

Research Article

Room Temperature Optical Constants and Band Gap Evolution of Phase Pure M_1 -VO₂ Thin Films Deposited at Different Oxygen Partial Pressures by Reactive Magnetron Sputtering

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Spectroscopic ellipsometry study was employed for phase pure VO₂(M₁) thin films grown at different oxygen partial pressures by reactive magnetron sputtering. The optical constants of the VO₂(M₁) thin films have been determined in a photon energy range between 0.73 and 5.05 eV. The near-infrared extinction coefficient and optical conductivity of VO₂(M₁) thin films rapidly increase with decreasing O₂-Ar ratios. Moreover, two electronic transitions can be uniquely assigned. The energy gaps correlated with absorption edge (E_1) at varied O₂-Ar ratios are almost the same (~2.0 eV); consequently, the absorption edge is not significantly changed. However, the optical band gap corresponding to semiconductor-to-metal phase transition (E_2) decreases from 0.53 to 0.18 eV with decreasing O₂-Ar ratios.

1. Introduction

Vanadium dioxide (VO₂), one of the most interesting transition metal oxides, exhibits a reversible first-order semiconductor-to-metal phase transition (SMT) at a critical temperature $T_c = 68^\circ\text{C}$ (for bulk single crystal VO₂) [1]. VO₂ has a tetragonal rutile structure with the P4₂/mnm space group (R phase) above the phase transition temperature, where the partially filled $d_{//}$ band localized at the Fermi level and the rutile phase is metallic [2]. Below the phase transition temperature, it transforms to a monoclinic structure with the P2₁/c space group (M₁ phase), in which the partially filled $d_{//}$ band splits into an unoccupied part being pushed past the π^* band and a filled part with the $d_{//}$ band dropping below the Fermi level, thus opening up a bandgap of ~0.6 eV between π^* and the filled part of $d_{//}$ band [2]. Dramatic changes in the electrical and optical properties across the SMT make VO₂ thin films suitable for many applications, such as switching devices, sensors, and smart windows [3–6].

It has been noted that oxygen partial pressure has effects on the structural and resistivity transition behaviors of VO₂ [7]. Although the optimized oxygen partial pressure to fabricate VO₂ films on glass and the optical properties of those samples were investigated [8], the optical constants, especially extinction coefficient k , which is crucial in understanding band structures, are not involved. Moreover, two energy gaps E_1 and E_2 are not distinguished as well.

Low visible transparency and unfavorable yellowish colour, which are correlated with absorption edges, limit the application of VO₂ smart windows. For most practical applications the phase transition temperature T_c needs to be in the vicinity of room temperature (~25°C) and the T_c may be assumed to be correlated with the optical band gap E_2 . Consequently, the distinguishment of E_1 and E_2 plays an important role in improving the performance of VO₂.

In this research, we thoroughly investigated the effects of oxygen partial pressures on the optical constants and

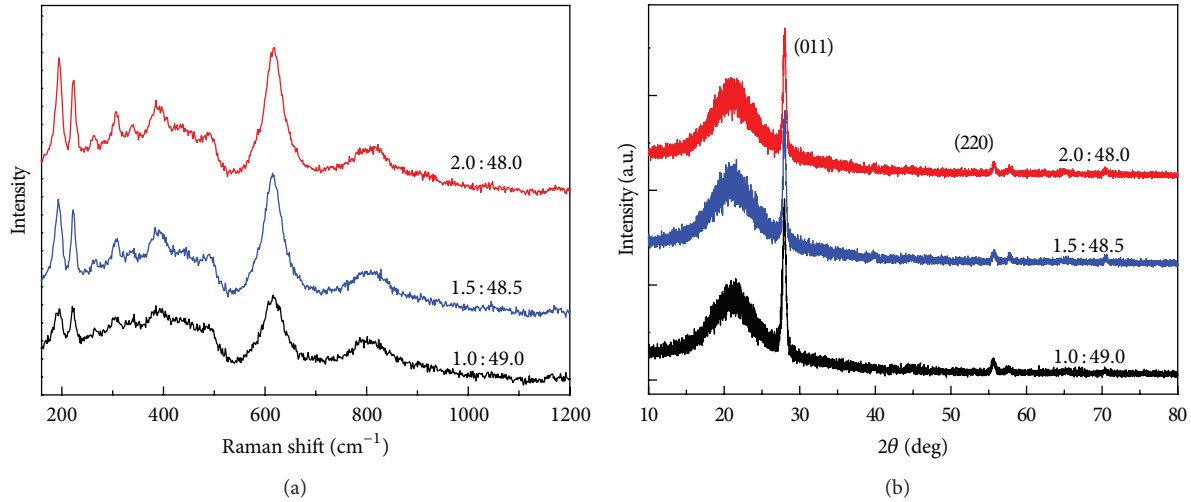


FIGURE 1: Raman (a) and XRD (b) spectra for VO_2 films deposited at different O_2 -Ar ratios.

