

## Research Article

# Drug Release Effect on Large-Clearance TiO<sub>2</sub> Nanotube Arrays Decorated with P25 Nanoparticles

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Large-clearance TiO<sub>2</sub> nanotube arrays (LTAs) were synthesized by anodization, and evenly distributed P25 nanoparticles (P25 NPs) decorated on this were prepared by the hydrothermal treatment. The effect of P25 concentrations on the morphology of LTAs was studied. LTAs decorated with P25 NPs (LTAs-P25) were loaded with ibuprofen (IBU), and the release properties were investigated. Interestingly, P25 NPs were successfully decorated on the surface of LTAs, with an optimal concentration of  $1.2 \times 10^{-3}$  M. The total amount of drug released from LTAs increased by 49.5% compared to that from TiO<sub>2</sub> nanotube arrays without large clearance. LTAs-P25 provided a lower burst and longer release time.

## 1. Introduction

Nanotechnology is the science that manipulates matter at molecular and atomic levels. Different application fields of components at the nanoscale level have been explored, such as nanomedicine [1–3] and electrochemical sensor [4–6]. Titania nanotube arrays (TNTs) formed by a self-ordering electrochemical anodization process on Ti have attracted considerable attention due to their biocompatibility, controlled nanotube dimensions, chemical stability, and osseointegration properties. TNTs have been recognized as a promising strategy to develop advanced implants and local drug delivery systems [7, 8]. Previous studies have demonstrated that TNTs as orthopedic implants can promote osteointegration [9, 10]. Moreover, TNTs are used as local drug loading and release systems in the field of bone implants to enhance antibacterial and anti-inflammatory properties, which are desirable for preventing implant-associated infections [11, 12]. However, the nanotube walls are so smooth that the loaded drugs are easily released. The burst release in a short period is not conducive to the sustained release

of a drug. Therefore, the surface modification of TNTs is an important means to improve the drug loading capacity of this kind of biomedical material and realize the sustained and controlled release mode [13].

P25 nanoparticles have a diameter of about 21 nm and a large specific surface area and are composed of 80% anatase titanium dioxide and 20% rutile titanium dioxide. P25 is widely used in biomedicine and has attracted much attention because of its high antibacterial activity, biocompatibility, and low toxicity or no toxicity in vitro and in vivo [14, 15]. In this paper, we synthesized the large-clearance TiO<sub>2</sub> nanotube arrays (LTAs) by anodizing Ti foils. The obtained LTAs have a large average pore diameter and big void spaces among the tubes. Then, we decorated P25 on the surface of LTAs to prepare a drug delivery system via the hydrothermal method and loaded it with ibuprofen (IBU) by the vacuum drying method. The large void spaces and pore diameter of LTAs facilitated the intercalation of the decorating and loading IBU into the nanotubes. The increased roughness and surface area of LTAs improved the adsorption capacity of the nanotube to achieve sustained release of the drugs. It

would be reasonable to believe that the LTAs-P25 will become promising candidates for drug delivery.

## 2. Experimental Details

**2.1. Preparation of LTAs.** The LTAs were obtained through a typical anodization approach, in which a two-electrode anodization setup was made. Ti was used as a working electrode, and a platinum sheet was used as a cathode; the two electrodes were separated by a distance of 10 mm. First, titanium foils were mechanically polished to achieve a mirrorlike finish and then ultrasonically cleaned in acetone for 15 min, followed by chemical etching for 10 s in nitric acid/hydrofluoric acid ( $\text{HNO}_3/\text{HF}$ ), rinsed with deionized water, and air-dried. The prepared Ti foil sample was then immersed in a lactic acid electrolyte and 10 vol% DMSO containing 0.35 wt%  $\text{NH}_4\text{F}$ , first under a voltage of 55 for approximately 15 min, followed by 50 V for 24 h at 30°C. After anodization, the LTAs were rinsed with deionized water and dried at room temperature.

**2.2. LTAs Decorated with P25 (LTAs-P25).** P25 nanoparticles were loaded on LTAs by the hydrothermal method. LTAs-P25 is prepared by the following steps: firstly, P25 nanoparticles with different concentrations are added into distilled water and then uniformly stirred for 15 min after ultrasonic treatment. The solution and LTAs were placed in a hydrothermal kettle for heat treatment (85°C, 12 h). After cooling and washing, annealing treatment followed (5°C/min, 400°C, 2 h).

