

## Research Article

# Effect of the Freeze-Thaw on the Suspension Stability and Thermal Conductivity of EG/Water-Based $\text{Al}_2\text{O}_3$ Nanofluids

Tae Jong Choi <sup>1</sup>, Seok Pil Jang <sup>1</sup>, Dae Soo Jung,<sup>2</sup> Hyung Mi Lim,<sup>2</sup> Young Man Byeon,<sup>3</sup> and Im Joo Choi<sup>2</sup>

<sup>1</sup>School of Aerospace and Mechanical Engineering, Korea Aerospace University, Goyang 10540, Republic of Korea

<sup>2</sup>Korea Institute of Ceramic Engineering and Technology, Jinju 52851, Republic of Korea

<sup>3</sup>Firstec Corporation, Changwon 51528, Republic of Korea

Correspondence should be addressed to Seok Pil Jang; [spjang@kau.ac.kr](mailto:spjang@kau.ac.kr)

Received 20 September 2018; Revised 19 November 2018; Accepted 11 December 2018; Published 12 February 2019

Academic Editor: Simo-Pekka Hannula

Copyright © 2019 Tae Jong Choi et al. This is an open access article distributed under the Creative Commons Attribution License, which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

This paper reports the effect of the freeze-thaw on the suspension stability, particle size distribution, and thermal conductivity of EG/water-based nanofluids containing  $\text{Al}_2\text{O}_3$  nanoparticles that can be used as improved working fluid for cooling systems. The EG/water-based  $\text{Al}_2\text{O}_3$  nanofluids were prepared using a two-step method with a nanodisperser and decanting processes. To investigate the effect of freeze-thaw on the suspension stability and thermal conductivity of nanofluids, the prepared nanofluids were frozen at  $-32^\circ\text{C}$  for 24 hours using a refrigerating chamber, and then they were completely thawed at room temperature for 24 hours. The suspension stability of the thawed nanofluids was quantitatively analysed for over a day using a Turbiscan. In addition, the particle size distributions and deformation of nanoparticles dispersed in the nanofluids were measured using a particle size analyzer (PSA) and TEM. Also, the thermal conductivity of the nanofluids was measured using a transient hot wire (THW) method in temperature from  $-10$  to  $70^\circ\text{C}$ . Based on the results, we show that the suspension stability, thermal conductivity, and particle size of EG/water-based  $\text{Al}_2\text{O}_3$  nanofluids were not affected by low temperature.

## 1. Introduction

The efficiency improvement of thermal management system in many industrial fields is one of the most challenging issues [1–6]. Especially, the efficiency of cooling system inherently has limitation because the working fluids such as EG and PG have relatively low thermal properties [7]. To overcome this problem, many researchers have put a lot of effort into enhancing the thermal properties and the convective heat transfer characteristics of working fluids. One of the solutions is nanofluids which is to disperse the nanoparticles into the basefluid with well suspension stability [8–30]. Since Choi [31], it has been reported that the thermal conductivity of working fluids can be enhanced by dispersing the nanoparticles and nanofluids' thermal conductivity can be controlled with geometry of nanoparticles. Recently, Kim et al. [26] experimentally showed the thermal conductivity of nanofluids

strongly depends on the suspension stability. They prepared three kinds of  $\text{Al}_2\text{O}_3$  nanofluids with different particle shapes to make a difference of nanofluids' suspension stability, and then suspension stability and thermal conductivity of manufactured nanofluids were measured. Based on the results, they showed that even though nanofluids have the same volume fraction, the thermal conductivity of nanofluids is more increased in accordance with suspension stability. Furthermore, the convective heat transfer characteristics of nanofluids have been one of the popular topics. Many researchers [11, 19, 20, 23–25, 28] experimentally presented that the convective heat transfer coefficient is enhanced by nanoparticles. Their results showed the engineering parameters which strongly affect the convective heat transfer characteristics of nanofluids such as volume fraction of nanoparticles, nanoparticles' size, motion of particles, and suspension stability. Especially, the convective heat transfer coefficient of

