

Research Article

Anti-Glycation, Anti- β -Amyloid Aggregation, and Antioxidant Effect of Cassia Seed-Derived Secondary Metabolites

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Aggregation and non-enzymatic glycation of amyloidogenic peptides, amyloid-beta ($A\beta$), and insulin are key features of neuropathology. In this study, we evaluate the anti-glycation effect of Cassia seed-derived secondary metabolites on human insulin and bovine serum albumin, as well as their anti- $A\beta$ aggregation and antioxidant effects using in vitro spectrofluorometric method, thioflavin T fluorescence, and peroxynitrite ($ONOO^-$) scavenging assay. Furthermore, molecular docking simulation was performed to investigate the binding characteristics of test compounds and $A\beta$ 42 peptide. Among 38 compounds, anthraquinones **9–12** and **14–18**; naphthopyrones **24**, **25**, **27**, and **33–36**; and **38** from the naphthalenes and naphthalenic lactone groups showed moderate-to-good inhibition of AGE formation with IC_{50} values ranging from 7.52 ± 1.19 to $155.86 \pm 0.79 \mu M$. Likewise, compounds **5**, **24**, **15**, **16**, **27**, and **20** showed good inhibition of D-ribose-mediated glycation of human insulin, with an IC_{50} value range of 46.37 ± 4.06 to $97.69 \pm 7.88 \mu M$. In the thioflavin-T assay, compounds **8** and **12** showed promising inhibition of $A\beta$ aggregation, comparable to that of the reference compound morin. Molecular docking simulations confirmed that these active compounds have strong potential to interact with $A\beta$ 42 peptides and interrupt their self-assembly and conformational transformation, thereby inhibiting $A\beta$ 42 aggregation. In addition, compounds **5**, **8**, **10**, **14**, and **35** scavenged $ONOO^-$ at low concentrations. Overall, Cassia compounds' anti-glycation, anti- $A\beta$ aggregation, and antioxidant effects warrant further in vivo studies to evaluate their potential neuroprotective effects against comorbid AD and diabetes.

1. Introduction

The non-enzymatic glycation or oxidation of the amino ($-NH_2$) groups of amino acids, proteins, peptides, and hormones by reducing sugars results in a set of different products, the advanced glycation end products (AGEs), that assemble in various parts of the human body and increase

the risk of diabetic complications, including neuropathy, nephropathy, and cardiovascular diseases. In addition to their role in diabetic complications, AGEs have also been implicated in the pathogenesis of Alzheimer's disease (AD) and other neurodegenerative diseases [1]. The presence of senile plaques and neurofibrillary tangles (NFTs) characterizes the pathology in AD, and amyloid β protein ($A\beta$) and

tau proteins are their major constituents. Glycation of tau protein enhances paired helical filament formation [2], and glycation of A β prolongs its half-life, which accelerates amyloid protein aggregation [3].

Similarly, glycation of insulin and proinsulin in the insulin-resistant diabetic state [4] and α -synuclein in Parkinson's disease (PD) [5] is well documented. Spheroproteins or normal globular proteins can be refolded into amyloid fibrils (a major component of proteinaceous plaques), comprising a cross- β structure, during glycation [6]. In addition, a highly reactive intermediate product, 3-deoxyglucosone, of the glycation reaction can reinforce the formation of intermolecular and intramolecular cross-links in AGE-modified protein monomers.

Glycation of proteins by glucose occurs via several pathways [7]: (a) through nucleophilic addition between an amino group of a protein and the carbonyl group of glucose to form a reversible Schiff base, (b) through oxidative degradation in the presence of reactive oxygen species (ROS), and (c) through the reaction of dicarbonyl compounds from glucose auto-oxidation with the amino groups of proteins. Hormones, peptides, and proteins are endogenous sources of AGEs that are responsible for disease development. A modern diet comprised of high-heat-processed meats, processed milk, and bakery products with enhanced flavor and taste is a major contributor to oxidative stress, inflammation, and liver fibrosis, along with several chronic disease states [8]. In regard to the role of AGEs in the development of diabetic complications and other diseases, various natural and synthetic compounds have been evaluated for their anti-glycation properties. Inhibiting any step of glycation and preventing the formation of intermediate and end products, scavenging of free radicals, detoxification of liver enzymes, etc. can help in the prevention of AGE formation and diabetic complications [9]. To date, numerous synthetic compounds with anti-glycation properties have been discovered including but not limited to urea and thiourea derivatives of glycine/proline conjugated benzoxazole analog [10], bis-Schiff bases of isatins [11], benzimidazole derivatives [12], oxindole derivatives [13], bergenin derivatives [14], and piroxicam derivatives [15]. However, due to fewer complications compared to synthetic compounds, natural compounds are considered a better treatment for inhibiting the glycation process and AGE formation. Therefore, there is an emerging interest in the discovery of new anti-glycation agents from natural sources. With this aim, we investigate *Cassia obtusifolia* Linn seeds-derived secondary metabolites.

Cassia obtusifolia Linn seed is a traditional Chinese medicine that is well known for its anti-inflammatory and antioxidant properties. Cassia seed is consumed as brewed tea in South Korea. Cassia seed extract has been reported to have numerous biological activities, including hypolipidemic [16, 17], antifungal [18], hypotensive [19], antioxidant [20], and neuroprotective [21] effects. Similarly, secondary metabolites from Cassia seeds have larvicidal [22], antioxidant [23], anti-AD [24], hepatoprotective [25], antidiabetic [26], antihypertensive [27], comorbid diabetes and depression [28] and antimutagenic effect [29]. However, there

are a limited number of studies on the anti-glycation properties of Cassia seed-derived metabolites. A few numbers of naphthopyrone glycosides [30] and anthraquinones [31] have previously been reported to have an inhibitory effect on BSA-glycation. This study reports the anti-glycation properties of other Cassia compounds in human insulin and bovine serum albumin, as well as anti-A β aggregation and antioxidant effect. To our knowledge, this is the first report of the anti-glycation and insulin-glycation inhibition properties of these metabolites.

