids. Folin-Denis method was used for determination of total phenolics in honey. Standard solutions of gallic acid (Sigma Aldrich, USA) (0–200 μ g/mL) were used for calibration curve. The total phenol content was expressed as milligrams of gallic acid equivalents (mg GAE)/kg of honey. 2.0 mL of honey solution (2.0 mg/mL in methanol) was mixed with 2.0 mL of 2% methanolic solution of aluminum trichloride (Sigma Aldrich, USA). The solution mixer was incubated for 30 min. at room temperature and the absorption was measured at 415 nm against methanol (Sigma Aldrich, USA) as blank reagent. Standard solutions of quercetin (Sigma Aldrich, USA) (0–200 μ g/mL) were used for calibration curve. The total flavonoids content was expressed as milligrams of quercetin equivalents (mg QE)/kg of honey [27].

2.3.3. Total Protein, Proline, and Ascorbic Acid Content. Protein contents were determined by Folin-Denis reagent method [28] and results of protein in honey samples are presented in mg/g of honey. For proline, 0.5 mL of honey solution (5.0 g/100 mL of water) was added in test tube, mixed with 1.0 mL of formic acid (Sigma, Aldrich, USA) and 1.0 mL of 3% Ninhydrin (Sigma, Aldrich, USA) in ethylene glycol monomethylether (Fisher Scientific, UK), and shaken vigorously for 15 min. The test tube was placed in water bath (TW8 Julabo, Germany) at 70°C for 10 min and then 5.0 mL of 50% propanol (Fisher Scientific, UK) solution was added in water. The test tube was cooled at temperature and the absorption was measured at 510 nm. Same procedure was adopted for reagent blank water and reference standard of 0.8 mg/25 mL of proline (Sigma, Aldrich, USA) [29]. The results of proline contents in honey samples are expressed in mg/kg of honey. 100 mg of honey sample was extracted with 10 mL of 1% metaphosphoric acid (Sigma, Aldrich, USA) at room temperature for 45 min and filtered through filter paper (Whatman number 4). 9.0 mL of 0.005% 2,6-dichlorophenolindophenol (Sigma, Aldrich, USA) was mixed with 1.0 mL of filtrate and absorption was measured at 515 nm within 30 min. Ascorbic acid (Sigma Aldrich, USA) in concentration range of $50-400 \,\mu\text{g/mL}$ was used for calibration curve [9]. The results of ascorbic acid contents in honey samples are presented in mg/kg of honey.

2.4. Antioxidant Activities

2.4.1. Ascorbic Acid Equivalent Antioxidant Content (AEAC). A 0.75 mL of methanolic honey solution (0.03 g/mL) was mixed with 2,2-diphenyl-1-picrylhydrazyl (Sigma, Aldrich, USA) (0.02 mg/mL in methanol). The mixture was incubated at room temperature for 15 min. and the absorption was measured at 517 nm. Ascorbic acid in concentration range of $1-8 \mu g/mL$ was used for calibration curve [10]. Antioxidant activity was expressed as milligrams of ascorbic acid equivalent antioxidant content (mg AEAC)/100 g of honey.

2.4.2. 1, 1-Diphenyl-2-Picrylhydrazyl (DPPH) Activity. DPPH free radical scavenging activity of honey was determined as reported earlier [30], methanolic DPPH (0.024 mg/mL, 2.7 mL) was mixed with 0.5 mL (0.2 g/mL) of honey extract, and the mixture was shaken vigorously and left at room temperature for 30 min. in dark. DPPH radical scavenging effect was determined by measuring the absorption at 517 nm. The percentage scavenging activity of DPPH radical was calculated as follows: % RSA = ($[A_{DPPH} - A_S]/A_{DPPH}$) × 100, where A_S is the absorbance of sample solution and A_{DPPH} is the absorbance of the DPPH solution. Butylated hydroxytoluene (BHT) was used as a reference standard.