the electronic transition behaviors of phase pure $\text{VO}_2(\text{M}_1)$ thin films deposited on quartz glass by reactive magnetron sputtering. Moreover, two electronic transitions related to absorption edge (E_1) and SMT (E_2) were distinguished.

2. Experimental Section

VO_2 thin films with a thickness of ~ 70 nm were deposited using a reactive rf magnetron sputtering system with a water-cooled vanadium metal target (50 mm in diameter, 99.9% purity). Quartz glasses ($20 \times 20 \times 1$ mm) were used as substrates and they were ultrasonically cleaned in acetone and subsequently in ethanol for 15 min, respectively, and then dried with pure nitrogen flow.

After being pumped down to a base pressure of 5×10^{-4} Pa, the deposition chamber was filled with high purity (99.999%) Ar and O_2 mixture gas. The O_2 -Ar ratio was fixed as 1.0:49.0, 1.5:48.5, and 2.0:48.0, respectively (the unit is sccm). The total gas pressure was maintained at ~ 1.0 Pa. An rf power of 200 W was applied to the V target. During deposition, the substrate temperature was kept at 450°C for the better crystallinity of VO_2 thin films. To improve the film homogeneity, the substrates were rotated along the vertical axis at a speed of 10 rpm.

The structure of the films was characterized by Raman spectrometer (Renishaw inVia Raman microscope) using a 514.5 nm laser. The optical transmittance was measured at a photon energy range of 0.73–5.05 eV at 26°C and 95°C by a spectrophotometer (Hitachi Corp., Model UV-4100). Temperature was measured using a PT100 temperature sensor in contact with the films and was controlled via a temperature controlling unit. Heating was controlled through a temperature-controlling unit. Hysteresis loops were measured by collecting the transmittance of films at a fixed photon energy (0.83 eV) at a temperature interval of $\sim 2.0^\circ\text{C}$. Spectroscopic ellipsometry (SE J. A. Woollam M-2000) measurements were carried out between photon

energies of 0.73 and 5.05 eV at 75° angle of incidence and the results were modeled using a commercial software.

3. Results and Discussion

3.1. Structural Characterization. Raman measurements were conducted to examine the effect of O_2 -Ar ratio on the microstructure of VO_2 (Figure 1(a)). The Raman spectrum at room temperature shows characteristic vibration modes for the M_1 semiconducting phase of VO_2 . Comparing to previous works, Raman peaks are identified as 194(Ag), 223(Ag), 262(Bg), 307(Bg), 391(Ag), 492(Ag), and 618(Ag) cm^{-1} [9–11]. No Raman shifts for other kinds of vanadium oxides and any types of other polymorphs of VO_2 (M_2/T) [12, 13] were identified within the measurement accuracy ($\pm 0.2 \text{ cm}^{-1}$). The XRD spectra are shown in Figure 1(b). All peaks can be indexed to $\text{VO}_2(\text{M})$ and (011) was the prominent plane for VO_2 thin film. No reflections due to other VO_x phases such as V_4O_7 , V_6O_{13} and V_3O_7 were observed [14, 15].

3.2. Optical Properties of the VO_2 Films. Figures 2(a) and 2(b) show the transmittance and absorptivity spectra of the prepared VO_2 thin films at different O_2 -Ar ratios. The SMT transition is clearly observed with a dramatic change in the infrared transmittance with varied temperature ranges. The absorption edge, luminous (lum) transmittance (T_{lum} , 1.64–3.27 eV), and near-infrared (NIR) transmittance (0.73–1.64 eV) at the high temperature of 95°C were almost the same for all of the samples studied here. However, low temperature (below 26°C) phase $\text{VO}_2(\text{M}_1)$ showed a gradually decreased NIR transmittance but increased absorptivity with decreasing O_2 -Ar ratios.

Thermooptical hysteresis curves were deduced by measuring the transmittance at 0.83 eV with varying temperatures, which are shown in Figure 2(c). For comparison, the vertical axis of this figure has been normalized as X . From the X -temperature (T) data, a plot of dX/dT was obtained, yielding one peak with a well-defined maximum.