2. Materials and Methods

2.1. Chemicals and Reagents. Bovine serum albumin (BSA), L-penicillamine, D-(–)-fructose, D-(+)-glucose, D-(–)-ribose, human insulin, aminoguanidine hydrochloride, dihydro-rhodamine (DHR) 123, and diethylenetriamine-pentaacetic acid (DTPA) were purchased from Sigma-Aldrich (St. Louis, MO, USA). The A β 42 peptide was purchased from Bachem AG (Bubendorf, Switzerland). Peroxynitrite was obtained from Calbiochem/Merck Millipore (Darmstadt, Germany). Solvents for column chromatography were of reagent grade and were used as received from commercial sources.

2.2. Isolation of Compounds. Details on plant material, extraction, fractionation, and isolation have been described in our recent work [32]. Chemical structures of isolated compounds are shown in Figure 1.

2.3. Bovine Serum Albumin Glycation Inhibition Assay. Following some modifications in the spectrofluorometric method previously described by Vinson and Howard [33], the inhibitory potential of isolated compounds in AGE formation was evaluated. At first, the AGE reaction mixture was prepared by mixing 10 mg/ml BSA in 50 mmol/L sodium phosphate buffer (pH 7.4), 0.2 M fructose, and 0.2 M glucose. To prevent bacterial growth, the reaction mixture was maintained in 0.02% sodium azide. Subsequently, 950 μ L of reaction buffer was mixed with 50 μ L of the test sample (2.5 μ M to 250 μ M) solution in vials and incubated at 37°C for 7 days. Color controls were prepared similarly and incubated at 4°C for 7 days. The inhibition percentage was calculated by measuring the fluorescence intensity of the reaction product at 450 nm after excitation at 350 nm in a microplate spectrofluorometer (Molecular Devices, Sunnyvale, CA, USA). The inhibition result was validated with aminoguanidine HCl as a positive control.

2.4. Human Insulin Glycation Inhibition Assay. The inhibition potential of test compounds on insulin glycation was evaluated according to a recently described method [34], with slight modifications. First, monomeric insulin was obtained by dissolving human insulin in ultra-pure water at pH 2.0. Then, the monomeric insulin solution was neutralized to pH 7.0 and kept in phosphate buffer (50 mM; pH 7.0). To obtain a glycated insulin control, 10 μ L of monomeric human insulin (final concentration:

0.5 mg/mL), 10 μ L of 0.5 M D-ribose in phosphate buffer (50 mM; pH 7.0), and 10 μ L of ultra-pure water were mixed, filtered through 0.2 μ m filter, and incubated at 37°C for 7 days. The anti-glycation properties of test compounds were evaluated by adding 10 μ L of the test sample (2.5 μ M to 250 μ M; replacing 10 μ L water from the control) to a mixture of insulin and D-ribose and incubating at 37°C for 7 days. For the protein control, human insulin was incubated with buffer only. Then, the fluorescence signal was measured at λ_{ex} 320 nm/ λ_{em} 410 nm in a microplate spectrofluorometer (Molecular Devices, Sunnyvale, CA, USA).

2.5. Anti-Aggregation Assay. ThT is known to exhibit fluorescence upon binding to the β -sheet structure [35]. We used ThT analysis to investigate the effect of test compounds on the aggregation of A β that undergoes structural conversion into a β -sheet-rich fiber. For A β aggregation, we added 3rd distilled water to synthetic A β 42 to produce A β 42 at a concentration of 25 μ M. The A β peptide chemically treated with aqueous ammonia, purchased from Bachem AG (Bubendorf, Switzerland), is monomeric and has no fluorescence when measured by the ThT assay. We used a 25 μ M concentration that was obtained from 50% toxicity among different concentrations in the cell viability test. The A β 42 peptide, purchased from Bachem AG (Bubendorf, Switzerland), was dissolved in 0.1 M aqueous ammonia. To investigate the effects of Cassia compounds on the aggregation of A β , ThT assays with non-aggregated A β 42 were performed. ThT assays were carried out in black polypropylene 96-well plates (SPL Life Science, Republic of Korea). For A β self-aggregation, triple distilled water was added to the A β 42 stock until it reached a concentration of 25 μ M. Various concentrations of Cassia compounds (2.5, 25, and 250 μ M) and the positive control, morin (100 μ M), were added together with a ThT solution (15 μ M in 50 mM glycine buffer, pH 8.9) into the 25 μ M A β 42 solution. All solutes used in the experiment were well soluble in the solvent. They were then incubated for 3 h at 37°C (Figure 2). Finally, the ThT fluorescence intensity was measured at excitation wavelengths of 440 nm and emission wavelengths of 484 nm using a SpectraMax iD3 Multi-Mode Microplate Reader (Molecular Devices, San Jose, CA, USA). Experiments were performed in triplicate.

2.6. Molecular Docking Simulation. We used AutoDock 4.2 to investigate the binding characteristics of test compounds and A β 42 peptide. To determine the receptor structures, we obtained the three-dimensional structures of A β 42 peptides from the Protein Data Bank (PDB code: A β 42: 1Z0Q and 2BEG). Two- and three-dimensional structures of the test compounds and morin were obtained from the PubChem database, and energy minimization and ligand structure conversion to the PDB file format were performed using Chimera 1.15 (<https://www.cgl.ucsf.edu/chimera>). For molecular docking purposes, we prepared receptors and ligands using AutoDockTools, whereas to conduct molecular docking, we applied the prepared macromolecules and

ligands using AutoDockTools [36]. In the case of the monomer structures of A β , we used Autogrid to generate grid parameters centered on the reference ligand and created a grid box of 60 \times 60 \times 60 Å. In the case of A β 42 pentamer (2BEG), the grid parameters were generated using the mean coordinates of the hydrophobic core (16KLVFFA 21) and C-terminal (29GAIIGLMVGGVVIA42) segments, and grid boxes of 90 \times 90 \times 90 Å were set at the established coordinates using Autogrid. One hundred Lamarckian generic algorithms were run for docking, and the other docking parameters were set to default values. The docking results were analyzed using AutoDockTools and visualized using Discovery Studio (v17.2, Accelrys, San Diego, CA, USA). The docking pose with the lowest binding energy and the highest number of cluster populations was selected for the analysis of docking results.