**2.3. Drug Loading and Releasing on LTAs-P25.** The loading of IBU was realized by the vacuum drying method. A certain amount of IBU/ethanol was absorbed and spread on the LTA surface, put in a vacuum drying oven, and treated at 18°C for 2 h, and the step was repeated five times. LTAs-P25 loaded with IBU was placed in PBS, and a drug release experiment was carried out at 37°C for 15 days. A 500  $\mu\text{L}$  sample was sucked regularly, and then, the same volume of blank PBS was added to measure the drug release by ultraviolet-visible spectroscopy at 264 nm, and a drug release curve was drawn to judge drug release performance.

**2.4. Characterization.** Field emission scanning electron microscopy (FE-SEM, JSM-7500F) was used to analyze the surface morphology of the samples. Philips X'Pert MPD using Cu K radiation generated at 40 kV and 30 mA was used to characterize the crystalline phases of the samples. The amount of IBU released into the PBS at 264 nm was detected by UV-vis spectroscopy (TU1900). The detection of sample vibrations was measured on Nicolet Avatar 360 FTIR.

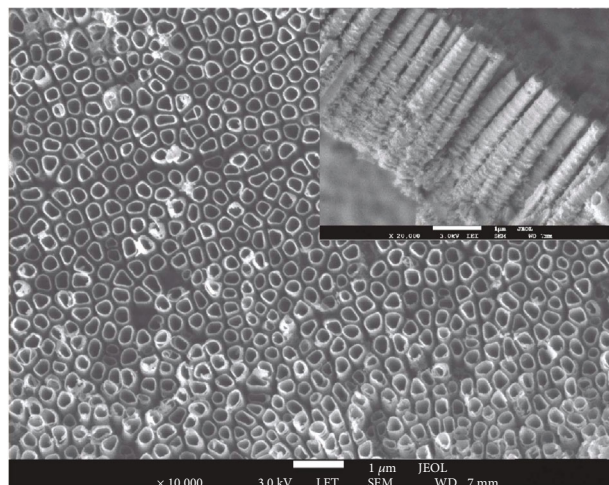
## 3. Results and Discussion

**3.1. The Morphology and XRD Analysis of LTAs and LTAs-P25.** Figures 1(a)–1(d) show the SEM images of the morphology of the prepared LTAs before (Figures 1(a) and 1(b)) and after (Figures 1(c) and 1(d)) decoration with P25 NPs. Specifically, Figure 1(a) is the top view SEM image with a low magnification, with the inset as its cross-section before deco-

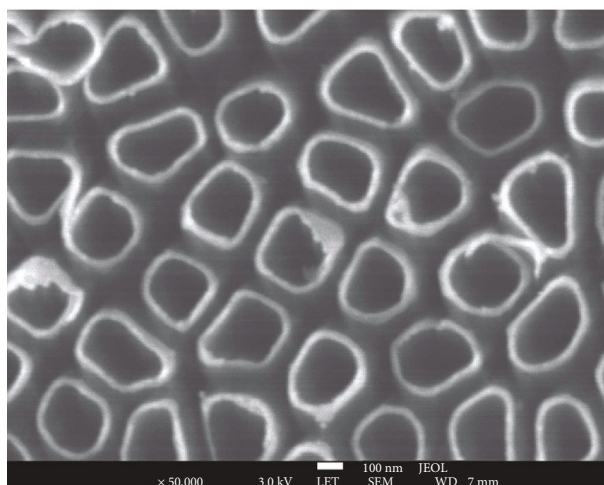
ration with P25. Figure 1(b) shows the image with a high magnification before decoration with P25. Figures 1(a) and 1(b) clearly show that the LTAs not only exhibit perfect morphologic characterization (such as self-organized, highly ordered, and free-standing) but also show the optimized geometrical architectures, the large void spaces, large pore diameters of  $\sim 320$  nm, long tube lengths of  $\sim 3.3$   $\mu\text{m}$ , and large intertubular spaces of 100–150 nm. This suggests that the special structure of the LTAs provides both the highly facilitated entrance for P25 nanoparticles and very easy growth points for homogenous modification [16]. Compared with conventional  $\text{TiO}_2$  nanotube arrays prepared in the glycerol/ $\text{H}_2\text{O}$  system, the LTAs have shown an increase of almost 110 nm in diameter and a significantly shorter length. Figures 1(c) and 1(d) present the morphologies of the as-prepared LTAs-P25. It indicates that P25 NPs are successfully decorated on the surface of TNTs, with the pore diameter decreasing to  $100 \pm 10$  nm after decoration (Figure 1(c)). The previous study has shown that the prepared nanotube has a relatively smooth tube wall when the organic electrolyte is involved in the anodic oxidation process [17]. In contrast, the specific surface area and roughness of the nanotubes were increased after the P25 nanoparticles are loaded on LTAs by a hydrothermal method.