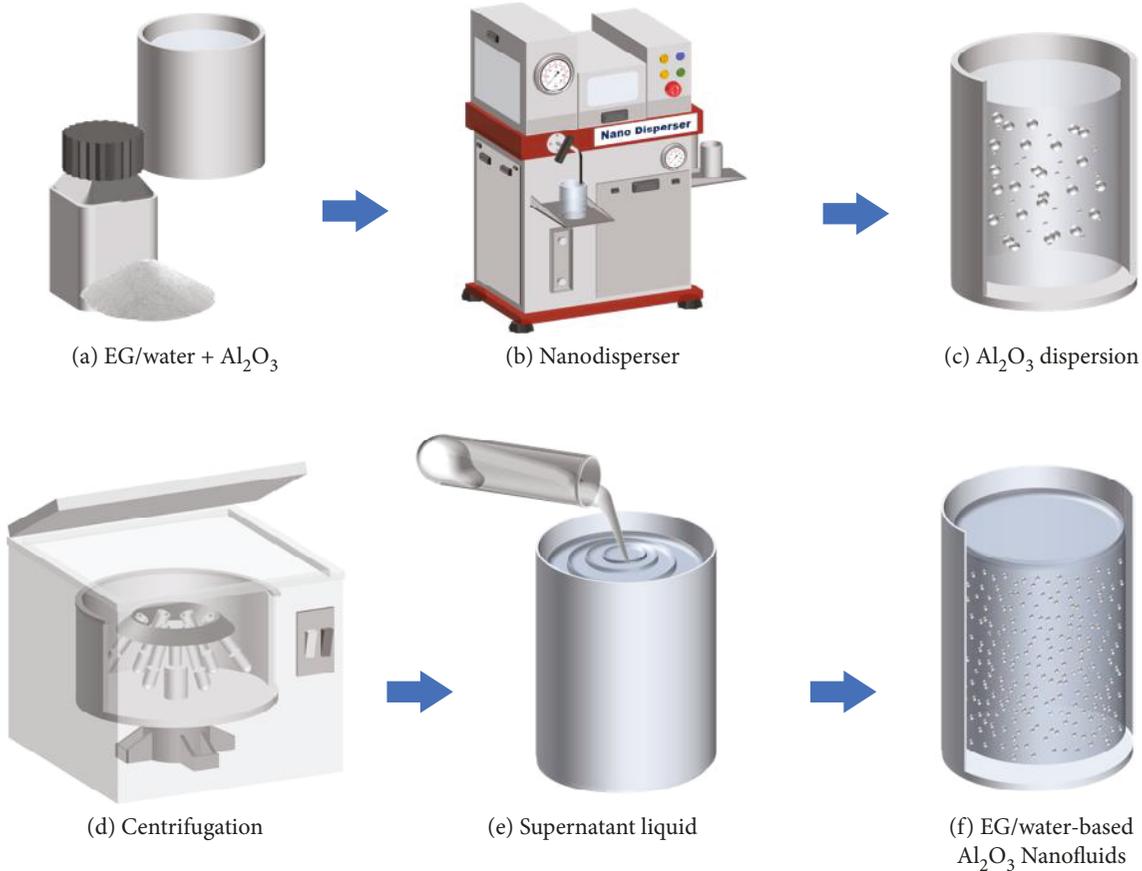


FIGURE 1: Manufacturing process of EG/water-based Al<sub>2</sub>O<sub>3</sub> nanofluids.

nanofluids is higher than that of thermal conductivity enhancement because of nanoparticle's motion [12, 17]. However, when the aggregation and sedimentation of nanoparticles are occurred due to low suspension stability, the convective heat transfer characteristics of nanofluids cannot be improved. From the previous results, it is well known that the key factor of nanofluids with improved thermal characteristics is the suspension stability [21, 22, 26].

For this reason, many researchers have tried to maintain the suspension stability of nanofluids using a variety of methods such as bath sonication, tip sonication, surface treatment, and use of surfactants [26, 32–35]. Despite these efforts, the suspension stability of nanofluids still has problem according to the temperature. Choi et al. [35] reported the effect of temperature on suspension stability of nanofluids. Water-based CNT nanofluids are prepared with four kinds of surfactants, and they reported that the water-based CNT nanofluids with Triton X-100 dramatically decreased suspension stability at high temperature (85°C). Also, the nanofluids with SDS and CTAB formed precipitates in the bottom of the bottle at low temperature (10°C). Wen and Ding [36] also reported that the water-based CNT nanofluids have poor suspension stability at 69.7°C because of strong aggregation, so they did not evaluate the thermal conductivity of nanofluids due to low suspension stability. Based on their results [26, 32–36], several researchers [35–39] explained the reason why temperature

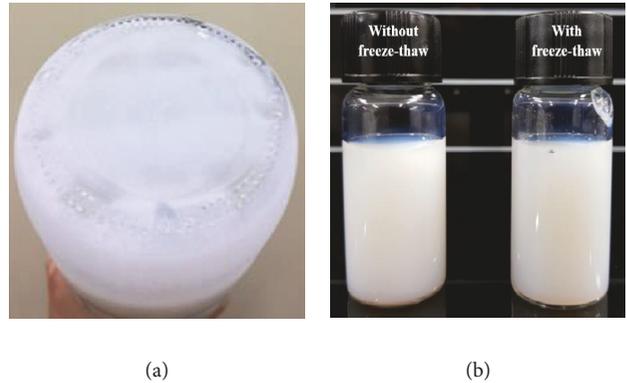


FIGURE 2: The frozen nanofluids at -32°C for 24 hours (a) and two types of EG/water-based Al<sub>2</sub>O<sub>3</sub> nanofluids at room temperature (b).

affects the stability of nanofluids in high temperature and low temperature, respectively. At the high temperature region, the suspension stability of nanofluids depends on the aggregation of nanoparticles which is occurred by the particle collision due to the intense movement of individual nanoparticles [37–39]. They reported that the viscosity of basefluid is decreased with temperature and collision between fluid molecules and nanoparticles increases. However, at the low temperature region, surfactants used for dispersion process strongly affect the suspension

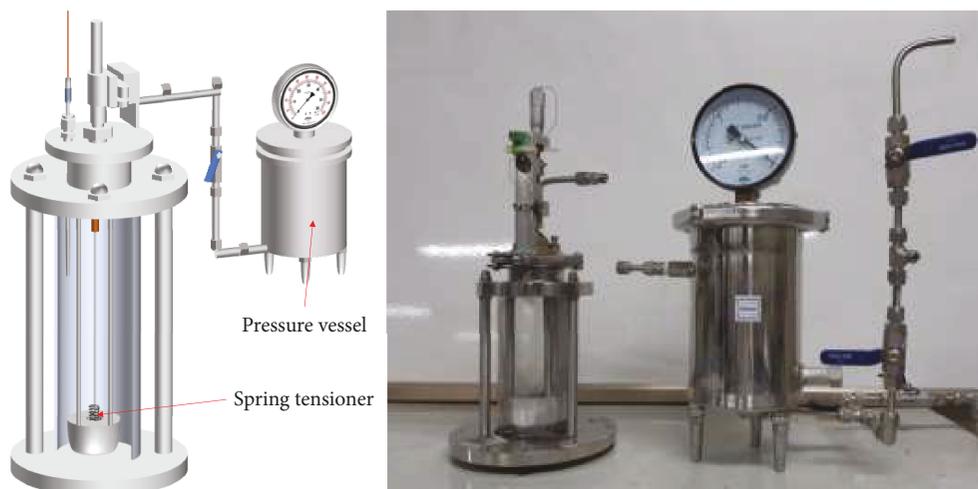


FIGURE 3: Images of transient hot wire system.

