

### Research Article

## Risk Assessment of Trace Element Contamination in Drinking Water and Agricultural Soil: A Study in Selected Chronic Kidney Disease of Unknown Etiology (CKDu) Endemic Areas in Sri Lanka

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Unexplained or unclear etiology of chronic kidney disease (CKDu) has been reported in Sri Lanka's North Central Province (NCP) for more than two decades. Meanwhile, high exposure to heavy metals/metalloids and their accumulation are recognized as the origin of many acute and chronic diseases in certain vulnerable human tissues including kidneys. This study evaluates the contamination status of heavy metals/metalloids of the drinking water and agricultural soil in two CKDu endemic areas compared with a reference area in Sri Lanka based on common indexes and attribute of the commonly used fertilizers evaluated to identify the basic sources of toxic metals in the agricultural soil. Mean concentrations of heavy metals/metalloids such as Mn, Co, As, Cd, Pb, Cu, Zn, and Fe in drinking water of CKDu endemic areas were far below Sri Lankan water quality standards (permissible limits). In addition, all sampling locations dropped below the medium range of the heavy metal pollution index of water (HPI 15–40). Geoaccumulation indexes ( $I_{geo}$ ) of soil reveal that paddy soil in CKDu endemic areas is being moderately polluted with toxic metals, could be the driving force for agricultural soil pollution, and limitless application of low-quality fertilizer would lead to more soil contamination with heavy metals. Hence, hazardous metals can be incorporated into the food chains via contaminated paddy soil.

#### 1. Introduction

Chronic kidney disease unknown etiology (CKDu) was first discovered in Sri Lanka in the mid-1990s, mainly found among farmers in Sri Lanka's North Central Province (NCP); since then, the disease has spread dramatically over a span of two decades, including other agricultural areas of the country such as the North, North West, Eastern, Uva, and Central Provinces [1–3]. Chronic kidney disease (CKD) is a noncommunicable disease that is related to risk factors such as diabetes or hypertension, past snakebites, and urinary tract infections. The aforementioned risk factors of CKD or family history of kidney

failures are not responsible for the occurrence of chronic kidney disease of unknown etiology [4]. Geographic "hotspots" of CKDu have emerged in a number of countries, including El Salvador, Guatemala, Mexico, Nicaragua, Bulgaria, Croatia, Serbia, India, and Sri Lanka [5], and it is prevalent among agricultural workers with several symptoms including fatigue, panting, lack of appetite, nausea, and anemia [6]. The studies of CKDu in Sri Lanka are significant because of the increasing number of Sri Lankans who are suffering from CKDu since the 1990s, e.g., more than 50,000 patients diagnosed with late-stage kidney disease, and the majority of these patients are reported from the North Central Province (NCP) [7]. Hence, potentially

causative agents for the prevalence of kidney disease must be present in drinking water and soils of the endemic areas in sufficient quantities [4].

Although some of the heavy metals are known to be enhancing human body functions, the same heavy metals above their critical limits and some others are toxic for human beings [8]. There are so many major pathways to contamination of groundwater and soil with nephrotoxic contaminants such as heavy metals and counterions with the chemical applications due to the intense agriculture. Those toxicants exist as clusters in the environment [9]. And exposure to those may be linked to conditions such as neurotoxicity in human kidneys. Exposure to metals such as As, Cd, Hg, Cr, Ni, Mn, and Fe present in the drinking water significantly collaborated with the functional and structural integrity of kidneys [10, 11]. According to the toxicological studies on nephrotoxicity, exposure to cation species of Pb, As, and Cd can be associated with renal tubular necrosis, and exposure to heavy metals such as Pb, Hg, and Cd is directly associated with the collapse of glomeruli [12]. Kidney dysfunction was correlated with high creatinine level which results from exposure to a low dose of Pb, As, and Cd. High creatinine levels indicate a weak glomerular filtration process which means reduced ability to excrete waste products from the blood through urine [9].

Agrochemicals and fertilizers are the most sensible source for heavy metal distribution in the paddy soil [13]. Insecticides, herbicides, and fungicides are the primary sources of Cu, Zn, Cd, Pb, and As in the agricultural soil. Phosphate, nitrate, potash, and lime fertilizers are the main sources of Cr, Cd, Cu, Ni, Zn, Mn, and Pb [14].