Figure 2 shows the XRD patterns of LTAs (Figure 2(a)), pure P25 (Figure 2(b)), and LTAs-P25 (Figure 2(c)). Notably, all the samples exhibited the characteristic diffraction peaks of anatase  $\text{TiO}_2$  at  $2\theta = 25.3^\circ$  (JCPDS Card No. 21-1272) [18]. This may be due to the transformation of the TNT phase from amorphous to anatase phase after annealing treatment of LTAs at 400, and then, the characteristic peak of anatase  $\text{TiO}_2$  appeared. As shown in Figure 2(b), compared to pure P25, the diffraction peak observed at  $2\theta = 27.4, 36.1$  was attributed to the characteristic diffraction peak of rutile-type  $\text{TiO}_2$  (JCPDS Card No. 21-1276) [19]. Although it has been found by SEM that P25 was successfully modified to LTAs, a diffraction peak attributed to rutile-type  $\text{TiO}_2$  did not appear. Only about 20% of rutile titanium dioxide is present in P25. These results indicate that very little rutile  $\text{TiO}_2$  was successfully modified to LTAs, resulting in P25 providing a less intense diffraction peak of rutile  $\text{TiO}_2$  not observed [20].

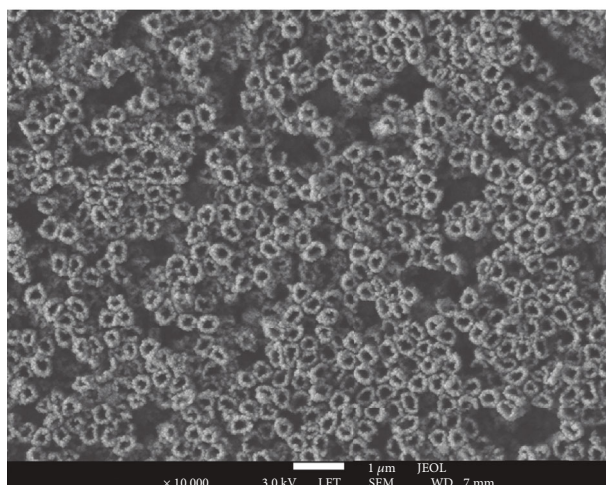
**3.2. The Impact of Different Concentrations of Decoration Solution.** Figure 3 shows the microtopography of LTAs prepared by the hydrothermal method at different P25 concentrations ( $7.5 \times 10^{-4}$ ,  $1.2 \times 10^{-3}$ , and  $1.6 \times 10^{-3}$  M) at 85°C for 12 h. The results showed that the amount of P25 loaded on the nanotube surface was positively correlated with the concentration of P25 solution, and only the optimal batch of  $1.2 \times 10^{-3}$  M resulted in the uniform decoration of P25 on the LTAs (Figure 3(b)). Figure 3(a) shows that only a small quantity of P25 NPs coated the LTAs, with no noticeable decrease in diameters because the P25 solution concentration was negligible ( $7.5 \times 10^{-4}$  M). In contrast, a significant amount of P25 NPs coated the pores and void spaces of nanotubes when a high concentration of P25 solution was used ( $1.6 \times 10^{-3}$  M), leading to occlusion of most of the nanotubes, as shown in Figure 3(c). The solution



(a)