2.7. Peroxynitrite (ONOO⁻) Scavenging Assay. ONOO-scavenging activity was assessed using the modified Kooy's method, which involves the monitoring of highly fluorescent rhodamine 123, which is rapidly produced from non-fluorescent DHR 123 in the presence of ONOO⁻ [37]. In brief, the rhodamine buffer (pH 7.4) consisted of 50 mM sodium phosphate dibasic, 50 mM sodium phosphate monobasic, 90 mM sodium chloride, 5 mM potassium chloride, and 100 μ M DTPA. The final DHR 123 concentration was 5 μ M. The assay buffer was prepared before use and placed on ice. Test samples were dissolved in 10% DMSO (final conc., 100 μ M). The background and final fluorescent intensities were measured 5 min after treatment with and without the addition of authentic ONOO⁻ (10 μ M), dissolved in 0.3 N sodium hydroxide. The fluorescence intensity of the oxidized DHR 123 was evaluated using a microplate spectrofluorometer (Bio-Tek Instruments Inc., Winooski, VT, USA) at excitation and emission wavelengths of 480 and 530 nm, respectively. The values of ONOO⁻ scavenging activity were calculated as the final fluorescence intensity minus the background fluorescence, via the detection of DHR 123 oxidation. L-Penicillamine was used as the positive control.

2.8. Statistical Analysis. All statistical analyses were conducted using the GraphPad Prism 7.0 software (GraphPad Software, La Jolla, CA, USA). All statistical data are displayed as mean \pm standard error of the mean (SEM). An independent *t*-test was used to evaluate the normality of the data. Statistical significance was set at $P < 0.05$.

3. Results

3.1. Inhibition of D-Glucose/D-Fructose-Mediated Glycation of Bovine Serum Albumin. The protective effect of anthraquinones, naphthopyrones, naphthalenes, and naphthalenic lactones (Figure 1) isolated from Cassia seeds against protein glycation was evaluated in vitro, and the results are tabulated in Table 1. As shown in the table, most of the anthraquinones showed good inhibition potential on AGE formation.

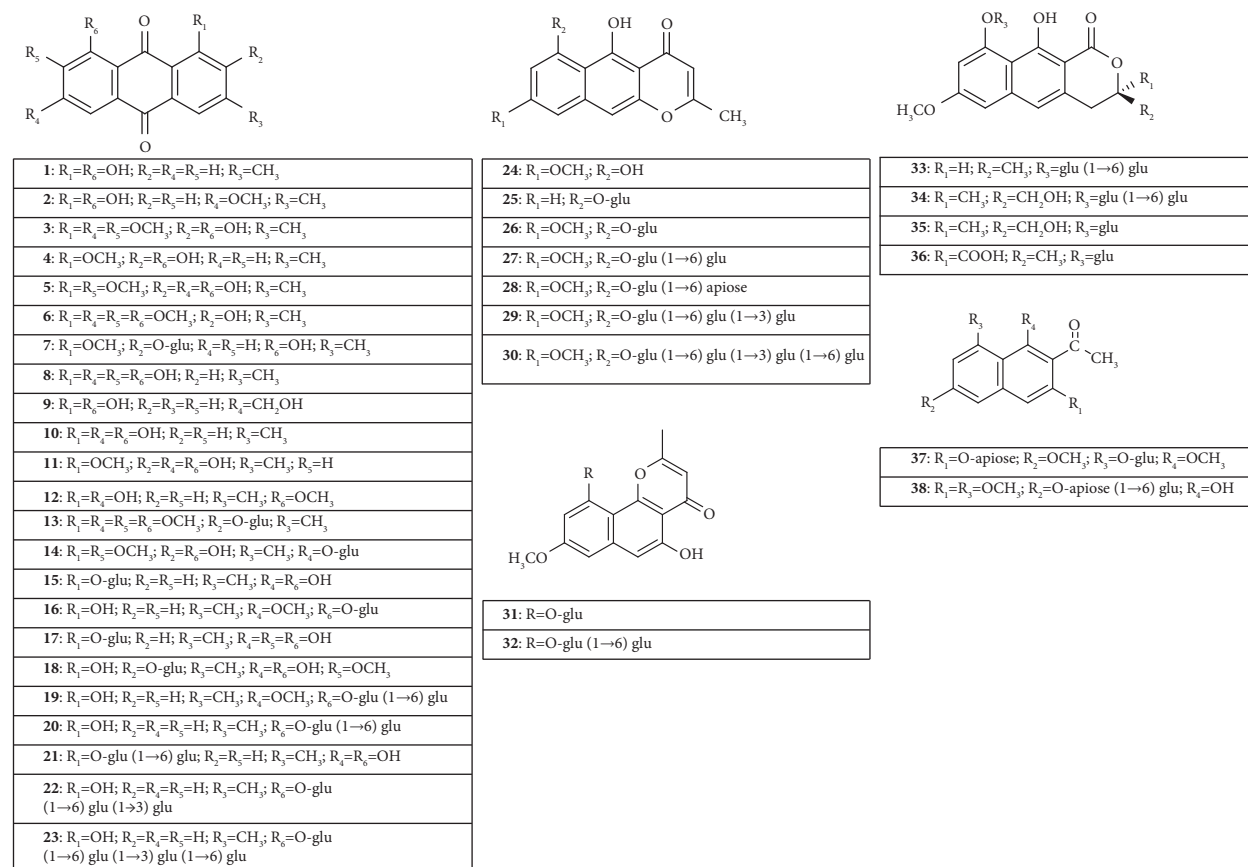


FIGURE 1: Chemical structures of compounds isolated from *Cassia obtusifolia* Linn seeds.