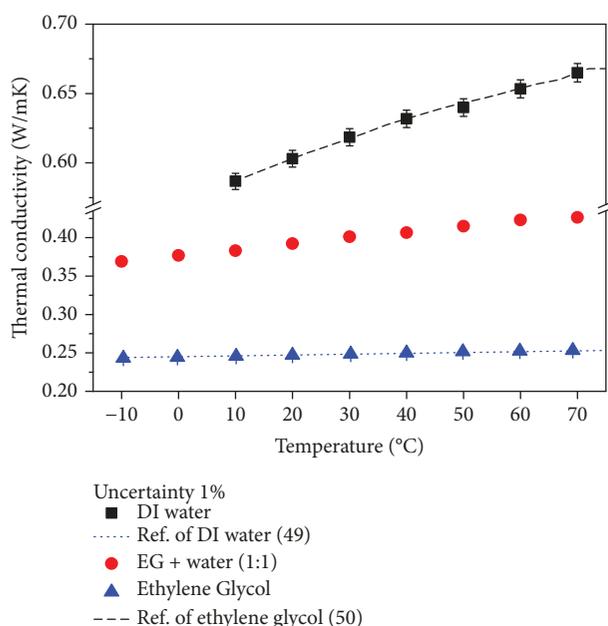
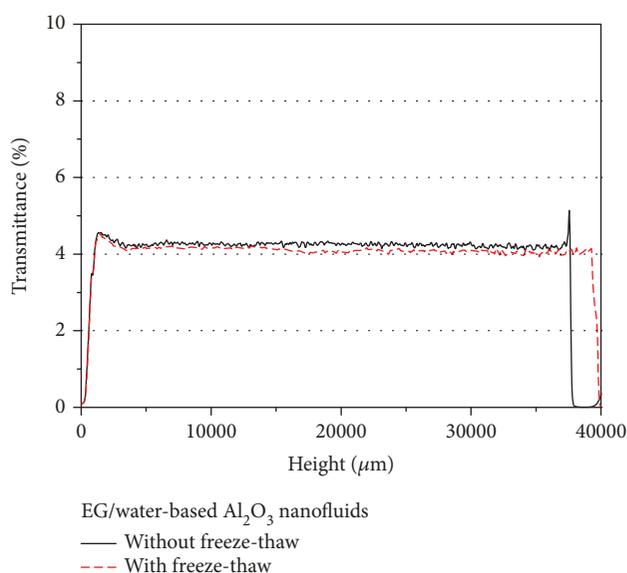


FIGURE 4: Validation results of transient hot wire system.

stability because the surfactants such as CTAB and SDS formed precipitates in the nanofluids because of reduced solubility [35].

Moreover, when the thermal management devices such as automobile and chiller using nanofluids are exposed to temperatures below the freezing point due to winter season or unexpected situations, it is very important to observe the behavior of suspension stability of nanofluids. To the author's knowledge, there are no experimental data of the effects of freeze-thaw on the suspension stability and thermal properties of nanofluids. Therefore, it is necessary to investigate the effect of freeze-thaw on the suspension stability of nanofluids.

So, in this paper, we experimentally investigated the effects of freeze-thaw on the suspension stability and thermal properties of nanofluids. For this purpose, we

FIGURE 5: Transmittance of two types of EG/water-based  $\text{Al}_2\text{O}_3$  nanofluids.

prepared the nanofluids using  $\text{Al}_2\text{O}_3$  nanoparticles, and antifreeze coolant (ethylene glycol and water at a volume ratio of 1:1) was employed as the basefluid. Especially, we used a two-step method with a nanodisperser and decanting processes to disperse the  $\text{Al}_2\text{O}_3$  nanoparticles into the basefluid. To observe the effect of freeze-thaw on nanofluids' suspension stability and thermal conductivity, the prepared nanofluids were frozen at  $-32^\circ\text{C}$  for 24 hours, and then stored for 24 hours at room temperature in order to thaw it completely. Suspension stability, particle size distribution, deformation of nanoparticles, and thermal conductivity of nanofluids without freeze-thaw and with freeze-thaw were measured by Turbiscan, a particle size analyzer, TEM, and the transient hot wire method, respectively. The comparison results between the nanofluids without freeze-thaw and with freeze-thaw were experimentally presented.

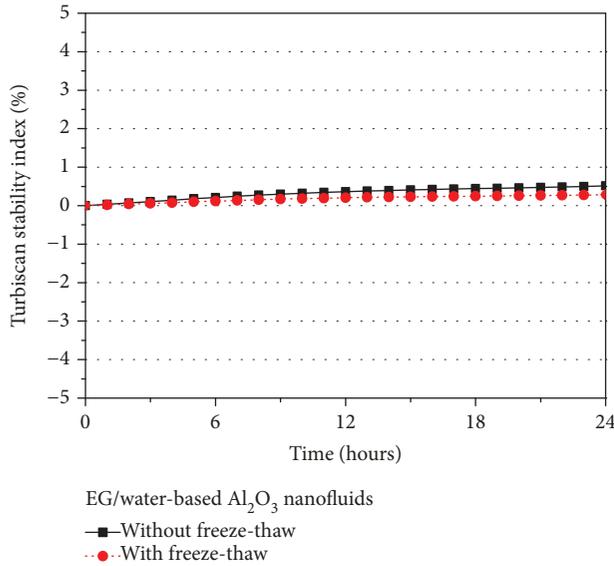


FIGURE 6: TSI values of two types of EG/water-based  $\text{Al}_2\text{O}_3$  nanofluids with elapsed time.