Moreover, identifying the current status of the toxic metal contamination in drinking water and agricultural soil is more significant because toxic metals can directly enter the human body via drinking water and human food chains can be contaminated with toxic metals via polluted agricultural soil. In order to identify the sources of the toxic metals, assessment of the frequently use fertilizers is more essential as selected areas are fully agricultural areas. Along with it, this study is expected to update the water and agricultural soil contamination levels and the current status of heavy metal pollution in two CKDu endemic areas in Sri Lanka with respect to the reference area (CKDu nonendemic) and illustrate and give a better understanding on heavy metal exposure using index-based assessment of drinking water and the agricultural soil. This comparison mainly focused on revealing a piece of evidence of the contribution of toxic metals to the prevalence of CKDu in selected endemic areas.

#### 2. Materials and Methods

2.1. Study Areas and Selection of Sampling Sites. Two (02) CKDu endemic areas including Ambagaswewa GND  $(8^{\circ}11'30.70''N, 81^{\circ}0'57.08''E)$  in Polonnaruwa District and Eppawala GND  $(8^{\circ}8'40.10''N, 80^{\circ}24'37.73''E)$  in Anuradhapura District and a reference area that is Dambethalawa GND  $(7^{\circ}17'1.68''N, 81^{\circ}32'52.35''E)$  in Ampara District in Sri Lanka were selected for the sampling based on the recent hospital data obtained from the Ministry of

Health, Sri Lanka. GPS coordinates were recorded at each and every sampling location using a GPS device. All sampling procedures were done in April 2019 (dry season).

2.2. Sampling. Sampling points for the drinking water sample collection were selected from shallow drinking water wells (dug wells) located in the home gardens of the residents which are the main sources of water consumption in their daily life. Thirty drinking water samples were collected from each CKDu hot spot and the reference site for the analysis of cations and anions. All the water samples were collected into prewashed Teflon bottles for analysis and acidified with ultrapure nitric acid (2% v/v) (Sinopharm, Shanghai, China). All the water samples were stored at 4°C during transport to the laboratory. Agricultural lands were selected for soil sampling, and twenty composite topsoil samples were collected from the paddy cultivated fields according to the random sampling method in each selected CKDu hotspot and the reference site. Composite soil samples were prepared by a combination of five samples from each location. As frequently used fertilizers in paddy cultivations, urea, potash, and triple superphosphate fertilizer samples were collected from the stores in selected CKDu endemic areas. Fifteen composite fertilizer samples were collected in each type of fertilizers.

2.3. Analysis. Concentrations of metal elements in drinking water including Mn, Co, As, Cd, Pb, Cu, Zn, Na, K, Al, Ca, Mg, Fe, and Ni were determined using inductively coupled plasma mass spectrometry (ICP-MS-7800-Agilent, Germany). Multielement ICP-MS standards (AccuStandard, USA) were used for the instrumental calibration. Two calibration series (0.1 ppb-10 ppb and 10 ppb to 1000 ppb) were prepared using multielement standard. Acidified water samples with conc. HNO<sub>3</sub> (69% purity, Sigma-Aldrich, India) were filtered through  $0.45 \,\mu m$  syringe filters before the insertion to the ICP-MS instrument. The concentrations of anions in water samples (including fluoride, chloride, nitrate, phosphate, and sulfate) and fluoride, chloride, nitrate, and sulfate contents in soil samples were measured using ion chromatography. Sodium bicarbonate (Sigma-Aldrich, India, CASRN 144-55-8) and sodium carbonate (Sigma-Aldrich, India, CASRN-497-19-7) were used as elution solution, and sulfuric acid (Sigma-Aldrich, India, CASRN-7664-93-9) was used as regeneration solution. ACS reagent grade 1000 mg/L stock solutions of considered counterions were used to prepare the standards and prepared for a range of concentrations encompassing expected sample concentrations. Standards and collected samples were filtered. A small volume of the sample (5.0 mL) was introduced, under the flow rate of 0.7 mL/min into the ion chromatograph (Metrohm Eco IC, Switzerland), and the anions of interest were separated and measured, using a system comprised of a guard column, analytical column, suppressor device, and conductivity detector.

0.200 g of each soil sample was digested using a microwave digester (ETHOS EASY, Italy) with adding 10.00 mL of concentrated nitric acid. Digested solutions were diluted up to 25.00 mL with ultrapure water, and 10.00 mL of each digested soil solution was filtered through 0.45  $\mu$ m nylon syringe filters. The determination of cation concentrations (Cr, Mn, Fe, Co, As, Cd, Pb, Cu, Zn, Mg, Ca, Na, K, and Ni) in soil samples was carried out by the ICP-MS method (ICP-MS-8000-Agilent, Germany). The calibration series (1 ppm–50 ppm) was prepared using multielement standard, and 0.200 g of each fertilizer sample (Urea, Potash, and Triple Superphosphate) was also used for the analysis of heavy metal/metalloids (Cr, Mn, Fe, Cu, Zn, As, Cd, and Pb) concentrations by the ICP-MS method.

