

Research Article

Biodistribution and Acute Toxicity of Intravenous Multifunctional ^{125}I -Radiolabeled Fe_3O_4 -Ag Heterodimer Nanoparticles in Mice

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Fe_3O_4 -Ag ^{125}I heterostructured radionuclide nanoparticles (NPs) have been developed as a novel type of dual-modality imaging agents for single-photon emission computerized tomography (SPECT) and magnetic resonance imaging (MRI). However, the biodistribution and toxicity of Fe_3O_4 -Ag ^{125}I NPs remain largely unknown. Therefore, we investigated the biodistribution and biological action of Fe_3O_4 -Ag ^{125}I NPs in mice by acute toxicity experiments (exposures over 7 days). The bioaccumulation of Fe_3O_4 -Ag ^{125}I NPs was studied via *in vivo* experiments. The serum biochemistry and hematology were analyzed to reveal potential functional changes. The histopathological changes were observed by using an electron microscope. Biodistribution analysis revealed that Fe_3O_4 -Ag ^{125}I NPs were mainly accumulated in the liver and spleen. The activities of liver enzymes (ALT and AST) were increased in Fe_3O_4 -Ag ^{125}I NP-challenged groups compared with the control groups. Collectively, liver and spleen were the major target organs for accumulation of Fe_3O_4 -Ag ^{125}I NPs. Damage of liver tissue was observed in the Fe_3O_4 -Ag ^{125}I NP-challenged groups compared with the control groups. Further studies on surface coating of Fe_3O_4 -Ag with targeted materials are highly necessary for safe medical applications of Fe_3O_4 -AgNPs as dual-modality imaging agents.

1. Introduction

In recent years, a great deal of attention has been paid to silver nanoparticles (AgNPs) since they are used as popular antibacterial and antifungal agents in the light of an enormously increasing bacterial resistance against repeatedly and excessively used classical antibiotics. AgNPs can effectively eliminate bacteria at a relatively low concentration [1–3]. Besides antimicrobial ability, AgNPs are effective in the field of photothermal cancer therapy and/or surface-enhanced Raman spectroscopy [4].

Magnetic iron oxide (Fe_3O_4) NPs have been widely used in many important fields due to their unique characteristics, such as biochemical properties, superparamagnetism and low price [5–8]. Fe_3O_4 -Ag heterodimer NPs

possess magnetic functionality and antimicrobial ability at the same time [2].

Our group has successfully developed Fe_3O_4 -Ag ^{125}I heterostructured radionuclide NPs as novel dual-modality imaging agents for magnetic resonance imaging (MRI) and single-photon emission computerized tomography (SPECT) [9]. The Fe_3O_4 -Ag ^{125}I heterostructured radionuclide NPs demonstrate high radiolabeling efficiency and clearly reduced T_2 -MRI signal intensity.

We aimed to apply this material to medical imaging. However, no study has investigated the distribution and toxicity of Fe_3O_4 -AgNPs in animals. Moreover, previous studies show inconsistent results, indicating that the distribution and toxicity of Fe_3O_4 or Ag NPs are highly dependent on the various factors, such as shape, size, coating agent of the

NPs, duration after drug administration, and animal gender [10–15]. Therefore, we investigated the biodistribution and toxicity of $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ NPs in mice after intravenous injection. The bioaccumulation of $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ NPs was studied via *in vivo* experiments. The serum biochemistry and hematology were analyzed to reveal potential functional changes. The histopathological changes were observed by using an electron microscope.

2. Materials and Methods

2.1. Ethics Statement. Male Kunming mice (6 weeks of age) were purchased from the Center for Experimental Animal of Soochow University. Animal experiments were preapproved by the institutional review board and the Experimental Animal Center of the First Affiliated Hospital of Soochow University. All SPECT scans were performed under general anesthesia, and all efforts were made to minimize animal suffering.

2.2. Materials. All the reagents for the synthesis of $\text{Fe}_3\text{O}_4\text{-Ag}$ NPs were purchased from Sigma-Aldrich. ^{125}I was obtained from Chengdu Gaotong Isotope Corporation (Chengdu, China). All other chemicals were prepared with analytical-grade reagents dissolved in deionized water prepared by LabWater (Shanghai Hejie Technology Co. Ltd.).