## 2. Experimental Study

**2.1. Manufacturing Processes of EG/Water-Based  $\text{Al}_2\text{O}_3$  Nanofluids.**  $\text{Al}_2\text{O}_3$  nanoparticles were used ( $D = 40$  to  $50$  nm, Nanophase) to manufacture the EG/water-based  $\text{Al}_2\text{O}_3$  nanofluids. The EG/water-based  $\text{Al}_2\text{O}_3$  nanofluids were produced using the two-step method with a nanodisperser (NH500-Y100, Ilshinautoclave) and decanting process as shown in Figure 1. The dispersion method using a nanodisperser is a method of passing the fluid and nanoparticles through a micro-orifice nozzle while being pressurized at ultra-high pressure (up to 1500 bar). Basefluid passing through the fine orifice nozzle were atomized and mixed uniformly with the nanoparticles due to the cavitation, turbulence, and high shear stress. Using this method, we obtained 2 liters of nanofluids at 6 volume fractions per unit minute. After that, a three-step decanting method was employed to enhance the suspension stability and thermal properties of nanofluids [40]. Finally, the manufactured nanofluids at 1.43 volume fractions were evaluated for suspension stability, thermal conductivity, and particle size distribution.

**2.2. Freeze-Thaw Process of EG/Water-Based  $\text{Al}_2\text{O}_3$  Nanofluids.** In this study, a thermostatic chamber was used to freeze the nanofluids at  $-32^\circ\text{C}$  for 24 hours. The freezing temperature of  $-32^\circ\text{C}$  is based on the US military standard (MIL-STD-810G method 502.5). It provides temperature criteria for evaluating the effect of low temperature conditions on material safety, integrity and performance during storage, operation, and manipulation. Figure 2 shows the frozen EG/water-based  $\text{Al}_2\text{O}_3$  nanofluids. To completely thaw the frozen nanofluids, they were stored for 24 hours at room temperature. The nanofluids with freeze-thaw process were compared with the nanofluids without the process. It is difficult to determine whether the suspension stability of nanofluids has changed due to the freeze-thaw process. Because

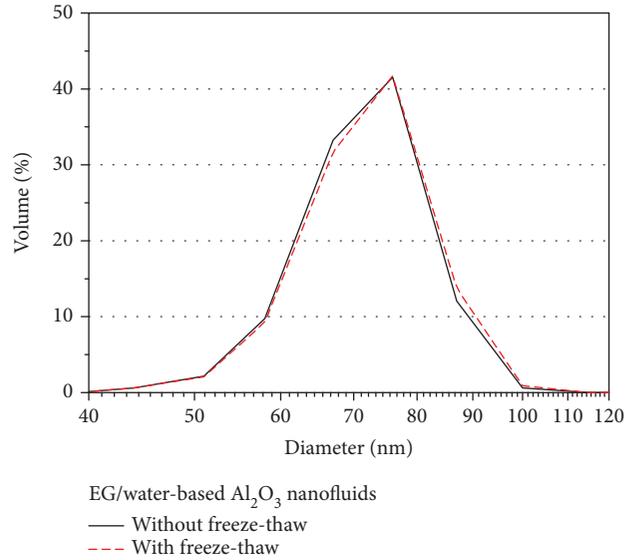


FIGURE 7: PSA results of two types of EG/water-based  $\text{Al}_2\text{O}_3$  nanofluids.

of this, we quantitatively evaluated the effect of the freeze-thaw process on the suspension stability of nanofluids using Turbiscan.

**2.3. Suspension Characterization of EG/Water-Based  $\text{Al}_2\text{O}_3$  Nanofluids.** The suspension stability of EG/water-based  $\text{Al}_2\text{O}_3$  nanofluids was quantitatively evaluated by Turbiscan (Turbiscan Lab, Fullbrook Systems Ltd.) which can be widely used to evaluate the suspension stability of nanofluids in previous researches [41–46]. Turbiscan can measure the intensity of the transmitted light through the nanofluids with elapsed time using the emitted wavelength of 880 nm. Based on the data, TSI (Turbiscan stability index) as given by equation (1) was calculated according to the sample's height with elapsed time.

$$\text{TSI} = \sum_i \frac{\sum_h |T_i(h) - T_{i-1}(h)|}{H}, \quad (1)$$

where  $i$ ,  $h$ ,  $T$ , and  $H$  are measured time, measured position, transmittance of light, and total sample height, respectively. If the nanofluids have good stability, the TSI is nearly not changed with the elapsed time. Moreover, the large values of TSI indicate lower suspension stability.

**2.4. Thermal Conductivity of EG/Water-Based  $\text{Al}_2\text{O}_3$  Nanofluids.** To investigate the effect of the freeze-thaw on the thermal conductivity of the nanofluids, it is measured by transient hot wire method in temperature range from  $-10^\circ\text{C}$  to  $70^\circ\text{C}$  [47, 48]. To measure the thermal conductivity over a wide temperature range, including high temperatures, two challenge issues must be solved. The first is to maintain the tension of a hot wire because the tension of a hot wire is easily changed due to the temperature variation in measurement device. So, it is very important to maintain constant tension of a hot wire (platinum wire) when