2.4.3. Ferric Reducing Antioxidant Power (FRAP) Assay. FRAP assay was performed to reduce the Fe³⁺-TPTZ complex (yellow) to Fe²⁺-TPTZ (blue). FRAP reagent was prepared by mixing 0.3 M acetate buffer (pH 3.6), 0.010 M 2,4,6tripyridyl-S-triazine (TPTZ) (Sigma, Aldrich, USA) solution in 0.040 M HCl (Merck, Germany), and 0.020 M FeCl₃·6H₂O (Merck, Germany) in ratio of 10:1:1.1.5 mL of FRAP reagent was mixed in 0.2 mL of honey solution (0.1 g/mL) and incubated for 4 min at 37°C and the absorption was measured at 593 nm against a reagent blank containing high pure water. Ferrous sulfate (Sigma Aldrich, USA) in concentration range of 100–1000 μ m/L was used for calibration curve and FRAP values were expressed as micromoles of ferrous equivalent (μ M Fe [II]) per 100 g of honey [31].

2.5. Mineral Contents. Digest 1.0 g of honey in 9.0 mL 65% $\rm HNO_3$ (Merck, Germany) and 1.0 mL 30% $\rm H_2O_2$ (Merck, Germany). Microwave oven (W-1900 Continuous System 220–240 V, Thomas Scientific, USA) was used for digestion procedure. Elemental standard solution each of Ca, Cd, Co, Cr, Cu, Fe, K, Mg, Mn, Na, Ni, Pb, and Zn was prepared from stock solution (1000 mg/L) (Fisher Scientific, UK). High pure water was used for diluting solutions. PG-990 atomic absorption spectrometer was used for mineral contents under the optimized instrumental conditions [32, 33].

2.6. Statistical Analysis. Statistical analysis was performed on SPSS version 15 (SPSS Inc., Chicago, IL, USA) for windows and analysis of variance (ANOVA) was performed for significant difference using post hoc test (p < 0.05). Correlation among the different parameters was established by Pearson's correlation coefficient (r) in bivariate linear correlations

Code name	рН	TDS (ppm)	EC (mS/cm)	Moisture (%)	Free acidity (meq/kg)	Lactone acidity (meq/kg)	Total acidity (meq/kg)
PAKH-1	4.01 ± 0.01	315.2 ± 0.91	0.493 ± 0.008	17.1 ± 0.12	17.1 ± 0.12	11.1 ± 0.11	28.2 ± 0.48
PAKH-2	3.84 ± 0.00	250.4 ± 0.81	0.391 ± 0.009	12.5 ± 0.23	33.7 ± 0.09	3.4 ± 0.12	37.1 ± 0.63
PAKH-3	4.17 ± 0.02	435.8 ± 0.33	0.681 ± 0.034	14.3 ± 0.13	12.3 ± 0.11	12.1 ± 0.32	24.2 ± 0.41
PAKH-4	3.88 ± 0.01	450.2 ± 0.31	0.703 ± 0.072	16.1 ± 0.11	19.6 ± 0.21	7.9 ± 0.33	27.5 ± 0.33
PAKH-5	3.93 ± 0.02	302.2 ± 0.55	0.472 ± 0.007	14.2 ± 0.21	21.7 ± 0.92	8.2 ± 0.84	29.9 ± 0.78
PAKH-6	3.97 ± 0.00	365.1 ± 0.43	0.571 ± 0.005	10.1 ± 0.33	25.6 ± 0.32	9.5 ± 0.33	35.1 ± 0.91
PAKH-7	4.11 ± 0.01	347.7 ± 0.52	0.544 ± 0.011	13.8 ± 0.14	24.6 ± 0.21	9.8 ± 0.32	34.4 ± 0.53
PAKH-8	3.93 ± 0.02	295.4 ± 0.23	0.462 ± 0.001	17.4 ± 0.21	17.8 ± 0.11	11.6 ± 0.42	29.4 ± 0.31
PAKH-9	3.84 ± 0.01	278.3 ± 0.22	0.435 ± 0.003	16.5 ± 0.16	33.7 ± 0.22	3.4 ± 0.33	37.1 ± 0.53
PAKH-10	4.07 ± 0.02	395.5 ± 0.54	0.618 ± 0.006	17.3 ± 0.21	12.3 ± 0.31	12.1 ± 0.43	24.4 ± 0.44
PAKH-11	4.28 ± 0.00	378.3 ± 0.55	0.591 ± 0.007	16.8 ± 0.25	17.6 ± 0.42	3.2 ± 0.33	20.8 ± 0.61
PAKH-12	3.83 ± 0.01	332.3 ± 0.76	0.519 ± 0.098	14.7 ± 0.17	21.3 ± 0.25	8.1 ± 0.24	29.4 ± 0.43
GERH-13	4.21 ± 0.01	315.2 ± 0.77	0.493 ± 0.001	12.1 ± 0.31	125.6 ± 0.43	9.5 ± 0.43	135.1 ± 0.57
GERH-14	3.84 ± 0.01	327.2 ± 0.34	0.511 ± 0.041	18.8 ± 0.22	24.6 ± 0.22	9.8 ± 0.33	34.4 ± 0.73
FRNH-15	4.11 ± 0.02	325.1 ± 0.33	0.508 ± 0.001	19.2 ± 0.31	17.1 ± 0.87	11.1 ± 0.44	28.2 ± 0.64
FRNH-16	3.97 ± 0.02	281.2 ± 0.87	0.439 ± 0.002	15.5 ± 0.15	32.6 ± 0.83	3.3 ± 0.45	35.9 ± 0.52