2.3. Synthesis of $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ NPs. $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ NPs were synthesized as previously reported [9]. Briefly, the Fe_3O_4 NPs were synthesized by thermal decomposition of ironoleate complex, and then the AgNPs were grown onto the cubic Fe_3O_4 NPs by adding the silver acetate into the reaction system. Subsequently, the $\text{Fe}_3\text{O}_4\text{-Ag}$ NPs were functionalized by hydrophilic mPEG-LA polymers and phase transferred from hexane to water. Finally, $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ NPs were produced by reacting the Ag component of the heterostructured NPs with ^{125}I .

The labeling efficiency and radiochemical purity were analyzed using paper chromatography. The fractions containing ^{125}I -labeled $\text{Fe}_3\text{O}_4\text{-Ag}$ were determined using a gamma counter to calculate the radiolabeling yield (%). The solution was filtered through a $0.22\ \mu\text{m}$ pore-size membrane in order to avoid potential bacterial and dust particles for *in vivo* studies.

2.4. Biodistribution of $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ NPs. Kunming mice ($n=5$ per time point) were intravenously injected with $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ NPs ($100\ \mu\text{L}/4.92\text{--}6.99\ \text{MBq}$) via the tail vein once daily and sacrificed by exsanguination under ether anesthesia at 1, 2, 8, 24, and 48 h after injection. Blood samples (approximately $100\ \mu\text{L}$ each) were collected via retroorbital bleeding, and main organs, such as the blood, lung, brain, kidney, liver, pancreas, spleen, stomach, thyroid, intestine, bone, and muscle, were dissected from anesthetized mice and weighed at 1, 2, 8, 24, and 48 h postinjection. The radioactivity of the tissue was measured in a γ -counter (Shanghai Nucleus Research Institute Rihuan Photoelectric Instrument Co. Ltd.). The uptake in organs was calculated as the proportion of injected dose per gram of tissue (%ID/g).

2.5. In Vivo SPECT Imaging. SPECT scans were performed using the IRIX (Philips, Netherlands) equipped with high-resolution low-energy parallel-hole collimator. Briefly, after injection of $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ NPs, mice were anesthetized using isoflurane. The SPECT scans were performed at various time points. Images were acquired with 1×10^5 counts on a 128×128 matrix. The energy peak for the camera was set to 37 keV, and the energy window was set to peak energy $\pm 30\%$, which was 26–48 keV.

2.6. Serum Biochemistry and Hematology. The mice were sacrificed after injection of $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ NPs ($40\ \text{mg/mL}$) for seven consecutive days. The blood was collected from the retroorbital sinus. For hematological analysis, the blood samples were combined with EDTA-3K for anticoagulation. The hematological measurements were performed using an automated hematology analyzer (BC-5800, Mindray Co., Shenzhen, China) following the standard protocols.

For serum biochemistry analysis, the blood samples were centrifuged at 3000 rpm for 15 min within 1 h, and the supernatant was collected. All the biochemical parameters were determined on a clinical automatic chemistry analyzer (Chemray360, Rayto Co., Shenzhen, China) following the standard protocols.

2.7. Transmission Electron Microscopy (TEM). For TEM analysis of the spleen, heart, liver, and kidney, small pieces of tissue samples ($\sim 1\ \text{mm}^3$) were fixed in 2.5% glutaraldehyde solution overnight and washed with phosphate-buffered saline (PBS). Postfixation was performed with 1% osmium tetroxide for 2 h. Then, the samples were washed with PBS and dehydrated with a graded series of alcohols (50%, 70%, 80%, 95%, and 100%), followed by rinsing with acetone. Ultrathin sections from each tumor sample were prepared and examined under JEOL-JEM-2100F TEM operating at 200 kV.