In order to determine the total phosphate concentration in the soil, 0.02 M Truog extracting solution was prepared, 26.54 g of (NH<sub>4</sub>)<sub>6</sub>Mo<sub>7</sub>O<sub>24</sub>·4H<sub>2</sub>O (Sigma-Aldrich, India) was dissolved in 200 mL of distilled water (warmed to 60°C), and 280.0 mL Conc. H<sub>2</sub>SO<sub>4</sub> (95-97% purity, EMPARTA® ACS, India) was diluted with 750 ml of distilled water. After both solutions cooled down to room temperature, ammonium molybdate solution was added slowly to the sulfuric acid solution with shaking to prepare the molybdate reagent (2.5%). 2.5 g of SnCl<sub>2</sub>·2H<sub>2</sub>O (Sigma-Aldrich, India) was dissolved in 10.0 mL of concentrated HCl and diluted to 100 mL with distilled water by rapid stirring. A few pieces of metallic tin (Sn) were added to the solution after filtration to prepare stannous chloride solution. A 100.0 mg/L standard phosphorus solution was prepared by dissolving 0.4394 g of KH<sub>2</sub>PO<sub>4</sub> (Sigma-Aldrich, India) in 1 L of H<sub>2</sub>SO<sub>4</sub> (0.01 mol/ L), and a 4.0 mg/L solution was prepared as a phosphorus working solution.

Using the phosphorus working solution, a calibration curve was constructed for the UV-visible spectrophotometer and measured the absorbance at 660 nm using a UV-visible spectrophotometer (Agilent Cary 3500, Germany). Soil samples for the phosphate analysis were prepared by adding 3.0 g of a soil sample to the 100.00 mL extracting solution. The mixture was shaken for about 10 minutes and allowed to stand for 15 minutes followed by filtration. Both reagents were mixed with the 50.00 mL aliquot of each soil solution and measured the absorbance at 660 nm.

2.4. Indexing Approach. The present study attempts to use the weighted arithmetic average mean method for calculating the HPI for the purpose of monitoring drinking water contamination levels in CKDu affected areas with respect to the reference combining the concentrations of Cr, Mn, Cu, Fe, Pb, Cd, As, and Zn. A set of equations were incorporated in calculating HPI for the drinking water samples. The critical pollution index value is 100, where higher HPI values indicate the greater damage to human health.

The HPI model [15] was calculated as

$$HPI = \frac{\sum_{i=1}^{n} W_i * Q_i}{\sum_{i=1}^{n} W_i},$$
 (1)

where Q is the subindex of the  $i^{\text{th}}$  parameter,  $W_i$  is the unit weight of the  $i^{\text{th}}$  parameter, and n is the number of parameters considered.

$$W_i = \frac{k}{S_i},\tag{2}$$

where  $W_i$  is the unit of weightage, k is the constant of proportionality (k = 1), and  $S_i$  is the recommended standard for the i<sup>th</sup> parameter according to the Sri Lankan standards for the drinking water.

The subindex  $(Q_i)$  of the *i*<sup>th</sup> parameter was calculated according to Reza and Singh [6]:

$$Q_{i} = \frac{|M_{i} - I_{i}|}{S_{i} - I_{i}} * 100,$$
(3)

where  $M_i$  is the monitored value of the heavy metal of  $i^{\text{th}}$  parameter in ppb,  $I_i$  is the maximum desirable value (ideal) of the  $i^{\text{th}}$  parameter, and  $S_i$  refers to the standard or permissible limit for the  $i^{\text{th}}$  parameter.

2.5. Geoaccumulation Index ( $I_{geo}$ ) for Agricultural Soil. Geoaccumulation value was developed for the determination of the degree of metal concentrations and the pollution which is caused by the metals in soil segregates. Geoaccumulation index ( $I_{geo}$ ) was determined using the following equation [16] (Muller, 1979):

$$I_{\text{geo}} = \frac{C_n}{1.5 \times B_n},\tag{4}$$

where  $C_n$  is the measured concentration of the element in soil dust,  $B_n$  is the geochemical background value, and the constant 1.5 allows us to analyze natural fluctuations in the content of a given substance in the environment and to detect a very small anthropogenic influence.

