

## Research Article

# Seasonal Changes and Spatial Variation in Water Quality of a Large Young Tropical Reservoir and Its Downstream River

# Teck-Yee Ling,<sup>1</sup> Norliza Gerunsin,<sup>2</sup> Chen-Lin Soo,<sup>1</sup> Lee Nyanti,<sup>1</sup> Siong-Fong Sim,<sup>1</sup> and Jongkar Grinang<sup>3</sup>

<sup>1</sup>Faculty of Resource Science and Technology, Universiti Malaysia Sarawak, 94300 Kota Samarahan, Sarawak, Malaysia
 <sup>2</sup>Universiti Teknologi MARA, Kota Samarahan Campus, Jalan Meranek, 94300 Kota Samarahan, Sarawak, Malaysia
 <sup>3</sup>Institute of Biodiversity and Environmental Conservation, Universiti Malaysia Sarawak, 94300 Kota Samarahan, Sarawak, Malaysia

Correspondence should be addressed to Teck-Yee Ling; teckyee60@gmail.com

Received 18 January 2017; Revised 18 May 2017; Accepted 19 June 2017; Published 26 July 2017

Academic Editor: Wenshan Guo

Copyright © 2017 Teck-Yee Ling et al. This is an open access article distributed under the Creative Commons Attribution License, which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

This study examined the water quality of the large young tropical Bakun hydroelectric reservoir in Sarawak, Malaysia, and the influence of the outflow on the downstream river during wet and dry seasons. Water quality was determined at five stations in the reservoir at three different depths and one downstream station. The results show that seasons impacted the water quality of the Bakun Reservoir, particularly in the deeper water column. Significantly lower turbidity, SRP, and TP were found during the wet season. At 3–6 m, the oxygen content fell below 5 mg/L and hypoxia was also recorded. Low  $NO_2^{-}$ -N,  $NO_3^{-}$ -N, and SRP and high BOD<sub>5</sub>, OKN, and TP were observed in the reservoir indicating organic pollution. Active logging activities and the dam construction upstream resulted in water quality deterioration. The outflow decreased the temperature, DO, and pH and increased the turbidity and TSS downstream. Elevated organic matter and nutrients downstream are attributable to domestic discharge along the river. This study shows that the downstream river was affected by the discharge through the turbines, the spillway operations, and domestic waste. Therefore, all these factors should be taken into consideration in the downstream river management for the health of the aquatic organisms.

## 1. Introduction

The creation of a large-scale dam and its associated reservoir often has significant negative impacts on the hydrological, biological, and chemical processes of the reservoir, upstream, and downstream of the dam [1–9]. The Bakun hydroelectric dam, which was impounded from 2010 to 2012 on the Balui River in Malaysia, produced these effects. The dam which is one of the tallest concrete rock filled dams (205 m) in the world created a reservoir covering a surface area of 695 km<sup>2</sup>. A few pre- and postimpoundment studies on the physicochemical parameters of the Bakun Dam reservoir have been performed [10–12]. However, the reservoir water quality is likely changing as the reservoir is receiving loads of pollutants from adjacent anthropogenic activities during its operation [13, 14]. Water quality deterioration is a common

problem in reservoirs surrounded with anthropogenic activities receiving high loads of suspended solids, organic matter, and nutrients [15, 16].

The water quality of reservoirs has been observed to vary seasonally in tandem with changes in temperature and rainfall [17–19]. The low and high precipitation during dry and wet seasons in a tropical country like Malaysia can greatly change the water quality of the reservoir. The high precipitation during the wet season can either decrease the pollutant concentration by dilution or deteriorate the reservoir water quality due to increased surface runoff from anthropogenic activities. Reference [20] demonstrated that the levels of total phosphorus in Batang Ai Reservoir during the rainy season and high water levels were lower than those observed during the dry season and low water levels. Besides, high volume of inflow following heavy rainfall promotes mixing and disturbs stratification in the reservoir. The increase of bottom



FIGURE 1: The study area and sampling stations in the present study.

dissolved oxygen level in the well-mixed reservoir inhibits the release of nutrients from sediments causing a rapid reduction of phytoplankton concentration in the reservoir [17].

On the other hand, the reservoir outflow has a great influence on the downstream river. Studies have shown that the downstream river is subjected to major environmental impacts which range from downstream morphology changes to loss of biodiversity of the ecosystem [1, 5, 7, 8, 21]. The reservoir outflow is often controlled by the electrical demand and operation cost, independent of ecological considerations in the downstream river. Differences in structure and operation scheme of a dam may result in differences in water quality downstream. Recently, [22] demonstrated that the physicochemical characteristics of the river downstream of the Bakun Dam changed when the spillway was opened.

As a young reservoir in a tropical country, changes continue to occur in the reservoir and it is important to monitor the water quality in order to evaluate its suitability for secondary purposes such as aquaculture and recreation. The knowledge of the seasonal variation of the reservoir's water quality is important for dam operation and management decision. The impact of the dam on the water quality of its downstream river during the wet and dry seasons remains unknown. Hence, the aim of this study was to assess the water quality of the Bakun Reservoir and the influence of its outflow on the water quality of the downstream river during wet and dry seasons.

## 2. Materials and Methods

2.1. Study Area and Sampling Stations. The present study was conducted at Bakun Reservoir and its downstream river in Sarawak, Malaysia, as illustrated in Figure 1. The Bakun hydroelectric dam was built across the Batang Balui with a total of eight installed turbines and a spillway weir located at 209 m above sea level. The reservoir covers mainly the Balui River that is fed by three major tributaries, namely,

the Murum River, Linau River, and Bahau River. A total of five stations were selected at the Bakun Reservoir and one station was selected at the downstream river. Stations 1 and 2 were located at the Batang Balui and Linau River, respectively. Stations 3 and 4 were located at the Murum River where the upstream Murum hydroelectric dam was under ongoing construction during the time of sampling. Station 5 was located in the proximity of the Bakun hydroelectric dam and downstream of active logging activities while Station 6 was located at the downstream river approximately 4.3 km from the dam.

Sampling was conducted in February and September 2014 corresponding to the wet and dry seasons in Sarawak (Table 1). There was no rain recorded during the two and three weeks prior to the first and second samplings, respectively. The water level during the second sampling in the dry season was approximately 7 m lower than the water level during the wet season. The water release during hydropower generation is drawn from the top 10 m of the reservoir using selective withdrawal intake structures. Occasionally, additional water is released from the spillway with intake at a depth of approximately 15 m. At the end of the spillway, the water hits the concrete barrier before entering Balui River downstream. Sampling was conducted during electrical power generation where the downstream river received the water discharged from the reservoir after the water passed through the turbines. During the first sampling, additional water was discharged from the spillway at a rate of 501 m<sup>3</sup>/s in addition to the turbine outflow  $(536 \text{ m}^3/\text{s})$ . The spillway was closed during the second sampling; hence, Station 6 was receiving solely the turbine outflow at a rate of  $730 \text{ m}^3/\text{s}$ .

2.2. Field Collection and Laboratory Analysis. Depth profiles of temperature and dissolved oxygen (DO) were measured using a YSI 6820 V2 multiparameter water quality sonde during the first sampling in February 2014. The pH and turbidity were measured at 0 m, 10 m, and 20 m depths in Bakun Reservoir in both samplings by using a pH meter (EcoScan, Eutech) and a turbidity meter (Martini Instruments, Mi415), respectively. Triplicate water samples were collected at 0 m, 10 m, and 20 m depths in Bakun Reservoir (Stations 1 to 5) using a Van Dorn water sampler whereas triplicate water samples were collected at 0 m depth at the downstream river of the dam (Station 6). The depth of the reservoir was measured using a portable depth sounder (Speedtech). All sampling bottles were acid-washed, cleaned, and dried before use. Water samples were acidified to pH < 2 for total phosphorus (TP) analysis. All samples were placed in an ice box and transported to the laboratory for further analysis [23].

All the analyses were conducted according to standard methods [23, 24]. Chlorophyll *a* (Chl *a*) was determined from adequate samples filtered through 0.45  $\mu$ m glass fiber filter (Whatman GF/F) and extracted for 24 h using 90% (v/v) acetone. The absorbance was read using a DR 2800 spectrophotometer and concentration of Chl *a* was calculated according to [25]. Total suspended solid (TSS) was calculated as the difference between the initial and final weights of the

Station	Coordinates	Date	Location
Bakun hydr	oelectric reservoir		
St. 1	N 02°43′34.4″	26 Feb. 2014, 1:15 p.m.	Batang Balui
	E 114°01′44.2″	24 Sept. 2014, 8:15 a.m.	Sunny during both sampling trips
St. 2	N 02°39′32.2″	26 Feb. 2014, 9:45 a.m.	Linau River
	E 114°03′29.5″	24 Sept. 2014, 10:45 a.m.	Sunny during both sampling trips
St. 3	N 02°42′59.8″ E 114°09′43.8″	27 Feb. 2014, 12:55 p.m. 25 Sept. 2014, 9:51 a.m.	Upper part of Murum River Sunny during both sampling trips Soil erosion was observed in the upper Murum River bank
St. 4	N 02°44′15.3″	26 Feb. 2014, 3:06 p.m.	Lower part of Murum River
	E 114°05′16.6″	24 Sept. 2014, 1:42 p.m.	Sunny during both sampling trips
St. 5	N 02°45′09.8″	27 Feb. 2014, 3:00 p.m.	Near the intake point and the dam
	E 114°02′32.9″	25 Sept. 2014, 2:08 p.m.	Cloudy during both sampling trips
Downstream	n river of Bakun hydroele	ctric dam	
St. 6	N 02°46′21.8″	26 Feb. 2014, 3:00 p.m.	Long Baagu (4.3 km downstream of the Bakun hydroelectric dam)
	E 114°01′41.6″	24 Sept. 2014, 1:35 pm	Sunny during both sampling trips

TABLE 1: The details of the sampling location and sampling regime in the present study.