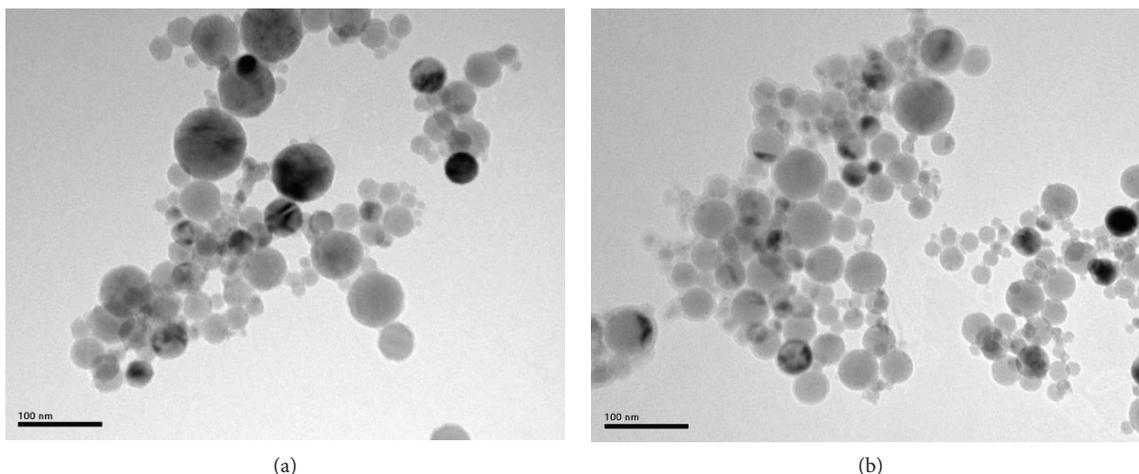


FIGURE 8: TEM images of two types of EG/water-based  $\text{Al}_2\text{O}_3$  nanofluids: without freeze-thaw (a) and with freeze-thaw (b).

measuring thermal conductivity by using transient hot wire method. Second, the generation of non-condensable gases inside the fluid should be restrained. Non-condensable gases inside the fluid are generated and attached at the hot wire surface. Bubbles on the surface of the hot wire dramatically decrease the accuracy of thermal conductivity measurement at high temperature ( $>40^\circ\text{C}$ ). So, two challenge issues must be solved to measure the accurate thermal conductivity.

To overcome these challenges for measuring the accurate thermal conductivity of nanofluids, we employed spring tensioner and pressure vessel as shown in Figure 3. A spring tensioner maintains the tension of the hot wire (platinum wire) even under a wide temperature change. Moreover, pressure vessel pressurizes the fluid to 5 bars to suppress bubbles at high temperatures.

The validation of the transient hot wire system was conducted using EG as well as DI water, and it had 1% uncertainty as shown in Figure 4 [49, 50].

### 3. Results and Discussion

**3.1. Suspension Characterization of EG/Water-Based  $\text{Al}_2\text{O}_3$  Nanofluids.** Figure 5 shows the measured transmittances for the two kinds of nanofluids (without freeze-thaw and with freeze-thaw) according to the sample's height at room temperature (approximately  $25^\circ\text{C}$ ). The transmittances of two types of nanofluids were not significantly different according with the sample's height. It means that the freeze-thaw process does not have a sharp influence on the suspension stability of the nanofluids.

Moreover, we measured transmittances of two kinds of nanofluids with elapsed time to evaluate the long-term suspension stability of nanofluids. With the data, TSI values are shown in Figure 6. Both nanofluids have a TSI variation of less than 1% and their suspension stability is well maintained for 24 hours. Based on the result, the freeze-thaw process does not affect the suspension stability of nanofluids.

**3.2. Particle Size Distribution and Particle Deformation of EG/Water-Based  $\text{Al}_2\text{O}_3$  Nanofluids.** Many researchers reported that suspension stability of nanofluids and particle agglomeration are closely related [35, 36, 51–53]. Moreover, they noted that particle agglomeration can affect the properties of nanofluids such as thermal conductivity, viscosity, and optical properties. Because of this, we also measured the particle size distribution suspended in the nanofluids with particle size analyzer (Zetasizer Nano S90, Malvern). Figure 7 shows the particle size distributions of two types of the nanofluids, respectively. The average particle size of EG/water-based  $\text{Al}_2\text{O}_3$  nanofluids before freeze-thaw process was 67.63 nm and the size of nanofluids after freeze-thaw was 68.07 nm. Moreover, the particle size distribution is similar between the two nanofluids. This result indicates that the freeze-thaw process of nanofluids did not affect the particle agglomeration nor the suspension stability.

Moreover, TEM images were employed to observe the deformation of particles. We prepared TEM samples using special method. First, the nanofluids were diluted with base fluid, and then moisture of diluted nanofluids were eliminated by paper towel. The remaining moisture was completely removed through natural drying at room temperature. This process can minimize the agglomeration of nanoparticles during the drying process. As shown in Figure 8, any particle deformation between the two types of nanofluids was not observed.

**3.3. Thermal Conductivity of EG/Water-Based  $\text{Al}_2\text{O}_3$  Nanofluids.** To evaluate the effect of freeze-thaw on thermal conductivity of nanofluids (without freeze-thaw and with freeze-thaw), we measured the thermal conductivity of the two types of nanofluids using transient hot wire method. As shown in Figure 9, the thermal conductivity of both nanofluids nearly is the same and within the error range of the transient hot wire system. It clearly shows that the freeze-thaw process does not affect the thermal conductivity of the EG/water-based  $\text{Al}_2\text{O}_3$  nanofluids. In addition, it is reasonable because the particle size distribution as well as suspension stability was not affected by the freeze-thaw process of nanofluids.

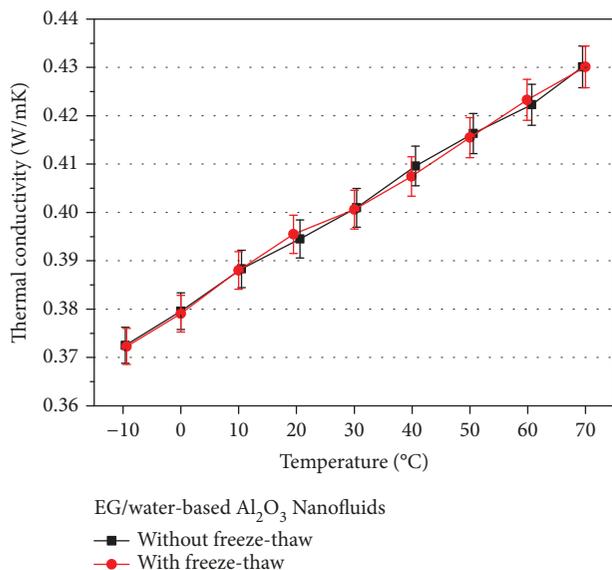


FIGURE 9: Thermal conductivity of two types of EG/water-based  $\text{Al}_2\text{O}_3$  nanofluids.