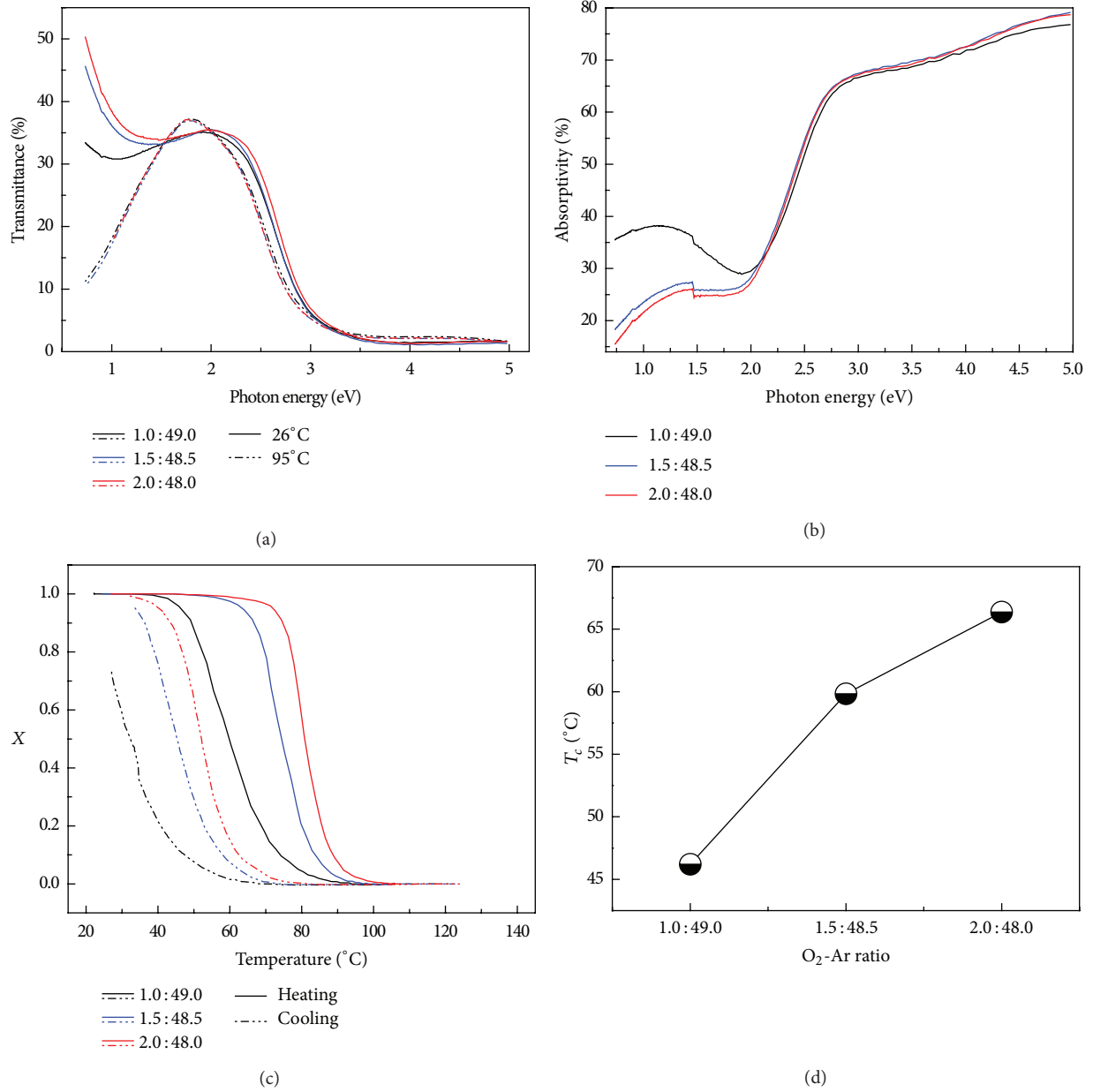


FIGURE 2: (a) Transmittance spectra of VO₂ films deposited at different O₂-Ar ratios. Solid and dashed lines refer to low temperature (26°C) and high temperature (95°C), respectively. (b) Low temperature (M₁ phase) absorptivity spectra. Thermal hysteresis loops of optical transmittance at photon energy of 0.83 eV and phase transition temperature T_c are shown in (c) and (d).