 $0.45 \,\mu\text{m}$  glass fiber filter (Whatman GF/F), after filtration of an adequate sample volume and drying at 105°C. Fiveday biochemical oxygen demand (BOD<sub>5</sub>) was determined as the difference between the initial and five-day DO content, after five-day-long incubation of the sample. The initial DO content was determined in the field and increased by vigorous aeration if the DO value was low. NO<sub>2</sub><sup>-</sup>-N and NO<sub>3</sub><sup>-</sup>-N levels were determined by the diazotization method (low range) and the cadmium reduction method, respectively, after filtering through a 0.45  $\mu$ m glass fiber filter (Whatman GF/F). Organic Kjeldahl nitrogen (OKN) was determined by the Macro-Kjeldahl Method where ammonia was removed from the water sample before digestion and distillation, followed by Nessler's method. SRP was determined by the colorimetric ascorbic acid method after filtering through a 0.45  $\mu$ m glass fiber filter (Whatman GF/F). TP was determined by the ascorbic acid method after persulfate digestion of samples. The estimated detection limits of NO<sub>2</sub><sup>-</sup>-N, NO<sub>3</sub><sup>-</sup>-N, and SRP were  $0.005 \text{ mg/L NO}_2^-$ -N,  $0.01 \text{ mg/L NO}_3^-$ -N, and 0.02 mg/L $PO_4^{3-}$ , respectively.

Quality control steps were taken throughout the study. Sample bottles and glassware were washed using phosphatefree detergent followed by the standard acid wash procedure. Sample preparation and storage were performed according to the standard methods [23]. Triplicate blank water that was free of the analytes of interest was used in the same procedure for each of the aforementioned analyses.

2.3. Statistical Analysis. Comparison of water quality parameters between the stations and the depths in the Bakun hydroelectric reservoir was conducted using one-way ANOVA and Tukey's pairwise comparisons with 5% significance level. Student's *t*-test was used to compare the water quality of the reservoir between the wet and dry seasons. Pearson's correlation analysis was performed to determine the relationship among all the parameters in the reservoir during each season. The water quality of the downstream river between the wet and dry seasons and the results between the intake point of the dam and the downstream river were compared using Student's *t*-test. Cluster analysis (CA) was used to investigate the grouping of the sampling stations with different depths by using the water quality parameters collected in the reservoir and the downstream river. *Z*-score standardization of the variables and Ward's method using Euclidean distances as a measure of similarity were used. All the statistical analyses were carried out by using the Statistical Package for the Social Sciences (SPSS Version 22, SPSS Inc., 1995).

## 3. Results and Discussion

3.1. Water Quality of Bakun Reservoir. Figure 2 illustrates the vertical stratification in Bakun Reservoir, indicating poor water mixing in the reservoir. Among the five sampling stations in the Bakun Reservoir, Station 2, which is located at Linau River, is stratified into three distinct layers of different temperatures. The thermocline layer observed at 3 m to 7 m separates the epilimnion ( $\approx 30.5^{\circ}$ C) and hypolimnion ( $\approx 25.5^{\circ}$ C) at Station 2. Similarly, [10, 11] reported that the thermocline started at a depth of 4-5 m and between 6 m and 9 m in Bakun Reservoir during the filling phase and 13 months after reaching the full-supply level, respectively. Thermal stratification in reservoirs has been widely reported in tropical and subtropical reservoirs [19, 26-28]. The temperature gradient within the thermocline layer in the Bakun Reservoir is in agreement with the range of thermal stratification of 0.5°C to 5°C for a tropical reservoir [29].

Dissolved oxygen was relatively consistent in the surface water of the Bakun Reservoir, with a mean value of 7.22 mg/L. The DO level started to decrease rapidly from a depth of 2 m to less than 0.2 mg/L at a depth of 4 m at Station 1 which is located at Batang Balui. The DO level at Stations 2, 3, and 5 started to decrease rapidly from the depth of around 3 m whereas the DO level at Station 4 started to decrease from 5 m depth. In other words, the healthy level of DO content above 5 mg/L was only observed at the water column above 3–6 m in Bakun Reservoir. Similarly, [26]



FIGURE 2: Depth profile of temperature and DO in Bakun Reservoir in February 2014.

showed that oxygen depletion is a common phenomenon in the hypolimnia of Indonesian lakes and reservoirs with different oxycline depths. The authors attributed the shallow oxycline depth and thick anoxic layer in the Cirata Reservoir to the weak wind-induced mixing and high organic loads that lead to rapid decomposition and oxygen depletion in the reservoir. On the other hand, the DO concentration never fell below 2 mg/L in Qiandaohu Lake, China, where the DO depth profiles were closely linked to the water temperature depth profiles [19]. The decrease of DO with depth is commonly observed in reservoirs as photosynthesis increases oxygen level in the surface water while respiration of bacteria decomposing dead organic matter consumes all the dissolved oxygen in the bottom water column coupled with insufficient exchange with oxygenated surface water [30]. However, a slight increase of DO content was observed at the water column of the Bakun Reservoir between 12 m and 20 m which is most likely due to the additional water discharged from the spillway where the water intake was at a depth of approximately 15 m. The rapid water movement due to the additional water withdrawal at the particular water column promotes the mixing of the low DO water with a large volume of oxygenated colder water inflow from tributaries around the reservoir [14]. This phenomenon was not observed in the study in [11] where the DO content was reported as undetectable from a depth of 7 m up to a depth of 30 m as the reservoir water was not discharged from the spillway during this study.

The pH value of the Bakun Reservoir ranged from  $4.93 \pm 0.06$  to  $8.06 \pm 0.05$  during the wet season with the lowest and highest pH value being observed at Station 5 and Station 2, respectively. On the other hand, the pH value of the Bakun Reservoir is relatively consistent during the dry season with a mean value of 7.30. Vertical distribution of pH values in Bakun Reservoir differed between the wet and dry seasons although this was not significantly different (*p* value > 0.05) (Table 2). During the dry season, the pH value of the Bakun Reservoir decreased as depth increased up to a depth of

10 m and remained at a similar value up to a depth of 20 m as illustrated in Figure 3. The vertical distribution of pH values during the dry season in the present study is in good agreement with the previous study in the Bakun Reservoir [11] and the Batang Ai Reservoir [31] where the pH value of the reservoir water decreased as depth increased. However, the pH value tends to increase with depth when the surface pH value is low as demonstrated by Stations 3 and 5 during the wet season. The results showed that the low pH value at the surface water was diluted by the reservoir water with higher pH value as depth increased. The dilution in the water column improved the pH at Station 3 from 6.3 to 6.8. However, despite the dilution in the water column, Station 5, which was the closest station to the dam, still showed pH values of less than 6.5 mg/L. On the other hand, when the pH value was high (>7), the pH value decreased as depth increased which is similar to vertical pH distribution during the dry season. The surface pH value was classified as Class I but was changed to Class II as depth increased according to the National Water Quality Standard (NWQS) for Malaysia [32] during the dry season. During the wet season, the pH values of the Bakun Reservoir were classified as Class I except for Stations 3 and 5. The surface water at Station 3 was classified as Class II while the extremely low surface pH value of 4.9 at Station 5 exceeded the NWQS. Besides, the pH values at Station 5 at depths of 10 m and 20 m were classified as Classes II and III, respectively.

Table 3 shows that no significant correlation (p value > 0.05) was found between the pH value of the Bakun Reservoir and the other parameters during the wet season suggesting that the pH value of the Bakun Reservoir, particularly Stations 3 and 5, was mainly influenced by the low pH surface runoff from the anthropogenic activities in the surrounding area. Stations 3 and 5 were located downstream of the construction site of the Murum hydroelectric dam and active logging activities. The decomposition of organic matter derived from anthropogenic activity acidified the upstream rivers that flow into the reservoir and caused the acidification at the stations.