TABLE 1: pH, TDS, EC, moisture, free acidity, lactone acidity, and total acidity of honey samples (mean \pm SD, n = 3).

(p < 0.01). All the assays were performed in triplicate and results expressed in mean \pm SD.

3. Results and Discussion

3.1. Physiochemical Analysis. Acidic pH is characteristic of honey; the pH value (Table 1) of twelve Pakistani and four exotic tested honey samples was found to be between 3.83 and 4.28 while free acidity of multifloral honey was between 12.3 and 33.7 meq/kg and the lactone acidity was observed between 3.4 and 12.1 meq/kg which are comparable with already reported values in different region of world. Free and lactone acidity value for acacia honey (Monofloral-Germany) was obtained at 125.6 and 9.5 meq/kg which is comparable with monofloral acacia honey from Saudi Arabia (free acidity: 122.3 meq/kg; lactone acidity: 9.5 meq/kg). Jujube honey from Pakistan (Monofloral) has free and lactone acidity value 19.6 and 2.9 meq/kg, respectively, which is similar to honey of same flora from Yemen (free acidity: 17.0 meq/kg; lactone acidity: 3.5 meq/kg). The total acidity (Table 1) values ranging between 24.2 and 37.1 meq/kg except the acacia honey (GERH-13: 135.1 meq/kg) and these values in compliance with international standard (50 meq/kg). The acidic pH of honey is irrespective of its geographical region, but due to presence of organic acids formed after the fermentation of sugars and inorganic anions, for example, phosphates, sulfates. These organic acids are contributor of flavor, acidic pH of honey which stabilize it and prevent it from bacterial growth because these grow in neutral to slightly alkaline media. Blossom honeys (lower pH) and honeydew (higher pH) can be distinguished by the pH values [34]. TDS and EC are interlinked with each other and important parameters to determine the physical characteristics of honey. TDS were between 250 and 450 ppm and EC of tested honey was

between the ranges of 0.391 and 0.703 mS/cm (Table 2) which is lower than the recommended value of 0.8 mS/cm given by the European Commission [9]. TDS is a measure of all the inorganic and organic substance present in honey either in molecular or in ionized form. Our study indicated the good correlation between TDS and EC; these two parameters can be used to determine the honey purity. The moisture content (Table 1) in Pakistani honey was ranged between 10.1 and 17.3% while 12.1-19.2% water content was found in exotic honey. The lower level of water content attributed to dry weather in area of honey production. The moisture content in tested honey samples was in range of codex standard ($\leq 20\%$). The moisture present in honey stabilizes and prevents it from fermentation and granulation. Various beekeeping organizations in Germany, Belgium, Austria, Italy, and Switzerland recommended the moisture content maximum 18-18.5%. The moisture content has important role in shelf life of honey, lower the moisture content longer will be shelf life because it prevent the fermentation of osmotolerant yeast. PAKH-2 and PAKH-6 have the moisture content of 12.5% and 10.1%, respectively, so it might be attributed for its long shelf life and its ability to be stored for long time. The ash contents were between 0.04 and 0.11% (Table 2), since the ash content determination has been omitted from standard and replaced by TDS and EC measurement [34–36]. Color is an important characteristic of honey and varied from region to region. Naturally different tones of honey available like light yellow to amber, dark amber, black, and so forth and sometimes reddish and green hue may be present. The colors of honey are classified according to USDA based on the mm Pfund values; the values less than 8 are classified as white, extra white (9-17 mm), white (18-34 mm), extra light amber (35-50 mm), light amber (51–85 mm), amber (86–114 mm), and more than 114 mm is characterized as dark amber [34]. Among the tested