#### 4. Conclusions

This investigation experimentally presents the effect of freeze-thaw on the suspension stability, the particle size distribution, and the thermal conductivity of EG/water-based nanofluids containing  $\text{Al}_2\text{O}_3$  nanoparticles that have been used as a working fluid for various cooling systems. The EG/water-based  $\text{Al}_2\text{O}_3$  nanofluids were prepared using a two-step method with a nanodisperser and decanting processes. To evaluate the suspension stability of the nanofluids with freeze-thaw process and without the process quantitatively, Turbiscan was employed to measure the intensity of the transmitted light through the nanofluids with elapsed time and TSI value was used to analyze it. It is shown that both nanofluids with and without the freeze-thaw process have well maintained their suspension stability for 24 hours. Also, particle size distributions and particle deformation in both nanofluids are experimentally presented with PSA and TEM. We observed that the particle size distribution of nanofluids were almost similar regardless of the freeze-thaw process. Moreover, particle deformation also was not observed. The thermal conductivity of the two types of nanofluids were measured using the transient hot wire method at the temperature range -10 to 70°C, respectively. The thermal conductivities of both nanofluids also did not changed by freeze-thaw process. With the experimental results, we clearly show that the freeze-thaw process did not affect the particle size dispersed in nanofluids, the suspension stability, nor the thermal conductivity of nanofluids.

#### Data Availability

The data in Figures 4–9 used to support the findings of this study are included within the supplementary information files as follows: Supplementary 1 (data in Figure 4);

Supplementary 2 (data in Figure 5); Supplementary 3 (data in Figure 6); Supplementary 4 (data in Figure 7); and Supplementary 5 (data in Figure 9).

#### Conflicts of Interest

The authors of this work declare that there are no conflicts of interest regarding the publication of this paper.

#### Acknowledgments

This work was supported by Defense Acquisition Program Administration and Agency for Defense Development under the contract UC150011ID, Korea.

#### Supplementary Materials

*Supplementary 1.* It includes experimental data for thermal conductivities measured by the transient hot wire method as shown in Figure 4.

*Supplementary 2.* It includes experimental data for transmittance measured by the Turbiscan as shown in Figure 5.

*Supplementary 3.* It includes experimental data for TSI values measured by the Turbiscan as shown in Figure 6.

*Supplementary 4.* It includes experimental data for PSA results measured by the particle size analyzer as shown in Figure 7.

*Supplementary 5.* It includes experimental data for thermal conductivities measured by the transient hot wire method as shown in Figure 9.

#### References

- [1] Z. Zhneghuo, X. Tao, and F. Xiaoming, "Experimental study on heat transfer enhancement of a helically baffled heat exchanger combined with three-dimensional finned tubes," *Applied Thermal Engineering*, vol. 24, no. 14-15, pp. 2293–2300, 2004.
- [2] L. D. Tijing, B. C. Pak, B. J. Baek, and D. H. Lee, "A study on heat transfer enhancement using straight and twisted internal fin inserts," *International Communications in Heat and Mass Transfer*, vol. 33, no. 6, pp. 719–726, 2006.
- [3] P. Naphon, "Effect of coil-wire insert on heat transfer enhancement and pressure drop of the horizontal concentric tubes," *International Communications in Heat and Mass Transfer*, vol. 33, no. 6, pp. 753–763, 2006.
- [4] B. Sahin and A. Demir, "Performance analysis of a heat exchanger having perforated square fins," *Applied Thermal Engineering*, vol. 28, no. 5-6, pp. 621–632, 2008.
- [5] M.-Y. Wen and C.-Y. Ho, "Heat-transfer enhancement in fin-and-tube heat exchanger with improved fin design," *Applied Thermal Engineering*, vol. 29, no. 5-6, pp. 1050–1057, 2009.
- [6] S. M. Peyghambarzadeh, S. H. Hashemabadi, M. S. Jamnani, and S. M. Hoseini, "Improving the cooling performance of automobile radiator with  $\text{Al}_2\text{O}_3$ /water nanofluid," *Applied Thermal Engineering*, vol. 31, no. 10, pp. 1833–1838, 2011.
- [7] J.-H. Lee, S.-H. Lee, C. Choi, S. Jang, and S. Choi, "A review of thermal conductivity data, mechanisms and models for