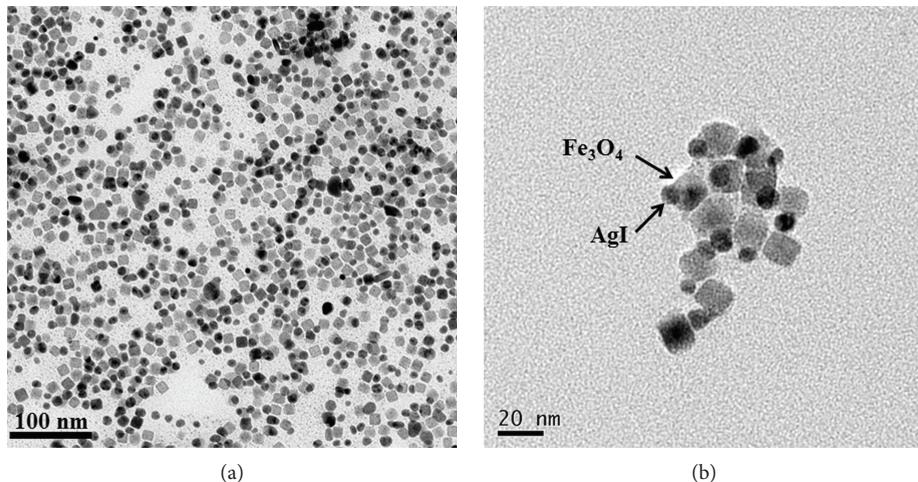
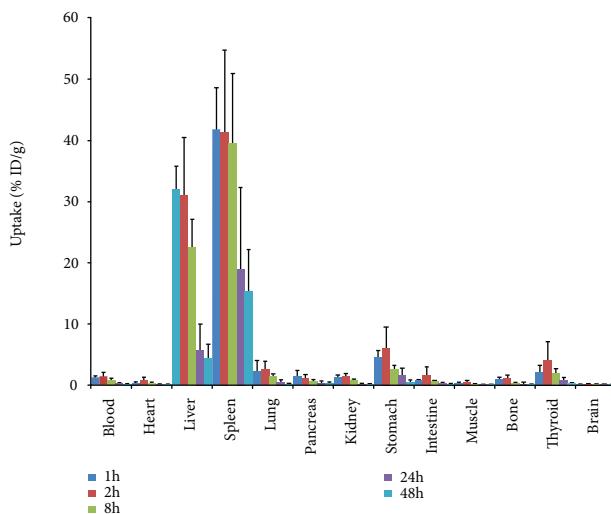
2.8. Statistical Analysis. The results were expressed as the mean \pm standard deviation (SD). Data were analyzed by one-way ANOVA and Student's *t*-test. $p < 0.05$ was considered as statistically significant. All statistical tests were two sided.

3. Results

3.1. Radioiodination of $\text{Fe}_3\text{O}_4\text{-Ag}$ NPs. The radiolabeling efficiency of $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ heterostructured NPs was $95.57\%\pm 2.06\%$, and the radiochemical purity was $91.99\%\pm 0.32\%$ after 24 h.

A TEM image (Figure 1) confirmed that the average size of $\text{Fe}_3\text{O}_4\text{-Ag}$ NPs was $24.53\pm 2.99\ \text{nm}$. The addition of a radionuclide into the $\text{Fe}_3\text{O}_4\text{-Ag}$ NPs did not change the morphology of the samples.

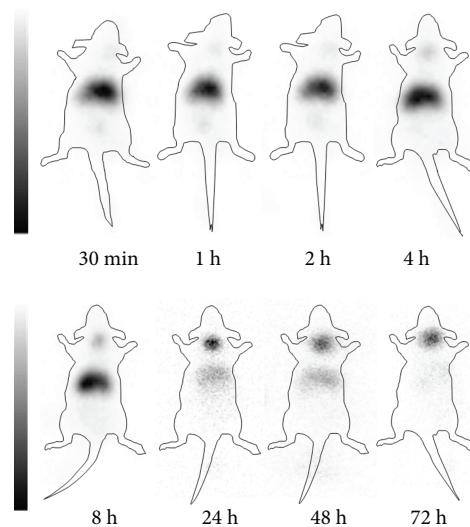
3.2. Biodistribution of $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ NPs in Mice. Figure 2 presents the biodistribution data of $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ NPs in different organs at various time points postinjection. The uptake of $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ was high in the liver ($31.98\pm 3.74\%\text{ID/g}$ at 1 h after injection, $31.00\pm 9.42\%\text{ID/g}$ at 2 h after injection, $22.51\pm 4.57\%\text{ID/g}$ at 8 h after injection, $5.79\pm 4.24\%\text{ID/g}$ at 24 h after injection, and $4.48\pm 2.20\%\text{ID/g}$ at

FIGURE 1: TEM images of $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ heterodimer NPs.FIGURE 2: Biodistribution of $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ in mice over time reported as % ID ($n = 5$).

48 h after injection) and spleen ($41.87 \pm 6.73\%$ ID/g at 1 h after injection, $41.41 \pm 13.32\%$ ID/g at 2 h after injection, $39.49 \pm 11.37\%$ ID/g at 8 h after injection, $19.07 \pm 13.22\%$ ID/g at 24 h after injection, and $15.34 \pm 6.82\%$ ID/g at 48 h after injection). These findings indicated that the injected ^{125}I -labeled conjugates were mainly taken up by the reticuloendothelial system (RES).