5-HMF (mg/kg)	21.23 ± 0.11	23.34 ± 0.21	25.65 ± 0.09	31.76 ± 0.24	14.54 ± 0.08	19.81 ± 0.11	25.92 ± 0.73	23.44 ± 0.71	31.32 ± 0.73	25.33 ± 0.43	26.54 ± 0.41	34.53 ± 0.47	37.09 ± 0.08	22.16 ± 0.29	16.31 ± 0.33	22.33 ± 0.11
Color intensity (mAU)	495.22 ± 1.34	699.71 ± 2.12	504.82 ± 0.98	614.31 ± 1.22	386.23 ± 2.34	708.32 ± 3.32	399.14 ± 3.45	382.22 ± 2.11	748.83 ± 3.22	584.82 ± 2.33	589.31 ± 2.13	498.23 ± 2.11	642.32 ± 2.43	493.82 ± 2.33	632.22 ± 2.11	685.71 ± 3.21
Ash (g/100 g)	0.06 ± 0.01	0.04 ± 0.00	0.11 ± 0.01	0.11 ± 0.01	0.06 ± 0.00	0.08 ± 0.03	0.07 ± 0.02	0.04 ± 0.00	0.04 ± 0.01	0.10 ± 0.01	0.22 ± 0.02	0.09 ± 0.01	0.22 ± 0.03	0.21 ± 0.02	0.11 ± 0.01	0.21 ± 0.02
Proline (mg/kg)	175.13 ± 0.09	378.25 ± 0.06	203.44 ± 0.06	192.62 ± 0.01	187.34 ± 0.04	372.11 ± 0.11	153.31 ± 0.09	162.65 ± 0.04	294.43 ± 0.11	303.76 ± 0.08	287.75 ± 0.09	194.33 ± 0.10	353.87 ± 0.11	248.34 ± 0.08	378.66 ± 0.10	354.22 ± 0.07
Protein (mg/g)	2.14 ± 0.13	3.48 ± 0.11	2.83 ± 0.12	3.11 ± 0.09	1.83 ± 0.06	4.32 ± 0.34	1.98 ± 0.08	2.81 ± 0.11	4.76 ± 0.45	3.82 ± 0.32	3.68 ± 0.22	2.91 ± 0.65	4.73 ± 0.76	3.28 ± 0.11	2.74 ± 0.21	3.28 ± 0.23
Total sugar (g/100 g)	65.78 ± 0.86	66.87 ± 0.66	64.45 ± 0.71	64.93 ± 0.43	67.42 ± 0.52	66.51 ± 0.54	65.32 ± 0.55	64.71 ± 0.53	64.97 ± 0.55	64.47 ± 0.32	66.74 ± 0.87	63.41 ± 0.71	62.63 ± 0.92	65.47 ± 0.56	67.75 ± 0.55	65.43 ± 0.43
Reducing sugar (g/100 g)	63.21 ± 0.87	63.53 ± 1.03	62.42 ± 0.98	61.84 ± 0.73	65.34 ± 0.71	63.83 ± 0.68	62.91 ± 0.82	62.22 ± 0.93	62.45 ± 0.74	62.73 ± 0.19	63.82 ± 0.34	61.37 ± 0.78	60.81 ± 0.77	63.32 ± 0.85	65.29 ± 0.65	63.15 ± 0.66
Sucrose (g/100 g)	2.42 ± 0.23	3.21 ± 0.11	1.84 ± 0.22	2.88 ± 0.16	1.93 ± 0.15	2.51 ± 0.27	2.29 ± 0.35	2.32 ± 0.31	2.46 ± 0.45	1.66 ± 0.18	2.77 ± 0.92	1.89 ± 0.33	1.73 ± 0.32	1.98 ± 0.22	2.32 ± 0.37	2.17 ± 0.97
Code name	PAKH-1	PAKH-2	PAKH-3	PAKH-4	PAKH-5	PAKH-6	PAKH-7	PAKH-8	PAKH-9	PAKH-10	PAKH-11	PAKH-12	GERH-13	GERH-14	FRNH-15	FRNH-16

TABLE 2: Sugars, total protein, proline, ash, color intensity, and 5-HMF in sixteen honey samples produced in Pakistan and two other countries (mean \pm SD, n = 3).