TABLE 2. INICALL		iici quanty p			e uir wei season ann e	ما كم عدمة ما المسلمات	TIT T.COT HAT & ATTA OCH	hremon zor.	T) Traphentivery (	·(c - k)
Parameter	Sampling	Depth	1	2	Station 3	4	IJ	Mean	Difference	<i>p</i> value
		0 m	$7.52 \pm 0.02^{d,3}$	$8.06 \pm 0.05^{e,3}$	$6.34 \pm 0.02^{b,1}$	$7.37 \pm 0.02^{c,3}$	$4.93 \pm 0.06^{a,1}$			
	Wet season	10 m	$6.81 \pm 0.02^{\mathrm{b},2}$	$6.83 \pm 0.01^{\mathrm{bc,2}}$	$6.78 \pm 0.02^{b,2}$	$6.90 \pm 0.02^{c,2}$	$6.22 \pm 0.06^{a,3}$	6.67		
Нч		$20 \mathrm{m}$	$6.52 \pm 0.01^{\mathrm{b,1}}$	$6.63 \pm 0.02^{c,1}$	$6.76 \pm 0.01^{d,2}$	$6.66 \pm 0.02^{c,1}$	$5.65 \pm 0.02^{a,2}$		000+	0 477
hıı		0 m	$7.21 \pm 0.01^{b,3}$	$7.01 \pm 0.02^{a,2}$	$7.56 \pm 0.01^{d,3}$	$7.36 \pm 0.01^{c,3}$	$7.36 \pm 0.00^{c,3}$		10.20	771.0
	Dry season	10 m	$5.96 \pm 0.01^{ m a,1}$	$5.95 \pm 0.00^{ m a,l}$	$6.17 \pm 0.00^{d,1}$	$6.11 \pm 0.01^{c,1}$	$6.03 \pm 0.01^{b,2}$	6.47		
		20 m	$6.03 \pm 0.00^{\mathrm{b},2}$	$5.95 \pm 0.01^{ m a,1}$	$6.24 \pm 0.00^{ m d,2}$	$6.14 \pm 0.00^{c,2}$	$5.95 \pm 0.01^{\mathrm{a},1}$			
		0 m	$0.06 \pm 0.01^{a,1}$	$2.65 \pm 0.04^{\mathrm{b,1}}$	$5.16 \pm 0.29^{d,1}$	$3.77 \pm 0.12^{c,1}$	$3.49 \pm 0.19^{c,1}$			
	Wet season	10 m	$34.63 \pm 0.12^{b,2}$	$33.81 \pm 0.49^{\mathrm{b},2}$	$40.63 \pm 0.82^{c,2}$	$63.00 \pm 1.00^{ m d,2}$	$16.09 \pm 2.27^{a,2}$	38.89		
Turbidity FNII		20 m	$48.29 \pm 0.72^{b,3}$	$51.00 \pm 0.00^{b,3}$	$128.00 \pm 4.36^{c,3}$	$131.33 \pm 0.58^{c,3}$	$21.46 \pm 1.56^{a,3}$		72 22	0.000
o tra chanty, the		0 m	$2.39 \pm 0.05^{a,1}$	$3.66 \pm 0.01^{ m b,1}$	$4.07 \pm 0.03^{c,1}$	$4.35 \pm 0.03^{ m d,1}$	$3.84 \pm 0.13^{ m b,1}$		CC:71	07070
	Dry season	10 m	$63.33 \pm 0.58^{\mathrm{b},2}$	$38.33 \pm 0.07^{a,2}$	$261.33 \pm 1.53^{e,2}$	$102.00 \pm 1.00^{ m d,2}$	$67.00 \pm 0.00^{c,2}$	81.24		
		20 m	$78.33 \pm 0.58^{b,3}$	$45.58\pm 0.16^{a,3}$	$263.67 \pm 1.53^{e,2}$	$199.67 \pm 0.58^{ m d,3}$	$81.00 \pm 0.58^{c,3}$			
		0 m	$1.41 \pm 0.23^{\mathrm{ab,2}}$	$1.45 \pm 0.22^{\mathrm{ab,1}}$	$2.30 \pm 0.81^{b,2}$	$0.62 \pm 0.21^{ m a,1}$	$0.64 \pm 0.15^{a,2}$			
	Wet season	10 m	$0.90 \pm 0.00^{\mathrm{b},1}$	$1.50 \pm 0.45^{c,1}$	$0.26 \pm 0.06^{a,1}$	$0.71 \pm 0.06^{\mathrm{ab,1}}$	$0.37 \pm 0.10^{\mathrm{ab},12}$	0.93		
		20 m	$0.76 \pm 0.06^{\mathrm{b},1}$	$1.59 \pm 0.02^{c,1}$	$0.57 \pm 0.22^{\mathrm{b},\mathrm{l}}$	$0.75 \pm 0.03^{\mathrm{b},\mathrm{l}}$	$0.17 \pm 0.12^{ m a,1}$		0.18	0 655
ош и, <i>р</i> 8/ г		0 m	$1.30 \pm 0.09^{\mathrm{ab,3}}$	$1.76 \pm 0.57^{\mathrm{ab,2}}$	$2.57 \pm 0.55^{\mathrm{b},3}$	$0.64 \pm 0.02^{\mathrm{a},3}$	$5.87 \pm 0.99^{c,2}$		01.0	CC0.0
	Dry season	10 m	$0.62 \pm 0.04^{ m d,2}$	$0.38 \pm 0.06^{ m bc,1}$	$0.18 \pm 0.05^{a,1}$	$0.35 \pm 0.07^{ m ab,2}$	$0.55\pm0.11^{\mathrm{cd,1}}$	1.11		
		20 m	$0.40 \pm 0.08^{\mathrm{b,1}}$	$0.33 \pm 0.01^{\mathrm{b,1}}$	$1.20 \pm 0.11^{c,2}$	$0.10 \pm 0.00^{\mathrm{a},1}$	$0.39 \pm 0.07^{\mathrm{b,1}}$			
		0 m	$7.8 \pm 1.0^{a,1}$	$6.7 \pm 1.7^{a,1}$	$13.3 \pm 0.0^{\rm b,1}$	$8.0 \pm 2.0^{a,1}$	$9.1\pm0.8^{a,1}$			
	Wet season	10 m	$43.2 \pm 1.1^{a,2}$	$49.9 \pm 1.8^{b,2}$	$55.3 \pm 1.2^{c,2}$	$64.5 \pm 0.8^{ m d,2}$	$40.3 \pm 0.6^{a,2}$	40.1		
TSS ma/I		20 m	$63.3 \pm 1.2^{bc,3}$	$60.7 \pm 1.2^{b,3}$	$60.7 \pm 1.2^{b,3}$	$66.0 \pm 0.7^{c,2}$	$53.0 \pm 1.0^{a,3}$		-206	0 147
100, mg/ L		0 m	$6.0\pm0.0^{\mathrm{b},1}$	$10.2\pm0.3^{ m c,1}$	$14.2 \pm 0.5^{d,1}$	$4.1\pm0.2^{\mathrm{a},1}$	$6.1 \pm 0.5^{b,1}$		0.77-	1110
	Dry season	10 m	$45.4 \pm 0.9^{\mathrm{b},2}$	$30.4 \pm 1.2^{a,2}$	$170.2 \pm 1.1^{c,2}$	$82.7 \pm 2.3^{c,2}$	$48.6 \pm 1.1^{ m b,2}$	69.7		
		20 m	$60.2 \pm 0.9^{b,3}$	$38.1 \pm 0.3^{a,3}$	$328.9 \pm 1.9^{e,3}$	$131.8 \pm 1.4^{ m d,3}$	$68.5 \pm 0.7^{c,3}$			
		0 m	$3.24 \pm 0.19^{a,2}$	$3.42 \pm 0.26^{a,1}$	$3.97 \pm 0.08^{b,1}$	$3.36 \pm 0.22^{a,1}$	$4.30 \pm 0.15^{b,1}$			
	Wet season	10 m	$3.79 \pm 0.19^{a,3}$	$4.11 \pm 0.06^{\mathrm{ab,2}}$	$5.32 \pm 0.12^{c,3}$	$4.30 \pm 0.19^{b,2}$	$4.44 \pm 0.06^{\mathrm{b},1}$	4.02		
BOD_ma/L		$20 \mathrm{m}$	$2.23 \pm 0.01^{ m a,1}$	$4.63 \pm 0.24^{c,3}$	$4.55 \pm 0.03^{c,2}$	$4.11 \pm 0.07^{b,2}$	$4.50 \pm 0.08^{c,1}$		-0.26	0 135
		0 m	$3.90 \pm 0.10^{\mathrm{b},2}$	$4.27 \pm 0.06^{cd,1}$	$4.13 \pm 0.06^{c,1}$	$3.47 \pm 0.06^{\mathrm{a},1}$	$4.43 \pm 0.06^{d,1}$			
	Dry season	10 m	$5.13 \pm 0.06^{d,3}$	$4.70 \pm 0.10^{c,2}$	$4.27 \pm 0.06^{\mathrm{b},\mathrm{l}}$	$3.87 \pm 0.06^{\mathrm{a},2}$	$4.63 \pm 0.06^{c,2}$	4.28		
		20 m	$3.60 \pm 0.10^{a,1}$	$4.50 \pm 0.10^{c,2}$	$5.07 \pm 0.06^{d,2}$	$3.87 \pm 0.10^{\mathrm{b},2}$	$4.40 \pm 0.10^{c,1}$			