- nanofluids,” *International Journal of Micro-Nano Scale Transport*, vol. 1, no. 4, pp. 269–322, 2010.
- [8] J. A. Eastman, S. U. S. Choi, S. Li, L. J. Thompson, and S. Lee, “Enhanced thermal conductivity through the development of nanofluids,” in *Nanophase and Nanocomposite Materials II*, S. Komarneni, J. C. Parker, and H. J. Wollenberger, Eds., pp. 3–11, MRS, Pittsburg, PA, USA, 1997.
- [9] S. Lee, S. U. S. Choi, S. Li, and J. A. Eastman, “Measuring thermal conductivity of fluids containing oxide nanoparticles,” *Journal of Heat Transfer*, vol. 121, no. 2, pp. 280–289, 1999.
- [10] X. Wang, X. Xu, and S. U. S. Choi, “Thermal conductivity of nanoparticle - fluid mixture,” *Journal of Thermophysics and Heat Transfer*, vol. 13, no. 4, pp. 474–480, 1999.
- [11] Y. Xuan and Q. Li, “Heat transfer enhancement of nanofluids,” *International Journal of Heat and Fluid Flow*, vol. 21, no. 1, pp. 58–64, 2000.
- [12] Q. Li and Y. Xuan, “Convective heat transfer and flow characteristics of Cu-water nanofluid,” *Science in China Series E: Technological Science*, vol. 45, no. 4, pp. 408–416, 2002.
- [13] S. P. Jang and S. U. S. Choi, “Role of Brownian motion in the enhanced thermal conductivity of nanofluids,” *Applied Physics Letters*, vol. 84, no. 21, pp. 4316–4318, 2004.
- [14] D. Wen and Y. Ding, “Experimental investigation into convective heat transfer of nanofluids at the entrance region under laminar flow conditions,” *International Journal of Heat and Mass Transfer*, vol. 47, no. 24, pp. 5181–5188, 2004.
- [15] D. J. Faulkner, D. R. Rector, J. J. Davidson, and R. Shekarriz, “Enhanced heat transfer through the use of nanofluids in forced convection,” in *ASME 2004 International Mechanical Engineering Congress and Exposition*, pp. 219–224, Anaheim, CA, USA, 2004.
- [16] M. S. Liu, M. C. C. Lin, C. Y. Tsai, and C. C. Wang, “Enhancement of thermal conductivity with Cu for nanofluids using chemical reduction method,” *International Journal of Heat and Mass Transfer*, vol. 49, no. 17–18, pp. 3028–3033, 2006.
- [17] B. J. Cox, N. Thamwattana, and J. M. Hill, “Electrostatic force between coated conducting spheres with applications to electrorheological nanofluids,” *Journal of Electrostatics*, vol. 65, no. 10–11, pp. 680–688, 2007.
- [18] J. Garg, B. Poudel, M. Chiesa et al., “Enhanced thermal conductivity and viscosity of copper nanoparticles in ethylene glycol nanofluid,” *Journal of Applied Physics*, vol. 103, no. 7, 2008.
- [19] W. Duangthongsuk and S. Wongwises, “An experimental study on the heat transfer performance and pressure drop of TiO<sub>2</sub>-water nanofluids flowing under a turbulent flow regime,” *International Journal of Heat and Mass Transfer*, vol. 53, no. 1–3, pp. 334–344, 2010.
- [20] A. R. Sajadi and M. H. Kazemi, “Investigation of turbulent convective heat transfer and pressure drop of TiO<sub>2</sub>/water nanofluid in circular tube,” *International Communications in Heat and Mass Transfer*, vol. 38, no. 10, pp. 1474–1478, 2011.
- [21] A. Ghadimi, R. Saidur, and H. S. C. Metselaar, “A review of nanofluid stability properties and characterization in stationary conditions,” *International Journal of Heat and Mass Transfer*, vol. 54, no. 17–18, pp. 4051–4068, 2011.
- [22] W. Yu and H. Xie, “A review on nanofluids: preparation, stability mechanisms, and applications,” *Journal of Nanomaterials*, vol. 2012, Article ID 435873, 17 pages, 2012.
- [23] A. M. Hussein, K. V. Sharma, R. A. Bakar, and K. Kadrigama, “The effect of nanofluid volume concentration on heat transfer and friction factor inside a horizontal tube,” in *Journal of Nanomaterials*, vol. 2013, Article ID 859563, 12 pages, 2013.
- [24] W. H. Azmi, K. V. Sharma, P. K. Sarma, R. Mamat, S. Anuar, and V. Dharma Rao, “Experimental determination of turbulent forced convection heat transfer and friction factor with SiO<sub>2</sub> nanofluid,” *Experimental Thermal and Fluid Science*, vol. 51, pp. 103–111, 2013.
- [25] D. Madhesh, R. Parameshwaran, and S. Kalaiselvam, “Experimental investigation on convective heat transfer and rheological characteristics of Cu-TiO<sub>2</sub> hybrid nanofluids,” *Experimental Thermal and Fluid Science*, vol. 52, pp. 104–115, 2014.
- [26] H. J. Kim, S.-H. Lee, J.-H. Lee, and S. P. Jang, “Effect of particle shape on suspension stability and thermal conductivities of water-based bohemite alumina nanofluids,” *Energy*, vol. 90, pp. 1290–1297, 2015.
- [27] H. J. Kim, S.-H. Lee, S. B. Kim, and S. P. Jang, “The effect of nanoparticle shape on the thermal resistance of a flat-plate heat pipe using acetone-based Al<sub>2</sub>O<sub>3</sub> nanofluids,” *International Journal of Heat and Mass Transfer*, vol. 92, pp. 572–577, 2016.
- [28] W. H. Azmi, K. Abdul Hamid, R. Mamat, K. V. Sharma, and M. S. Mohamad, “Effects of working temperature on thermo-physical properties and forced convection heat transfer of TiO<sub>2</sub> nanofluids in water – ethylene glycol mixture,” *Applied Thermal Engineering*, vol. 106, pp. 1190–1199, 2016.
- [29] T. J. Choi, B. Subedi, H. J. Ham, M. S. Park, and S. P. Jang, “A review of the internal forced convective heat transfer characteristics of nanofluids: experimental features, mechanisms and thermal performance criteria,” *Journal of Mechanical Science and Technology*, vol. 32, no. 8, pp. 3491–3505, 2018.
- [30] N. Ali, J. A. Teixeira, and A. Addali, “A review on nanofluids: fabrication, stability, and thermophysical properties,” *Journal of Nanomaterials*, vol. 2018, Article ID 6978130, 33 pages, 2018.
- [31] S. U. S. Choi, “Enhancing thermal conductivity of fluids with nanoparticles,” in *Congress and Exposition*, pp. 99–105, ASME, FED 231/MD 66, San Francisco, CA, USA, 1995.
- [32] M. P. Beck, Y. Yuan, P. Warriar, and A. S. Teja, “The thermal conductivity of alumina nanofluids in water, ethylene glycol, and ethylene glycol + water mixtures,” *Journal of Nanoparticle Research*, vol. 12, no. 4, pp. 1469–1477, 2010.
- [33] S. Sen Gupta, V. Manoj Siva, S. Krishnan et al., “Thermal conductivity enhancement of nanofluids containing graphene nanosheets,” *Journal of Applied Physics*, vol. 110, no. 8, 2011.
- [34] S.-H. Lee and S. P. Jang, “Efficiency of a volumetric receiver using aqueous suspensions of multi-walled carbon nanotubes for absorbing solar thermal energy,” *International Journal of Heat and Mass Transfer*, vol. 80, pp. 58–71, 2015.
- [35] T. J. Choi, S. P. Jang, and M. A. Kedzierski, “Effect of surfactants on the stability and solar thermal absorption characteristics of water-based nanofluids with multi-walled carbon nanotubes,” *International Journal of Heat and Mass Transfer*, vol. 122, pp. 483–490, 2018.
- [36] D. Wen and Y. Ding, “Effective thermal conductivity of aqueous suspensions of carbon nanotubes (carbon nanotube nanofluids),” *Journal of Thermophysics and Heat Transfer*, vol. 18, no. 4, pp. 481–485, 2004.
- [37] M. E. Woods and I. M. Krieger, “Rheological studies on dispersions of uniform colloidal spheres I. Aqueous dispersions in steady shear flow,” *Journal of Colloid and Interface Science*, vol. 34, no. 1, pp. 91–99, 1970.