Among the active anthraquinone aglycones, **8** (IC₅₀; 11.50 ± 0.07 μM) was the most active, followed by **11** (IC₅₀; 86.29 ± 7.09 μM) and **12** (IC₅₀; 118.12 ± 3.11 μM). Aglycones **9**, **10**, and **5** had a similar effect (IC₅₀ values approx. 170 μM), while **6** was slightly weaker (IC₅₀; 189.61 ± 1.70 μM). However, aglycones **1–4** were ineffective up to 200 μM under the given experimental conditions.

Similarly, among anthraquinone glucosides, only single glucose-bearing anthraquinones (**14–18**, except **7** and **13**) were active. The most potent among these compounds was **18** (IC₅₀; 32.51 ± 1.40 μM). Other glucosides **14–17** had an IC₅₀ in the range of 105 to 155 μM. Anthraquinone glycosides with two or more glucose moieties (**19–23**) were ineffective up to 200 μM.

From the naphthopyrone series, only **24**, **25**, **27**, and **33–36** were active. The most potent compound in this series was **36** (IC₅₀; 7.52 ± 1.19 μM), followed by **38** (IC₅₀; 11.63 ± 0.12 μM), **37** (IC₅₀; 23.40 ± 1.05 μM), **27** (IC₅₀; 38.89 ± 2.05 μM), **35** (IC₅₀; 40.36 ± 1.36 μM), **24** (IC₅₀; 42.58 ± 1.79 μM), and **25** (IC₅₀; 101.37 ± 4.72 μM). Other glycosides **26** and **28–32** showed no observable effect up to 200 μM.

Similarly, between the two compounds from naphthalenes and naphthalenic lactone group, only **38** showed a moderate effect, with an IC₅₀ value of 89.03 ± 3.08 μM. Interestingly, **33** was ineffective up to 200 μM. The overall results of this in vitro anti-glycation assay revealed that naphthopyrones from Cassia seeds are promising inhibitors

of protein glycation. Furthermore, all of the active compounds had better activity than the positive control, aminoguanidine HCl (IC₅₀; 590.94 ± 2.81 μM).

3.2. Inhibition of D-Ribose-Mediated Glycation of Human Insulin. Next, we evaluated the effect of Cassia compounds on D-ribose-mediated glycation of human insulin. As shown in Table 1, among twenty-three anthraquinones, only four (**5**, **15**, **16**, and **20**) showed moderate inhibition of insulin glycation with an IC₅₀ value range of 46.37 ± 1.19 to 83.23 ± 0.38 μM. Rubrofusarin (**24**) and its gentiobioside (**27**) also showed moderate inhibitory activity with IC₅₀ values of 87.03 ± 3.93 and 97.69 ± 7.88 μM, respectively. None of the naphthalene and naphthalenic lactones exhibited inhibitory activity.

3.3. Inhibitory Effect on Aβ42 Self-Aggregation. Some anthraquinones, naphthopyrone, and flavonoids have been reported to inhibit the formation of Aβ fibrils in a ThT study [38–40]. In this study, the effect of aurantio-obtusin (**5**), chryso-obtusin (**6**), aloë-emodin (**9**), 2-hydroxyemodin (**8**), questin (**12**), glucoaurantio-obtusin (**14**) and naphthopyrones – nor-rubrofusarin glucoside (cassiaside, **25**), rubrofusarin apiosylglucoside (**28**), and toralactone gentiobioside (**33**) on Aβ42 aggregation was evaluated. We observed that various concentrations of these