## Journal of Chemistry

Parameter	Sampling	Depth	-	6	Station 3	4	S	Mean	Difference	<i>p</i> value
NO <sub>2</sub> <sup>-</sup> -N, mg/L	Wet season Dry season	0 m 10 m 20 m 0 m 20 m	$\begin{array}{c} 0.002 \pm 0.001^{a,1}\\ 0.001 \pm 0.001^{a,1}\\ 0.002 \pm 0.001^{a,1}\\ 0.002 \pm 0.001^{a,1}\\ 0.001 \pm 0.001^{a,1}\\ 0.001 \pm 0.001^{a,1}\\ \end{array}$	$\begin{array}{c} 0.003 \pm 0.001^{a,1} \\ 0.004 \pm 0.001^{b,1} \\ 0.003 \pm 0.001^{ab,1} \\ 0.001 \pm 0.000^{a,1} \\ 0.001 \pm 0.000^{a,1} \\ 0.001 \pm 0.001^{a,1} \end{array}$	$\begin{array}{c} 0.004 \pm 0.001^{a,l} \\ 0.007 \pm 0.001^{c_2} \\ 0.005 \pm 0.001^{b_{c,12}} \\ 0.007 \pm 0.001^{b_2} \\ 0.001 \pm 0.001^{a,l} \\ 0.002 \pm 0.001^{a,l} \end{array}$	$\begin{array}{c} 0.003 \pm 0.001^{a,l}\\ 0.003 \pm 0.001^{b,l}\\ 0.003 \pm 0.001^{ab,l}\\ 0.008 \pm 0.001^{b,2}\\ 0.001 \pm 0.001^{b,2}\\ 0.009 \pm 0.001^{b,2} \end{array}$	$\begin{array}{c} 0.003 \pm 0.001^{a,l}\\ 0.004 \pm 0.000^{b,l}\\ 0.007 \pm 0.001^{c,2}\\ 0.002 \pm 0.001^{a,l}\\ 0.001 \pm 0.000^{a,l}\\ 0.001 \pm 0.000^{a,l} \end{array}$	0.004	+0.001	0.282
NO <sub>3</sub> <sup>-</sup> -N, mg/L	Wet season Dry season	0 m 10 m 20 m 0 m 20 m	$\begin{array}{c} 0.018 \pm 0.001^{a,1}\\ 0.012 \pm 0.005^{a,1}\\ 0.012 \pm 0.006^{a,1}\\ 0.008 \pm 0.001^{ab,1}\\ 0.009 \pm 0.001^{a,1}\\ 0.009 \pm 0.001^{a,1}\\ \end{array}$	$\begin{array}{c} 0.037\pm0.001^{b,2}\\ 0.009\pm0.007^{a,1}\\ 0.050\pm0.005^{b,3}\\ 0.009\pm0.0005^{b,3}\\ 0.016\pm0.006^{a,1}\\ 0.010\pm0.001^{a,1}\\ 0.009\pm0.001^{a,1} \end{array}$	$\begin{array}{c} 0.026 \pm 0.009^{ab,l} \\ 0.023 \pm 0.009^{a,l} \\ 0.018 \pm 0.007^{a,l} \\ 0.006 \pm 0.006^{a,l} \\ 0.009 \pm 0.001^{a,l} \\ 0.008 \pm 0.001^{a,l} \end{array}$	$\begin{array}{c} 0.014 \pm 0.006^{a,l} \\ 0.020 \pm 0.005^{a,l} \\ 0.027 \pm 0.011^{a,l} \\ 0.022 \pm 0.011^{b,l} \\ 0.059 \pm 0.011^{b,l} \\ 0.051 \pm 0.010^{b,l2} \end{array}$	$\begin{array}{c} 0.020 \pm 0.006^{a,l} \\ 0.019 \pm 0.006^{a,l} \\ 0.016 \pm 0.005^{a,l} \\ 0.008 \pm 0.001^{ab,l} \\ 0.009 \pm 0.000^{a,l} \\ 0.009 \pm 0.000^{a,l} \end{array}$	0.021	+0.006	0.233
OKN, mg/L	Wet season Dry season	0 m 10 m 20 m 0 m 20 m 20 m	$\begin{array}{c} 0.39 \pm 0.01^{6,2} \\ 0.22 \pm 0.01^{a,1} \\ 0.22 \pm 0.01^{a,1} \\ 0.35 \pm 0.00^{c,3} \\ 0.30 \pm 0.01^{a,2} \\ 0.26 \pm 0.00^{a,1} \end{array}$	$\begin{array}{c} 0.35\pm0.01^{b,2}\\ 0.32\pm0.01^{b,1}\\ 0.43\pm0.01^{d,3}\\ 0.29\pm0.00^{a,1}\\ 0.33\pm0.01^{b,2}\\ 0.35\pm0.01^{b,2}\\ 0.35\pm0.00^{b,3}\end{array}$	$\begin{array}{c} 0.32 \pm 0.01^{a,1} \\ 0.36 \pm 0.01^{c,2} \\ 0.38 \pm 0.01^{c,3} \\ 0.33 \pm 0.01^{b,1} \\ 0.34 \pm 0.01^{b,1} \\ 0.34 \pm 0.01^{b,1} \\ 0.41 \pm 0.01^{d,2} \end{array}$	$\begin{array}{c} 0.34 \pm 0.01  ^{ab,1} \\ 0.35 \pm 0.00^{c,1} \\ 0.34 \pm 0.01^{b,1} \\ 0.32 \pm 0.00^{b,1} \\ 0.35 \pm 0.01^{c,2} \\ 0.35 \pm 0.01^{c,2} \\ 0.37 \pm 0.01^{c,3} \end{array}$	$\begin{array}{c} 0.42 \pm 0.01^{d,3} \\ 0.31 \pm 0.01^{b,1} \\ 0.33 \pm 0.01^{b,2} \\ 0.34 \pm 0.01^{b,2} \\ 0.34 \pm 0.01^{b,1} \\ 0.34 \pm 0.01^{b,1} \\ 0.34 \pm 0.01^{b,1} \end{array}$	0.34	+0.01	0.783
SRP, µg/L	Wet season Dry season	0 m 10 m 20 m 10 m 20 m	$\begin{array}{c} 8.0 \pm 1.8^{a,12} \\ 5.9 \pm 1.8^{a,1} \\ 13.2 \pm 3.1^{a,2} \\ 44.7 \pm 3.1^{b,3} \\ 14.3 \pm 1.8^{a,1} \\ 14.3 \pm 1.8^{a,1} \\ 22.6 \pm 3.1^{a,2} \end{array}$	$\begin{array}{c} 6.9 \pm 0.0^{a,1} \\ 4.8 \pm 1.8^{a,1} \\ 10.1 \pm 3.1^{a,1} \\ 54.1 \pm 3.1^{c,3} \\ 32.1 \pm 3.1^{c,3} \\ 32.1 \pm 3.1^{b,2} \\ 22.6 \pm 3.1^{a,1} \end{array}$	$\begin{array}{c} 10.1\pm3.1^{a,1}\\ 12.2\pm1.8^{b,1}\\ 10.1\pm3.1^{a,1}\\ 54.1\pm3.1^{c,3}\\ 36.3\pm1.8^{b,2}\\ 36.3\pm1.8^{b,2}\\ 17.4\pm1.8^{a,1} \end{array}$	$\begin{array}{c} 5.9\pm1.8^{a,1}\\ 6.9\pm3.1^{ab,1}\\ 9.0\pm1.8^{a,1}\\ 17.4\pm1.8^{a,1}\\ 34.2\pm1.8^{b,2}\\ 32.1\pm3.1^{b,2}\\ 32.1\pm3.1^{b,2}\end{array}$	$\begin{array}{c} 10.1\pm3.1^{a,1}\\ 6.9\pm0.0^{ab,1}\\ 9.0\pm1.8^{a,1}\\ 107.5\pm3.1^{d,3}\\ 44.7\pm3.1^{c,2}\\ 35.2\pm3.1^{b,1}\end{array}$	8.6 37.9	-29.4	0.000
TP, µg/L	Wet season Dry season	0 m 10 m 20 m 0 m 20 m	$78.2 \pm 3.6^{a,1}$ $78.2 \pm 13.1^{a,1}$ $107.5 \pm 6.3^{a,2}$ $212.8 \pm 19.3^{bc,1}$ $270.6 \pm 19.3^{a,12}$ $270.1 \pm 19.3^{a,12}$ $232.1 \pm 19.3^{a,12}$	$\begin{array}{c} 107.5 \pm 12.6^{b,1} \\ 95.0 \pm 6.3^{a,1} \\ 109.6 \pm 13.1^{a,1} \\ 193.5 \pm 19.3^{b,1} \\ 289.9 \pm 19.3^{b,2} \\ 212.8 \pm 19.3^{a,1} \end{array}$	$\begin{array}{c} 136.9\pm9.6^{c,1}\\ 149.5\pm3.6^{b,1}\\ 191.4\pm9.6^{b,2}\\ 135.6\pm19.3^{a,1}\\ 244.9\pm11.1^{a,2}\\ 232.1\pm19.3^{a,2}\\ \end{array}$	$\begin{array}{c} 109.6 \pm 9.6^{b,1} \\ 90.8 \pm 15.8^{a,1} \\ 113.8 \pm 6.3^{a,1} \\ 232.1 \pm 19.3^{b,1} \\ 322.1 \pm 11.1^{c,2} \\ 322.6 \pm 29.5^{a,1} \end{array}$	$\begin{array}{c} 109.6 \pm 9.6^{b.2}\\ 88.7 \pm 6.3^{a,1}\\ 120.1 \pm 6.3^{a,2}\\ 244.9 \pm 11.1^{c,12}\\ 270.6 \pm 19.3^{ab,2}\\ 270.6 \pm 22.3^{a,1}\\ 206.4 \pm 22.3^{a,1}\\ \end{array}$	112.4 235.1	-122.6	0.000
Means in the same ro the wet season where	w with the same as the negative va	letters or colu ilue indicates	umn with the same num that the parameter stuc	thers are not significantly lied is higher during the	r different at 5% level. Th dry season. The significa	e positive value of mear unt difference at $p$ value :	i difference indicates tha ≤ 0.05 is indicated in bo	at the param ad.	eter studied is hig	cher during

TABLE 2: Continued.

6

Journal of Chemistry



FIGURE 3: The distribution of pH and turbidity at three different depths (0, 10, and 20 m) of Bakun Reservoir in February (a) and September (b) 2014.

TABLE 3: Correlation of water quality parameters of the Bakun Reservoir during the wet season (N = 15).

