

Chemosensing Test Paper Based on Aggregated Nanoparticles of a Barbituric Acid Derivative

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^1H NMR (400 MHz, Chloroform-*d*) δ 8.35 (s, 1H), 8.14 (d, J = 8.8 Hz, 2H), 7.36 (t, J = 7.8 Hz, 4H), 7.21 (d, J = 7.6 Hz, 6H), 6.95 (d, J = 8.7 Hz, 2H), 4.74 (dd, J = 14.0, 10.5 Hz, 2H), 2.37 (q, J = 12.5 Hz, 4H), 1.84 (s, 3H), 1.67 (s, 4H), 1.42 – 1.33 (m, 4H), 1.31 – 1.22 (m, 3H).

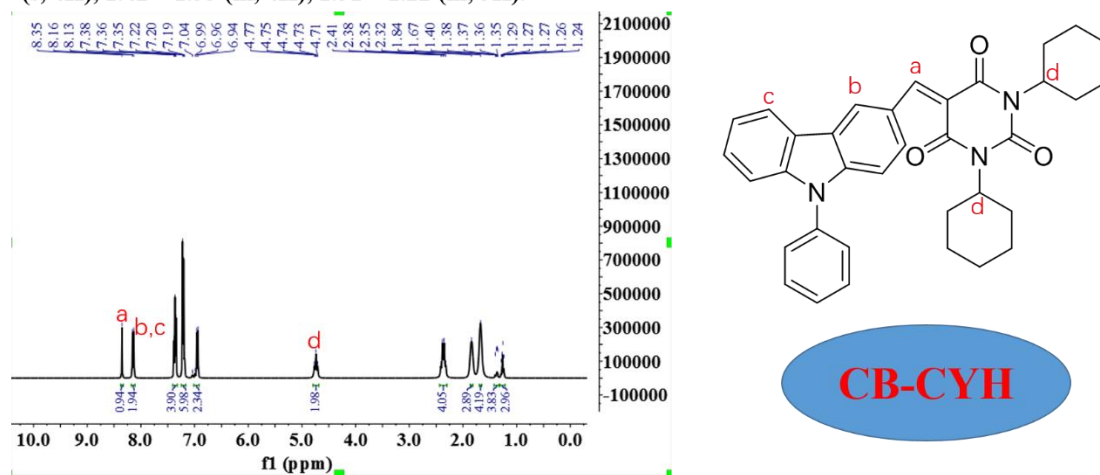


Fig. S1. ^1H NMR of **CB-CYH** in Chloroform-*d*

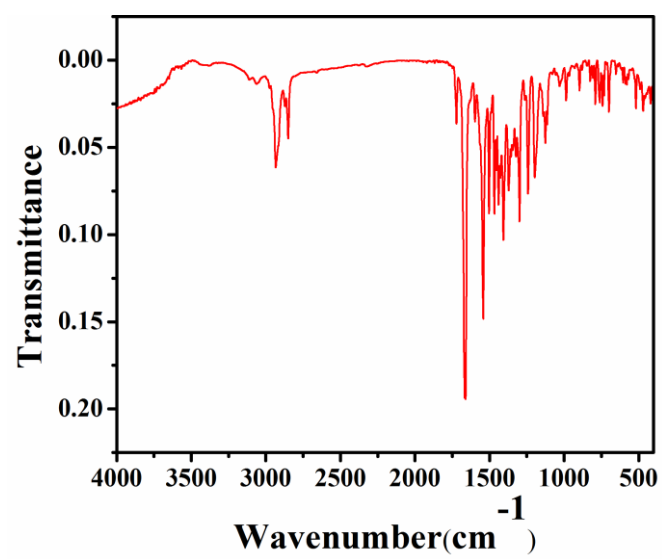