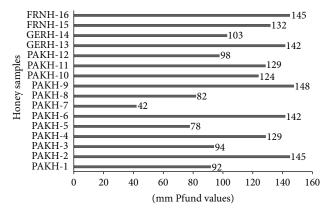


FIGURE 1: Color characteristics of local and exotic honey samples.

honey samples, nine samples were dark amber (124-148 mm Pfund), five samples were amber (92-103 mm Pfund), two samples were light amber (78 and 82 mm Pfund), and one sample was extra light amber (42 mm Pfund). The color characteristics are presented in Figure 1. The color intensities (ABS₄₅₀) of tested honey samples ranged between 382.22 and 748.83 mAU (Table 2). The presence of pigments like carotenoids and flavonoids is related color of honey; dark color honey has large amount of these pigments and known to have highest oxidation potential which is proved in our study. Fresh honey contains trace amount of 5-HMF and this is essential parameter to indicate the purity of honey. The values of 5-HMF in tested honey samples are presented in Table 2; the data indicates that both local and exotic honey samples have 5-HMF value ranging between 14.54 and 37.09 mg/kg. These values are in compliance with standard requirement of 80 mg/kg [32, 35]. Since the label claimed of tested honey samples did not exceed the three-month storage, lower values of 5-HMF obtained in honey samples proved their claim of storage. Various factors are involved in higher values of 5-HMF like prolonged storage or overheating (high temperature storage). 5-HMF values also depend on pH, higher 5-HMF results from acid-catalyzed (low pH) dehydration of hexose sugars with particular more susceptibility of fructose. Overall, low values of 5-HMF in both local and exotic tested honey samples confirm their good quality.

3.2. Biochemical Analysis. Data in Table 2 indicated that the sucrose, reducing sugar, and total sugar contents ranged between 1.73 and 3.21%, 61.37 and 65.34%, and 62.63– 62.75%, respectively. These obtained values are comparable with the results reported for sugars in honey samples in different countries like Nepal, Algeria, and Bangladesh [2, 25, 36]. Our data presented that reducing sugar results are as per recommendation of EC Directive 2001/110 (\geq 60 g/100 g) except the honeydew (\geq 45 g/100 g). The results of sucrose are also in compliance with codex standard [34] that should be lower than 5%. In our study, it is also indication of tested honey samples that are nonadulterated and high values of reducing sugar might be due to its nonconversion into 5-HMF. The protein contents (Table 2) are found in range between 1.83 and 4.76 mg/g of honey. Different enzymes are