Parameter	pН	Turbidity	Chl a	TSS	BOD <sub>5</sub>	$NO_2^{-}-N$	NO <sub>3</sub> <sup>-</sup> -N	OKN	SRP	TP
рН	1.000									
Turbidity	0.005	1.000								
Chl a	0.314	-0.254	1.000							
TSS	-0.162	0.755*	-0.368	1.000						
BOD <sub>5</sub>	-0.345	0.242	-0.263	0.310	1.000					
$NO_2^{-}-N$	-0.288	0.093	-0.383	0.247	$0.674^{*}$	1.000				
NO <sub>3</sub> <sup>-</sup> -N	0.161	0.067	0.389	-0.011	0.272	-0.002	1.000			
OKN	-0.081	0.023	0.091	-0.208	0.499	0.265	0.552*	1.000		
SRP	-0.365	0.237	-0.130	0.281	0.044	0.288	0.223	0.093	1.000	
ТР	-0.128	0.474	-0.144	0.217	0.400	0.620*	0.121	0.271	0.539*	1.000

\* The significant correlation at *p* value  $\leq 0.05$  is indicated in bold.

On the other hand, during the dry season, the pH value of the Bakun Reservoir was significantly positively correlated with Chl *a* and SRP but negatively correlated with turbidity and TP (*p* value  $\leq 0.05$ ) as shown in Table 4. The relationship revealed that the pH value of the Bakun Reservoir was regulated by the process of photosynthesis and decomposition of organic

matter in the reservoir during the dry season. Photosynthesis increases pH value in the surface water but the rate decreases with depth due to light limitation in the water column [33, 34].

The surface turbidity value of the Bakun Reservoir was low (<6 FNU) and increased significantly (p value  $\leq 0.05$ )

Parameter	pН	Turbidity	Chl a	TSS	BOD <sub>5</sub>	NO <sub>2</sub> <sup>-</sup> -N	NO <sub>3</sub> <sup>-</sup> -N	OKN	SRP	ТР
pН	1.000									
Turbidity	-0.516*	1.000								
Chl a	0.676*	-0.385	1.000							
TSS	-0.407	0.923*	-0.259	1.000						
BOD <sub>5</sub>	-0.378	0.217	0.090	0.346	1.000					
NO <sub>2</sub> <sup>-</sup> -N	0.465	-0.008	0.041	-0.048	-0.479	1.000				
NO <sub>3</sub> <sup>-</sup> -N	-0.192	0.197	-0.293	0.085	-0.405	0.289	1.000			
OKN	-0.085	$0.524^{*}$	0.001	0.622*	0.283	0.166	0.276	1.000		
SRP	0.556*	-0.361	$0.884^{*}$	-0.359	-0.024	-0.041	-0.163	-0.009	1.000	
TP	-0.539*	0.205	-0.255	0.128	0.176	-0.406	0.513	0.103	-0.168	1.000

TABLE 4: Correlation of water quality parameters of the Bakun Reservoir during the dry season (N = 15).

\* The significant correlation at p value  $\leq 0.05$  is indicated in bold.

as depth increased at all stations which agrees with the previous study on the reservoir [10, 11]. The turbidity value increased up to 131.33 FNU and 263.67 FNU at a depth of 20 m during the wet season and dry season, respectively. The surface turbidity value in the Bakun Reservoir was classified as Class I during both trips except for Station 3 during the wet season which was classified as Class II. The turbidity value exceeded the NWQS as depth increased where the turbidity value was more than 50 FNU. The turbidity value was significantly higher (p value  $\leq 0.05$ ) at Stations 3 and 4 which are located at Murum River. Figure 3 illustrates that the turbidity value at Station 4 increased linearly during both trips while the turbidity value at Station 3 increased linearly during the wet season. The turbidity value at Station 3 increased up to 261.3 FNU at a depth of 10 m and became stagnant up to a depth of 20 m during the dry season. The significant positive correlation between turbidity and TSS (p value  $\leq 0.05$ ) suggested that the turbidity resulted from the suspended solids in the water column. The land clearing coupled with the construction at the upstream area of the Murum River accelerated the soil erosion and sedimentation and the resulting suspended solids were transported into the reservoir during surface runoff and were deposited at the bottom of the reservoir. The high turbidity value which increased with depth was most likely due to the settling and resuspension of settled solids. The turbidity value at Station 3 was recorded up to 1000 FNU at a depth of 15 m and 30 m in the year 2013 [13]. The present study did demonstrate an improvement in the water turbidity over time although the value still exceeded the standard. Turbidity was also significantly positively correlated with OKN (*p* value  $\leq 0.05$ ) during the dry season. Many pollutants are attaching to the particles; thus, an increase in particles in the reservoir results in an increase in OKN in the present study.

The surface Chl *a* concentration ranged from  $0.62 \pm 0.21 \,\mu g/L$  to  $2.30 \pm 0.81 \,\mu g/L$  and from  $0.64 \pm 0.02 \,\mu g/L$  to  $5.87 \pm 0.99 \,\mu g/L$  in the Bakun Reservoir during the wet and dry seasons, respectively. The vertical distribution of Chl *a* in the Bakun Reservoir shows that Chl *a* decreased with depth or remained at similar concentrations in the water column (Figure 4). Sufficient light availability on the

surface water promoted the growth of phytoplankton leading to the highest concentration of Chl a in the surface water whereas light limitation as depth increased reduced the growth of the phytoplankton in the present study. However, in studies such as [27], the Chl a concentration exhibited a different trend where the Chl a concentration was the highest at a depth of 10 m compared to the surface water. The authors attributed this observation to the unfavorable high temperature and irradiance in the surface water for the phytoplankton. Nevertheless, the authors reported that the Chl a concentration was the lowest at a depth of 30 m due to the light limitation in the reservoir.

Previously, the highest concentration of 7.25  $\mu$ g/L of Chl *a* was reported in the surface water of Bakun Reservoir [11] whereas the Chl *a* concentration was reported up to 4.58 mg/m<sup>3</sup> [35] and 6.02 mg/m<sup>3</sup> [28] in the surface water of the Batang Ai Reservoir. The high Chl *a* in the Batang Ai Reservoir was attributed to the cage culture activities in the reservoir [35] whereas the high Chl *a* concentration in the present study was most likely due to the nutrient availability from the anthropogenic activities in the adjacent area. In the present study, Chl *a* was significantly positively correlated with SRP during the dry season (*p* value ≤ 0.05). The positive correlation in the Bakun Reservoir was not limited by the phosphorus.

All surface TSS values in Bakun Reservoir were classified as Class I which is less than 25 mg/L. The surface TSS concentration in the present study was lower than the previously reported surface TSS concentration (66.7–100.0 mg/L) in the year 2013 [11]. The improvement of TSS concentration demonstrated the settling of the suspended solids in the reservoir over time. The old Batang Ai Reservoir also contained low TSS values which are less than 25 mg/L even at a depth of 30 m [35]. The vertical distribution of TSS exhibited a similar trend with turbidity where the TSS value increased significantly (*p* value  $\leq$  0.05) as depth increased in the present study. Figure 4 shows that Station 3 contained the highest TSS concentration as depth increased during the dry season, followed by Station 4 where both stations were located at the Murum River. The extremely high TSS concentrations at



FIGURE 4: The distribution of Chl a, TSS, and BOD<sub>5</sub> at three different depths (0, 10, and 20 m) of Bakun Reservoir in February (a) and September (b) 2014.

Stations 3 and 4 revealed the impact of the land clearing and dam construction upstream of the river that carries eroded soil particles into the reservoir.

Surface BOD<sub>5</sub> concentration at the Bakun Reservoir ranged from  $3.24 \pm 0.19$  mg/L to  $4.30 \pm 0.15$  mg/L and from  $3.47 \pm 0.06$  mg/L to  $4.43 \pm 0.06$  mg/L during the wet and dry seasons, respectively. Table 2 shows that Stations 3 and 5 often contained significantly higher BOD<sub>5</sub> concentrations

(*p* value  $\leq 0.05$ ) at the three different depths of the water column. The high BOD<sub>5</sub> concentrations at the two stations downstream of the construction of the Murum hydroelectric dam and logging activities show organic matter loading from the anthropogenic activities into the reservoir. BOD<sub>5</sub> concentrations decreased to a value of 2.23 mg/L at a depth of 20 m at Station 1, classified as Class II. Other than that, BOD<sub>5</sub> concentrations in the Bakun Reservoir were classified

as Class III. In contrast to the TSS concentration, the surface  $BOD_5$  concentration in the present study was higher than the surface  $BOD_5$  concentration (<2 mg/L) in the year 2013 [13]. The elevated  $BOD_5$  concentration in the reservoir indicates high loading and accumulation of organic matter in the Bakun Reservoir over time. However, the present surface  $BOD_5$  concentration was lower than the  $BOD_5$  concentration in the Batang Ai Reservoir where up to 12 mg/L of  $BOD_5$  concentration was reported [35].