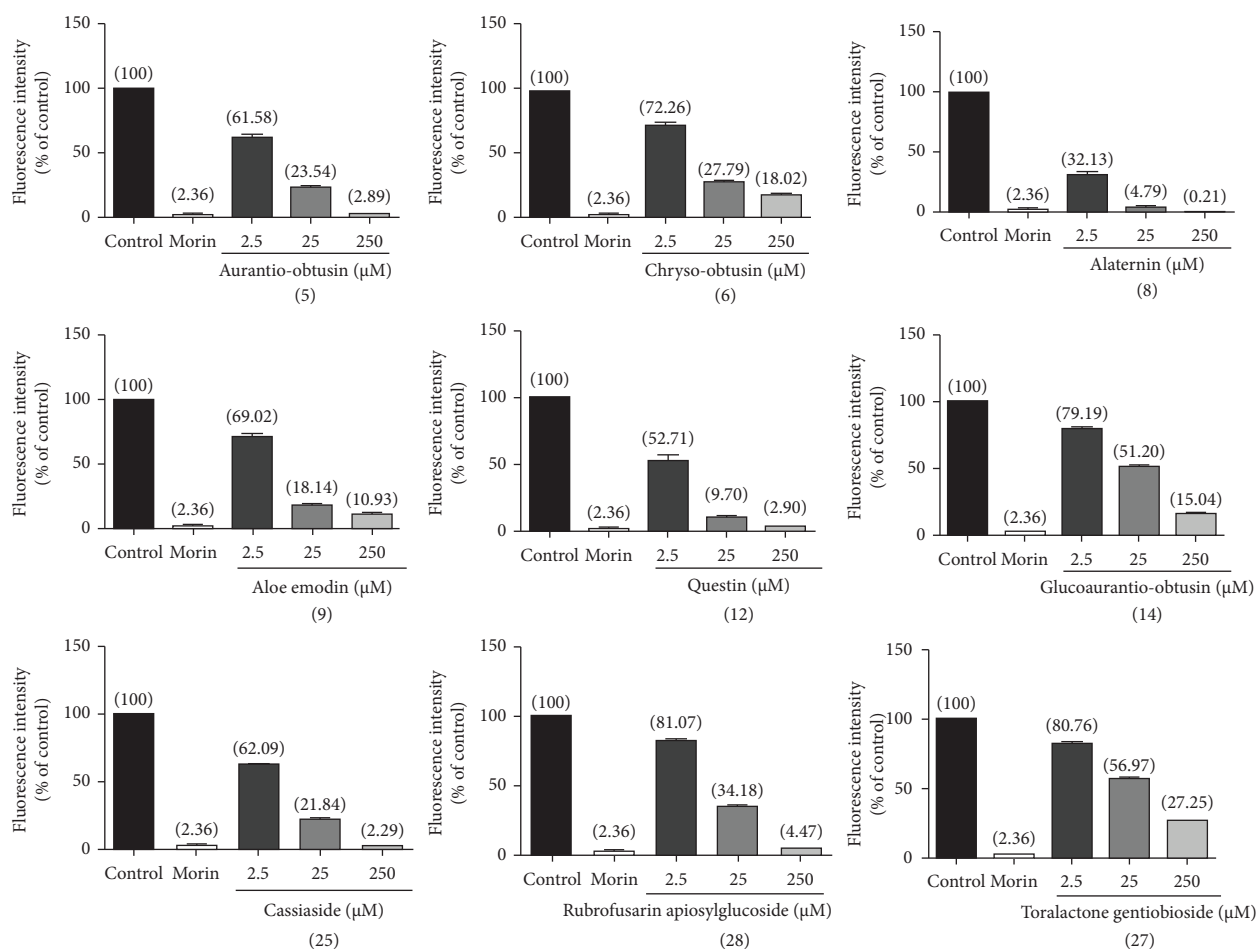


FIGURE 2: Anti-aggregation effect of different kinds of Cassia compounds on $A\beta_{42}$. The molar ratios of $A\beta$: Cassia compounds are 10:1 (2.5 μM), 1:1 (25 μM), and 1:10 (250 μM). The vehicle group was treated with 3rd distilled water. Morin (100 μM) was used as a positive control as an $A\beta$ aggregation inhibitor. Values in parentheses are percentages of the control value. Numbers below the compound name in each graph represent the assigned compound number.

TABLE 1: Anti-glycation effect of major anthraquinones and naphthopyrones from *Cassia obtusifolia* Linn seeds.

Compounds	BSA	Human insulin IC ₅₀ (μM) ^a
<i>Anthraquinones</i>		
Chrysophanol (1)	>200	ND
Physcion (2)	>200	ND
Obtusin (3)	>200	ND
Obtusifolin (4)	>200	ND
Aurantio-obtusin (5)	166.33 ± 0.83	55.69 ± 1.19
Chryso-obtusin (6)	189.61 ± 1.70	135.26 ± 2.98
Obtusifolin 2-glucoside (7)	>200	>200
7-Hydroxyemodin (alaternin) (8)	11.50 ± 0.07	>200
Aloe-emodin (9)	169.79 ± 2.97	122.27 ± 5.49
Emodin (10)	169.56 ± 1.09	>200
2-Hydroxyemodin 1-methyl ether (11)	86.29 ± 7.09	140.82 ± 2.56
Questin (emodin-8-methyl ether) (12)	118.12 ± 3.11	194.80 ± 5.01
Chryso-obtusin 2-glucoside (13)	>200	>200
Glucoaurantio-obtusin (14)	155.86 ± 0.79	146.60 ± 4.69
Emodin 1-O- β -D-glucopyranoside (15)	105.61 ± 6.32	67.45 ± 3.02
Physcion 8-O- β -D-glucopyranoside (16)	120.64 ± 2.97	83.23 ± 0.38
Alaternin 1-O- β -D-glucopyranoside (17)	113.49 ± 3.05	159.12 ± 9.61
1-Desmethyaurantio-obtusin 2-O- β -D-glucopyranoside (18)	32.51 ± 1.40	>200
Physcion 8-O- β -gentiobioside (19)	>200	>200
Chrysophanol 1-O- β -gentiobioside (20)	>200	46.37 ± 4.06

TABLE 1: Continued.

Compounds	BSA	Human insulin IC ₅₀ (μM) ^a
Emodin 1-O-β-gentiobioside (21)	>200	>200
Chrysophanol triglucoside (22)	>200	>200
Chrysophanol tetraglucoside (23)	>200	>200
<i>Naphthopyrones</i>		
Rubrofusarin (24)	42.58 ± 1.79	87.03 ± 3.93
Cassiaside (25)	101.37 ± 4.72	>200
Rubrofusarin 6-O-β-D-glucopyranoside (26)	>200	>200
Rubrofusarin 6-O-β-D-gentiobioside (27)	38.89 ± 2.05	97.69 ± 7.88
Rubrofusarin 6-O-β-D-apiofuranosyl-(1→6)-O-β-D-glucopyranoside (28)	>200	175.38 ± 6.62
Rubrofusarin triglucoside (29)	>200	128.35 ± 4.39
Rubrofusarin tetraglucoside (cassiaside B2) (30)	>200	>200
Demethylflavesperone 10-O-β-D-glucopyranoside (31)	>200	>200
Isorubrofusarin gentiobioside (32)	>200	163.79 ± 5.97
Toralactone 9-O-β-gentiobioside (33)	40.36 ± 1.36	>200
Cassialactone 9-O-β-gentiobioside (34)	7.52 ± 1.19	>200
(3R)-Cassialactone 9-O-β-D-glucopyranoside (35)	23.40 ± 1.05	>200
(3S)-9,10-Dihydroxy-7-methoxy-3-methyl-1-oxo-3,4-dihydro-1H-benzo[g]isochromene-3-carboxylic acid 9-O-β-D-glucopyranoside (36)	11.63 ± 0.12	>200
<i>Naphthalenes and Naphthalenic lactones</i>		
Cassitoroside (37)	>200	>200
1-Hydroxyl-2-acetyl-3,8-dimethoxy-naphthalene-6-O-β-D-apiofuranosyl-(1→2)-β-D-glucopyranoside (38)	89.03 ± 3.08	>200
Aminoguanidine ^b	590.94 ± 2.81	—

^aThe 50% inhibitory concentration (IC₅₀) values (μM) were calculated from a log dose inhibition curve and are expressed as mean ± SEM of triplicate experiments. ^bPositive control.

compounds (2.5, 25, and 250 μM) significantly inhibited Aβ self-aggregation in a dose-dependent manner. Morin, which was used as a positive control, significantly reduced Aβ self-aggregation (Figure 2).