*2.6. Natural Background Concentration (NBC).* NBC can be calculated by the method described by Ander et al., 2013 [17].

$$NBC = \frac{x + (1.96 * SD)}{\sqrt{n}},$$
(5)

where x is the sample mean, SD is the standard deviation, and n is the number of samples.  $I_{\text{geo}}$  was distinguished into seven classes by Muller: Table 1 represents the seven classes of geoaccumulation index by Muller.

2.7. Geographical Data Treatment. Geographic Information System (GIS) is widely used for collecting diverse spatial data and for overlay analysis in the spatial register domain to represent spatially variable phenomena [18–20]. GIS, which synthesizes different and important quality data into an easily understood format, provides a way to summarize overall water or soil quality conditions in a manner that can be clearly communicated to policymakers [21] and finally can be incorporated with the decision-making process. The maps which show the spatial distribution of heavy metal pollution in selected sampling areas were interpolated by the Inverse Distance Weighted (IDW) tool and the Spatial

TABLE 1: Classes of geoaccumulation index (Muller, 1979).

Igeo	I <sub>geo</sub> class	Soil/sediment quality
0-0	0	Unpolluted
0-1	1	Unpolluted to moderately polluted
1-2	2	Moderately polluted
2-3	3	Moderately polluted to highly polluted
3-4	4	Highly polluted
4-5	5	Highly polluted to very highly polluted
5-6	>5	Very highly polluted

Autocorrelation tool (Moran's index) using ArcGIS 10.2 software.

#### 3. Results and Discussion

3.1. Assessment of Drinking Water. In the drinking water scenario, the priority should be given to the fluoride toxicity because there have been many cases reported which were related to human kidney function failure due to fluoride toxicity. In accordance with the literature, an investigation of 210 children in China found that drinking water with more than 2.00 mg/L fluoride had increased levels of N-acetyl glucosamine (NAG) and y-glutamyl transpeptidase (yGT) in their urine, both of which are markers of renal tubular damage [22]. The present study has found that mean fluoride contents in the CKDu endemic areas have exceeded the permissible limits for drinking water fluoride (1.00 mg/L) (SLS 614:2013) (Figure 1).

The mean concentrations of other counterions which were analyzed in drinking water samples such as  $Cl^-$ ,  $Br^-$ ,  $NO_3^-$ ,  $PO_4^{3-}$ , and  $SO_4^{3-}$  (Table 2) have not exceeded permissible limits defined in the Sri Lankan drinking water quality standards (SLS 614:2013) in both endemic and nonendemic areas. However, Chandrajith et al., 2010, mentioned that even though no single geochemical parameter could be clearly and directly correlated with the etiology of CKDu, the unique hydrogeochemistry of the drinking water is closely associated with the incidence of the disease. In affected areas, water quality needs to be assessed, particularly combinations of various constituents, such as metal elements and the hardness in combination [4].

According to the primary data gathered from both CKDu affected areas (Table 3), Cd, As, Pb, Cr, Cu, and Zn concentrations of drinking water were below the Sri Lankan drinking water quality guidelines [23] in the average, median, maximum, and minimum. According to the literature review of the study, another drinking water quality analysis has been conducted by a research group from Iran, and their studies also revealed that the concentrations of As, Cd, Cr, Pb, Mn, Zn, Cu, and Fe in drinking water were also lower than the maximum allowed concentrations advised by the USEPA and WHO [22].

Additionally, average cadmium and lead concentrations in drinking water of both CKDu endemic areas were significantly higher (p < 0.05) than those of the reference area. Although the metal contents were found in low levels in drinking water in CKDu endemic areas, long-term exposure via drinking water may have harmful effects of etiological significance for CKDu due to bioaccumulation and dehydration.

In recent years, much attention has been given toward the evaluation of heavy metal pollution in ground and surface water with the development of a heavy metal pollution index (HPI) [24]. In order to evaluate the suitability of the water for drinking, the HPI of the drinking water in both CKDu endemic areas can be considered. The calculated HPI values were high in water from Eppawala GND which was ranging between 21.4 and 30.7. The higher values of HPI may be attributed due to the natural Apatite ore which is located in Eppawala, Sri Lanka. HPI for Ambagaswewa GND ranged from 25.2 to 27.8, and according to Moran's index, values resulted from the Spatial Autocorrelation tool showed that the heavy metal pollution in sampled drinking water sources tends to be clustered throughout Ambagaswewa GND with the higher concentrations. Lower HPI values were recorded in Dambethalawa GND in Ampara district (reference site) where no CKDu patients were recorded (Figure 2).