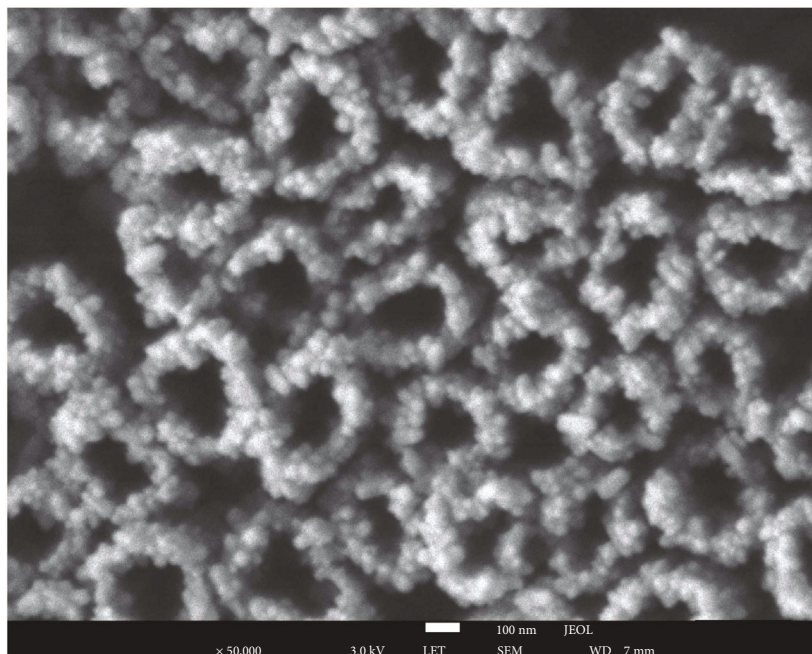
(b)



(c)

FIGURE 1: Continued.





(d)

FIGURE 1: SEM images of TiO<sub>2</sub> nanotube arrays: (a) top view before decorated P25; (b) high magnification before decorated P25; (c) top view after decorated P25; (d) high magnification after decorated P25.

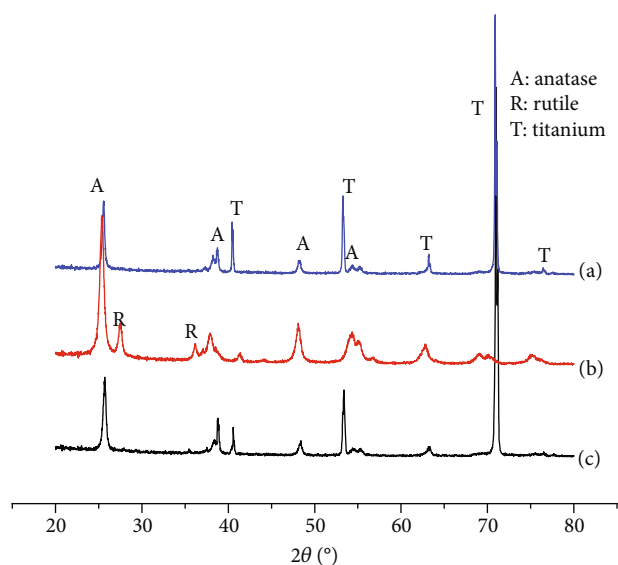


FIGURE 2: XRD patterns of two samples: (a) LTAs; (b) pure P25 particle; (c) LTAs-P25.

concentration is at a high level, resulting in a large amount of P25 deposited in the nozzle and gap of LTAs, which is not conducive to a large amount of drug loading. Therefore, excessive P25 NPs sintered and agglomerated after heating. Obviously, the loading capacity and effect of Figure 3(b) are better than those of Figures 3(a) and 3(c), and it is a sample with the best morphology and larger surface area. It can be judged that the optimum concentration for loading P25 in LTAs more uniformly is  $1.2 \times 10^{-3}$  M (Figure 3(b)).

### 3.3. The Morphology of the Samples after Drug Loading.

Figures 4(a) and 4(b) present the SEM images of the morphology of the prepared LTAs and LTAs-P25 after IBU loading, indicating that the drug was successfully loaded onto TiO<sub>2</sub> nanotube arrays by the vacuum drying method. A significant amount of the drug is clustered, and most nanotubes are entirely covered (Figure 4(a)). In contrast, Figure 4(b) shows that IBU distributed on the pores homogeneously, and void spaces of LTAs-P25 are entirely covered by IBU.

Furthermore, Figure 4(b) illustrates that IBU is adsorbed on the surface and is incorporated partly into the nanotubes, indicating that the increased specific surface area and roughness of LTAs after hydrothermal treatment are conducive to drug dispersion and absorption.