A moderate level of radioactivity was accumulated in the thyroid ($2.15 \pm 1.04\%$ ID/g at 1 h after injection, $4.21 \pm 2.90\%$ ID/g at 2 h after injection, $1.94 \pm 0.74\%$ ID/g at 8 h after injection, $0.83 \pm 0.44\%$ ID/g at 24 h after injection, and $0.29 \pm 0.10\%$ ID/g at 48 h after injection) and stomach ($4.52 \pm 1.15\%$ ID/g at 1 h after injection, $6.16 \pm 3.29\%$ ID/g at 2 h after injection, $2.67 \pm 0.51\%$ ID/g at 8 h after injection, $1.58 \pm 1.16\%$ ID/g at 24 h after injection, and $0.56 \pm 0.24\%$ ID/g at 48 h after injection). These accumulations were probably attributed to free ^{125}I released *in vivo*.

A low level of radioactivity was present in the brain ($0.11 \pm 0.04\%$ ID/g at 1 h after injection, $0.15 \pm 0.11\%$ ID/g

FIGURE 3: Representative static whole-body SPECT imaging of $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ localization in mouse at 0.5, 1, 2, 4, 8, 24, 48, and 72 h after injection.

at 2 h after injection, $0.07 \pm 0.02\%$ ID/g at 8 h after injection, $0.04 \pm 0.02\%$ ID/g at 24 h after injection, and $0.02 \pm 0.01\%$ ID/g at 48 h after injection, respectively) and muscle ($0.35 \pm 0.17\%$ ID/g at 1 h after injection, $0.50 \pm 0.26\%$ ID/g at 2 h after injection, $0.20 \pm 0.06\%$ ID/g at 8 h after injection, $0.08 \pm 0.04\%$ ID/g at 24 h after injection, and $0.05 \pm 0.02\%$ ID/g at 48 h after injection).

3.3. SPECT Imaging Studies. Mice administered with $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ NPs were subjected to SPECT imaging. Figure 3 shows representative images of mice obtained at 0.5, 1, 2, 4, 8, 24, 48, and 72 h postinjection.

The activity level in the abdominal region (particularly the spleen and liver) was high in the first five static images, which was generally consistent with the results of *in vivo* biodistribution studies, indicating that the injected $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ NPs were mainly sequestered by the RES.

TABLE 1: Haematological and serum biochemistry parameters of the mice exposed to $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$. Data represent means \pm SD ($n = 5$).

	ALB (g/L)	ALT (U/L)	AST (U/L)	ALP (U/L)	Ur (mmol/L)	Cr ($\mu\text{mol}/\text{L}$)	WBC ($10^9/\text{L}$)	RBC ($10^{12}/\text{L}$)	HB (g/L)	PLT ($10^9/\text{L}$)
Control	27.32 ± 0.98	15.32 ± 10.34	189.33 ± 32.11	71.12 ± 22.96	8.03 ± 1.69	21.52 ± 4.74	3.94 ± 0.74	8.40 ± 0.93	143.00 ± 5.66	759.50 ± 65.76
$\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$	26.14 ± 2.48	30.93 $\pm 14.41^*$	256.35 $\pm 56.28^*$	69.63 ± 15.05	10.27 ± 1.98	35.16 ± 30.21	3.81 ± 1.39	7.21 ± 0.42	120.50 ± 3.54	629.00 ± 131.52

* $p < 0.05$ compared with control group. ALB: albumin; ALT: alanine aminotransferase; AST: aspartate aminotransferase; ALP: alkaline phosphatase; Ur: urea; Cr: creatinine; WBC: white blood cell count; RBC: red blood cell count; HB: hemoglobin; PLT: platelet.