However, considering the classes of HPI, all the sampling locations fall under the medium range (HPI 15–40). This indicates that water is not critically polluted with respect to nephrotoxic heavy metals because the critical value of the HPI is 100 [15].

3.2. Evaluation of Agricultural Soil. In the soil, metals are found in different forms such as inorganic compounds, metal complexes, and organometallic compounds. When these metal compounds dissolve in water, they dissociate into ions and tend to behave like cations; they become part of the exchange complex and are available for absorption in plants, by displacing the essential cations [25, 26]. In paddy cultivated agricultural soils in Eppawala GND, the total average concentrations of toxic metals or metalloid/s carried a sequence of Cr > Zn > Cu > Pb > As > Cd. The total average concentrations of toxic metals or metalloid/s in the agricultural soils in Ambagaswewa GND carried a sequence of Zn > Cu > Cr > Pb > As > Cd (Table 4). However, the results indicate the total amount of toxic metals/metalloids in the paddy soil samples, and some fraction may be absorbed by the crop plants from the soil solution depending on the conditions of the soil environment.

According to the paired *t*-test outcomes between CKDu prevalence areas and the reference, the concentrations of Cr, As, and Cd of paddy soil in both CKDu hotspots were significantly higher than those of Dambethalawa GND (the reference) (at 95% confidence interval). All those three contaminants are considered nephrotoxic contaminants, and synthetic agrochemicals are known to be the main sources of heavy metal pollution in agricultural areas [27].

Mineral fertilizers used as a source of nutrients for plants may sometimes have a negative impact on the environment, mainly on soil and water. Soil pollution with heavy metals is particularly dangerous [28]. Small contents of these metals in nitrogen and potassium fertilizers do not pose any hazard of soil or plant contamination; however, phosphorus and multicomponent fertilizers used for soil deacidification are



Ambagaswewa GND

FIGURE 1: Graphical representation of lead, cadmium, arsenic concentrations, and fluoride content in collected drinking water samples in both CKDu hotspots: Eppawala GND and Ambagasewewa GND.

TABLE 2: Statistical evaluation of counterions of the drinking water sources in the selected CKDu prevalence areas: Eppawala (EP) and Ambagaswewa (AM) GNDs and the reference (RE).

Variable	Sampling site	Mean	SE mean	Minimum	Maximum
	EP	1.87	0.17	0.09	3.98
Fluoride (mg/L)	AM	1.72	0.15	0.05	4.00
C C	RE	0.40	0.03	0.10	1.10
	EP	112	13.3	18.2	357
Chloride (mg/L)	AM	42.0	5.73	0.70	135
-	RE	69.8	48.3	1.20	117
	EP	0.20	0.06	ND	1.48
Bromide (mg/L)	AM	0.12	0.06	ND	1.70
-	RE	ND	0.00	ND	0.00
	EP	3.47	0.66	ND	18.4
Nitrate (mg/L)	AM	3.24	0.20	0.39	6.08
-	RE	1.70	0.49	0.16	11.8
	EP	ND	_	ND	ND
Phosphate (mg/L)	AM	ND	_	ND	ND
	RE	ND	—	ND	ND
	EP	37.2	5.60	1.30	258
Sulfate ( $\mu$ g/L)	AM	31.6	3.75	ND	99.3
	RE	30.1	19.2	ND	67.0

ND: not detected.

usually a significant factor in heavy metal balance in the environment [10, 29].

Therefore, Pearson's correlation was done on the available metals present in the sampled agricultural soils and phosphorus content in those soil samples to check whether there is a possible relationship between soil phosphorus content and the available metal contents. The correlation matrix of soil phosphate and heavy metals in selected paddy grown agricultural areas is shown in Table 5. According to the results of the correlation matrix, there was a significant positive correlation between available soil phosphate content and cadmium and chromium concentrations in agricultural soils (significant at p = 0.05 and p = 0.01, respectively). This correlation may be a result of the sulfuric acid used at manufacturing fertilizers containing phosphorus in a watersoluble form. During chemical processing of these minerals,

mostly cadmium and chromium pass into the soluble phase and then, as a result of the technological process, to the fertilizers. Therefore, the rates of usage of phosphorus and multicomponent fertilizers are usually determined on the basis of their phosphorus contents, and there is a direct connection between phosphorus content and an increase in heavy metal concentrations in agricultural soil. Even worse, the fertilizers do not replace the trace minerals resulting in mineral-depleted soil. When paddy grows in mineral-depleted soil, it easily absorbs metals like Cd which is too toxic to the human kidneys [30].