### 3.4. XRD Analysis of the Samples after Drug Loading.

Figure 5 shows the XRD pattern of the drug-loaded samples. Compared with the XRD pattern of pure IBU, the characteristic diffraction peak credit belongs to IBU-loaded LTAs and LTAs-P25 appeared at  $2\theta = 6.08, 12.21, 16.70,$  and  $20.19$  (JCPDS card number 32-1723), respectively. The successful loading of the drug on the two samples is confirmed by the vacuum drying method. Compared with Figure 5(b), a stronger characteristic peak of IBU appears in Figure 5(a), which indicates that the modified sample has stronger adsorption capacity, so that a large amount of IBU is effectively loaded on LTAs-P25.

### 3.5. In Vitro Release of IBU.

Figure 6 presents comparative drug release curves of ibuprofen loaded into LTAs and LTAs-P25. It indicated that the release of IBU from LTAs

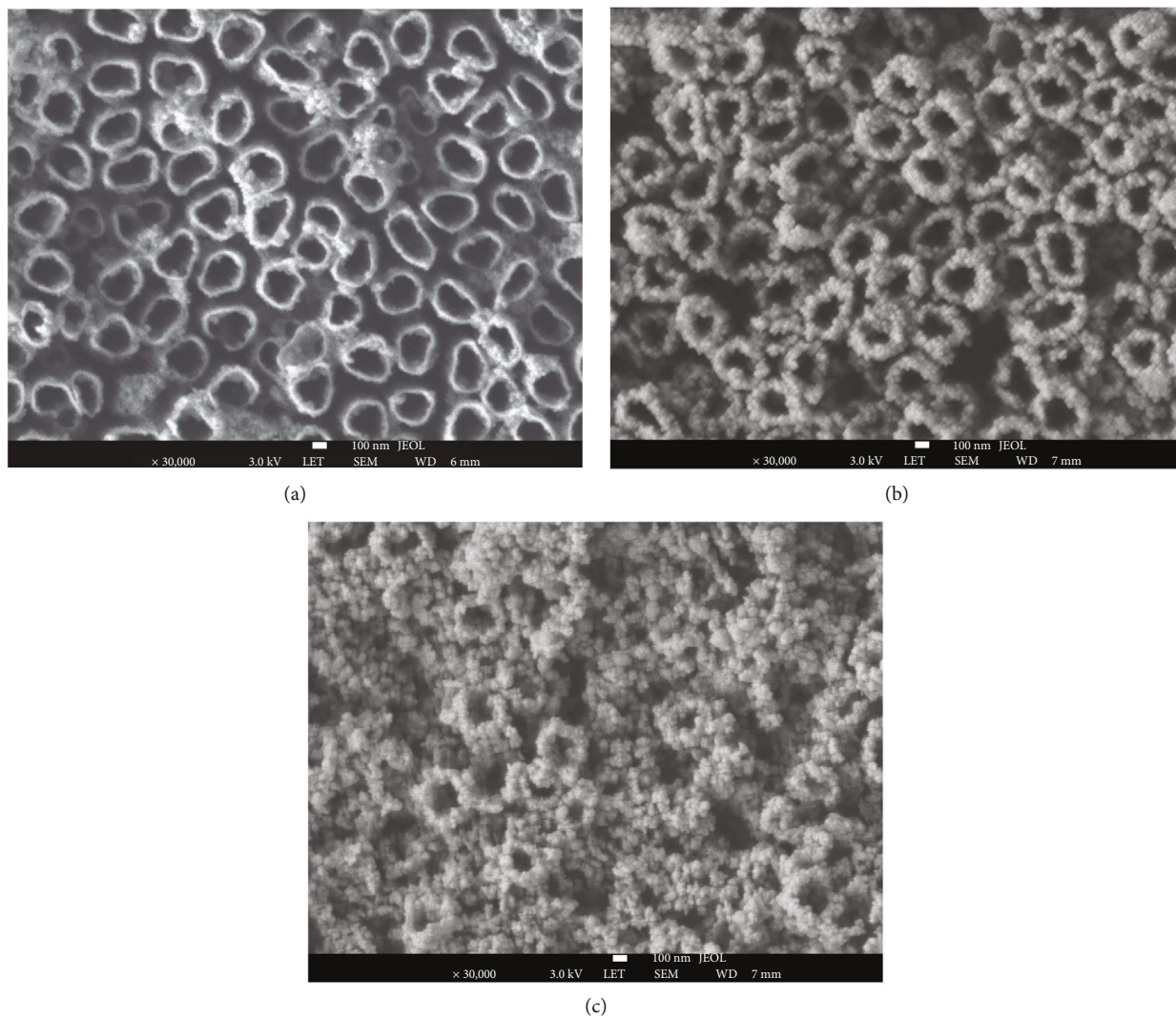


FIGURE 3: The SEM images of LTAs decorated by different concentrations: (a)  $7.5 \times 10^{-4}$  M; (b)  $1.2 \times 10^{-3}$  M; (c)  $1.6 \times 10^{-3}$  M.