responsible for protein in honey. These enzymes are added by bees during the ripening process of honey. The protein results for both monofloral and multifloral in this study are similar to results obtained for honey from Bangladesh, Malaysia, and Algeria. Generally protein contents ranged from 2 to 5 mg/g in honey [2, 10, 22, 25, 26]. The tested honey showed proline contents (Table 2) ranging between the 153.31 and 378.66 mg/kg; there is significant difference between these two values. Carboxylic acid which is found most abundantly in honey is proline. Proline is produced by salivary secretion of honey bees during conversion of nectar into honey. Ripeness and sugar adulteration in honey are determined by amount of proline in honey. Generally proline content ranged between 202 and 680 mg/kg and minimum acceptable limit is 180 mg/kg. The results showed that three samples have lower proline contents than the minimum acceptance level. Some author reported the proline content in wide range (343-1118 mg/kg). Antioxidant activity of honey also correlates with proline contents [22, 37]. In the present study, total phenolic contents (Table 3) were found in tested honey samples between the ranges of 501.42 and 611.62 mg GAE/kg for dark amber color while 362.11-418.43 mg GAE/kg were found for amber color. The phenolic contents of acacia (GERH-14) and jujube (PAKH-11) honey were found to be 595.53 mg GAE/kg and 558.45 mg GAE/kg, respectively, comparable with honey found in Saudi Arabia and Yemen [37]. Quality and curative properties of honey can be determined from its total phenolic contents, antioxidant potential, and total phenolic contents which are in strong correlation with each other. Quality and quantity of phenolics are found in honey according to floral region. Dark color honeys have high phenolic contents than lighter ones and possess high antioxidation potential as indicated in our studies in Pakistani and exotic honey samples [10]. Low molecular weight phenolic compounds are flavonoids and these are vital components for antioxidant properties and aroma of honey. In present study, flavonoids contents in honey samples were determined on base of yellow color complex formation between the Al (III) and carbonyl oxygen and hydroxyl group present in flavonols and flavones. The flavonoids contents (Table 3) found in tested honey samples ranged between 11.83 and 57.66 mg QE/kg. The dark amber color honey has higher flavonoids contents and vice versa. The flavonoids found in both Pakistani and exotic honey samples are similar as reported earlier in Brazilian honey and lower than Burkina Fasan honey (17-83.5 mg QE/kg) by using this method [7, 38]. The ascorbic acid content (Table 3) was found between the ranges of 128.41 and 147.28 mg/kg of honey. The results obtained in tested honey samples of Pakistani and two other countries (Germany and France) are similar to results obtained in Bangladesh, Malaysia, Algeria, Portuguese, and India honey. In addition to phenolics, ascorbic acid present in honey also acts as antioxidant.

3.3. Antioxidant Activities. AEAC was determined by using the standard ascorbic acid calibration curve ($r^2 = 0.9968$) in mg AEAC/100 g of honey. In data in Table 3, the AEAC values ranged from 17.11 to 31.76 mg AEAC/100 g of honey.

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TABLE 3: Total phenolics, total flavonoids, ascorbic acid, AEAC, DPPH, and FRAP of sixteen honey samples produced in Pakistan and two other countries samples (mean \pm SD, n = 3).

Code name	Total phenolics (mg GAE/kg)	Total flavonoids (mg QE/kg)	Ascorbic acid (mg/kg)	AEAC (mg/100 g)	DPPH (% Inhibition)	FRAP (µM Fe(II)/100 g)
PAKH-1	362.11 ± 0.23	45.87 ± 0.41	143.28 ± 0.23	17.11 ± 0.13	57.31 ± 0.97	454.34 ± 1.13
PAKH-2	541.64 ± 0.11	57.66 ± 0.43	143.75 ± 0.32	22.64 ± 0.23	82.23 ± 1.11	646.34 ± 1.11
PAKH-3	373.23 ± 0.43	47.86 ± 0.22	138.32 ± 0.45	25.05 ± 0.13	58.33 ± 0.78	465.74 ± 1.03
PAKH-4	515.42 ± 0.43	51.32 ± 0.45	141.34 ± 0.71	31.76 ± 0.24	66.87 ± 0.54	502.02 ± 0.91
PAKH-5	295.11 ± 0.52	28.92 ± 0.64	126.78 ± 0.92	17.04 ± 0.28	44.23 ± 0.11	255.33 ± 1.52
PAKH-6	588.44 ± 0.78	49.87 ± 0.53	147.23 ± 0.33	19.31 ± 0.15	82.13 ± 0.32	595.33 ± 1.18
PAKH-7	217.33 ± 0.92	11.83 ± 0.65	128.41 ± 0.26	25.42 ± 0.13	39.91 ± 0.54	238.21 ± 0.12
PAKH-8	301.33 ± 0.11	18.32 ± 0.71	131.34 ± 0.15	22.94 ± 0.11	47.22 ± 0.37	311.82 ± 0.41
PAKH-9	611.62 ± 0.42	52.15 ± 0.57	147.28 ± 0.44	30.12 ± 0.23	88.45 ± 0.82	672.02 ± 1.42
PAKH-10	521.34 ± 0.76	48.32 ± 0.78	142.21 ± 0.54	26.03 ± 0.81	63.46 ± 0.11	511.77 ± 1.16
PAKH-11	558.45 ± 0.83	42.11 ± 0.34	139.36 ± 0.67	26.66 ± 0.71	65.82 ± 0.23	548.15 ± 1.13
PAKH-12	401.38 ± 0.65	38.23 ± 0.31	131.42 ± 0.81	32.53 ± 0.87	61.37 ± 0.88	491.33 ± 1.47
GERH-13	595.53 ± 0.51	48.86 ± 0.43	145.58 ± 0.43	31.09 ± 0.08	81.81 ± 0.43	598.11 ± 1.21
GERH-14	418.43 ± 0.52	39.83 ± 0.44	129.72 ± 0.55	22.18 ± 0.09	62.42 ± 0.55	498.77 ± 1.12
FRNH-15	501.42 ± 0.73	56.21 ± 0.32	138.46 ± 0.86	18.31 ± 0.48	67.45 ± 0.33	521.02 ± 0.91
FRNH-16	553.37 ± 0.78	53.11 ± 0.34	139.58 ± 0.18	23.36 ± 0.81	64.15 ± 0.72	553.88 ± 1.18