Fig. S2. FT-IR spectra of **CB-CYH**

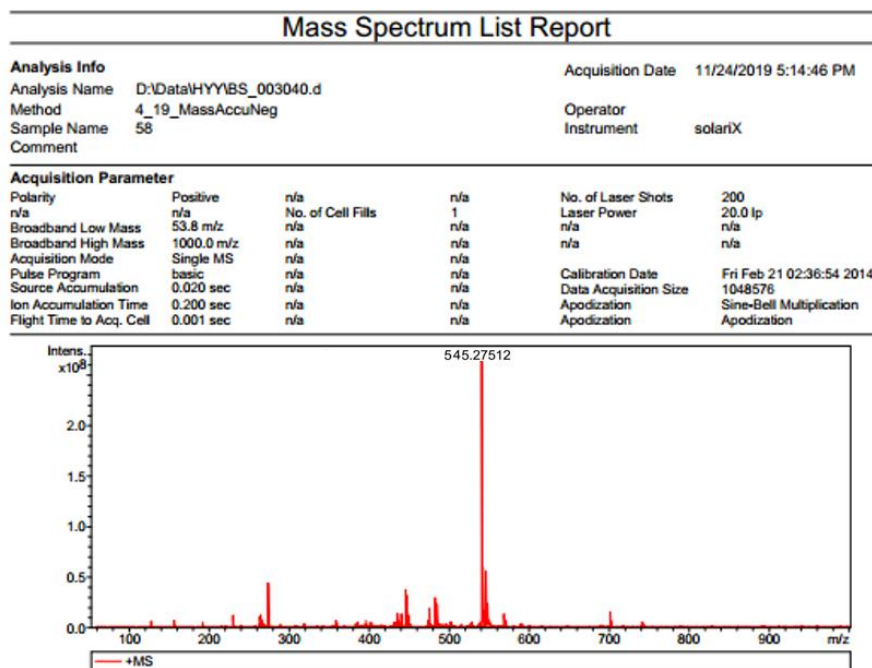


Fig. S3. HRMS spectra of **CB-CYH**

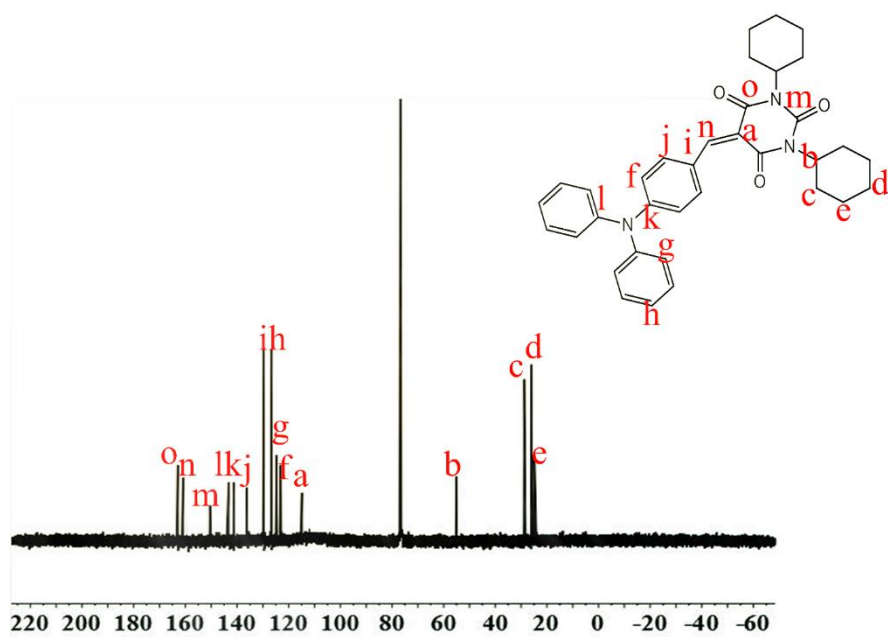


Fig. S4. ^{13}C NMR of CB-CYH in $\text{Chloroform-}d$.

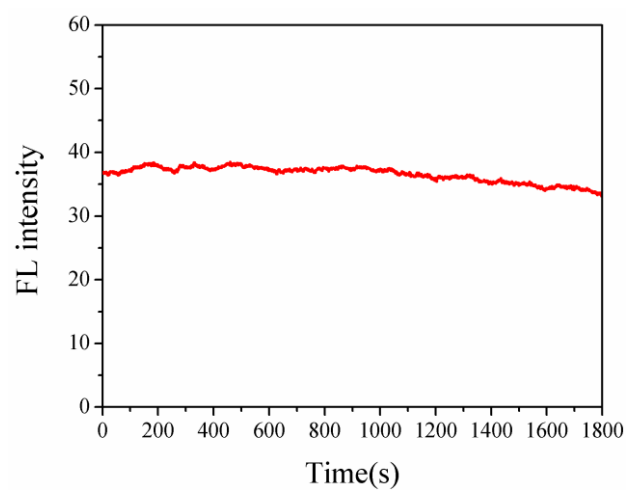


Fig. S5. CB-CYH light stability in nano-aggregates state.

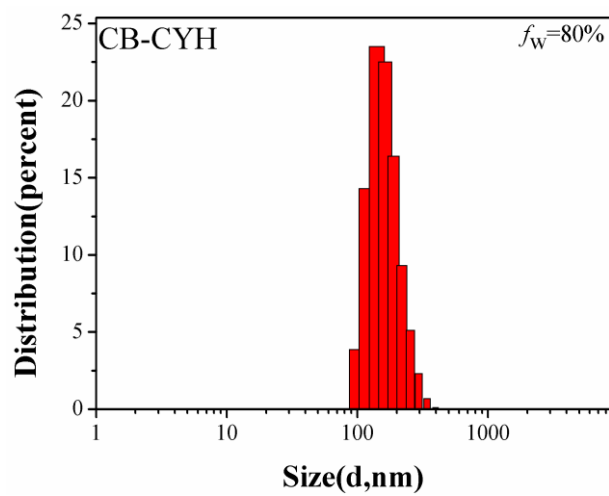


Fig. S6. Particle size distribution histograms of **CB-CYH** in a THF/H₂O mixture ($f_w = 80\%$) was left at room temperature for half an hour (Solution concentration: 10^{-4}M).