The NO<sub>2</sub><sup>-</sup>-N concentration was low in the Bakun Reservoir. The surface NO2<sup>-</sup>-N concentrations were not significantly different (p value > 0.05) among the stations during the wet season with a mean of 0.003 mg/L. During the dry season, the highest value of surface NO<sub>2</sub><sup>-</sup>-N was observed at Station 4 (0.008  $\pm$  0.001 mg/L) followed by Station 3  $(0.007 \pm 0.001 \text{ mg/L})$  which were significantly higher (p value  $\leq 0.05$ ) than other stations ( $\approx 0.002 \text{ mg/L}$ ). Similarly, the surface NO<sub>3</sub><sup>-</sup>-N concentration was generally low in the Bakun Reservoir, ranging from 0.014  $\pm$  0.006 mg/L to 0.037  $\pm$ 0.001 mg/L during the wet season. The mean concentration of surface NO<sub>3</sub><sup>-</sup>-N in the Bakun Reservoir during the dry season was 0.008 mg/L except at Station 4 which exhibited a peak of  $0.022 \pm 0.011 \text{ mg/L}$ . The highest concentrations of NO<sub>3</sub><sup>-</sup>-N in the reservoir were observed at Station 2 at a depth of 20 m (0.050 mg/L) and Station 4 (0.059 mg/L)during the wet and dry seasons, respectively. Comparisons of the results with NWQS indicated that NO2<sup>-</sup>-N and NO3<sup>-</sup>-N in the Bakun Reservoir were classified as Class I. Surface OKN concentrations of the Bakun Reservoir ranged from  $0.32 \pm 0.01 \,\mathrm{mg/L}$  to  $0.42 \pm 0.01 \,\mathrm{mg/L}$  and from  $0.29 \pm$ 0.01 mg/L to  $0.35 \pm 0.01 \text{ mg/L}$  during the wet and dry seasons, respectively. Significantly high concentrations of OKN (p value  $\leq 0.05$ ) were observed at Stations 4 and 5 during the wet and dry seasons which are predominantly from the surface runoff from the anthropogenic activities as mentioned above. Similar to the BOD<sub>5</sub> concentration, no obvious trend was observed in the vertical distribution of OKN concentrations in the Bakun Reservoir (Figure 5).

The NO<sub>2</sub><sup>-</sup>-N concentration in the present study is within the range of NO<sub>2</sub><sup>-</sup>-N (0.0003–0.0083 mg/L) reported in the year 2013 [11] and lower than the NO<sub>2</sub><sup>-</sup>-N concentration (0.001-0.053 mg/L) in the Batang Ai Reservoir [35]. On the other hand, the NO<sub>3</sub><sup>-</sup>-N concentration in the present study is within the range of  $NO_3^{-}$ -N (0.01–0.06 mg/L) in the Batang Ai Reservoir [35] but the  $NO_3^{-}$ -N concentrations in Station 2 during the wet season and Station 4 during the dry season were higher than the range of  $NO_3^--N$  (0.003–0.027 mg/L) reported in the Bakun Reservoir in the year 2013 [11]. NO<sub>2</sub><sup>-</sup>-N was significantly positively correlated with TP and BOD<sub>5</sub> (*p* value  $\leq$  0.05), and OKN was significantly positively correlated with NO<sub>3</sub><sup>-</sup>-N (*p* value  $\leq 0.05$ ) during the wet season. The relationship indicated the active decomposition and nitrification process in the reservoir. The relatively higher NO<sub>3</sub><sup>-</sup>-N concentration in the reservoir compared to the year 2013 indicated that nitrogen in the reservoir is being converted to  $NO_3^{-}$ -N which is less toxic to aquatic organisms in the reservoir.

The concentration of SRP was low and relatively consistent in the Bakun Reservoir during the wet season with a mean value of 8.6  $\mu$ g/L. The highest concentration of SRP  $(13.2 \,\mu\text{g/L})$  was observed at Station 1 at a depth of 20 m. The SRP concentration was significantly higher during the dry season in the Bakun Reservoir (p value  $\leq 0.05$ ) with a mean value of 37.9  $\mu$ g/L. The lowest and the highest concentrations of surface SRP were observed at Station 4 (17.4  $\pm$  1.8  $\mu$ g/L) and Station 5 (107.5  $\pm$  3.1  $\mu$ g/L), respectively, and significantly differed (*p* value  $\leq 0.05$ ) from other stations in the reservoir. Figure 6 illustrates that most stations in the Bakun Reservoir exhibited similar vertical distributions of SRP concentration during the dry season except for Station 4. The SRP concentration significantly decreased (p value  $\leq 0.05$ ) as depth increased in the Bakun Reservoir except at Station 4 where the surface SRP concentration was significantly lower (p value  $\leq 0.05$ ) than the SRP concentration at depths of 10 m and 20 m. Reference [26] reported that phosphate was lower in hypolimnion (>2.5  $\mu$ M) compared to the surface water (0.05–0.23  $\mu$ M) in the Cirata Reservoir which could be caused by enhanced loading from the sediment in the anoxic condition. The SRP concentration in the present study was lower than the SRP concentration in the year 2013 where the highest SRP concentration was  $652.2 \,\mu g/L$  [11].

Surface TP of the Bakun Reservoir ranged from 78.2  $\pm$  $3.6 \,\mu\text{g/L}$  to  $136.9 \pm 9.6 \,\mu\text{g/L}$  and from  $135.6 \pm 19.3 \,\mu\text{g/L}$  to  $244.9 \pm 11.1 \,\mu$ g/L during the wet and dry seasons, respectively. Similar to the SRP, TP concentration was significantly higher during the dry season (*p* value  $\leq 0.05$ ) in the Bakun Reservoir. This shows that high precipitation during the wet season and the elevated reservoir water volume diluted both SRP and TP substantially in the reservoir. A similar observation where TP showed lower concentrations during the rainy season and high water level (24.90–38.59  $\mu$ g/L) than the dry season and low water level (45.94–67.28  $\mu$ g/L) was reported in the Batang Ai Reservoir [20]. The present TP concentration in the Bakun Reservoir was higher than the TP concentration in the Batang Ai Reservoir during both seasons. Figure 6 illustrates that the vertical distribution of TP concentration is relatively consistent in the Bakun Reservoir but shows an opposite trend between the wet and dry seasons. The TP concentration was the lowest at the depth of 10 m during the wet season but became the highest during the dry season. Station 3 contained significantly higher TP concentrations at 0 m, 10 m, and 20 m (p value  $\leq 0.05$ ) in the reservoir during the wet season, suggesting that the phosphorus originates from the surface runoff from the Murum Dam construction. The intensity of the impact increased substantially during the wet season as more phosphorus is washed down into the reservoir. In the present study, SRP concentrations complied with the 200  $\mu$ g/L standard in accordance with the NWQS [32] during both trips. The TP concentration complied with the NWQS during the wet season but was noncompliant with the standard when the TP concentration increased substantially during the dry season.

3.2. Water Quality of the Downstream River of the Bakun Hydroelectric Dam. Table 5 summarizes the in situ and ex situ water quality of the downstream river of the Bakun Dam during the wet and dry seasons. The result demonstrated that



FIGURE 5: The distribution of  $NO_2^--N$ ,  $NO_3^--N$ , and OKN at three different depths (0, 10, and 20 m) of Bakun Reservoir in February (a) and September (b) 2014.

the reservoir water has altered the surface water temperature of the downstream river. When the cooler reservoir water at a depth of 10 m is released into the downstream river, it decreased approximately 4°C to 5°C of the surface water temperature of its downstream river. The mean value of DO in the downstream river was 9.40 mg/L and 2.59 mg/L during the wet and dry seasons and was classified as Class I and Class IV, respectively. Both DO values in the downstream river were higher than the DO value of the reservoir water at a depth of 10 m (<1 mg/L), particularly the DO content during the wet season because the spillway of the dam was open and additional water was discharged from the spillway during the sampling. The strong water current from the spillway coupled with the water flow from the turbines promotes aeration and increases the DO content substantially. On the other hand, the DO content was low when the spillway was



FIGURE 6: The distribution of SRP and TP at three different depths (0, 10, and 20 m) of Bakun Reservoir in February (a) and September (b) 2014.

TABLE 5: Summary of the mean and standard deviation of the in situ and ex situ water quality parameters in the downstream river of the Bakun Dam (Station 6) and the mean difference of the parameters between the wet and dry seasons (N = 3).

Parameter	Wet season	Dry season	Mean difference	<i>p</i> value
In situ				
Temperature, °C	$25.0 \pm 0.0$	$27.1\pm0.1$	-2.1	0.000
DO, mg/L	$9.40\pm0.10$	$2.59\pm0.05$	+6.81	0.000
pН	$6.2 \pm 0.01$	$6.0 \pm 0.01$	+0.2	0.001
Turbidity, FNU	$77.00 \pm 1.00$	$113.67 \pm 0.58$	-36.67	0.000
Ex situ				
Chl <i>a</i> , $\mu$ g/L	$0.64\pm0.06$	$0.52 \pm 0.06$	+0.12	0.179
TSS, mg/L	$45.0 \pm 1.7$	$61.7 \pm 1.4$	-16.7	0.011
BOD <sub>5</sub> , mg/L	$5.70\pm0.03$	$3.10\pm0.10$	+2.60	0.000
NO <sub>2</sub> <sup>-</sup> -N, mg/L	$0.007\pm0.001$	$0.001\pm0.001$	+0.006	0.009
NO <sub>3</sub> <sup>-</sup> -N, mg/L	$0.023\pm0.010$	$0.009\pm0.001$	+0.014	0.115
OKN, mg/L	$0.40\pm0.01$	$0.38 \pm 0.00$	+0.01	0.057
SRP, µg/L	$10.1 \pm 3.1$	$34.2 \pm 1.8$	-24.1	0.008
TΡ, μg/L	99.2 ± 9.6	$386.4 \pm 19.3$	-287.2	0.001

The positive value of mean difference indicates that the parameter studied is higher during the wet season whereas the negative value indicates that the parameter studied is higher during the dry season.