These values are similar to already reported values for honey from different countries. The higher values have higher oxidation potential [2, 5, 10, 25, 26, 39]. DPPH scavenging activity was determined by using 0.2 g/mL concentration of each of honey sample. The percentage free radical inhibition (Table 3) of tested samples ranged between 39.91 and 88.45%. Those honey samples having dark color had higher oxidation potential with higher percentage of inhibition of DPPH. The dark amber color honeys have percentage inhibition values 62.42-88.45% while amber color honeys have 57.31-61.37% inhibition. The lowest % inhibition is achieved by the extra light amber honey (39.91%). Free radical scavenging ability of various samples is determined by DPPH-stable nitrogen centered radical. Since antioxidant potential of honey is associated with phenolics and flavonoids contents, DPPH is frequently used for determining free radical scavenging ability. Higher the % inhibition value superior will be the antioxidant activity of sample. The diseases with free radical origin can be treated with honey because it has ability to scavenge the free radical [2]. The FRAP values (Table 3) of tested honey samples (both monofloral and multifloral) ranged from 255.33 to $672.02 \,\mu\text{M}$ Fe (II)/100 g. These values are similar to that already reported in honey samples of different countries such as Slovenia, Cuba, Algeria, India, Malaysia, and Bangladesh. Again the honey samples having higher % inhibition values of DPPH attributed to higher FRAP values and vice versa. Basically FRAP is reducing test which reflects the ability of substance to break the free radical chain reaction. Breaking the free radical chain reaction is indicator of antioxidant capacity. FRAP assay directly estimates the presence of reductant or antioxidant in samples which reduce the Fe (III) into Fe (II) [2].

3.4. *Mineral Contents.* The mineral contents (macro, micro, and trace) found in honey samples are presented in Table 4. The values obtained in our study are similar to results obtained in Brazilian and Romanian honey. The composition of mineral contents largely depends on the climatic conditions, geographical area, and floral sources. Other than the phenolics and flavonoids, copper and iron also have reducing or antioxidant properties [40].

3.5. Correlation between Antioxidant Properties, Biochemical Parameters, and Color Intensity. There is significant correlation (Table 5) found between the antioxidant properties, biochemical properties, and color intensity. A strong correlation was observed for flavonoids and phenolics with color intensity. As it was earlier reported in studies, the dark color honey contains higher phenolic and flavonoids contents and also our study on twelve Pakistani and four exotic honey samples proved the same result; that is, dark color honeys have strong antioxidant properties. The proline content is also a contributor in antioxidant properties of honey [31] which is in strong correlation (r = 0.824 and 0.727) with flavonoids and phenolic contents. Phenolic contents have strong positive correlation with DPPH and FRAP (r = 0.914 and 0.930) as compared to flavonoids for same correlation (r = 0.799 and 0.876). Another antioxidant ascorbic acid is in strong positive correlation with DPPH and FRAP (r = 0.832 and 0.816) but not more potent than flavonoids and phenolics. This correlation demonstrated the overall antioxidant properties of tested honey samples. Color pigments, phenolics, flavonoids, and proline are attributed to antioxidant properties of honey as expressed in correlation values. In conclusion, we investigated the physiochemical, biochemical, mineral contents,