- [38] I. M. Krieger, "Rheology of monodisperse latices," *Advances in Colloid and Interface Science*, vol. 3, no. 2, pp. 111–136, 1972.
- [39] H. Chang, C. H. Lo, T. T. Tsung et al., "Temperature effect on the stability of CuO nanofluids based on measured particle distribution," *Key Engineering Materials*, vol. 295–296, pp. 51–56, 2005.
- [40] J.-H. Lee, S.-H. Lee, and S. Pil Jang, "Do temperature and nanoparticle size affect the thermal conductivity of alumina nanofluids?," *Applied Physics Letters*, vol. 104, no. 16, article 161908, 2014.
- [41] M. Wisniewska, "Influences of polyacrylic acid adsorption and temperature on the alumina suspension stability," *Powder Technology*, vol. 198, no. 2, pp. 258–266, 2010.
- [42] W. Kang, B. Xu, Y. Wang et al., "Stability mechanism of W/O crude oil emulsion stabilized by polymer and surfactant," *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, vol. 384, no. 1–3, pp. 555–560, 2011.
- [43] M. Wiśniewska, K. Terpiłowski, S. Chibowski, T. Urban, V. I. Zarko, and V. M. Gun'ko, "Effect of polyacrylic acid (PAA) adsorption on stability of mixed alumina-silica oxide suspension," *Powder Technology*, vol. 233, pp. 190–200, 2013.
- [44] G. Colangelo, E. Favale, P. Miglietta, M. Milanese, and A. de Risi, "Thermal conductivity, viscosity and stability of Al<sub>2</sub>O<sub>3</sub>-diathermic oil nanofluids for solar energy systems," *Energy*, vol. 95, pp. 124–136, 2016.
- [45] M. Luo, X. Qi, T. Ren et al., "Heteroaggregation of CeO<sub>2</sub> and TiO<sub>2</sub> engineered nanoparticles in the aqueous phase: application of turbiscan stability index and fluorescence excitation-emission matrix (EEM) spectra," *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, vol. 533, pp. 9–19, 2017.
- [46] X. Qi, Y. n. Dong, H. Wang, C. Wang, and F. Li, "Application of turbiscan in the homoaggregation and heteroaggregation of copper nanoparticles," *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, vol. 535, pp. 96–104, 2017.
- [47] S. H. Lee and S. P. Jang, "Note: effect of the tilting angle of the wire on the onset of natural convection in the transient hot wire method," in *Review of Scientific Instruments*, vol. 83, no. 7, 2012.
- [48] S. H. Lee, *Transient hot wire method for measuring thermal conductivity of nanofluids*, [M.S. thesis], Korea Aerospace University, 2010.
- [49] F. P. Incropera and D. P. Dewitt, *Fundamentals of Heat and Mass Transfer*, Wiley, 5th edition, 2002.
- [50] D. Bohne, S. Fischer, and E. Obermeier, "Thermal, conductivity, density, viscosity, and Prandtl-numbers of ethylene glycol-water mixtures," *Berichte der Bunsengesellschaft für physikalische Chemie*, vol. 88, no. 8, pp. 739–742, 1984.
- [51] B. P. Singh, R. Menchavez, C. Takai, M. Fuji, and M. Takahashi, "Stability of dispersions of colloidal alumina particles in aqueous suspensions," *Journal of Colloid and Interface Science*, vol. 291, no. 1, pp. 181–186, 2005.
- [52] Y. He, Y. Jin, H. Chen, Y. Ding, D. Cang, and H. Lu, "Heat transfer and flow behaviour of aqueous suspensions of TiO<sub>2</sub> nanoparticles (nanofluids) flowing upward through a vertical pipe," *International Journal of Heat and Mass Transfer*, vol. 50, no. 11–12, pp. 2272–2281, 2007.
- [53] K. B. Anoop, S. Kabelac, T. Sundararajan, and S. K. Das, "Rheological and flow characteristics of nanofluids: influence of electroviscous effects and particle agglomeration," *Journal of Applied Physics*, vol. 106, no. 3, 2009.



**Hindawi**  
Submit your manuscripts at  
[www.hindawi.com](http://www.hindawi.com)