Each of the dX/dT curves has been analyzed with a Gaussian function using the single peak fitting module of Originpro 8.0 software. The temperature corresponding to the maximum of dX/dT was defined as the phase transition temperature during a heating/cooling cycle; T_1 and T_2 represent the SMT temperature of heating and cooling branches, respectively. The SMT temperature was defined as $T_c = (T_1 + T_2)/2$. As shown in Figure 2(d), the SMT temperatures were 46.2°C, 59.8°C, and 66.4°C for VO₂ films deposited at the O₂-Ar ratio of 1.0:49.0, 1.5:48.5, and 2.0:48.0, respectively. The phase transition temperature T_c consistently decreased as the O₂-Ar ratio decreased. It was pointed out that vanadium interstitials

and/or oxygen vacancies could reduce the SMT temperature of VO₂ [16], which is also responsible for the decreased SMT temperatures at low O₂-Ar ratios of this work. Low O₂-Ar ratios can introduce both extra electrons and internal strains in nonstoichiometric VO₂ (related data was revealed in a paper submitted to Thin Solid Films, Manuscript Number: TSF-D-13-00641).

3.3. Optical Constant. The electronic transitions, optical constants, and optical band gap (OBG) in the vicinity of the phase transition temperature have been investigated by Li et al. [17]. For a standard bulk VO₂ sample, when