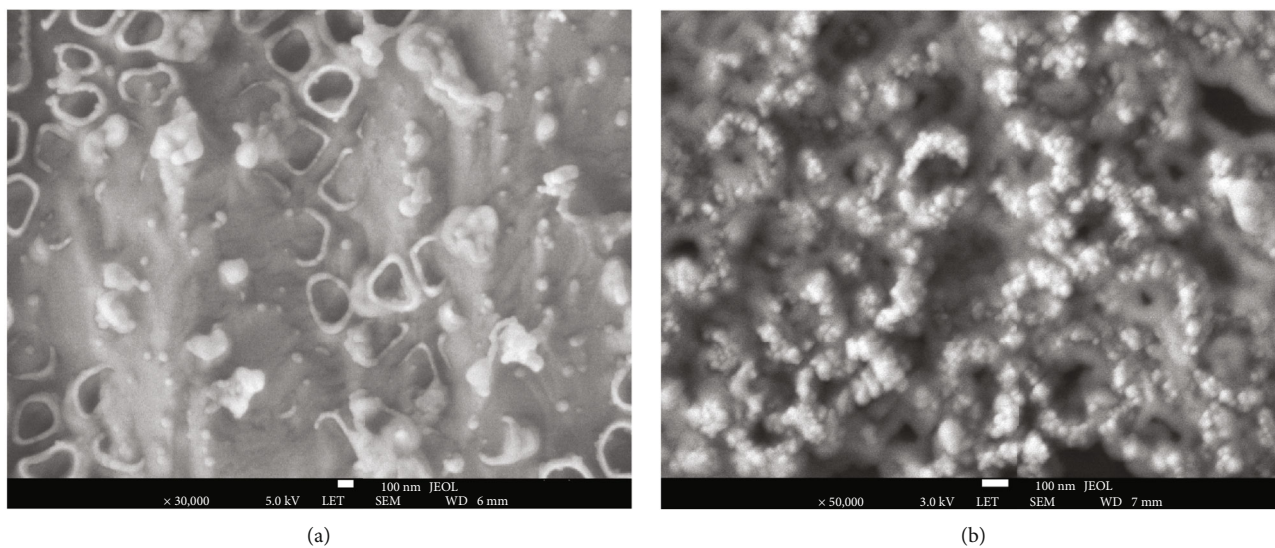


FIGURE 4: SEM images of two samples: (a) LTAs after drug loading; (b) TNTs-P25 after drug loading.

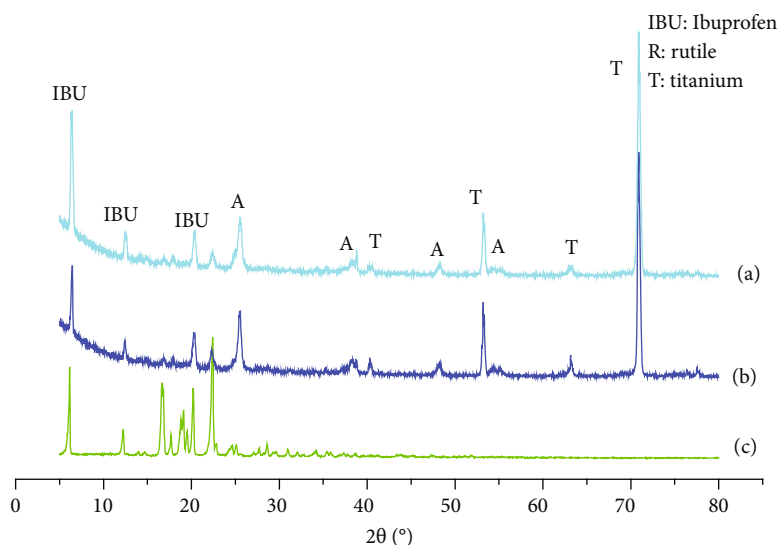


FIGURE 5: XRD pattern of the samples: (a) LTAs-P25 after drug loading; (b) LTAs after drug loading; (c) pure IBU.