3.4. In Silico Molecular Docking Simulation of Aβ Monomers.

We conducted an *in silico* study to elucidate the mechanism underlying the anti-aggregation effects of different kinds of Cassia compounds on Aβ42. There are two types of Aβ monomers: α-helical and β-sheet. Therefore, we performed a molecular docking study to evaluate the binding between these Cassia compounds and the two types of Aβ chains (Tables S1–S3; Figures S1–S3). Since some of the anthraquinones, naphthopyrones, and flavonoids have already been reported for inhibition of Aβ aggregation [38–40], we chose unreported compounds for Aβ42 aggregation assay and predicted their binding affinities via docking study. Cassia compounds interacted with the Aβ α-helical monomer (1Z0Q) and the β-sheet monomer and pentamers (2BEG) with a binding affinity of about –4.00 to –5.00 kcal/mol, involving hydrogen bonds and hydrophobic and electrostatic interactions (Tables S1–S3; Figures S1–S3). Morin, a reference ligand, and 1Z0Q Aβ formed a complex with a binding affinity of –4.66 kcal/mol through three hydrogen bond interactions (Gln15, Glu11, Glu22) and a π-anion interaction with Glu22 (Table S1 and Figure S1). Morin interacted with the 2BEG Aβ pentamer via three hydrogen bonds (Leu17^{C,E}, Phe19^D) and six hydrophobic bonds with residues Leu17^E (π-alkyl), Leu17^D (π-σ, π-alkyl), Phe19^C (π-π T-shaped), Val40^C (π-σ, π-alkyl),

and Val40^{D,E} (π-alkyl) with a binding affinity of –7.42 kcal/mol (Table S3 and Figure S3). Anthraquinones and naphthopyrone glycosides from Cassia seeds bound to the α-helical and β-sheet monomers in a similar manner as the reference compound morin (Tables S1 and S2; Figures S1 and S2), demonstrating an anti-aggregation effect on Aβ42, which was verified with the ThT assay. We also conducted a molecular docking study on Cassia compounds and the β-sheet pentamer (2BEG) to determine the interaction of the test compounds with Aβ aggregates. We found that these compounds conjugated with Aβ42 β-sheet pentamers with a binding affinity of –6.39 to –9.02 kcal/mol via a hydrogen bond interaction, hydrophobic interactions, and a π-lone pair interaction (Table S3 and Figure S3).

3.5. Peroxynitrite (ONOO[–]) Scavenging Activity. In the case of the ONOO[–] radical scavenging assay, compounds **8** and **33** showed strong scavenging activity with EC₅₀ values of 0.83 ± 0.03 and 2.05 ± 0.04 μM, respectively (Table 2). Compounds **5** and **14** also showed potent activity with EC₅₀ values of 4.33 ± 0.12 and 9.34 ± 0.86 μM, respectively. Interestingly, the ONOO[–] radical scavenging activity of compound **8** was two-fold higher than that of the positive control, L-penicillamine (IC₅₀ = 1.61 ± 0.09 μM). Compound **33** showed a similar scavenging activity as L-penicillamine. Other test compounds (Table 2) showed moderate to mild scavenging effects. Altogether, our study demonstrates that compounds **8**, **33**, **5**, and **14**, in descending order, are promising antioxidants.

TABLE 2: Peroxynitrite scavenging activity of compounds isolated from *C. obtusifolia*.

Compound	Concentration ($\mu\text{g mL}^{-1}$) ^a	Inhibition (%) (mean \pm SD)	IC ₅₀ ^b	
			$\mu\text{g mL}^{-1}$	μM
Obtusifolin (4)	10	17.51 \pm 8.02	38.99 \pm 1.85	137.28 \pm 2.45
	50	75.37 \pm 4.34		
Aurantio-obtusin (5)	0.4	33.64 \pm 3.88	1.43 \pm 0.24	4.33 \pm 0.12
	2	59.05 \pm 3.57		
	10	71.72 \pm 1.74		
	50	87.75 \pm 1.67		
Obtusifolin 2-glucoside (7)	10	9.02 \pm 3.93	50.68 \pm 1.89	113.63 \pm 3.14
	50	49.50 \pm 2.73		
7-Hydroxyemodin (alaternin) (8)	0.08	29.60 \pm 2.80	0.24 \pm 0.02	0.83 \pm 0.03
	0.4	69.31 \pm 2.45		
	2	95.86 \pm 0.84		
Emodin (10)	2	5.21 \pm 7.83	14.83 \pm 3.00	54.92 \pm 1.22
	10	42.79 \pm 5.02		
	50	96.14 \pm 41.20		
2-Hydroxyemodin 1-methyl ether (11)	10	16.87 \pm 7.29	30.78 \pm 3.21	102.60 \pm 7.11
	50	79.16 \pm 2.42		
Questin (12)	10	2.65 \pm 2.00	33.95 \pm 0.50	119.54 \pm 4.81
	50	85.21 \pm 0.54		
Glucourantio-obtusin (14)	2	32.59 \pm 8.05	4.60 \pm 0.12	9.34 \pm 0.86
	10	79.51 \pm 0.26		
Chrysophanol triglucoside (22)	10	4.37 \pm 2.29	59.27 \pm 1.55	80.09 \pm 0.97
	50	41.70 \pm 0.74		
Toralactone 9-O- β -gentiobioside (33)	0.4	36.05 \pm 1.59	1.22 \pm 0.06	2.05 \pm 0.04
	2	63.07 \pm 0.38		
	10	82.92 \pm 0.80		
	50	93.22 \pm 0.50		
Penicillamine	0.4	56.10 \pm 1.54	0.24 \pm 0.04	1.61 \pm 0.09
	2	82.92 \pm 1.51		
	10	97.39 \pm 1.60		

^aFinal compound concentration of the test samples was 0.4–50 $\mu\text{g mL}^{-1}$, which were dissolved in DMSO. ^bThe concentrations producing 50% inhibition (IC₅₀, in $\mu\text{g mL}^{-1}$ and μM) were calculated from the log dose inhibition curve and expressed as mean \pm SEM of duplicate experiments.