Apart from that, the prolonged exposure of crop plants in soils with heavy metals increases their absorption capacity and depends on factors such as pH, cation exchange capacity, organic matter content, clay content, and redox potential; these determine the soil capacity to retain or

TABLE 3: Statistical evaluation of concentrations of selected toxic metals/metalloids of the drinking water sources in the selected areas: Eppawala (EP) and Ambagaswewa (AM) GNDs and the reference (RE).

Metal element	Sample	Mean	±SE mean	Minimum	Median	Maximum
	EP	0.27	0.03	0.02	0.13	0.66
Cr ( $\mu$ g/L)	AM	0.26	0.05	0.01	0.13	1.13
	RE	0.22	0.03	0.05	0.25	0.65
	EP	157	25.6	0.30	25.0	485
Mn (µg/L)	AM	62.0	9.46	0.10	19.5	163
	RE	129	19.5	0.20	189	122
	EP	135	38.9	0.00	89.5	867
Fe ( $\mu$ g/L)	AM	60.6	13.7	0.40	16.2	305
	RE	260	41.3	0.60	189	299
	EP	1.37	0.16	0.49	0.96	4.59
Cu (µg/L)	AM	0.67	0.05	0.21	0.65	1.25
	RE	0.98	0.09	ND	0.88	2.88
	EP	0.25	0.30	0.03	0.19	0.68
As $(\mu g/L)$	AM	0.19	0.02	ND	0.15	0.57
	RE	0.55	0.08	0.05	0.72	2.29
	EP	0.07	0.00	ND	0.06	0.07
Cd ( $\mu$ g/L)	AM	0.02	0.00	ND	0.01	0.13
	RE	0.01	0.02	ND	0.34	0.45
	EP	0.32	0.07	ND	ND	1.78
Pb (µg/L)	AM	0.20	0.02	0.05	0.14	0.66
	RE	0.15	0.16	0.07	0.21	0.48
	EP	40.0	10.3	2.60	18.3	289
Zn (µg/)	AM	12.0	1.73	1.32	6.53	38.9
	RE	30.7	8.16	0.91	6.58	106

ND: not detected.

mobilize heavy metals [31]. Heavy metals in trace amounts can accumulate in soils of agricultural areas, due to their characteristics and buffering capacity [32, 33].

The risks, the degree of toxicity, and the persistence of the metals depend on the impact which soils receive by different anthropogenic activities. The use and application of geoaccumulation indexes ( $I_{geo}$ ) will identify the source of pollutants and the degree of bioaccumulation in soil [34] (Figure 3).

For all considered metals in the studied samples (for CKDu endemic areas), the  $I_{geo}$  values presented the decreasing order of As > Pb > Cu > Ni > Cr > Zn > Cd for Eppawala GND and for Ambagaswewa GND, and the  $I_{geo}$  values presented the decreasing order of As > Cu > Cr > Ni > Cd > Pb > Zn. The mean  $I_{geo}$  values of all the studied metals in all sampling sites indicate that the soils are slowly contaminated with heavy metals. A trend like this was also revealed in another study of agricultural soil analysis. Rostami et al. depict that agricultural soil was contaminated by Cr, Cu, Ni, Pb, and Zn but was moderately contaminated by Cd and As when considering the  $I_{geo}$  for the soil [34].

According to the findings, soil samples from both CKDu prevalence areas and the reference area have moderately contaminated/polluted with the metal contaminants and belong to class 1. However, none of the selected area's paddy soil has exceeded the class 2 contamination level. And also most of the  $I_{\text{geo}}$  values of the paddy soil in the CKDu nonendemic area (reference) were much low with respect to CKDu endemic areas, and agricultural soil in CKDu

endemic areas tends to convert to the contaminated position with some heavy metals.

Those metal contaminants are important since they are capable of decreasing crop production due to the risk of bioaccumulation and biomagnification in the food chain, and there is also a risk of superficial and groundwater contamination [8]. Therefore, remediation of agricultural soil contaminated by heavy metals is necessary in order to reduce the associated health risks, make the land resource available for agricultural production, and enhance food security.