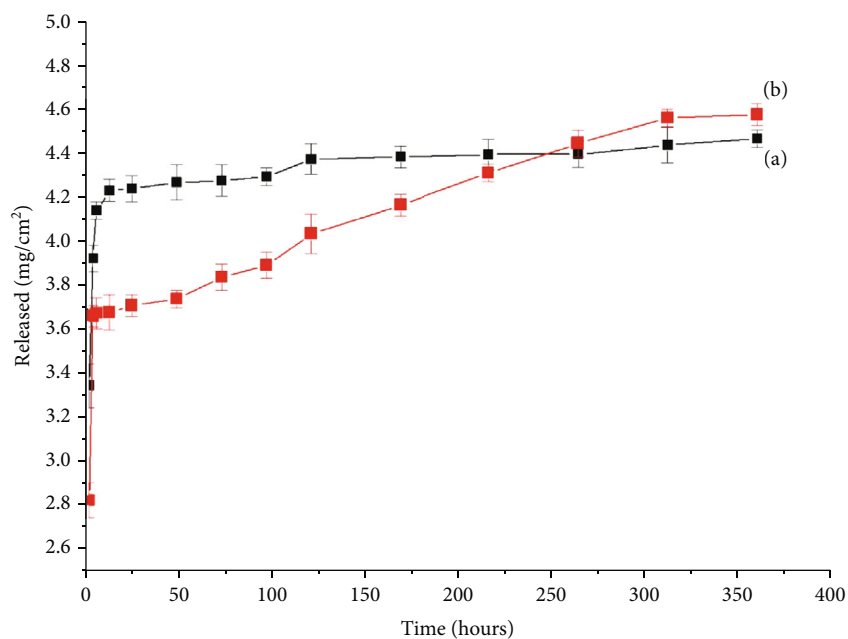


FIGURE 6: Drug released profiles from (a) LTAs loaded with IBU; (b) TNTs-P25 loaded with IBU.

and the drug release phase of LTAs-P25 consist of the initial 5 h drug burst release and the subsequent 15-day slow-release process. During the 5 h burst release period, the released drugs mainly came from covering the surface and the nozzle, and these drug molecules will diffuse into PBS very quickly. At this stage, the amount of IBU released was about 92.7%. However, the IBU released by LTAs-P25 decreased to 80.2% in the same time period. Subsequent sustained release for up to 15 days mainly comes from IBU loaded on the tube wall and inside the nanotubes. It was concluded that LTAs-P25 provided a lower burst and longer release time. A comparison with the literature [13] showed that the total amount

of drug released from LTAs increased by 49.5% compared to that from TiO<sub>2</sub> nanotube arrays without large clearance (from 2.99 mg cm<sup>-2</sup> to 4.47 mg cm<sup>-2</sup>). Although the specific surface area and roughness of the LTAs significantly increased, which was conducive to improving the load effect of IBU, the released amount of IBU did not significantly increase for LTAs-P25 in the release stage due to its large void spaces.

#### 4. Conclusions

In this paper, P25 NPs were successfully decorated on the surface of LTAs via hydrothermal treatment under the



optimal conditions, with the pore diameter decreasing to  $100 \pm 10$  nm after decoration. The specific surface area and roughness of the nanotubes were increased after the P25 nanoparticles are loaded on LTAs by a hydrothermal method. The optimum concentration for loading P25 in LTAs more uniformly is  $1.2 \times 10^{-3}$  M. The increased specific surface area and roughness of LTAs after hydrothermal treatment are conducive to drug dispersion and absorption, which was beneficial to improve the load degree and quantity of IBU. The total amount of drug released from LTAs increased by 49.5% compared to that from TiO<sub>2</sub> nanotube arrays without large clearance. Notably, the increase in roughness and surface area of LTAs gave rise to the sustained release of drugs. This is indispensable for the implantation of biomedical materials, especially plastic surgery. These results demonstrate the fact that LTAs-P25 has potential application value in constructing drug delivery systems applied in the biomedical field.

### Data Availability

No data were used to support this study.

### Conflicts of Interest

The authors declare that they have no conflicts of interest.

### Authors' Contributions

Tao Zhang and Nannan Liu contributed equally to this work.

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