4. Discussion

Advanced glycation end products (AGEs) are a heterogeneous class of molecules (hydroimidazolones, 3-deoxyglucosone derivatives, bis(lysyl)imidazolium cross-links, and monolysyl adducts) formed due to non-enzymatic protein glycation by glucose, which have detrimental effects on the normal physiology and functions of a biological system. The formation and accumulation rate of AGEs is elevated in diabetes, renal failure, and aging. Considerable evidence links accelerated AGE formation with hyperglycemia, which leads to the development of long-term vascular complications due to subsequent accumulation in vessel wall proteins [41].

The ability of insulin to regulate plasma glucose homeostasis is impaired by glycation. Once insulin is glycated, insulin receptors do not recognize the hormone, prolonging the half-life, which leads to hyperinsulinemia and insulin resistance [42]. A recent study by Iannuzzi et al. [43] demonstrated that D-ribose-induced insulin glycation affects endothelial cell viability through the mitochondrial pathways of apoptosis (caspase 9 and 3/7), intracellular ROS activation, and the transcription factor NF- κ B activation.

ROS is one signaling molecule for signal transduction of several receptors, including the receptor for advanced glycation end products (RAGE). The interaction of this receptor with AGEs causes oxidative stress by enhancing ROS generation and NF- κ B activation.

AD and T2D share many pathophysiological features, including increased oxidative stress and amyloid aggregation [44], and cerebrovascular disease is one of the major complications of T2D. Of the various forms of insulin, a key hormone regulating glucose homeostasis, monomeric and dimeric forms are less stable than hexamers and tend to aggregate, forming amyloid fibrils [45, 46]. Pathological conditions related to insulin fibril formation can occur in diabetic patients. In addition, insulin is susceptible to non-enzymatic glycation under diabetic conditions. Accumulation of the end products of glycation, AGEs, is the main factor responsible for the development and progression of several diabetic complications, including nephropathy, retinopathy, and neuropathy [47]. In addition, increasing evidence suggests that depending on the glycating agent, glycation differentially affects the amyloid aggregation [48, 49]. Thus, the study of anti-amyloidogenic and anti-AGE agents to develop new potential therapeutic strategies

in amyloid-based neurodegenerative disease is increasing [50, 51].

In light of these considerations, we designed the current study to investigate the anti-glycation effect of Cassia seed-derived secondary metabolites in human insulin and bovine serum albumin, as well as their anti-A β aggregation and antioxidant effects.

Thioflavin T fluorescence enhancement indicates the structural changes and formation of fibrils as a result of glycation. Natural products have been shown to offer protection from structural changes caused by glycation [52]. In the A β aggregation assay, anthraquinones **5**, **8**, **9**, and **12** and naphthopyrone glycosides **25** and **28** showed promising inhibition of A β self-aggregation in a dose-dependent manner. Several low-molecular-weight compounds have recently been discovered that can alter the A β -aggregation process. However, detailed descriptions of their interactions with oligomers and fibrils have so far been lacking. In a previous study, molecular dynamics simulations were used to investigate the effect of two relatively similar planar tricyclic compounds, 9,10-anthraquinone and anthracene, on the early stages of segmental A β -aggregation [39]. Simulations demonstrated that 9,10-anthraquinone hinders the heptapeptide segment H14QKLVFF20, a hydrophobic stretch that promotes A β self-assembly more than anthracene. In particular, 9,10-anthraquinone intercalates into β -sheets because polar interactions between compounds and peptide backbones disrupt interchain hydrogen bonds and promote their failure. Recently, a similar study demonstrated that emodin, daunorubicin, and Adriamycin inhibit the aggregation of tau protein into paired helical filaments [53]. Likewise, carmine, an anthraquinone core attached to a glucose moiety, effectively reduced aggregation of the model 41 GCWMLY 46 peptide fragment (GDC6 peptide) in a dose-dependent manner [54]. Natural quinones remain an important source of novel anti-amyloid inhibitors. Emodin and purpurin, two natural anthraquinones, have been reported to inhibit tau protein aggregation and have neuroprotective effects [53, 55]. The phenolic hydroxyl groups on the anthraquinone backbone can bind to hydrophobic residues to prevent oligomerization caused by hydrophobic interactions. This feature may explain why anthraquinone derivatives inhibit aggregation more effectively than anthraquinones [56]. The prime interactions leading to the activity of anthraquinone molecules are hydrogen bonds, aromatic contacts, and charge-based contacts [39].

To gain further insights into the molecular interaction of Cassia compounds with A β aggregates, *in silico* studies were undertaken. Molecular docking studies of Cassia compounds with the fibrillar form of α -helical A β 42 monomer (PDB ID: 1Z0Q), β -sheet A β 42 monomer, and A β 42 pentamer (PDB ID: 2BEG) were performed to determine the likely mode of interaction. Test compounds bound to α -helical monomers (1Z0Q) and β -sheet monomers (2BEG) with binding affinities of approximately -4.00 to -5.00 kcal/mol, via hydrophobic, electrostatic, and hydrogen bonding interactions (Tables S1–S3; Figures S1–S3).