3.3. Assessment of Toxic Metals in Commonly Used Fertilizers in Sri Lanka. Another problematic case is agrochemicals including fertilizers and pesticides which were introduced to Sri Lanka in the 1970s. Sri Lankan scientists reported that paddy soils in Mahaweli development areas (most of the CKDu endemic areas) are polluted with potentially toxic metals, and paddy soils in Sri Lanka are highly modified by artificial fertilizer applications [35]. Table 6 shows the contribution of the commonly used fertilizers to toxic metal contamination in the soil in Sri Lanka. According to that, urea which is commonly used in paddy cultivation as well as other cultivations in Sri Lanka has contained toxic metals such as Cd, As, Cr, Pb, Zn, and Cu in noticeable amounts. Apart from that, potash and triple superphosphate that are applied in the paddy cultivation also have contained significant amounts of the aforementioned toxic metals except



FIGURE 2: Interpolation of heavy metal pollution index (HPI) in sampling location via drinking water analysis: (a) Eppawala GND in Anuradhapura district, (b) Ambagaswewa GND in Polonnaruwa district, and (c) Dambethalawa GND in Ampara district (the reference).

TABLE 4: Descriptive statistics on concentrations of environmental toxicants including toxic metals/metalloids and phosphates of paddy soil samples (CKDu hotspots: Eppawala (EP) and Ambagaswewa (AM) GNDs, and the reference: Dambethalawa (RE) GND).

Variable	Sample	Mean	SE mean	Minimum	Median	Maximum
	EP	49.9	2.12	32.6	49.9	64.2
Cr (mg/kg)	AM	33.0	2.01	19.6	30.6	48.4
	RE	8.87	1.19	3.59	7.80	24.1
	EP	302	29.9	164	273	611
Mn (mg/kg)	AM	590	203	173	397	3325
	RE	125	24.6	18.5	96.3	417
	EP	23765	1142	15052	24075	30728
Fe (mg/kg)	AM	30461	5103	12995	25735	95805
	RE	7117	1178	2627	5799	22581
	EP	8.08	0.51	4.73	7.81	12.0
Co (mg/kg)	AM	13.8	1.50	5.90	14.1	21.6
	RE	3.35	0.64	0.93	2.90	11.5
	EP	5.03	0.12	4.04	5.08	5.65
As (mg/kg)	AM	3.30	2.23	0.57	0.97	34.4
	RE	0.48	0.04	0.22	0.45	0.84
	EP	0.11	0.01	0.08	0.09	0.25
Cd (mg/kg)	AM	0.10	0.01	0.04	0.10	0.22
	RE	0.04	0.000	0.02	0.04	0.08
	EP	4290	171	3090	4376	5355
Ca (mg/kg)	AM	2288	1895	136	268	28762
· -	RE	59.35	9.71	28.7	50.0	177

Variable	Sample	Mean	SE mean	Minimum	Median	Maximum
	EP	2359	103	1495	2339	2827
Mg (mg/kg)	AM	1659	185	710	1654	3190
0 0 0	RE	651	165	209	497	2870
	EP	9.88	0.32	6.60	10.2	11.2
Pb (mg/kg)	AM	6.80	1.20	3.52	6.26	22.7
	RE	36.5	8.14	1.35	35.9	107
	EP	24.6	0.87	15.7	25.0	28.5
Cu (mg/kg)	AM	36.4	19.9	7.70	17.8	313
0 0	RE	2.90	0.38	1.20	2.10	5.66
	EP	34.6	1.95	27.4	32.4	59.8
Zn (mg/kg)	AM	39.4	8.74	15.1	32.3	156
0 0	RE	14.6	2.43	8.32	11.2	39.6
	EP	288	14.6	207	294	383
Na (mg/kg)	AM	337	161	84.0	186	2581
0 0	RE	58.5	5.23	30.1	54.8	90.9
	EP	26866	974	17959	28426	30312
Al (mg/kg)	AM	22735	2322	9511	19625	44042
0.0	RE	1276	509	ND	ND	4971
	EP	1106	46.9	539	1125	1350
K (mg/kg)	AM	625	167	235	394	2799
0 0	RE	320	62.7	135	245	1155
	EP	16.9	0.63	11.1	17.4	20.2
Ni (mg/kg)	AM	11.9	1.23	5.07	10.4	24.2
0 0	RE	1.03	0.36	ND	ND	3.21
	EP	110	6.24	61.0	111	141
$PO_4^{3-}$ (mg/kg)	AM	8.19	0.90	3.62	7.56	16.6
	RE	7.94	1.48	3.00	4.96	20.5

TABLE 4: Continued.

TABLE 5: Correlation matrix of phosphate and heavy metal concentrations of soil in paddy grown agricultural areas in selected sampling areas.