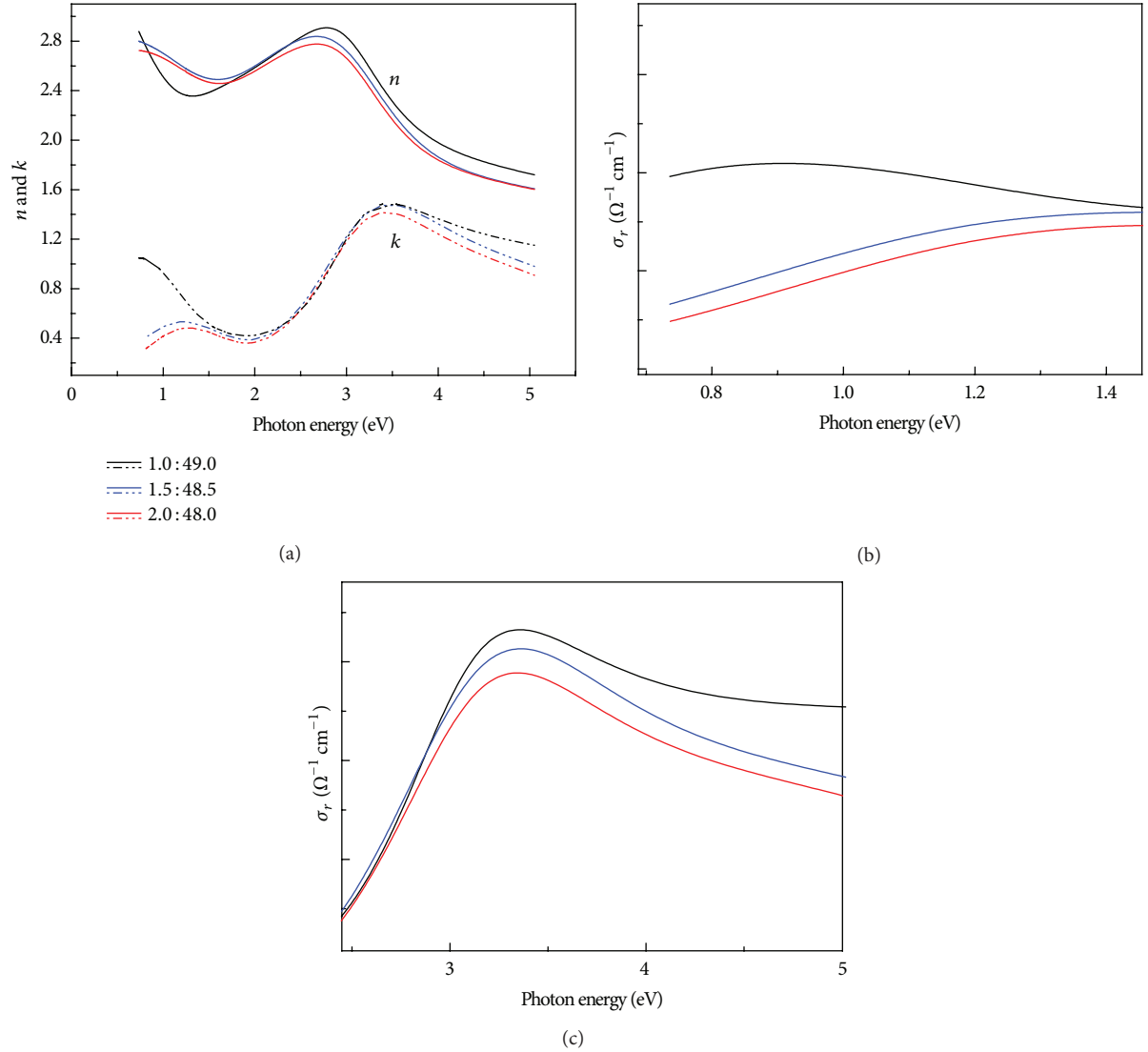


FIGURE 3: Refractive index n , extinction coefficient k , and optical conductivity σ_r of low temperature phase VO₂ deposited at different O₂-Ar ratios.

temperature was increased to 67°C, contributions from the Drude response become more prominent. The extinction coefficient k and the optical conductivity σ_r values at the NIR region rapidly increase with increasing temperature. In this research, electron concentration increases with decreasing O₂-Ar ratios and the Drude response is also responsible for the increased k and σ_r value at the NIR region (Figure 3). However, refractive index n was not significantly changed.

3.4. Band Gap. The indirect OBG can be estimated using the power law:

$$(\alpha E)^{1/2} \propto (E - E_g), \quad (1)$$

where $\alpha = 4\pi k/\lambda$ is the absorption coefficient and E_g is the OBG energy. The E_g value is extrapolated by the linear portion of the plot to $(\alpha E)^{1/2} = 0$.

Two electronic transitions can be uniquely assigned, as shown in Figures 4(a) and 4(b). The gaps correlated with absorption edge (E_1) at varied O₂-Ar ratios are almost the same (~2.0 eV); consequently, the absorption edge was not significantly changed, as shown in Figure 2(a). However, the optical band gap corresponding to SMT (E_2) decreases from 0.53 to 0.18 eV with decreasing O₂-Ar ratios, as shown in Figures 4(b) and 4(d). The OBG E_2 of semiconducting VO₂ can be assigned to the indirect transition from the top of filled d_{||} bands to the bottom of empty π^* band, as shown using a red arrow in Figure 4(c). Note that E_2 at an O₂-Ar ratio of 2.0:48.0 is similar to that from a previous report by theoretical calculation (0.6 eV) [18] and experimental results by photoemission spectroscopy (0.6 eV) [2]. Besides, the 0.41 eV band gap of the VO₂ film prepared at an O₂-Ar ratio of 1.5:48.5 agrees to a calculated intermediate structure at 339.8 K (0.36 eV) [18]. E_2 at 1.0:49.0 O₂-Ar ratio further decreased to 0.18 eV.