A natural flavonol, morin, exhibits a neuroprotective effect by inhibiting AB aggregation via fibril destabilization

and the production of off-pathway intermediates [57], inhibition of the formation of IAPP aggregates, and disaggregation of preformed fibrils [58]. In addition, morin binds to monomeric, oligomeric, and fibrillary insulin, thereby preventing conformational changes and avoiding the release of oligomeric species [59]. Therefore, morin is used as a reference natural drug. In the docking study, reference ligand morin and 1Z0Q formed a complex with a binding affinity of -4.66 kcal/mol through three hydrogen-bonding interactions (Gln15, Glu11, Glu22) and a hydrophobic interaction with Glu22 (Table S1 and Figure S1). Morin has three hydrogen bonds (Leu17^{C, E}, Phe19^D) and six hydrophobic residues, Leu17^E (π -alkyl), Leu17^D (π - σ , π -alkyl), Phe19^C (π - π , T form), Val40^C (π - σ , π -alkyl), and Val40^{D, E} (π -alkyl), for 2BEG pentamer through binding with a binding affinity of 7.42 kcal/mol (Table S3 and Figure S3). In particular, Cassia anthraquinone and naphthopyrone glycosides bound to α -helix or β -sheet monomers like that of the reference compound morin (Tables S1 and S2; Figures S1 and S2) which also corroborate a recent report [60]. Likewise, with binding affinities of -6.39 to -9.02 kcal/mol via hydrogen bonding, hydrophobic, and π lone pair interactions (Table S3 and Figure S3), these compounds bound to the A β 42 β -sheet pentamers.

The *in vitro* anti-glycation assay results show that Cassia compounds inhibit the protein glycation reaction in human insulin. In particular, compounds **5**, **15**, **16**, **20**, **24**, and **27** inhibited the glycation of human insulin with IC₅₀ values less than 100 μ M. Interestingly, compounds **8**, **18**, **24**, **26**, and **33–36** inhibited the glycation of bovine serum albumin with IC₅₀ values less than 50 μ M, whereas **11** and **38** showed IC₅₀ values between 50 and 100 μ M. A small change in the functional moiety in the structure greatly affects the bioactivity. Our graphical abstract relates differences in the structures of a series of molecules to differences in their biological activity. However, a detailed structure-activity relationship should be conducted to explore further.

An exposure time of the glycating agent affects the structural changes in the protein. In a study by Khan et al. [61], thymoquinone showed a much more protective effect on the glycation of superoxide dismutase by methylglyoxal at short incubation time (1 h) compared to longer incubation (10 days). In the same study, the band in SDS-PAGE appeared lighter with an increase in the duration of exposure to the sugar, presumably indicating cross-linking and/or degradation into small peptides. In our study, the incubation time is 7 days. So, based on the previous report [61], our test compounds could have a promising anti-glycation effect in a short incubation time.

Proteins exposed to glucose are cleaved and undergo conformational changes, forming fluorescent adducts called “sugar fluorophores.” Glycation is a functional and structural change in macromolecules resulting from exposure to high glycemic levels of glucose, caused by the covalent binding of glucose to amino groups of proteins, resulting in surface charges, hydrogen-bonding capacity, cellular recognition, and/or altered formation of cross-linkable complex products [62–64]. Protein fragmentation by

monosaccharides has been observed previously but was not considered equivalent to free radical damage [65]. A study by Hunt et al. demonstrates that monosaccharide incorporation into proteins can be independent of the fragmentation and conformational changes that occur when proteins are exposed to glucose [66]. They also showed that the level of sugar fluorophore formation does not match the extent of protein conformational changes. Thus, the conformational change induced by the exposure of the protein to glucose can be dissociated from the incorporation of the monosaccharide into the protein itself and confirms that free radical and peroxide production must be considered in any situation where biological structures are exposed to high levels of monosaccharides.

Due to the antioxidant properties, several natural compounds are used to treat depression, an example being polyphenolic compounds that modulate neurotransmitter systems and simultaneously induce anti-inflammatory, anti-apoptotic, antigenic toxicity, antimutagenic, and antioxidant effects that can protect against cellular damage [67]. In this study, the ONOO-radical scavenging assay showed that compounds **5**, **8**, **14**, and **33** are strong antioxidants (Table 2). As oxidative stress is directly involved in the pathogenesis of depressive disorders [68], the promising antioxidant effect of Cassia compounds via ONOO-radical scavenging could be neuroprotective, which warrants *in vivo* studies.

5. Conclusion

In summary, a total of thirty-eight metabolites from Cassia seeds comprising anthraquinones, naphthopyrones, naphthalene, and naphthalenic lactones and their glycosides were tested for anti-glycation, anti- β -amyloid aggregation, and antioxidant effects. Results showed that compounds **9–12**, **14–18**, **24**, **25**, **27**, **33–36**, and **38** exhibit good inhibition of AGE formation. Likewise, compounds **5**, **15**, **16**, **20**, **24**, and **27** showed good inhibition of D-ribose-mediated glycation of human insulin. In addition, compounds **8** and **12** showed promising inhibition of A β aggregation in the thioflavin-T assay, and compounds **5**, **8**, **14**, and **33** scavenged the peroxy nitrite anion (ONOO⁻) at lower concentrations. Molecular docking simulations confirmed that these active compounds have great potential to interact with the A β 42 peptide and interfere with its self-assembly and conformational transition, and the inhibition of A β aggregation prevents full peptide A β 42 aggregation. Overall, the present study highlights Cassia compounds as potential neuroprotective agents against comorbid AD and diabetes.

Still, *in silico* molecular dynamics studies predicting the stability of ligand-receptor complexes are lacking. Penetration into the central nervous system and stability of the ligand-receptor complex remain to be studied *in vivo*. The effects of the tested compounds at the cellular or organismal level remain unknown because *in silico* modeling cannot account for interactions between compounds and other unrelated targets. In the future, a more detailed understanding of amyloid β -aggregation, especially using *in vivo* models, will be critical to ensure the activity of these Cassia compounds in comorbid AD and diabetes.

Data Availability

The data supporting the findings of this study are available within the paper and its supplementary files.

Conflicts of Interest

The authors declare that they have no conflicts of interest.

Authors' Contributions

PP and JSC were responsible for conceptualization. PP, SS, and SHS were responsible for methodology. RP, HJL, and SHS were responsible for software. JSC and HAJ were responsible for investigation, validation, and review and editing. PP was responsible for original draft preparation. JSC was responsible for visualization and supervision. All authors have read and approved the final manuscript.

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Supplementary Materials

Molecular docking (Figures S1–S3 and Tables S1–S3) of compounds and reference drugs with amyloid-beta peptide. (*Supplementary Materials*)

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