	[PO <sub>4</sub> <sup>3-</sup> ]	[Cd]	[Cr]	[As]	[Pb]	[Cu]	[Zn]	[Fe]	[Mn]
[PO <sub>4</sub> <sup>3-</sup> ]		0.361*	0.712**	0.28	-0.245	0.061	0.171	0.147	-0.04
[Cd]	0.361*	_	0.473**	-0.197	0.576**	$0.250^{*}$	0.212	-0.165	-0.337
[Cr]	0.712**	0.473**	_	-0.070	$0.216^{*}$	0.064	-0.074	-0.062	-0.112
[As]	0.28	-0.197	-0.070	—	0.047	0.219	-0.089	-0.050	-0.202
[Pb]	-0.245	0.576**	0.216*	0.047	_	0.052	0.103	0.032	0.120
[Cu]	0.061	0.250*	0.064	0.219	0.052		0.016	0.213	0.020
[Zn]	0.171	0.212	-0.074	-0.089	0.103	0.016	—	0.301	-0.120
[Fe]	0.147	-0.165	-0.062	-0.050	0.032	0.213	0.301		0.045
[Mn]	-0.040	-0.337	-0.112	-0.202	0.120	0.020	-0.120	0.045	—

\*Correlation is significant at the 0.05 level (2-tailed). \*\*Correlation is significant at the 0.01 level (2-tailed).



FIGURE 3: Geoaccumulation index  $(I_{geo})$  for heavy metals in paddy cultivated soil of selected GN areas.

	Mean metal contents in fertilizers ± SD								
Fertilizer type	Cr (mg/kg)	Mn (mg/kg)	Fe (mg/kg)	Cu (mg/kg)	Zn (mg/kg)	As (mg/kg)	Cd (mg/kg)	Pb (mg/kg)	
Urea	102 (±4.17)	2211 (±128.1)	2301 (±108.1)	21.3 (±1.1)	723 (±96.5)	59.9 (±2.72)	1.74 (±0.14)	289 (±9.78)	
MOP (muriate of potash)	83.6 (±4.4)	4.20 (±0.75)	702 (±116.1)	0.58 (±0.17)	93.8 (±20.4)	ND	ND	6.34 (±1.64)	
TSP (triple superphosphate)	386 (±44.8)	4234 (±55.8)	2642 (±466)	141 (±15.9)	348 (±21.9)	ND	ND	124 (±14.5)	
ND and detected									

TABLE 6: Selected heavy metal/metalloid contents in fertilizer samples collected from the selected CKDu endemic area.

ND: not detected.

Cd and As. Furthermore, the risk of accumulation of toxic metals is augmented due to the limitless application of fertilizers in paddy cultivation. Owing to toxic element contaminated fertilizer application in the paddy areas, tank sediment is also contaminated as a long-term influence. Hence, those toxic metals can be incorporated into human food chains due to the consumption of freshwater fish [36–38].

#### 4. Conclusions

Incorporation of the toxic metals into the food chains mainly occurs via contaminated paddy soil in rice consuming regions in the world as major food items such as the Asian region. In this study, selected CKDu endemic areas in Sri Lanka are also identified as an agricultural area, and for the first time, this study compared the hazardous metal contamination of drinking water and agricultural soil in CKDu endemic areas with a nonendemic area in Sri Lanka. Even if the current study reveals that toxic metal contents in drinking water samples in CKDu endemic areas were far below the permissible limits, long-term exposure of the toxic metals via drinking water may generate a risk. Apart from that, the persistence of the toxic metals/metalloids such as Cr, As, Cd, and Pb in frequently applying fertilizers would be the major reason for paddy soil pollution with toxic metals/metalloids. Geoaccumulation indexes reveal that paddy soil in selected CKDu endemic areas is being transferred to a moderate contamination stage, and frequently, metal-contaminated fertilizer applications may augment the risk of entering toxic metals into the paddy plants and then rice grins from the soil solution. Hence, the long-term accumulation of toxic metals/metalloids definitely affects human kidney functions. There is an urgent need to expand the evaluation of the quality of the fertilizers' use in Sri Lanka.

#### **Data Availability**

No data were used to support this study.

#### **Conflicts of Interest**

The authors declare that they have no known competing financial interest or personal relationships that could have appeared to influence the work reported in this paper.

#### **Authors' Contributions**

W. P. R. T. Perera conceptualized the study and performed data analysis and investigations; D. M. U. C. Dissanayake, R. A. S. D. Rathnasekara, and K. A. M. Kularathne were responsible for formal analysis and methodology; M. D. N. R. Dayananada and W. S. M. Botheju prepared the original draft and further edited the manuscript; Janitha A. Liyanage was responsible for funding acquisition, resources, and supervision; S. K. Weragoda was responsible for resources and supervision in instrumentation.

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