PAKH-12

GERH-13

GERH-14

FRNH-15

FRNH-16

	TABLE	4: Minera	l contents	in sixteen	honey sa	mples proc	luced in Pa	akistan and	l two oth	er countri	es sample	es.		
Code name		Mineral contents (mg/kg)												
Code manie	Ca	Cd	Co	Cr	Cu	Fe	Κ	Mg	Mn	Na	Ni	Pb	Zn	
PAKH-1	57.23	0.91	0.015	0.021	0.21	19.11	400.2	54.54	0.73	150.2	0.11	0.075	2.14	
PAKH-2	70.11	0.83	0.017	0.031	0.23	23.32	394.2	67.78	0.78	144.4	0.13	0.084	3.23	
PAKH-3	53.65	0.76	0.009	0.012	0.26	21.35	432.5	43.25	0.76	122.8	0.16	0.065	2.32	
PAKH-4	61.76	0.87	0.008	0.023	0.27	11.56	432.6	51.36	0.97	152.4	0.17	0.047	2.14	
PAKH-5	76.11	0.92	0.007	0.034	0.22	16.75	456.7	44.45	0.83	166.6	0.21	0.032	3.19	
PAKH-6	85.24	0.97	0.018	0.032	0.27	15.33	511.8	55.33	0.73	141.3	0.11	0.034	3.64	
PAKH-7	72.22	0.83	0.019	0.034	0.23	12.32	345.9	72.02	0.83	145.4	0.25	0.025	3.11	
PAKH-8	82.82	0.73	0.009	0.025	0.23	22.32	346.9	56.12	0.73	176.6	0.24	0.028	2.54	
PAKH-9	72.32	0.78	0.007	0.026	0.28	22.72	445.4	56.32	0.78	155.6	0.17	0.029	2.32	
PAKH-10	82.11	0.76	0.018	0.025	0.26	25.66	421.7	62.76	0.76	161.7	0.18	0.067	2.45	
PAKH-11	52.93	0.92	0.015	0.024	0.22	34.33	441.8	55.87	0.83	181.5	0.18	0.078	2.67	

33.42

21.52

22.22

14.43

12.53

445.6

421.7

411.4

398.8

501.5

63.22

48.92

72.07

54.18

52.03

0.76

0.87

0.92

0.83

0.76

175.4

181.4

181.7

168.5

171.4

0.19

0.21

0.22

0.25

0.22

0.074

0.053

0.066

0.046

0.054

2.87

2.23

3.12

2.18

3.11

. . . .

TABLE 5: Correlation among biochemical parameters, antioxidant properties, and color intensity.

	Flavonoids	Ascorbic acid	Protein	Proline	DPPH	FRAP	ABS_{450}
Phenolics	0.835**	0.813**	0.866**	0.824**	0.914**	0.930**	0.770^{**}
Flavonoids		0.768^{**}	0.548^{*}	0.727**	0.799**	0.876**	0.648^{**}
Ascorbic acid			0.725**	0.621*	0.832**	0.816**	0.859**
Protein				0.665**	0.869**	0.821**	0.657**
Proline					0.771^{**}	0.767**	0.595*
DPPH						0.956**	0.757**
FRAP							0.718^{**}

** Correlation is significant at p < 0.01.</p>

83.42

88.22

82.37

74.48

72.43

0.98

0.84

0.76

0.87

0.95

0.014

0.019

0.008

0.009

0.007

0.022

0.035

0.016

0.018

0.032

0.28

0.24

0.26

0.27

0.25

*Correlation is significant at p < 0.05.

and oxidation potential of honey available in commercial market for first time in Pakistan. Both the monofloral (jujube and acacia) and multifloral honeys were tested. Tested honey samples have lower contents of 5-HMF which might be due to acidic pH and low moisture content which prevent its formation. Higher oxidation potential possessed by honey is indicated from their phenolics and flavonoids contents. Total sugar contents are found in range given by codex standard which is indication of nonadulterated honey. The mineral contents are also investigated because iron and copper also have antioxidant properties. Strong correlation was also found between the biochemical and antioxidant agents. Phenolics have most significant positive correlation with DPPH and FRAP. Overall both Pakistani and exotic honey samples have strong oxidation potential which can be used for treatment of free radical origin diseases.

Conflict of Interests

The authors declare that there is no conflict of interests regarding the publication of this paper.

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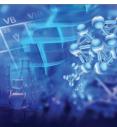
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