TABLE 6: Mean difference of in situ and ex situ water quality parameters between the intake point of the dam at 10 m (Station 5) and its downstream river (Station 6) during wet and dry seasons (N = 3).

Daramator	Wet seaso	on	Dry seaso	n
rarameter	Mean difference	p value	Mean difference	<i>p</i> value
In situ				
pН	-0.03	0.469	+0.01	0.288
Turbidity, FNU	+60.9	0.000	+46.7	0.000
Ex situ				
Chl <i>a</i> , $\mu$ g/L	+0.28	0.013	-0.02	0.755
TSS, mg/L	+4.67	0.011	+13.03	0.000
BOD <sub>5</sub> , mg/L	+1.26	0.000	-1.53	0.000
NO <sub>2</sub> <sup>-</sup> -N, mg/L	+0.003	0.001	+0.000	0.374
NO <sub>3</sub> <sup>-</sup> -N, mg/L	+0.003	0.631	-0.000	0.374
OKN, mg/L	+0.08	0.000	+0.05	0.000
SRP, µg/L	+3.2	0.157	-10.5	0.007
TP, µg/L	+10.5	0.189	+115.7	0.002

The positive value of mean difference indicates that the parameter studied is higher in the downstream river whereas the negative value indicates that the parameter studied is lower in the downstream river.

not open during the sampling in the dry season. When the oxygen-deprived reservoir water was released into the downstream river without additional aeration, it decreased the oxygen level of the downstream river below the minimum requirement of 5 mg/L for sensitive aquatic organisms.

Low pH values ( $\approx 6.1$ ) were observed at the downstream river and classified as Class II according to NWQS [32]. Table 6 shows that there was no significant difference in pH value between Station 6 and the dam intake point at 10 m for both seasons (p value > 0.05) revealing that the low pH in the downstream river is due to the low pH of the reservoir water that was released into the downstream river after passing through the turbines. The pH value of the downstream river was relatively lower than the pH value of tributaries that flow into the Bakun Reservoir (6.8-7.8) [14]. The turbidity value in the downstream river of the dam was high and exceeded the standard guideline of 50 FNU in Malaysia [32]. The turbidity values were also significantly higher (p value > 0.05) than the reservoir water during both seasons. When water is discharged from the spillway in addition to turbine outflow, resuspension of deposited sediments under the high flow rate increases the suspended solids downstream. The turbidity value (77.00  $\pm$  1.00 FNU) during the wet season was significantly lower (p value > 0.05) than the dry season (113.67  $\pm$  0.58 FNU) which is most probably due to more dilution from the tributaries along the downstream river during the wet season. Similar to the turbidity value, the TSS concentration during the wet season was significantly lower than the dry season (p value  $\leq 0.05$ ) and was classified as Class II and Class III, respectively. Both values were also significantly higher than the TSS concentration (p value  $\leq$ 0.05) at the intake point.

There was no significant difference in Chl *a* between the wet and dry seasons (*p* value > 0.05) in the downstream river with a mean value of 0.58  $\mu$ g/L. The Chl *a* concentration was significantly higher (*p* value  $\leq$  0.05) than the intake

point during the wet season but it was similar to the Chl a concentration at surface reservoir water (0.64  $\mu$ g/L). There was no significant difference in Chl a between downstream river and the intake point (p value > 0.05) during the dry season. The mean value of BOD<sub>5</sub> in the downstream river was 5.70 mg/L and 3.10 mg/L during the wet and dry seasons and was classified as Class III. The BOD<sub>5</sub> concentration during the wet season was significantly higher than the dry season (p value  $\leq 0.05$ ). Besides, the downstream BOD<sub>5</sub> concentration was significantly higher than the BOD<sub>5</sub> concentration at the intake point (p value  $\leq 0.05$ ) during the wet season whereas the BOD<sub>5</sub> concentration was significantly lower than the BOD<sub>5</sub> concentration at the intake point (p value  $\leq$  0.05) during the dry season. The higher downstream BOD<sub>5</sub> concentration during the wet season indicates that the high BOD<sub>5</sub> concentration is most likely attributed to other domestic discharge and runoff in addition to the reservoir water. Several longhouses and villages located along the downstream river may have contributed substantial organic matter to the downstream river.

NO<sub>2</sub><sup>-</sup>-N and NO<sub>3</sub><sup>-</sup>-N concentrations were also low in the downstream river, similar to the reservoir water, and were classified as Class I according to NWQS [32]. Significantly higher NO<sub>2</sub><sup>-</sup>-N concentration (*p* value ≤ 0.05) was found during the wet season whereas no significant difference of NO<sub>3</sub><sup>-</sup>-N concentration was found between the wet and dry seasons (*p* value > 0.05). The downstream NO<sub>2</sub><sup>-</sup>-N concentration at the intake point. There was no significant difference in OKN between the wet and dry seasons (*p* value > 0.05) in the downstream river with a mean value of 0.39 mg/L. OKN was significantly higher (*p* value ≤ 0.05) at the downstream river than the OKN concentration at intake point. The higher downstream NO<sub>2</sub><sup>-</sup>-N and OKN concentrations besides BOD<sub>5</sub> demonstrated



Wet season: Cases 1 to 16 Dry season: Cases 17 to 32 0 m depth: Cases 1, 4, 7, 10, 13, 16, 17, 20, 23, 26, 29, and 32 10 m depth: Cases 2, 5, 8, 11, 14, 18, 21, 24, 27, and 30 20 m depth: Cases 3, 6, 9, 12, 15, 19, 22, 25, 28, and 31

FIGURE 7: Clusters of the five sampling stations located in Bakun Reservoir at three different depths (0 m, 10 m, and 20 m) and one station located at its downstream river collected at 0 m during dry and wet seasons in Sarawak, Malaysia.

the organic pollutant contribution from adjacent domestic discharge and runoff in the downstream river.

SRP and TP concentrations in the downstream river exhibited a similar trend where the concentration during the wet season was significantly lower than during the dry season (*p* value  $\leq$  0.05). The downstream SRP concentration complied with the 200  $\mu$ g/L standard in accordance with the NWQS [32] in both trips. On the other hand, the downstream TP concentration also complied with the NWQS during the wet season but changed to noncompliance with the standard when the TP concentration increased substantially during the dry season. The TP concentration was found to be significantly higher (p value  $\leq 0.05$ ) at the downstream river than the TP concentration at the intake point. The high TP and low SRP concentration indicate that phosphorus concentration mainly consisted of organic phosphorus in the present study. The higher downstream TP concentration further confirms the organic pollutant contribution from the adjacent domestic discharge.

3.3. Cluster Analysis. Figure 7 demonstrates that the water quality in the reservoir can be grouped into three clusters according to the season and the water depth of the reservoir. Cluster 1 and Cluster 2 are mostly made up of sampling stations at depths of 10 m and 20 m conducted during the dry season and wet season, respectively, indicating that the dry and wet seasons have an influence on the deeper water

column of the reservoir. The surface water quality of the downstream river during the dry (Case 32) and wet (Case 16) seasons was also grouped to Clusters 1 and 2, respectively, as it was influenced by the deeper reservoir water discharged into the river. On the other hand, surface water quality of the reservoir except at Station 5 (Case 13) during the wet season was not influenced by the season where the surface water quality of the Bakun Reservoir during both seasons was categorized as Cluster 3. This phenomenon is the most apparent based on the turbidity and TSS in the reservoir. The surface turbidity and TSS values during the wet and dry seasons were relatively similar with a mean value of 3.02 FNU and 3.66 FNU and 9.0 mg/L and 8.1 mg/L, respectively, but the turbidity (76.02 FNU versus 133.65 FNU) and TSS (60.7 mg/L versus 125.5 mg/L) values at a depth of 20 m during the wet season were around two times lower than during the dry season.

## 4. Conclusions

The Bakun hydroelectric reservoir is a thermally stratified reservoir with a temperature gradient of approximately 5°C within the thermocline layer. The thickness of the well oxygenated water was around 3–6 m of the surface water, whereas the oxygen content of most of the water body was below 5 mg/L or even in hypoxia. The Bakun Reservoir showed signs of organic pollution with high BOD<sub>5</sub>, OKN,

and TP concentrations observed in the reservoir. Acidification was observed in parts of the reservoir, particularly downstream of active logging activities and the Murum hydroelectric dam construction during the wet season. The water quality of the reservoir was influenced by the wet and dry seasons particularly in the deeper water column. SRP and TP concentrations were discovered to be higher during the dry season in the reservoir. This result suggests the necessity of management and conservation of the reservoir to prevent further deterioration in the reservoir's water quality where different water quality parameters should be targeted during different seasons. The present study also demonstrated that the water discharged from the Bakun Reservoir has a great impact on the water quality at the downstream river. The water released from the reservoir decreased the temperature, DO, and pH of the downstream river whereas turbidity and TSS concentration increased in the downstream river. Nevertheless, the water quality of the downstream river, particularly BOD<sub>5</sub>, OKN, and TP concentrations, was also influenced by adjacent anthropogenic activities such as household wastewater. This result suggests that the downstream river of the Bakun Reservoir was not solely impacted by the reservoir's outflow. Therefore, all factors should be taken into account in decision-making of the management of the downstream river for the health of sensitive aquatic organisms.

## **Conflicts of Interest**

The authors declare that there are no conflicts of interest regarding the publication of this paper.

### Acknowledgments

The authors appreciate the financial support provided by the Malaysian Ministry of Higher Education through Grant no. FRGS/STWN01(04)/991/2013(32) and the facilities provided by Universiti Malaysia Sarawak.