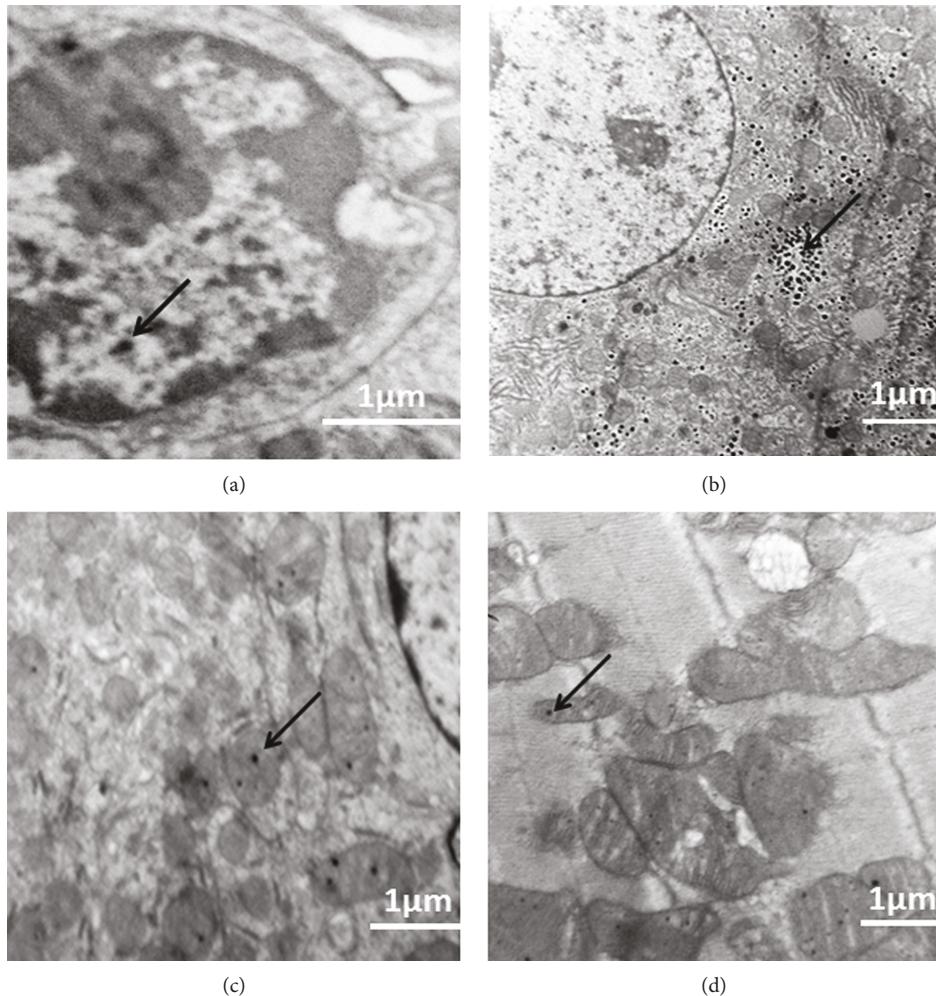


FIGURE 4: TEM images of the spleen, heart, liver, and kidney of $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ NPs-treated mice: (a) spleen, (b) liver, (c) kidney, and (d) heart. Arrows in black color show NPs.

Little radioactivity was observed in the thyroid region during the early imaging procedure. However, there were slight increases in thyroid at the end of the imaging procedure, suggesting that this compound was deiodinated *in vivo* just as the results of biodistribution.

3.4. Toxicity Evaluations. Haematological and serum biochemistry parameters were analyzed after exposure to $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ NPs. Table 1 lists the data.

Most parameters remained within the normal ranges at 7 days after the intravenous injection of $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ NPs.

Significant changes were only observed for alanine aminotransferase (ALT) and aspartate aminotransferase (AST).

3.5. TEM Analysis. TEM analysis was performed on the spleen, heart, liver, and kidney from the $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ NPs-administered mice and negative control mice (Figure 4). The results demonstrated that the $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ NPs aggregated in the spleen. In the liver, $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ NPs were scattered throughout the parenchyma. In line with the result of biodistribution, less $\text{Fe}_3\text{O}_4\text{-Ag}^{125}\text{I}$ NPs were detected in the heart and kidney.

4. Discussion

During the past few decades, there are increasing applications of AgNPs in various fields. However, AgNPs have several shortcomings, including agglomeration, easy oxidation, low penetration into tissue, and cytotoxicity [16, 17]. Iron oxide NPs can add a magnetic functionality and prevent agglomeration to AgNPs. It has been reported that the bactericidal efficiency of Fe_3O_4 -AgNPs is stronger than Fe_2O_3 -Ag heterodimers or plain Ag [18]. Despite the advantages of Fe_3O_4 -AgNPs, the biodistribution and toxicity of Fe_3O_4 -AgNPs remain largely unexplored.