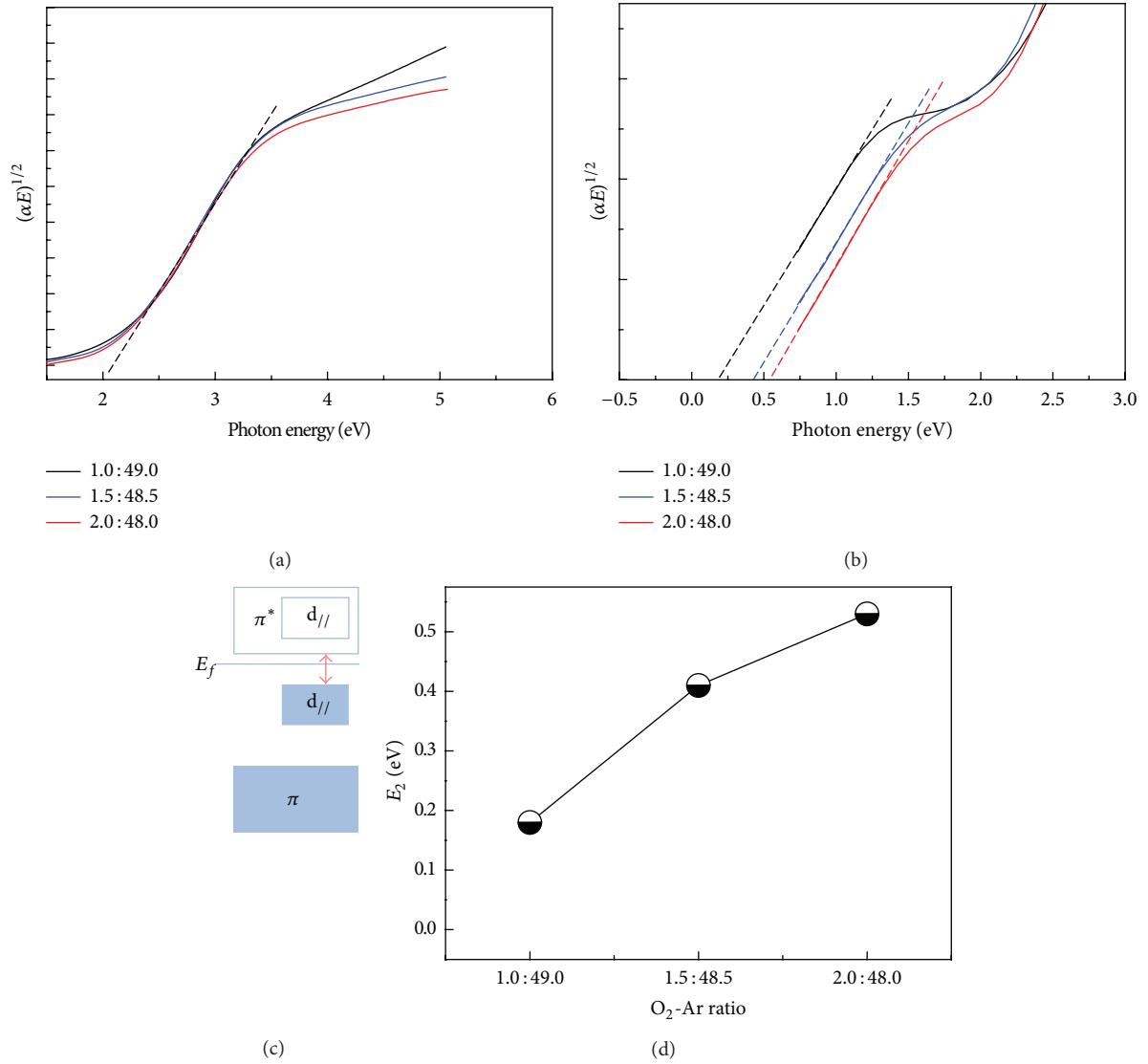


FIGURE 4: (a) and (b) $(\alpha E)^{1/2}$ versus E which is determined from extinction coefficient k and the extrapolating dashed lines are drawn to obtain the value of the band gap E_1 and E_2 , respectively. (b) is the enlarged low photon energy portion of $(\alpha E)^{1/2}$ versus E . (c) Schematic diagram of electronic band structures in the VO₂ film. (d) Variation of optical band gap E_2 of VO₂ thin films deposited at different O₂-Ar ratios.

When decreasing O₂-Ar ratios, the filled $d_{//}$ bands and the empty π^* band are shifted to the higher and lower energy, respectively. Both the $d_{//}$ and π^* bands gradually moved closely, resulting in a redshift of E_2 . The decreasing E_2 results in a decrement in the SMT energies; therefore the SMT temperature decreased with decreasing O₂-Ar ratios. Moreover, with decreasing the O₂-Ar ratio, the NIR transmittance of the film evidently decreases. This behavior is because the bandgap E_2 of the film, which is narrowing at low O₂-Ar ratios, is different at the distinct O₂-Ar ratio regions. It can enhance the electronic transitions and results in more interband absorptions at lower O₂-Ar ratios, as shown in Figure 2(b).

4. Conclusions

To summarize, diverse phase transformation properties are reported for phase pure VO₂(M₁) thin films grown at different oxygen partial pressures by reactive magnetron sputtering. The transmittance and absorptivity spectra below phase transition temperatures are closely related to O₂-Ar ratios. The phase transition temperature T_c decreased from 66.4°C to 46.2°C as the O₂-Ar ratio decreased from 2.0:48.0 to 1.0:49.0. The optical constants of the VO₂(M₁) thin films have been determined between 0.73 and 5.05 eV. The near-infrared extinction coefficient k and optical conductivity σ_r increase with decreasing O₂-Ar ratios. Moreover,

two electronic transitions were identified. The energy gaps correlated with absorption edge at varied O₂-Ar ratios are almost the same, while the optical band gap corresponding to semiconductor-to-metal phase transition decreases from 0.53 to 0.18 eV with decreasing O₂-Ar ratios.

Conflict of Interests

The authors declare that there is no conflict of interests regarding the publication of the article.

Acknowledgments

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