### References

- M. W. Beck, A. H. Claassen, and P. J. Hundt, "Environmental and livelihood impacts of dams: common lessons across development gradients that challenge sustainability," *International Journal of River Basin Management*, vol. 10, no. 1, pp. 73–92, 2012.
- [2] X. Li, S. Dong, Q. Zhao, and S. Liu, "Impacts of Manwan Dam construction on aquatic habitat and community in Middle Reach of Lancang River," *Procedia Environmental Sciences*, vol. 2, no. 5, pp. 706–712, 2010.
- [3] J. Li, S. Dong, S. Liu, Z. Yang, M. Peng, and C. Zhao, "Effects of cascading hydropower dams on the composition, biomass and biological integrity of phytoplankton assemblages in the middle Lancang-Mekong River," *Ecological Engineering*, vol. 60, pp. 316–324, 2013.
- [4] D. D. A. Cunha and L. V. Ferreira, "Impacts of the Belo Monte hydroelectric dam construction on pioneer vegetation formations along the Xingu River, Pará State, Brazil," *Revista Brasileira de Botanica*, vol. 35, no. 2, pp. 159–167, 2012.

- [5] Q. G. Wang, Y. H. Du, Y. Su, and K. Q. Chen, "Environmental impact post-assessment of dam and reservoir projects: a review," *Procedia Environmental Sciences*, vol. 13, pp. 1439–1443, 2012.
- [6] G. L. Wei, Z. F. Yang, B. S. Cui et al., "Impact of dam construction on water quality and water self-purification capacity of the Lancang River, China," *Water Resources Management*, vol. 23, no. 9, pp. 1763–1780, 2009.
- [7] Y. Yi, Z. Yang, and S. Zhang, "Ecological influence of dam construction and river-lake connectivity on migration fish habitat in the Yangtze River basin, China," *Procedia Environmental Sciences*, vol. 2, no. 5, pp. 1942–1954, 2010.
- [8] Q. Lin, "Influence of dams on river ecosystem and its countermeasures," *Journal of Water Resource and Protection*, vol. 03, no. 01, pp. 60–66, 2011.
- [9] W. L. Graf, "Downstream hydrologic and geomorphic effects of large dams on American rivers," *Geomorphology*, vol. 79, no. 3-4, pp. 336–360, 2006.
- [10] L. Nyanti, T. Y. Ling, and J. Grinang, "Physico-chemical characteristics in the filling phase of Bakun hydroelectric reservoir," vol. 2, pp. 92–101, Sarawak, Malaysia, 2012.
- [11] T.-Y. Ling, L. Nyanti, T. Muan, J. Grinang, S.-F. Sim, and A. Mujahid, "Physicochemical parameters of Bakun Reservoir in Belaga, Sarawak, Malaysia, 13 months after reaching full supply level," *Sains Malaysiana*, vol. 45, no. 2, pp. 157–166, 2016.
- [12] S. F. Sim, T. Y. Ling, L. Nyanti, N. Gerunsin, Y. E. Wong, and L. P. Kho, "Assessment of heavy metals in water, sediment, and fishes of a large tropical hydroelectric dam in Sarawak, Malaysia," *Journal of Chemistry*, vol. 2016, Article ID 8923183, 10 pages, 2016.
- [13] L. Nyanti, T. Y. Ling, and T. Muan, "Water quality of Bakun hydroelectric dam reservoir, the construction of Murum dam," *ESTEEM Academic Journal*, vol. 11, no. 1, pp. 81–88, 2015.
- [14] T. Y. Ling, L. Nyanti, and A. S. Masion, "Water quality of rivers that flow into Bakun hydroelectric dam reservoir, Sarawak, Malaysia," *ESTEEM Academic Journal*, vol. 11, no. 1, pp. 9–16, 2015.
- [15] V. Rossel and A. de la Fuente, "Assessing the link between environmental flow, hydropeaking operation and water quality of reservoirs," *Ecological Engineering*, vol. 85, pp. 26–38, 2015.
- [16] T. da Costa Lobato, R. A. Hauser-Davis, T. F. de Oliveira et al., "Categorization of the trophic status of a hydroelectric power plant reservoir in the Brazilian Amazon by statistical analyses and fuzzy approaches," *Science of the Total Environment*, vol. 506-507, pp. 613–620, 2015.
- [17] X. Li, T. Huang, W. Ma, X. Sun, and H. Zhang, "Effects of rainfall patterns on water quality in a stratified reservoir subject to eutrophication: Implications for management," *Science of the Total Environment*, vol. 521-522, pp. 27–36, 2015.
- [18] M. Varol, B. Gökot, A. Bekleyen, and B. Şen, "Spatial and temporal variations in surface water quality of the dam reservoirs in the Tigris River basin, Turkey," *Catena*, vol. 92, pp. 11–21, 2012.
- [19] Y. Zhang, Z. Wu, M. Liu et al., "Dissolved oxygen stratification and response to thermal structure and long-term climate change in a large and deep subtropical reservoir (Lake Qiandaohu, China)," *Water Research*, vol. 75, pp. 249–258, 2015.
- [20] T. Y. Ling, T. Z. E. Lee, and L. Nyanti, "Phosphorus in batang ai hydroelectric dam Reservoir, Sarawak, Malaysia," *World Applied Sciences Journal*, vol. 28, no. 10, pp. 1348–1354, 2013.
- [21] P. McCully, "Rivers no more: the environmental effects of dams," in *Silenced Rivers: The Ecology and Politics of Large Dams*, P. McCully, Ed., pp. 29–64, Zed Books, London, UK, 1996.

- [22] T.-Y. Ling, C.-L. Soo, T. L.-E. Heng, L. Nyanti, S.-F. Sim, and J. Grinang, "Physicochemical characteristics of river water downstream of a large tropical hydroelectric dam," *Journal of Chemistry*, vol. 2016, Article ID 7895234, 2016.
- [23] D. Jenkins, J. J. Connors, and A. E. Greenberg, *Standard Methods for the Examination of Water and Wastewater*, American Public Health Association, Washington, Wash, D.C., USA, 21st edition edition, 205.
- [24] Hach, *Hach Water Analysis Handbook*, Hach Company, USA, 2015.
- [25] D. F. Goerlitz and E. Brown, "Methods for analysis of organic substances in water," in *Techniques of Water-Resources Investigations of The United States Geological Survey*, R. L. Wershaw, M. J. Fishman, R. R. Grabbe, and L. E. Lowe, Eds., pp. 1–40, U. S. Geological Survey, United States, 1972.
- [26] Y. Hayami, K. Ohmori, K. Yoshino, and Y. S. Garno, "Observation of anoxic water mass in a tropical reservoir: the cirata reservoir in java, Indonesia," *Limnology*, vol. 9, no. 1, pp. 81–87, 2008.
- [27] C. Ariyadej, P. Tansakul, and R. Tansakul, "Variation of phytoplankton biomass as chlorophyll a in banglang reservoir, yala province," *Songklanakarin Journal of Science and Technology*, vol. 30, no. 2, pp. 159–166, 2008.
- [28] T.-Y. Ling, L. Nyanti, C.-K. Leong, and Y.-M. Wong, "Comparison of water quality at different locations at Batang Ai Reservoir, Sarawak, Malaysia," *World Applied Sciences Journal*, vol. 26, no. 11, pp. 1473–1481, 2013.
- [29] G. B. Sahoo and D. Luketina, "Modeling of bubble plume design and oxygen transfer for reservoir restoration," *Water Research*, vol. 37, no. 2, pp. 393–401, 2003.
- [30] Y. Zhou, D. R. Obenour, D. Scavia, T. H. Johengen, and A. M. Michalak, "Spatial and temporal trends in Lake Erie hypoxia, 1987-2007," *Environmental Science and Technology*, vol. 47, no. 2, pp. 899–905, 2013.
- [31] T. Y. Ling, D. P. Debbie, N. Lee, I. Norhadi, and J. J. E. Justin, "Water quality at Batang Ai Hydroelectric Reservoir (Sarawak, Malaysia) and implications for aquaculture," vol. 2, pp. 23–30, 2012.
- [32] Department of Environment, Malaysia Environmental Quality Report 2014, Department of Environment, Kuala Lumpur, Malaysia, 2015.
- [33] T. R. Fisher, L. W. Harding Jr., D. W. Stanley, and L. G. Ward, "Phytoplankton, nutrients, and turbidity in the Chesapeake, Delaware, and Hudson estuaries," *Estuarine, Coastal and Shelf Science*, vol. 27, no. 1, pp. 61–93, 1988.
- [34] P.-P. Shen, G. Li, L.-M. Huang, J.-L. Zhang, and Y.-H. Tan, "Spatio-temporal variability of phytoplankton assemblages in the Pearl River estuary, with special reference to the influence of turbidity and temperature," *Continental Shelf Research*, vol. 31, no. 16, pp. 1672–1681, 2011.
- [35] L. Nyanti, K. M. Hiii, A. Sow, I. Norhadi, and T. Y. Ling, "Impacts of aquaculture at different depths and distances from cage culture sites in batang Ai hydroelectric dam reservoir, Sarawak, Malaysia," *World Applied Sciences Journal*, vol. 19, no. 4, pp. 451– 456, 2012.



International Journal of Medicinal Chemistry







International Journal of Analytical Chemistry



Advances in Physical Chemistry

Spectroscopy



Chromatography Theoretical Chemistry Research International

Catalysts