In the present study, we systematically investigated the biodistribution of Fe_3O_4 -AgNPs in mice after intravenous injection by noninvasive nuclear imaging techniques. Our study confirmed that the majority of Fe_3O_4 -Ag¹²⁵I NPs were accumulated in the spleen and liver, and such pattern could be attributed to uptake by the B cells and macrophages in the spleen and the Kupffer cells in the liver, which are part of the mononuclear phagocyte system. These results were consistent with some previous studies on biodistribution of nontargeted AgNPs and Fe_3O_4 NPs [19–22]. Chrastina and Schnitzer have radiolabeled AgNPs with ¹²⁵I to track the *in vivo* tissue uptake of AgNPs after systemic administration by biodistribution analysis and SPECT imaging. Their results have also revealed the uptake of AgNPs in the liver and spleen [23].

Recently, toxicity of Fe_3O_4 NPs or AgNPs has been widely studied. Fe_3O_4 NPs are generally considered as biocompatible, safe, and nontoxic materials. Median lethal dose (LD-50) of the uncoated Fe_3O_4 NPs is 300–600 mg Fe/kg body weight [24]. However, the toxicity of AgNPs based on *in vivo* studies is controversial. Maneewattanapinyo et al. have investigated the acute oral toxicity of AgNPs by *in vivo* experiments and found that the LD-50 of colloidal AgNPs is greater than 5000 mg/kg body weight [25]. Another study has also revealed that no obvious changes in serum chemistry, hematology, and histopathology are found after SD rats are administered with up to 36 mg/kg AgNPs by oral gavage for 13 weeks [14]. However, other studies have demonstrated that short-term administration of AgNPs can significantly increase ALT or/and AST [15, 26, 27]. Tiwari et al. have investigated the toxic effect of various doses of AgNPs on Wistar rats and indicated that AgNPs at lower dose (<10 mg/kg) are safe, while its higher dose (>20 mg/kg) is toxic [28]. Recently, Ghaseminezhad et al. have compared the cytotoxicities of AgNPs and Ag/ Fe_3O_4 nanocomposites to human fibroblasts and found that Ag/ Fe_3O_4 nanocomposites are less cytotoxic than AgNPs [29]. The Ag/ Fe_3O_4 nanocomposites show lower release of Ag ions and less ROS production compared with AgNPs. In the present study, the activities of liver enzymes (ALT and AST) were increased in the Fe_3O_4 -Ag¹²⁵I NP-challenged groups compared with the control groups, indicating that liver tissues were damaged following administration of Fe_3O_4 -Ag¹²⁵I NPs.

Some studies have suggested that the toxicity of AgNPs depends on surface capping. It has been demonstrated that polysaccharide-coated AgNPs induce more severe damages

compared with uncoated AgNPs [30], whereas carbon-coated AgNPs are less cytotoxic towards macrophages [31]. Therefore, in order to develop the Fe_3O_4 -Ag¹²⁵I heterostructured radionuclide NPs as dual-modality imaging agents, NPs need to be coated with special compounds in the future. Additional studies are required in order to reshape the surface of Fe_3O_4 -Ag to modify their characteristics.

Collectively, our present study investigated the biodistribution and acute toxicity of ¹²⁵I-radiolabeled Fe_3O_4 -Ag heterodimer NPs in mice. We found that the liver and spleen were the major target organs for the accumulation of Fe_3O_4 -Ag¹²⁵I NPs. Damage of liver tissue was observed in the Fe_3O_4 -Ag¹²⁵I NP-challenged groups compared with the control groups. Further studies on surface coating of Fe_3O_4 -Ag with targeted materials are highly necessary for safe medical applications of Fe_3O_4 -AgNPs as dual-modality imaging agents.

Data Availability

The data used to support the findings of this study are available from the corresponding author upon request.

Conflicts of Interest

The authors declare no conflict of interest.

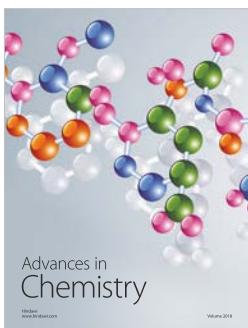
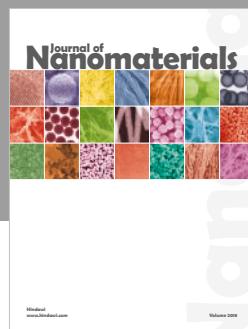
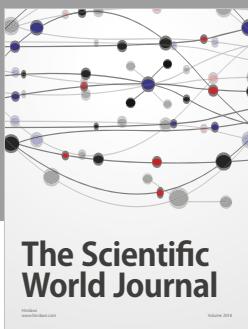
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